COMPREHENSIVE QUALITY ASSURANCE PLAN



HERITAGE LABORATORIES, INC

SUBMITTAL FORM PAGE 1 OF 1

ECC SITE ZIONSVILLE, INDIANA PROJECT NUMBER ____

Date of Submittal: Approval or Disappro Previous Submittal Di	oval By:/_	2 / 93 Submittal Number	-A -B -C -D
Title of Submittal: Manufacturer: Address:	LABORATORY	DUALITY ASSURANCE	PLAN
Supplier: Address:			
Specification Reference Specification Reference Specification Reference	ce Paragraph:	01392 AU	
Comments (additional	l space on back of th	is sheet)	
Deviations (additiona	l space on back of th	is sheet)	
field construction crit	ereby represent that I teria, materials, dime	have determined and verified all ensions, catalog numbers, and sinder applicable reviewed shop draw	nilar data and I have
		Items Included	Check with "X"
		Plan/Narrative	X
11		Shop Drawing(s)	
	The	Catalog Cut/Mfgr Data	
AWD Technologies	, Inc.	Technical Data	
Authorized Represen	ntative	Test Report	
		Certification	
		Specifications	

Other:

IN EXACT COPY OF

BY (BS DATE 8-2-93

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COMPREHENSIVE QUALITY ASSURANCE PLAN

Heritage Laboratories, Inc.

Indianapolis Division Two Environmental Plaza 7901 West Morris Street Indianapolis, Indiana 46231 (317) 243-8304

Charlotte Division 4132 Pompano Street Charlotte, NC 28216 (704) 393-1853

91GB1148.L5A

Chicago Division 1319 Marquette Drive Romeoville, Illinois 60441 (708) 378-1600

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Prepared by Heritage Laboratories, Inc. Indianapolis, Indiana

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C. Steven Gohmann, President Heritage Laboratories, Inc.	Date
Welesson	09-19-93
Dr. D. L. Peterson, QA Officer Heritage Laboratories, Inc.	Date
James Coplen, VP Operations	5/19/93 Date
Heritage Laboratories, Inc.	Jac
[DER QA Officer]	Date

EMS HERITAGE LABORATORIES, INC.



7901 West Morris Street Indianapolis, IN 46231 Phone: 317/243-0811 FAX: 317/243-0360

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November 1, 1991

To Whom It May Concern:

In order to more adequately serve the needs of our clients, EMS Heritage Laboratories, Inc. has adopted a Comprehensive Quality Assurance Plan (CompQAP) according to the guidelines established in EPA document number QAMS-005/80 and Florida DER-QA-001/90. This CompQAP will be implemented and all EMS Heritage divisions will be in substantial compliance by December 31, 1992.

This CompQAP covers such laboratory functions as data acquisition, review and reporting, and establishes QA objectives and mechanisms by which these objectives are measured, documented and reported. In addition, such functions as training and safety are covered.

It is our intention that all who use our services be provided with the opportunity to understand the procedures used in acquiring, reviewing and reporting laboratory results.

It is our requirement that all laboratory personnel read, review and understand the procedures and requirements established by this document.

Sincerely,

EMS HERITAGE LABORATORIES, INC.

C. Steven Gohmann President

CSG/ps

EMS HERITAGE LABORATORIES, INC.

HERITAGE,

7901 West Morris Street Indianapolis, IN 46231 Phone: 317/243-0811 FAX: 317/243-0360

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Comprehensive Quality Assurance Plan

EMS Heritage Laboratories, Inc.
Two Environmental Plaza
7901 West Morris Street
Indianapolis, Indiana 46231

Issued to:	AWD TECHNOLOGIES	
Date:	0Z-AUG-93	
Copy Number:	244	

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3.0 Statement of Policy

It is our policy that all who use our services be provided with the opportunity to understand the procedures used in acquiring, reviewing and reporting laboratory results.

It is our requirement that all laboratory personnel read, review, have free access to and understand the procedures and requirements established by this document as they relate to their job function. It is to be used as a basic reference in the laboratory. Each staff member is obligated to comply with the stated requirements of this Comprehensive Quality Assurance Plan (Comp QAP).

The United States Environmental Protection Agency (EPA) requires that laboratories generating data implement minimum procedures which assure that the precision, accuracy, completeness and representativeness of its data are known and documented. In addition, the laboratory must specify the minimum quality levels which data must meet in order to be acceptable for a specific regulatory program.

To insure that EMS Heritage Laboratories (EMS) meets these responsibilities, the following Comprehensive Quality Assurance Plan has been adopted by the company.

The EPA has established specific requirements for the development of Quality Assurance Project Plans. All Quality Assurance Project Plans must address sixteen specific areas which are outlined in QAMS-005/80 "Interim Guidelines and Specifications for Preparing Quality Assurance Project Plans". This Comprehensive Quality Assurance Plan covers those 16 points and also applies to environmental monitoring and measurement efforts which are outlined in Rule 17-160 Part III, F.A.C. of the State of Florida. This Comp QAP is generic in nature and summarizes the criteria/procedures used under standard operating conditions. Different Data Quality Objectives (DQO's) may require a specific project plan utilizing the basic procedures set forth in this comprehensive plan. This Comp QAP covers all samples submitted to the laboratory except those samples requiring different DQO's as specified by the client. All exceptions to this Comp QAP must be approved in writing by the Laboratory Director, QA Unit, Chief Chemist or the QA Officer. The certificate of analysis will contain a statement to reflect this deviation from the Comp QAP.

EMS Heritage Laboratories, Inc. is a nationally integrated network of commercial testing laboratories specializing in the analysis of groundwater, surface and wastewater as well as solid and hazardous wastes. EMS Heritage operates laboratories in Indianapolis, Indiana; Charlotte, North Carolina; Kansas City, Missouri; and Chicago, Illinois. Each of these laboratory divisions operates under the same Standard Operating Procedures Manuals (SOPs). All laboratory divisions are in direct communication with each other through a single Laboratory Information Management System (LIMS) which incorporates data from

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all divisions and generates the Quality Assurance Targets for all laboratories. All data from each laboratory is accessible through the LIMS to all other divisions.

EMS utilizes a system of project managers in charge of specific projects. Each project manager is responsible for receiving and approving completeness of final data reports relating to their assigned project even if the data is generated by another laboratory division. The Quality Assurance Unit at each division is responsible for reviewing and signing all final certificates of analysis. Employees from other branches are under supervision of the project manager in charge of the project, even if that manager is located in another division office. This system of dedicated project managers and quality assurance staff insures that the client's specific DQO's are met.

As outlined in the EMS Heritage Statement of Operations and Quality Assurance Policy, procedures must be uniquely defined in operation specific SOPs. These SOPs are subject to change without notice, but will not be less stringent than the requirements of this QAP. Where appropriate, these SOPs have been summarized in this document.

This Comp QAP will be reviewed and revised as necessary and at least annually. Major revisions will be forwarded to appropriate copy holders.

4.0 Organization and Responsibility

4.1 Introduction

EMS Heritage Laboratories is primarily an environmental analytical chemistry laboratory. Bacteriological testing is not specifically covered under this comp QAP. Field operations consist primarily of sample pick-up services. EMS Heritage does perform composite and grab wastewater sampling for those clients desiring this service. Other types of sampling may be offered depending on the equipment required and the hazards involved.

4.2 President's Duties

EMS Heritage Laboratories, Inc. currently operates laboratories in four locations. The President of EMS Heritage Laboratories, Inc. serves as the Chief Operating Officer (COO). Furthermore, the President shall:

- 4.2.1 Designate a Laboratory Director at each laboratory.
- 4.2.2 Replace Laboratory Directors promptly should such replacement become necessary.
- 4.2.3 Assure that there is a Corporate Quality Assurance Officer.
- 4.2.4 Assure that the Vice-President of Operations, Quality Assurance Officer and Corporate Sales Personnel clearly understand their functions and responsibilities.
- 4.2.5 Establish and revise company policy.

4.3 <u>Vice-President of Operations' Duties</u>

The Vice-President of Operations (VPO) is the only additional operating officer of the company. The VPO shall nominate Laboratory Directors to the President who will promptly act to confirm or reject their nomination. The VPO is charged with the overall coordination of all the laboratory locations so that they perform as one cohesive and integrated network of laboratories. Furthermore, the VPO shall:

- 4.3.1 Assure that personnel, resources, facilities, equipment, materials, and methodologies are known and available as scheduled.
- 4.3.2 Assure that any deviations from policy reported by the Quality Assurance Officer are communicated to the President and to the Laboratory Directors and that corrective actions are taken and documented.

4.4 Quality Assurance Officer Duties

The Corporate Quality Assurance Officer is responsible for assuring that all QA policies and procedures are followed and that the Data Quality Objectives of this QA Plan are met. Management will provide the QAO with the support and tools necessary to achieve this goal.

Several principles underlie the Quality Assurance Policy. These principles are:

- ♦ Defensible Data
- ♦ Documented Analytical Methods
- ♦ Sample Control
- ♦ Equipment Calibration
- ♦ Training and Training Documentation
- ♦ Internal and External Assessments
- ♦ QA/QC Policy

4.4.1 Defensible Data

Data shall be generated, recorded and archived in such a way that it will withstand challenge. All reported information shall be accurate and properly documented. By correct use of traceable calibration standards, verification standards, blanks, method blanks, duplicates, spikes and spike duplicates, the Company shall obtain and record data necessary to demonstrate that analytical systems remain in control for all sample data reported.

- 4.4.1.1 Computerized data records must follow strict guidelines for security and defensibility. These guidelines are established in the QAP.
- 4.4.1.2 Defensible data includes the use of analytical methods which are appropriate and adequate. Analytical methods used will be strictly adhered to and any deviations will be documented and reported; these deviations are subject to approval of the QAO and the Chief Chemist.

4.4.1.3 Defensible data includes adherence to pre-analytical holding time requirements. Wisconsin Administrative Code NR149.11(s) specifically requires flagging of any holding time violation. The State of Wisconsin further considers "recommended" holding times as required holding times.

EMS Heritage will adhere to the requirement for the flagging of any holding time violation for those states or organizations requiring such flagging.

4.4.1.4 Defensible data includes the laboratory analysts demonstration of their ability to generate data of acceptable precision and accuracy, in addition to the laboratory demonstration of detection limits. IDLs and MDLs must be approved, performed at the required frequency and kept on file by the QA Unit. See Section 11.0, Figure 11.3, IDL/MDL Reporting Form.

4.5 Organization Charts/Job Descriptions (Appendix E)

Job Descriptions of all key personnel within EMS Heritage Laboratories, Inc. are contained in Appendix E. Company Organization Charts are provided in the figures listed below.

Corporate Organization	Figure 4.1
Indianapolis Division	Figure 4.2
Charlotte Division	Figure 4.3
Chicago Division	Figure 4.4
Kansas City Division	Figure 4.5
Heritage Environmental Services, Inc.	Figure 4.6

4.6 Deputies to Cover Absences

In the event of the absence of a key person, the authority for that position will generally rest with the next in line, most senior staffer as per the EMS Heritage Organizational Charts.

Specific lines of authority exist for the following positions:

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- 4.6.1 In the absence of the Chief Chemist the Senior Chemists will be in charge until the return of the Chief Chemist, unless otherwise delegated by the Vice-President of Operations.
- 4.6.2 In the absence of the Quality Assurance Officer, the senior staffer from the Indianapolis QA Unit will be in charge until the return of the QA Officer. In the absence of all QA Unit staff from the Indianapolis Division, the QA Unit from another EMS Heritage Division as appointed by the President of EMS Heritage will be in charge until the return of the QA Officer or the QA Unit staff of the Indianapolis Division.
- 4.6.3 In the absence of all QA Unit staff from any EMS Heritage Division, the QA Unit from another EMS Heritage Division as appointed by the QA Officer or the President of EMS Heritage will be in charge until the return of the absent QA Unit staff.

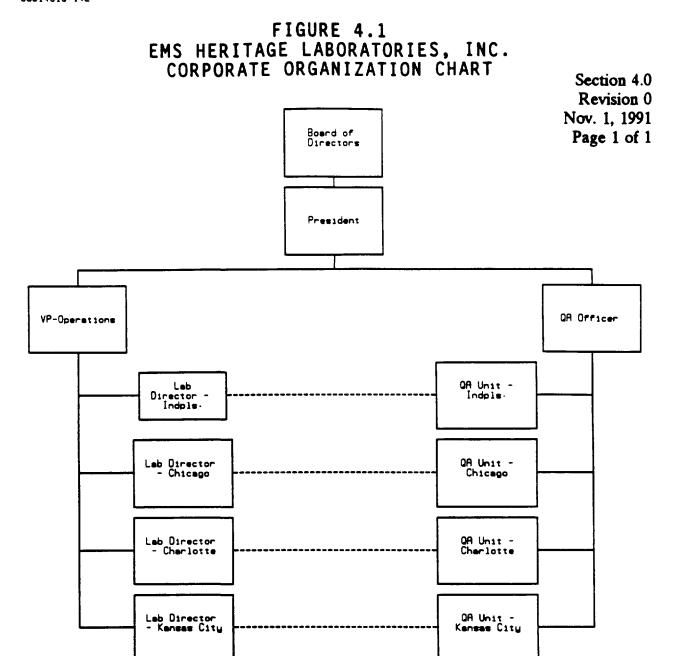


Figure 4-1 - Management and Quality Resurance Structure

FIGURE 4.2 ORGANIZATIONAL CHART EMS HERITAGE LABORATORIES, INC. INDIANAPOLIS, INDIANA

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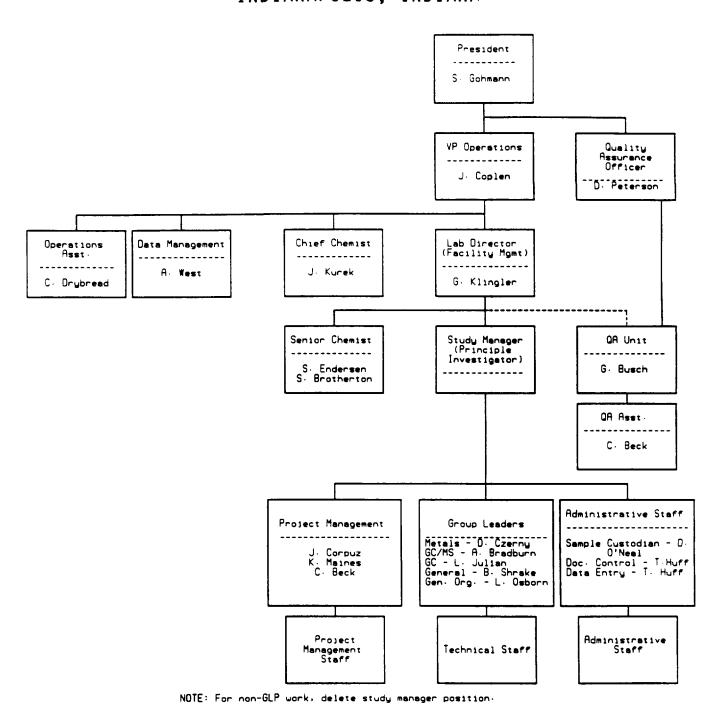
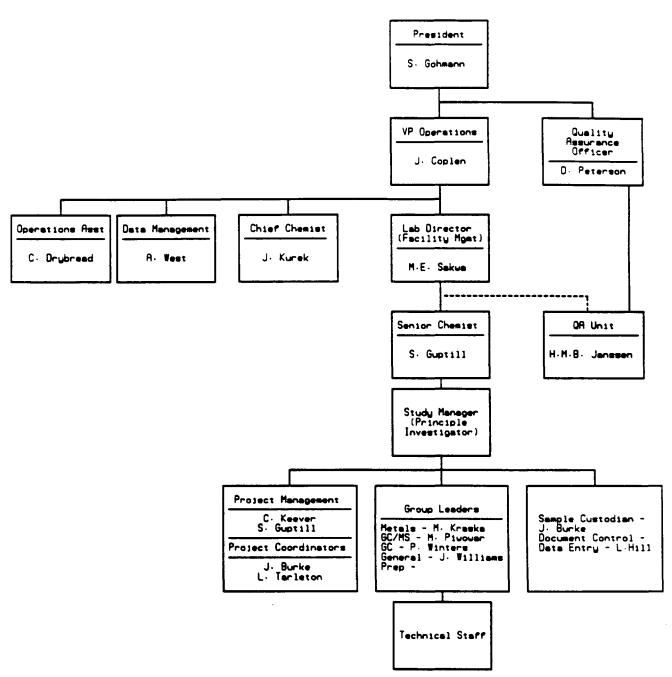
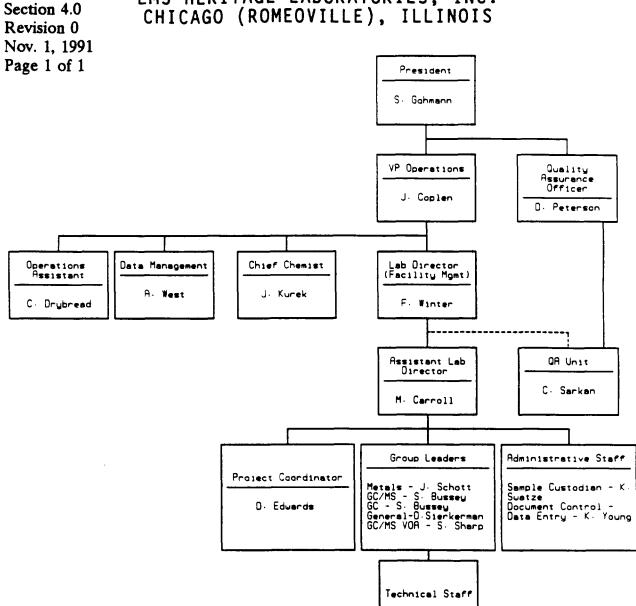


FIGURE 4.3 ORGANIZATIONAL CHART EMS HERITAGE LABORATORIES, INC. CHARLOTTE, NORTH CAROLINA



NOTE: For non-GLP work, delete study manager position-

FIGURE 4.4 ORGANIZATIONAL CHART EMS HERITAGE LABORATORIES, INC. CHICAGO (ROMEOVILLE), ILLINOIS



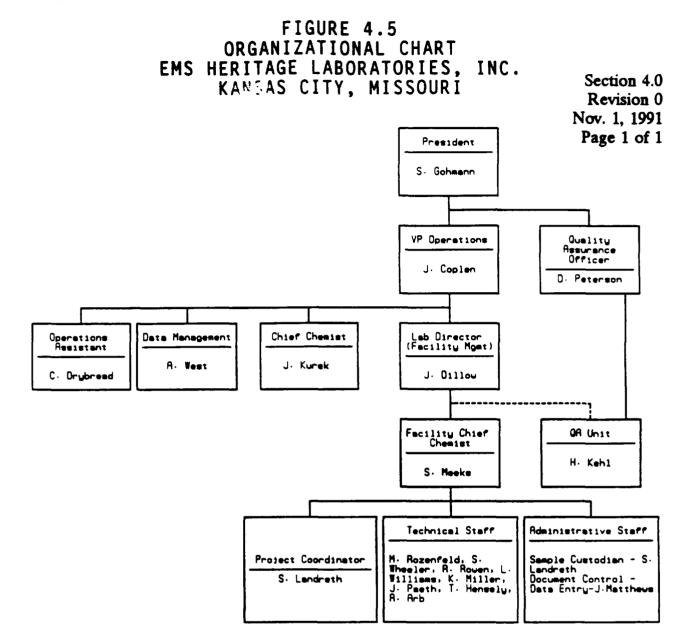
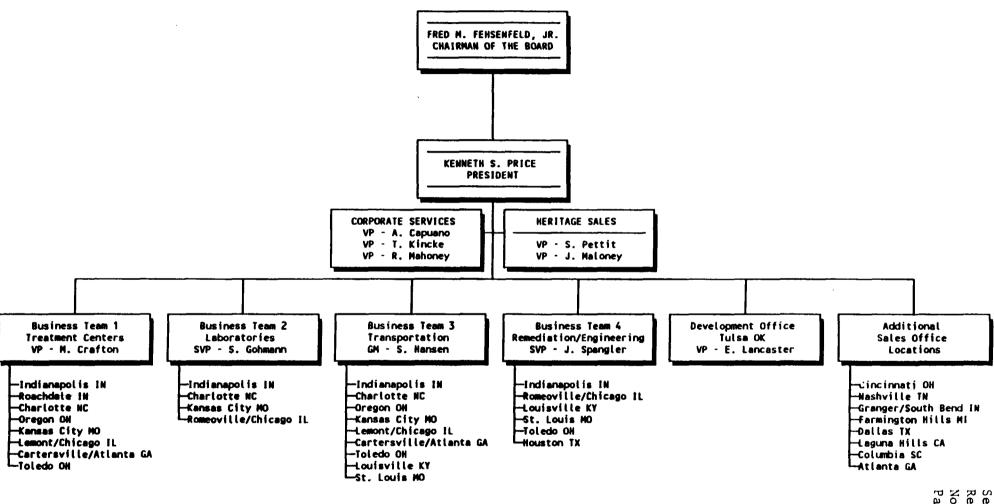


FIGURE 4.6





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5.0 OA Targets for Precision, Accuracy and Method Detection Limits

The targets for precision (duplicates), accuracy (spikes), and practical (reporting) detection limits for each sample matrix are given for each method performed by EMS Heritage in Table 5.1, Quality Assurance Targets for Precision, Accuracy and Method Detection Limits. Precision and accuracy targets and detection limits are developed from in-house data when possible. Precision and accuracy data are compiled continuously and reported quarterly using the Quality Control Tracking System (QCTS). The goals for precision and accuracy are reviewed quarterly by the QA Officer. The precision, accuracy, and detection limits presented in this CompQAP are routine target values. Deviations from these criteria must be addressed in individual project plans if required.

The QCTS specifies sample matrix as well as preparation procedures in summarizing QA data. Using this system, it is possible to insure that the QA Objectives for the analytical procedures using various sample matrices and preparation procedures are being met and are "representative of the media".

When precision and accuracy data are not available due to the newness or infrequency of use of a method, the literature or method values will be used until such time as in-house data can be accumulated sufficient to establish control limits. In any case those in-house control limits must be as stringent as the literature or method control limits given for the particular matrix. Some methods and testing programs specify control limits which will be used; these are identified in Section 12.0.

Since sample precision and accuracy data are matrix dependent, these criteria are "targets or goals". Out of control sample results may be reported with comments only after a determination is made by the QA Officer and/or QA Unit that based on the Laboratory Control Sample (LCS, an extracted or digested standard) results, a reanalysis of the out of control sample or best professional judgment that matrix problems prevent performance of the particular method within the established criteria. When out of control results are reported by the above protocol the QA Officer and/or the QA Unit will make the determination as to whether to enter the data into the QC data base (QCTS) or determine that the results are true outliers which will not be entered into the data base. The validity and the "completeness" of the results will be left for the sample submitter to evaluate in the few cases (less than 1% of all data reported) where interference and other matrix problems cannot be overcome by the methods employed.

The laboratory defines completeness as a measure of the amount of valid data obtained from a measurement system compared to the amount that was expected to be obtained under current normal conditions. The QA objective for completeness is 99%.

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The following tables are compiled from the QCTS where data is available and contain:

- a. Component
- b. Matrix
- c. Method
- d. Precision (RPD)
- e. Accuracy (% Recovery)
- f. PQL, MDL or other reporting limit

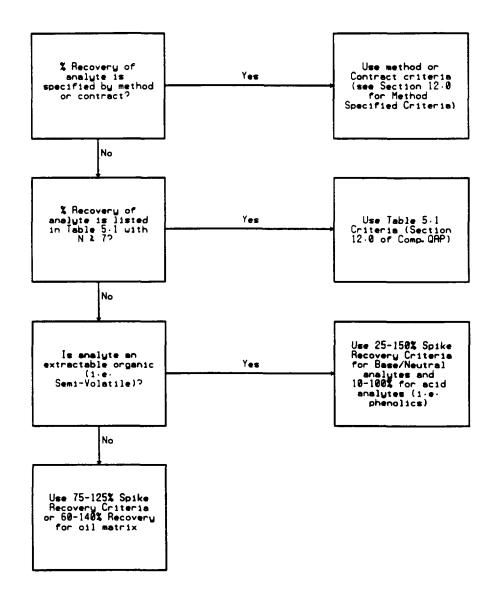
The variance of the analytical method changes as the analyte concentration increases from the MDL. Precision and accuracy will be evaluated as a function of concentration when comparing the QC data obtained for an analytical batch versus the QC criteria arrived at from the QCTS data base. Concentrations of analyte less than 10 times the method detection limit (MDL) present in non-spiked (lab duplicate) samples will not be considered as out-of-control based on the QCTS data. RPD's exceeding 20 percent for inorganic analyses and 50 percent for organic analyses will be considered as out-of-control for concentrations of analyte greater than 10 times the reporting detection limit.

Figure 5.1, Matrix Accuracy Control Criteria Usage is a flow chart which outlines the control limit criteria to be applied for accuracy measurements.

Figure 5.2, Matrix Precision Control Criteria Usage is a flow chart which outlines the control limit criteria to be applied for precision measurements.

EPA SW-846 methods that do not specify a sample preparation (extraction, digestion, cleanup, etc.) method are listed in Table 5.2, <u>Sample Preparation Method</u>, which includes a reference to the proposed sample preparation method.

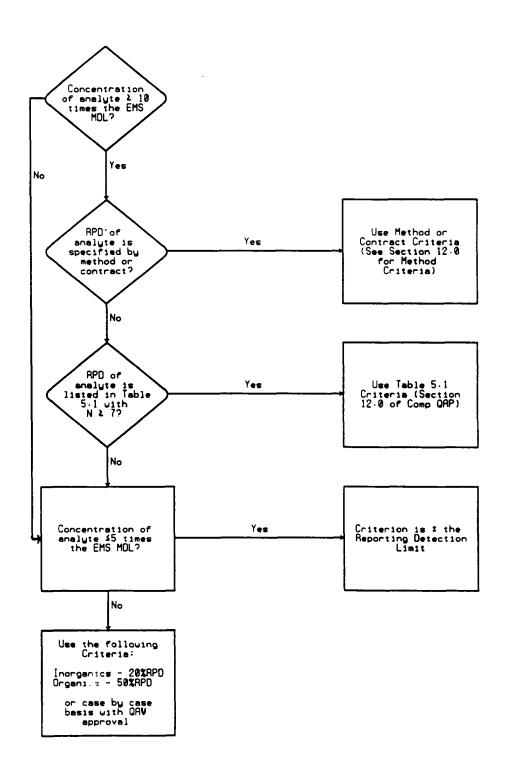
FIGURE 5.1 Matrix Accuracy (% Recovery) Control Criteria Usage



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FIGURE 5.2
Matrix Precision (RPD) Control Criteria Usage



1S F AGE ORA ES Table 5.1

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 THRU 30-SEP-91

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Method	Matrix	Parameter	N X REC N RPD Limit Page 1 Of 38
AOAC 57-3	\$7979	CATION EXCHANGE CAPACITY	2 0 - 43
APHA 209A	Non-Spec Water	SOL 1D6	20-0
	9/5/3	SOLIDS	15 0 - 4
APHA 213E	S/S/S	GRAVITY	. 10-0
APHA 3128	Non-Spec Water	HEXAVALENT CHROMIUM	88 74 - 187 85 9 - 19
	S/S/S	HEXAVALENT CHROMIUM	1 0 - 7
APHA 314A	Non-Spec Water	HARDNESS	1 0 - 3
APHA 412K	Non-Spec Water	THIOCYANATE	1 81 - 81 1 0 - 13,
APHA 415A (16E)	S/S/S	I OD I NE	2 81 - 97 1 0 - 5
APHA 509A	Non-Spec Water	ALDRIN 4,4'-DDD 4,4'-DDT HEPTACHLOR HEPTACHLOR EPOXIDE	2 0 - 32 2 0 - 52 2 0 - 17 2 0 - 41 2 0 - 5
APHA 909C (MOD)	Drinking Water	COLIFORM BACTERIA, FECAL	1 0 - 0
,	Non-Spec Water	COLIFORM BACTERIA, FECAL	3 0 - 0
ASTM D-129	Non-Spec Water	SULFUR	1 0 - 7
ASTM D-240	Non-Spec Water	HEAT OF COMBUSTION	1 0 - 13
	011	HEAT OF COMBUSTION	3 0 - 10
	S/S/S	HEAT OF COMBUSTION	40 0 - 22
ASTH D-244	Non-Spec Water	GRAVITY	3 0 - 19
	S/S/S	GRAVITY	8 0 - 0
ASTM D-482	011	RESIDUE, PERCENT ASH	1 0 - 2
	5/\$/\$	RESIDUE, PERCENT ASH	40 0 - 21
ASTM D-70	\$/\$/\$	DENSITY	30-4

N . Sample count

ان این ایست این

REC + Percent Recovery (observed/actual+100)

RPD = Relative Percent Difference (2+(rep1-rep2)/(rep1+rep2)+100)

EMS HERITAGE LAUDRATORIES Table 5.1

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91_THRU_30-SEP-91 Revision 0 Revision 0 New 1 1994

boriteM Matrix Parameter % REC RPD Limit ASTM D-808 011 TOTAL HALOGEN (TX) 1 0 -5/5/5 TOTAL HALDGEN (TX) 3 0 - 23 _____ S/S/S ASTM D-86 BOILING POINT ASTH D-92 5/5/5 FLASH POINT ASTM D-93 Non-Spec Water FLASH POINT S/S/S FLASH POINT ASTM D-95 011 UATER WATER 3/9/9 1 100 - 100 3 0 - 37 ASTM D-96 **3/5/5** BOTTOM SEDIMENT AND WATER 10 01L SOLIDS WATER CLP SOW 2/88 Non-Spec Water BENZENE 75 - 123 CHLOROBENZENE 93 - 121 1.1-DICHLOROETHENE 43 - 163PCB AROCLOR 1260 68 - 118 TOLUENE 121 - 58 TRICHLOROETHENE 9/9/9 ACENAPHTHENE 4-CHLORO-3-METHYLPHENOL 94 - 102 ___ 82 - 90 2-CHLOROPHENOL 1,4-DICHLOROBENZENE (P-DICHLOROBENZENE) 85 - 85 2,4-DINITROTOLUENE N-NITROSO-DI-N-PROPYLAMINE 72 - 98 4-NITROPHENOL 54 - 104 15 PENTACHLOROPHENOL 53 - 98 ..12. PHENOL 76 - 84 1 0 -3 PYRENE 51 - 132 1,2,4-TRICHLOROBENZENE 2 83 - 96 1 0 - 3 **EPA 110.3** Non-Spec Water COLOR 10 0 -EPA 120.1 Non-Spec Water CONDUCTIVITY 130 -EPA 130.2 Non-Spec Water HARDNESS R 100 - 100 8 0 - 5

N	=	S.	•	ţ,	1	c .	c	o	11	n	1

REC - Percent Recovery (observed/actual+100)

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RPD = Relative Percent Difference (2*(rep1-rep2)/(rep1+rep2)*100)

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Table 2.1

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 THRU 30-SEP-91 Revision 0 Nov 1, 1991

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Page 3of 38 Hethod Matrix Parameter X REC RPD Limit 5 0 -EPA 150.1 Brinking Water PH Non-Spec Water PH 304 0 - 2 5/5/5 PH EPA 160.1 Drinking Water SOLIDS 70- 15 Non-Spec Water SOLIDS 8 87 - 131 84 0 - 13 S/S/S SOLIDS 1 0 - 1 _____ EPA 160.2 Non-Spec Water SOLIDS SUSPENDED SOLIDS 21 62 - 128 112 0 - 34 EPA 160.3 Non-Spec Water SOLIDS S/S/S SOL 1 DS EPA 160.4 Non-Spec Water TOTAL SOLIDS, VOLATILE VOLATILE SUSPENDED SOLIDS S/S/S TOTAL SOLIDS, VOLATILE 60- 27 EPA 180.1 Drinking Water TURBIDITY 2 0 - 20 Non-Spec Water TURBIDITY EPA 200.7 Drinking Water BARIUM BORON 51 - 163CALCIUM 86 - B6 CHRONIUM 75 - 123 10 COPPER 72 - 128IRON - 110 HACNESIUM 84 - 11312 MANGANESE 75 - 124NICKEL 90 - 90 POTASSIUM 88 - 88 SILVER 73 - 127SODIUM 83 - 112 ZINC 63 - 131Non-Spec Water ALUMINUM B2 - 110 ANTIMONY 26 67 - 10212 0 -10 BARIUM 87 - 10815 0 -BERYLLIUM 69 - 134

N = Sample count

REC * Percent Recovery (observed/actual+100)

RPD = Relative Percent Difference (2*(rep1-rep2)/(rep1+rep2)+100)

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Table 5.1

Quality Assurance Objectives for Location ALL

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PERIOD: 01-APR-91 THRU 30-SEP-91 Page 4 of 38

Hethod	Matrix	Parameter	N	% REC	N	RPD	Limit	Units	
EPA 200.7	Non-Spec Water	BORON		73 - 128	1 0				
	2,22	CADHIUH	69	60 - 116	16 0	- 10			
		CALCIUM	43	80 - 106	28 0	- 4			
		CHRONIUM	128	76 - 111	40 0	- 44			
		COBALT	21	85 - 106	11 0	- 7			
		COPPER	182	77 - 121	86 0	- 27	***************************************		
		IRON	47	73 - 111	35 0	- 13			
		LEAD	75	69 - 115	16 0	- 25			
		LITHIUN			16 0	- 55			
		MAGNESIUM		<u>. 79 - 129</u> 87 - 111	25 0	- 4			
		MANGANESE	26					- Managharian	
				88 - 108	15 0	~ 5		···	
		MOLYBDENUM		66 - 124		<u>.</u>			
		NICKEL	73	77 - 108					
		POTASSIUM	26	86 - 104	17 0	- 8			
		SILICON			1_0.	4,			·
		SILVER	58	80 - 123	13 0			-	4
		SODIUM	26	89 - 108	17 0	- 19		-	₹,
		STRONTIUM	5	66 - 121	1.0	- 4			
		THALLIUM	4	50 - 101					
		TITANIUM	5	76 - 183					
		VANADIUM		84 - 104	100	- <u>3</u>			
		ZINC	132	70 - 118	107 0	- 56			
		ZIRCONIUM	2	64 - 145				-	
	5/5/5	ALUHINUM	2	80 - 101	2 0	- 10		remarkan de reference de la reconstrucción de la companya de la co	· · · · · · · · · · · · · · · · · · ·
		BARIUM	1	99 - 99	1 0	- 2			
		CADHIUH	4	69 - 110	3 0	- 65			
		CALCÍUN	1	94 - 94		- 1			
		CHRONIUM	4	61 - 126	5 0	- 38			29
		COBALT	1	93 - 93	1 0	- 1			•
		COPPER	5	87 - 105	<u>i . ă</u>	- 13			
		IRON	9	69 - 85	3 0	- '5			
		LEAD	4	62 - 111	3 0	- 17			
		MAGNESIUM		99 - 99					
		NICKEL	3	81 - 113	5 0	- 37		-	
		MUISSATOR	-		3 V	- •			
		SILVER		B7 - 94		<u></u>			
		SODIUM	1	93 - 93	20	- 55			
		ZINC	4	76 - 105	1 0 5 0	- 16 - 19		-	
÷ •				_ 10 _ 105_		! 2-			
EPA 202.1	Non-Spec Water	ALUHINUM	42	58 - 132	21 0	- 16			
EPA 204.1	Non-Spec Water	ANTIMONY	<u> </u>	92 - 134	1 0	- 5			
	•					_			
EPA 204.2	Non-Spec Water	ANTIMONY	2	0 - 173	7 0	- 25			
A 1 B									<u>f</u> .
N . Sample co	unt								

REC = Percent Recovery (observed/actual+100)
RPO = Relative Percent Oifference (2+(rep1-rep2)/(rep1+rep2)+100)

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Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91_THRU 30-SEP-91 Revision 0 Nov 1, 1991

Method	Hatrix	Parameter	N	% REC	N RPD Limit	Page 5 Of 38
EPA 206.2	Drinking Water	ARSENIC	13	71 - 127	4 0 - 14	
	Non-Spec Water	ARSENIC		65 - [33]	38 0 - RI	
	S/S/S	ARSENIC	1	95 - 95	1 0 - 200	······································
EPA 208.1	Drinking Water	BARIUM	3	106 - 113		
	Non-Spec Water	BARIUM	16	66 - 116	_ 90-18	
EPA 210.1	Non-Spec Water	BERYLLIUM	z	100 - 100	1 0 - 0	
EPA 210.2	Non-Spec Water	BERYLLIUM	45	53 - 161	23 0 - 21	-
EPA 212.3	Non-Spec Water	BORON			20-21	
EPA 213.1	Non-Spec Water	CADHIUH	133	74 - 123	58 0 - 7	
İ	S/S/S	CADHIUM	2	80 - 114	2 0 - 13	Charles and the control of the contr
EPA 213.2	Drinking Water	CADMIUM	20	54 - 138	1.00	
	Non-Spec Water	CADHIUM	23	54 - 150	7 0 - 10	
EPA 215.1	Drinking Water	CALCIUM			1 0 - 0	
	Non-Spec Hater	CALCIUM	3	90 - 100	1 0 - 0	
EPA 218.1	Drinking Water	CHROMIUM	4	78 - 108		
	Non-Spec Water	CHROMIUM	141	74 - 117	69 0 - 12	
	S/5/5	CHRONIUM	<u>.</u>	93 - 93	1 0 - 14	
EPA 218.2	Drinking Water	CHRONIUM	2	90 - 95	2 0 - 1	 -
	Non-Spec Water	CHROMIUM	8	77 - 123	4 0 - 12	
EPA 219.1	Non-Spec Water	COBALT	<u>2</u>	97 - 110	1 0 - 3	
EPA 220.1	Non-Spec Water	COPPER	139	81 - 117	72 0 - 10	
	5/5/5	COPPER	**************************************	-	1 0 - 0	
EPA 220.2	Non-Spec Water	COPPER		38 - 176_	30 0 - 26	

N : Sample count

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RFC = Percent Recovery (observed/actual+100)

RFO : Relative Percent Oifference (2*(rep1-rep2)/(rep1+rep2)*100)

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Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 [HRU 30-SEP-91

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Matrix Units Hethod Parameter % REC RPD Limit EPA 236.1 Drinking Water IRON 40 - 14110-0 Non-Spec Water IRON 13 0 - 20_____ 19 37 - 135EPA 239.1 Non-Spec Water LEAD 49 0 - 19 _____ 74 - 1235/5/5 LEAD 2 0 - 13 ____ 0 - 342 EPA 239.2 Drinking Water LEAD 21 67 - 131 4 9 - 24 Non-Spec Water LEAD 67 - 13474 0 - 25 _____ 3 0 - 57 _____ 5/5/5 LEAD 47 - 137 EPA 243.1 Drinking Water MANGANESE 78 - 112 Non-Spec Water MANGANESE 77 - 111 5 0 - 234 _____ 10-2____ 5/5/5 MANGANESE EPA 245.1 Drinking Water MERCURY 9 81 - 120 ____ Non-Spec Water HERCURY 110 58 - 129 53 0 - 32 _____ 5/5/5 MERCURY 2 78 - 120 EPA 246.2 Non-Spec Water MOLYBDENUM 69 - 129 ________ EPA 249.1 Non-Spec Water NICKEL 115 71 - 12358 0 - 9 _____ 5/5/5 NICKEL 2 - 143 2 0 - 20 ____ EPA 258.1 5/5/5 POTASSIUM 10-3_____ EPA 270.2 Drinking Water SELENIUM Non-Spec Water SELENIUM 56 - 142 14 0 - 21 _____ 5/5/5 SELENIUM 1 122 - 122 EPA 272.1 Drinking Water SILVER 40 - 40 Non-Spec Water SILVER 71 - 125 35 0 - 5 ____ 5/5/5 SILVER 1 82 - 82

N = Sample count

REC : Percent Recovery (observed/actual+100)

RPD = Relative Percent Difference (2+(rept-rep2)/(rept+rep2)+1001

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ALL	11	0 0 1 901	113 8 0 - 24	120 20-0	9 - 0 E	125 190 - 12	144 1 0 - 14	0 - 0 1 601	129	150 1 0 - 25	180	121 72 0 - 9	*** 2 0 - 23	81 - 0 -	6 0 - 14	10-2	2 - 0 99 651	PS 0 - 9	5 0 - 5	0 - 0 E 06	120 47 0 - 4	105 1 0 - 0	100 1 0 - 3	123 13 0 - 55
ES Location FP-91	×	101	- 68	120 -		10	- 91	103	129 -	1 0 4	120 -	73 -	**				ا پ	:		- 06	89	105 -	100	T
FOF L	2		<u>.</u>	a		37	വ	. ณ	-	a	-	4	-				•			. 🕶	95	αı	-	a.
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Quality Assura PERION:									:												1			
0u311t)		! !																						
	Parameter	SILVER	SILVER	SODIUM	SODIUM	THALLIUM	T I N	TITANIUM	TITANIUM	VANADIUM	ZINC	ZINC	ZINC	ACIDITY	ACIDITY	ALKALINITY	ALKALINITY	ALKALINITY	ALKALINITY	CHL OR 1 DE	CHLORIDE	CHLOR 1 DE	CHLORIDE	CHLORIDE
_		Vater	. Water	. Water	. Vater	. Uater	. Vater	. Water		. Vater	. Water	. Vater		. Water		Water	. Vater		Uater	Vater	Water		Water	Uat er
_	Matrix	Drinking Water	Non-Spec Water	Drinking Water	Non-Spec Water	Non-Spec Water	Non-Spec Water	Non-Spec Water	\$/\$/\$	Non-Spec Water	Drinking Water	Non-Spec Water	8/8/8	Non-Spec Water	8/8/8	Drinking Water	Non-Spec	8/8/8	Non-Spec Water	Drinking Water	Non-Spec Water	8/8/8	Drinking Water	Non-Spec Water
_	-	-	-	-	-	-	_	-		_	-	_	-,	_		-	-	υ,	-	J	~	G,	1	•
	Hethod	EPA 272.2		EPA 273.1		EPA 279.2	EPA 282.1	EPA 283.2		EPA 286.1	EPA 289.1			EPA 305.1		EPA 310.1			EPA 310.2	EPA 325 1			EPA 325 2	

N = Sample count REC : Percent Recovery (observed/actual+100) RPD = Relative Percent Difference (Petrep1-rep2)/(rep1+rep2)+100)

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Table 5.1

Quality Assurance Objectives for Location ALL PER100: 01-APR-91 THRU 30-SEP-91

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1	Hethod	Matrix	Parameter	N	% REC	N RPD Limit	Units
, 	EPA 325.3	Non-Spec Water	CHLORIDE			1 0 - Z	
	EPA 330.5	Non-Spec Water	CHLORINE			7 0 - 3	
	EPA 335.1	Non-Spec Water	CYANIDE, AMENABLE			2 0 - 26	
į	EPA 335.2	Non-Spec Water	CYANIDE	50	64 - 121	51 0 - 22	
		3/ \$/\$	CYANIDE	8 .	52 - 113,	10-0	
	EPA 335.3	Drinking Water	CYANIDE	1	100 - 100		
		Non-Spec Water	CYANIDE	ííÿ	67 - 129	18 0 - 47	
		S/S/S	CYANIDE	4	55 - 129		
	EPA 340.2	Drinking Water	FLUORIDE	5	67 - 122		*****
Ì		Non-Spec Water	FLUORIDE	25	59 - 125	23 0 - 39	
		\$/\$/\$	FLUORIDE	1	79 - 79	1 0 - 20	· · · · · · · · · · · · · · · · · · ·
	EPA 350.1	Non-Spec Water	NITROGEN, AMMONIA	8	67 - 132	8 0 - 11	
	EPA 350.3	Non-Spec Water	NITROGEN, AMMONIA	74	64 - 124	69 0 - 20	
		S/S/S	NITROGEN, AMMONIA	1 0	0 - 181	14 0 - 31	
	EPA 351.2	Non-Spec Water	NITROGEN, KJELDAHL	7	59 - 147	8 0 - 16	
	EPA 351.3	Non-Spec Water	NITROGEN, KJELDAHL	1	100 - 100	1 0 - 0	and the second section of the sectio
1		S/S/S	NITROGEN, KJELDAHL	!1 .	66 - 139_	14 0 - 11	
	EPA 353.1	Drinking Water	NITROGEN. NITRATE	3	82 - 108	2 0 - 26	distribution in research
i		Non-Spec Water		5	85 - 113	5 0 - 27	
			NITROGEN, NITRATE-NITRITE	4	86 - 140	40-4	
1		\$/\$/\$	NITROGEN, NITRATE	1	90 - 90	1 0 - 0	
	EPA 353.2		NITROGEN, NITRATE		120 - 120	3 0 - 15	and the second s
		Non-Spec Water	NITROGEN, NITRATE NITROGEN, NITRATE-NITRITE	37 18	69 - 130 89 - 128	59 0 - 15 11 0 - 17	
}			NITROGEN, NITRITE			20-20	
,							

N = Sample count

REC = Percent Recovery (observed/actual+100)

RPD = Relative Percent Difference (2*(rep1-rep2)/(rep1+rep2)+100)

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Quality Assurance Objectives for Location ALL PERIOD: 01 -APR-91 THRU 30-SEP-91

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		t .				rage 7 or 30
Method	Matrix	Parameter	N	% REC	N RPD Limit	Units
EPA 353.2	S/S/S	NITROGEN, NITRATE			3 0 - 14	
		NITROGEN, NITRITE		77 - 135_	20-27	
EPA 353.3	Non-Spec Water	NITROGEN, NITRATE	6	79 - 121	15 0 10	
	•	NITROGEN, NITRATE-NITRITE	9	85 - 113	11 0 = 30	
EPA 354.1	Non-Soas Haten	NITROGEN, NITRITE	1	97 - 97		
E	Holl Spec water	MITRODEN, MITRITE		31 - 31		
EPA 365.1	Non-Spec Water	ORTHO-PHOSPHORUS	•		20-4	
EPA 365.2	Non-Spec Water	ORTHO-PHOSPHORUS	12	69 - 128	28 0 - 18	
	opte natu.	PHOSPHORUS	74	67 - 129	99 0 - 19	- · · · · · · · · · · · · · · · · · · ·
			_			
	S/S/S	ORTHO-PHOSPHORUS PHOSPHORUS	5 3	87 - <u>117</u> 64 - 138	7 0 - 42	
			-			**************************************
EPA 365.4	Non-Spec Water	PHOSPHORUS	8	77 - 125	4 0 - 16	
EPA 375.4	Drinking Water	SULFATE	7	83 - 112	7 0 - 12	
!			1			
1	Non-Spec Water	SULFATE	50	79 - 127	55 0 - 10	
	\$/\$/\$	SULFATE	4	96 - 121	3 0 - 3	
: EPA 376.1	Non-Spec Water	CIN ETDE	1	82 - 82	4 0 - 0	
Era Jit.	Hon-spec water	SULFIDE	'	82 - 82	40-0	
.	S/S/S	SULFIDE	5	99 - 107	10-2	
EPA 377.1	Non-Spec Water	SULFITE	٤	104 - 109	1 0 - 0	
	•		-			
	S/ S /S	SULFITE	2	91 - 112	1 0 - 5	
EPA 405.1	Non-Spec Water	BIOCHEMICAL OXYGEN DEMAND		4.4	157 0 - 43	
						
EPA 410.2	Non-Spec water	CHEMICAL DXYGEN DEMAND			1 0 - 28	
EPA 410.3	Non-Spec Water	CHEMICAL DXYGEN DEMAND	1	104 - 104	1 0 - 32	
ÉPA 410.4	Drinking Water	CHEMICAL DXYGEN DEMAND	1	96 - 96	1 0 - 14	
	ar amaing wetter	AUTHENIE ON LACK AFHUMA	1	70 70		
	Non-Spec Water	CHEMICAL DXYGEN DEMAND	38	75 - 126	20 0 - 33	
	S/S/S	CHEMICAL DXYGEN DEMAND			6 0 - 13	
EPA 413.1	Non-Spec Water	OIL AND GREASE			19 0 - 13	

N : Sample count

N = Sample count
REC = Percent Recovery (observed/actual+100)
RPO = Relative Percent Oifference (2+(rep1-rep2)/(rep1+rep2)+100)
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EMS HERITAGE LABORATORIES

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Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 FHRU 30-SEP-91 Revision 0 Nov 1, 1991 Page 10 of 38

Hethod	Matrix	Parameter	N	% REC	N	RPD	Linit	Units
EPA 413.1	9/\$/\$	OIL AND GREASE			1 0	- 24		
EPA 415.1	Drinking Water	TOTAL ORGANIC CARBON (TOC)	Ē	109 - 109	1 0	- 9		
	Non-Spec Water	TOTAL ORGANIC CARBON (TOC)	10	81 - 116	11 0	- 9		
EPA 418.1 (MOD)	S/S/S	HYDROCARBONS	12	60 - 121	10 0	- 36		-
EPA 420.1	Non-Spec Water	PHENGL	8	28 - 159	12 0	- 38		
	5/5/5	PHENOL					· · · · · · · · · · · · · · · · · · ·	-
EPA 420.2	Drinking Water	PHENOI	1	95 - 95			-	
	Non-Spec Water		53		13 0	- 27		
	·		33	12 - 119	•			***************************************
	S/S/S	PHENOL						· · · · · · · · · · · · · · · · · · ·
EPA 425.1		SURFACTANTS, MBAS	3	2 - 176	3 0	- E0		
EPA 502.2	Brinking Water	BENZENE BROMODICHLOROMETHANE			2 0	- 39 - 1	-	#*************************************
		BROMOFORM CHLOROFORM			2 0	- 38		
		DIBROMOCHLOROMETHANE			i o	- 3		
: {		1,1-DICHLOROETHANE Dichloromethane (methylene chloride)		—ha	1 0	_		
		ETHYL BENZENE M/P-xylene			20			
		O-XYLENE Toluene			1 0 1 0	- 7 - 10		
		TOTAL TRIMALOMETHANES			1 0	- 6 08 -		
	Non-Spec Water	BENZENE			_	13	···	
· !	non open metal	ETHYL BENZENE M/P-XYLENE			1 0	- 10		
EPA 504	maianiaa inii							
EFR 344	Brinking Water	1,2-DIBROHO-3-CHLOROPROPANE (DBCP) 1,2-DIBROHOETHANE (EDB)	2 2	96 - 101 46 - 114	1 0	- 20		
	Non-Spec Water	1,2-D1BROMO-3-CHLOROPROPANE (DBCP) 1,2-D1BROMOETHANE (EDB)	<u>-</u> . 6	63 - 129 69 - 125	3 0 3 0	- 26 - 30		
EPA 507	Non-Spec Water			120 - 125	1 0			
İ								

N = Sample count

REC = Percent Recovery (observed/actual+100)

RPD = Relative Percent Difference (2*(rep1-rep2)/(rep1+rep2)*100)

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Table ... 1

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 THRU 30-SEP-91

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										Lage	11 01 30
Method	Matrix	Parameter	N	7	K REC	N	R	PD	Limit	Units	
	_ 		· · ·								
EPA 507	Non-Spec Water	ATRATON	2	0	- 328	1	0 -	128			
	•	ATRAZINE	2	0	- 199	-	_				
		BROMACIL	-	ō	- 248		0 -				
1						!		9!			
		CHLDROPROPHAN	5		- 213	1	0 -	40			
Ì		CYCLOATE	2	93	- 119	1	0 -	6			
		DIPHENAMIO	5	42	- 145	2	0 -	1			
		DISULFOTON	2	57	- 133	i	0 -	19			
		EPTC (ERADICANE)	5		- 126	è	0 -	24			
			-	0		_	-	-7			
		ETHOPROP		•	- 304	. 2	0 -	- ! !			· · · · · · · · · · · · · · · · · · ·
		HEXAZINONE	2	32	- 160	1	0 -	31			
		MERPHOS	2	31	- 183	1	0 -	34			
		HEVINPHOS	4	0	- 221	1	0 -	20			
		MOLINATE	Ċ	98	- 132	- :	ō -				
							-				
		PROMETON (PRAMITOL)	2		- 174	1	0 -	28			
		PROMETRYN (CAPAROL)	5	40	- 135	, è	0 -	5 .			
		PRONAMIDE	2	81	- 131	1	0 -	12			
i		PROPAZINE (MILOGARD)	6	0	- 207	2	0 -	4			
		STIROFOS		_	- 136		0 -	Ė			
			-	49		į	= .				•
		TERBUTRYN	>		- 125	2	•	10			
		TRIADEMEFON	6	0	- 248	2	0 -	8			
					4.5	4.4	g	. = = -			
EPA 508	Non-Spec Water	ETHALFLURALIN	14	23	- 52	7	0 -	26			
		FLUCHLORALIN	14	35	- 53	7	0 -	11			
		PENDIMETHALIN	14	37	- 63	6	0 -	14			
		PROFLURALIN	14	36		7	Ö -	٤ō			· •
			14	28	- 92	-	0 -				
		PROPACHLOR	14	2.0	- 76	•	v -	15			
EPA 515.1	Non-Spec Water	2,4-DB	16	14	- 167	7	0 -	95			
EIA 313.1	Holl-spec water		-	_		-	•	. –			
		DCPA (DACTHAL)	16		- 260	7	0 -	106			
		2,4-DICHLOROPHENOXYACETIC ACID	16	9	- 172	5	0 -	14.			
		4-NITROPHENOL	1.4	0	- 182	6	0 -	143			
		PICLORAM	15	0	- 180	7	0 -	115			
		2,4,5-T	16			ż	0 -	27			
								_=-			
		2,4,5-TRICHLOROPHENDXYACETIC ACID (SILVE	16	30	- 172	6	0 -	26			
		X)									1
EPA 524.2	Non-Spec Water	BENZENE	16		- 114	8	0 -	21			1
		BROMOBENZENE	16	75	- 115	8_	0	30			
		BROMOCHLOROMETHANE	16		- 125	8	0 -	18			
		BROMODICHLOROMETHANE	16		- 115	á	ā -	14			}
		BROMOFORM	16		- 126	a A	ă -	19			
		2 12				🦻	<u>*</u>				
		BROMOMETHANE	16		- 110	•	0 -	8			ĺ
		CARBON TETRACHLORIDE	16	_	- 118	7	0 -	6			
		CHLOROBENZENE	16	77	- 117	. 8	0 -	22			
		CHLOROETHANE	15	63	- 116	8	0 -	48			
			_		· · · -	•		-			

N : Sample count

REC = Percent Recovery (observed/actual+100)

RPO = Relative Parcent Oifference (2*(repi=rep2)/(repi+rep2)*100)

(1pa. Soap)

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Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 [HRU 30-SEP-91

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Method	Matrix	Parameter	N	X REC	N	RPD Limit	Units
EPA 524.2	Non-Spec Water	CHLORDFORM	16	77 - 110	8 0	- 12	
	-	CHLOROMETHANE	16	64 - 121	7 0	- 80	
		2-CHLOROTOLUENE (O-CHLOROTOLUENE)	16	73 - 113	8 0	- 16	
		4-CHLOROTOLUENE (P-CHLOROTOLUENE)	16	62 - 136		- 27	
		CIS-1,2-DICHLOROETHENE	16	73 - 116	8 0	- 16	
		DIBROMOMETHANE	16	76 - 124	7 0	- 19	
		1,3-DICHLOROBENZENE (M-DICHLOROBENZENE)	16	75 - 116	8 0	- 16	
		1,2-DICHLOROBENZENE (D-DICHLOROBENZENE)	16	76 - 120	8 0	- 88	
		1,4-DICHLOROBENZENE (P-DICHLOROBENZENE)	16	66 - 126	8 0	- 21	<u> </u>
		DICHLORODIFLUOROMETHANE	15	60 - 99	7 0	- 9	
		1,1-DICHLOROETHANE	16	72 - 114	6 0	- 6	
		1,2-DICHLORDETHANE	16	75 - 122	8 0	- 63	
		1,1-DICHLOROETHENE	16	58 - 117	8 0	- 28	
		DICHLOROMETHANE (METHYLENE CHLORIDE)	16	51 - 158	8 0	- 67	
		1,2-DICHLOROPROPANE	16	79 - 116	8 0	- 19	
		2,2-DICHLOROPROPANE	16	55 - 109	8 0	- 17'	
		1,3-DICHLOROPROPANE	16	82 - 114	8 0	- 12	
		1,1-DICHLOROPROPENE	16	65 - 119	8 0	- 14	***************************************
		ETHYL BENZENE	16	71 - 115	8 0	- 28	
		FLUDROTRICHLOROMETHANE	16	0 - 188	8 0	- 102	
		HEXACHLOROBUTADIENE	16	45 - 141	8 0	- 39	
		ISOPROPYLBENZENE (CUMENE)	16	62 - 123	6 0	- 4	
		4-ISOPROPYLTOLUENE (P-ISOPROPYLTOLUENE)	16	62 - 123	8 0	- 19	
		N-BUTYLRENZENE	16	66 - 115	8 0	- 19	
		N-PROPYLBENZENE	16	66 - 120	7 0	- 19	_
		NAPHTHALENE	16	71 - 140	7 0	- 16	
		O-XYLENE	16	78 - 111	7 0	- 8	
		SEC-BUTYLBENZENE	16	65 - 121	8 0	- 20	
		STYRENE	1.4	80 - 107	7 0	- 8	
		TERT-BUTYLBENZENE	16	43 - 133	8 0	- 7	
		1,1,1,2-TETRACHLOROETHANE	16	73 - 117	7 0	- 7	
		1,1,2,2-TETRACHLOROETHANE	16	81 - 124	8 0	- 22	
		TETRACHLOROETHENE	16	59 - 124	8 0	- 20	
		TOLUENE	16	78 - 110	8 0	- 15	
		TRANS-1,2-DICHLOROETHENE	16	66 - 116	8 0	- 20	
		1,2,3-TRICHLOROBENZENE	16	68 - 138	9 0	- 31	
		1,2,4-TRICHLOROBENZENE	16	69 - 133	8 0	- 17	
		1,1,1-TRICHLORGETHANE	16	66 - 117	8 0	- 18	
		1,1,2-TRICHLOROETHANE	16	79 - 119	8 0	- 21	
		TRICHLOROETHENE	16	72 - 111	8 0	- 21	
		1,2,3-TRICHLOROPROPANE	16	72 - 135	8 0	- 33	
		i,2,4-trimethylbenzene	16	53 - 133	7 0	- 80	
		1,3,5-TRIMETHYLBENZENE	16	41 - 132	8 0	- 12	
		VINYL CHLORIDE	16	68 - 123	80	- 14	
					.,	···	
EPA 525	Drinking Water	ACENAPHTHYLENE	1	36 - 36			

N = Sample count

N = Sample count

REC = Percent Recovery (observed/actual*100)

RPD = Relative Percent Difference (2*(rep1-rep2)/(rep1+rep2)*100)

(qao2.sq1)

EMS HERTTAGE LABORATORIES Tab. . . . 1

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 THRU 30-SEP-91

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Method	Matrix	Parameter	N		X REC	N	RPD	Limit	Units	_
EPA S25	Drinking Water	ALACHLOR	1	86	- 86	7777				
		ALDRIN	1	76	- 76					
		ALPHA-CHLORDANE	1	78	- 78					
		ANTHRACENE	1	38	- 38					
		ATRAZINE	1	82	- 82					
		BENZ (A) ANTHRACENE	1	80	- 80					
		BENZO(A)PYRENE	1	0	- 0					
		BENZO (B) FLUORANTHENE	1	220	- 220					
		BENZO(G,H,I)PERYLENE	1	260	- 260					
		BENZO(K)FLUDRANTHENE	1		- 550					
		(BENZYLBUTYLPHTHALATE) BUTYLBENZYLPHTHAL	i	88						
		ATE	•							
		BIS(2-ETHYLHEXYL)PHTHALATE	•	98	- 98					
		2-CHLOROBIPHENYL		88	- 88					
		CHRYSENE	•	98	~ ~	÷ ÷		1	· · ·	
		DI(2-ETHYLHEXYL)ADIPATE		92						
		DI-N-BUTYLPHTHALATE		88			-			
		DIBENZ(A, H)ANTHRACENE	•	300 84						
		2,3-DICHLOROBIPHENYL		•						
		DIETHYLPHTHALATE	1	86						
		DIMETHYLPHTHALATE	3	88						
		FLUDRENE	1	90	- 90					
		GAMMA-CHLORDANE		80	- 80					
		HEPTACHLOR	1		- 108					
		HEPTACHLOR EPOXIDE	1	80						
		2,2',3,3',4,4',6-HEPTACHLOROBIPHENYL	1	74						
		HEXACHLOROBENZENE	1	80	- 80					
		2,2',4,4',5,6'-HEXACHLOROBIPHENYL	1	B 0	- 80					
		HEXACHLOROCYCLOPENTADIENE	1	68	- 68					
		INDEND(1,2,3-CD)PYRENE	1		- 380					
		2,2',3,3',4,5',6,6'-OCTACHLDROBIPHENYL	1	72						
		2,2',3',4,6-PENTACHLOROBIPHENYL	1	78	- 78					
		PENTACHLOROPHENDL	1	108	- 108					
		PHENANTHRENE	1	84	- 84				<u> </u>	
		PYRENE	1	78	- 78					
		SIMAZINE (PRINCEP)	1	84	- 84					
		2,2',4,4'-TETRACHLOROBIPHENYL	1	82	- 82					
		TRANS-NONACHLOR	1	82	- 82					
		2,4,5-TRICHLOROBIPHENYL	1	84	- 84					
	Non-Spec Water	ACENAPHTHYLENE	21	17	- 126	8 0	- 13			
	•	ALACHLOR	11		- 126	5 0			· 	
		ALDRIN	9	57	- 98	4 0	- 29			
		ALPHA-CHLDRDANE	9	43	- 113	4 0	- 38			ı
		ANTHRACENE	Žİ		- 111	10 0	- 48			

N : Sample count

REC = Percent Recovery (observed/actual+100)
RFO = Relative Percent Difference (2+(rep1-rep2)/(rep1+rep2)+100)

(1pa. Sosp)

EMS HERITAGE LABORATORIES

Table 5.1

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91_IHRU_30-SEP-91

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Method Parameter % REC Units _____ Non-Spec Water ATRAZINE EPA 525 43 - 112 15 BENZ(A)ANTHRACENE 12 53 - 121 BENZO(A)PYRENE -_18 21 0 - 149 BENZO(B)FLUDRANTHENE 27 - 157 21 BENZOIG, H, I) PERYLENE 0 - 162 BENZO(K)FLUORANTHENE 32 - 135 21 (BENZYLBUTYLPHTHALATE) BUTYLBENZYLPHTHAL 39 - 165 100-24 25 AIE BIS(2-ETHYLHEXYL)PHTHALATE 15 2-CHLOROBIPHENYL 40 - 105 23 10 CHRYSENE 45 - 127 D1(2-ETHYLHEXYL)ADIPATE 1 - 208 DI-N-BUTYLPHTHALATE 23 42 - 156 43 DIBENZ(A, H) ANTHRACENE 21 19 - 155 2,3-DICHLOROBIPHENYL 23 42 - 114 42 - 138 DIETHYLPHTHALATE DIMETHYLPHTHALATE 22 41 - 124 ENDRIN 29 - 184 FLUDRENE 47 - 113GAMMA-BHC (LINDANE) 51 - 119 GAMMA-CHLORDANE 38 - 117 **HEPTACHLOR** 17 - 167HEPTACHLOR EPOXIDE 49 - 134 2,2',3,3',4,4',6~HEPTACHLOROBIPHENYL 23 27 - 123 29 HEXACHLOROBENZENE 23 39 - 11337 - 1132,2',4,4',5,6'-HEXACHLOROBIPHENYL 23 HEXACHLOROCYCLOPENTADIENE 23 0 - 125 INDENO(1,2,3-CD)PYRENE 21 0 - 186METHOXYCHLOR 9 24 - 154 2,2',3,3',4,5',6,6'-OCTACHLOROBIPHENYL 23 31 - 115 10 0 2,2',3',4,6-PENTACHLOROBIPHENYL 23 45 - 111 PENTACHLOROPHENOL 0 - 121 11 21 PHENANTHRENE 54 - 114 PYRENE 21 20 - 140 SIMAZINE (PRINCEP) 10 52 - 95 2,2',4,4'-TETRACHLOROBIPHENYL 23 42 - 115 23 41 - 116 TRANS-NONACHLOR 2,4,5-TRICHLOROBIPHENYL ____17 ... 43 - 107EPA 531.1 Non-Spec Water **ALDICARB** 36 - 111 16 ALDICARB SULFONE 16 52 - 128 _.15 - 13 ALDICARB SULFOXIDE 16 51 - 129 59 - 11380-12 CARBARYL 49 - 126 ... CARBOEURAN. 16 **3-HYDROXYCARBOFURAN** 45 - 132

N = Sample count

REC : Percent Recovery (observed/actual*100)

RPD = Relative Percent Difference (2*(rep1-rep2)/(rep1+rep2)*100)______

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Quality Assurance Objectives for Location ALL PIRIOD: 01-APR-91 THRU 30-SEP-91

Revision 0 Nov. 1,1991

		FIRTOR OF MER - 21 INK	0 34-25E	- 71			Page 15 of 38
Hethod	Matrix	Parameter	N	% REC		Limit	Units
EPA 531.1	Non-Spec Water	METHIOCARB	14	56 - 88	8 0 - 23		
2 22		METHONYL	16	45 - 125	80-10		
		OXAMYL	16	41 - 134	8 0 - 17		
		PROPOXUR (BAYGON)	16	49 - 126	89 - 28		
EPA 548.1	Non-Spec Water	ENDOTHALL	2	79 - IB9	1 0 - 12		
EPA 601/602 MOD	Drinking Water	BENZENE	12	35 - 143	5 0 - 74		
		DI-1SOPROPYL ETHER (IPE)	2	34 - 141			
		1,2-DIBROMOETHANE (EDB)	8	70 - 132	3 0 - 50		
		1,2-D1CHLORDETHANE	15	18 - 182	60 - 87		
		ETHYL BENZENE	12	56 - 124	5 0 - 37		***************************************
		M/P-XYLENE	, , ,	74 - 112	3 0 - 31		
		METHYL-T-BUTYL ETHER (MTBE)	ż	59 - 124	3 0 - 78		
		0-XYLENE	7	72 - 113			
		TOLUENE	12	53 - 135	3 0 - 35 5 0 - 62	r	
		IOLUENE	12	23 - 132	50 - 62		
	Non-Spec Water		1	97 - 97			
		DI-ISOPROPYL ETHER ([PE)					
		1,2-DIBROMOETHANE (EDB)	1	139 - 139			
		1,2-DICHLOROETHANE	1	96 - 96			
		ETHYL BENZENE	1	99 - 99			
		M/P-XYLENE	1	99 - 9 9			
		METHYL-T-BUTYL ETHER (MTBE)	1	65 - 65			
		0-XYLENE	1	100 - 100			
		TOLUENE	i	98 - 98			
EPA 602	Drinking Water	BENZENE	7	31 - 146	3 0 - 58		
		ETHYL BENZENE	7	43 - 144	3 0 - 18		
		METHYL-T-BUTYL ETHER (MTBE)	5	0 - 229	<u> 8 0 - 68</u>		
		TOLUENE	7		3 0 - 25		
			•	77 172	J V LE		
	Non-Spec Water	BENZENE	į	97 - 97			
		DI-ISOPROPYL ETHER ([PE)	1	99 - 99			
		ETHYL BENZENE	1	115 - 115			
		METHYL-T-BUTYL ETHER (MTBE)	1	98 - 98			
		TOLUENE	i	113 - 113			
		XYLENES (O/M/P-XYLENE)		116 - 116			-
ËPÄ 608	Drinking Water	ENDRÍN	2	110 - 110	1 0 - 0		
	• • •	GAMMA-BHC (LINDANE)	2	60 - 76	10-6		
	Non-Spec Water	ALDRÍN		FE _ 449	2 4 - 34		
	HOH-Spec Hater		4	55 - 112	20-34		
		CHLORDANE	1	78 - 78			
		4,4'-ODT	18	76 - 137	8 9 - 15		
		DIELDRIN	18	12 - 137	9 0 - 28		

N = Sample count

REC = Percent Recovery (observed/actual+100)

RPD = Relative Percent Difference (2+(rep1-rep2)/(rep1+rep2)+100)

(qao2.sq1)

Table 5.1

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 THRU 30-SEP-91 Revision 0
Nov. 1,1991

Page 16 of 38 bodish Matrix Parameter % REC Units RPD Limit **EPA 608** Non-Spec Water 77 - 1763 0 - 25 GAMMA-BHC (LINDANE) 69 - 123HEPTACHLOR 79 - 122 HEPTACHLOR EPOXIDE METHOXYCHLOR PCB AROCLOR 1242 PCB AROCLOR 1248 73 - 92 30 - 128 PCB AROCLOR 1260 48 TOXAPHENE 124 - 124 5/5/5 PCB AROCLOR 1260 50 - 174EPA 624 Non-Spec Water ACROLEIN 805 - 0 42 - 169 ACRYLONITRILE 11 5 0 - 71 BENZENE 75 - 133BROMODICHLOROMETHANE 19 83 - 146 4-BROMOFLUOROBENZENE B7 - 111BROMOFORM 72 - 138 19 33 BROMOMETHANE 18 50 - 14827 CARBON TETRACHLORIDE 19 80 - 143CHLOROBENZENE 16 88 - 116 CHLOROETHANE 64 - 14325 2-CHLOROETHYLVINYLETHER 15 0 - 19331 CHLOROFORM 17 102 - 123 CHLOROMETHANE 18 22 - 180 41 D1BROMOCHLOROMETHANE 19 79 - 14050-27 1,3-DICHLOROBENZENE (M-DICHLOROBENZENE) 69 - 138 69 - 138 1,2-D1CHLOROBENZENE (O-DICHLOROBENZENE) 3 1,4-DICHLOROBENZENE (P-DICHLOROBENZENE) 3 70 - 13318 1,1-DICHLOROETHANE 83 - 127 1,2-DICHLORDETHANE B1 - 151 17 DICHLOROETHANE-D4 8 82 - 1101,1-DICHLORDETHENE 65 - 1401,2-DICHLORDETHENE (CIS AND TRANS) 17 251 - 18 DICHLORDMETHANE (METHYLENE CHLORIDE) 19 5 - 193 1,2-DICHLOROPROPANE 19 69 - 146 ETHYL BENZENE 20 65 - 133FLUOROTRICHLOROMETHANE 19 71 - 156 20 1,1,2,2-TETRACHLOROETHANE 19 62 - 158TETRACHLOROETHENE 19 77 - 139 TOLUENE 14 86 - 121 7 0 TOLUENE-DB 87 - 110 TRANS-1,3-DICHLOROPROPENE 19 66 - 136 75 - 150 1,1,1-TRICHLORGETHANE 19 60 - 121,1,2-TRICHLOROETHANE 19 71 - 150TRICHLOROETHENE 63 - 134

N = Sample count

REC = Percent Recovery (observed/actual*100)

RPD = Relative Percent Difference (2*(rep1-rep2)/(rep1+rep2)*100)

(qao2.sq1)

THE HERITAGE LANGERTHRIES

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 THRU 30-SEP-91

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Method	Matrix	Parameter	N	% REC	N	RPD	Limit	Units	
EPA 624	Non-Spec Water	VINYL CHLORIDE	0.5	59 - 142	5 0	- 28			
EPA 625	Non-Spec Water	ACENAPHTHENE	şş	49 - 117	QQ	=_17			
		ACENAPHTHYLENE	11	73 - 108	5 0	- 18			
		ANTHRACENE	11	78 - 110	5 0	- 6			
		BENZ(A)ANTHRACENE	11	86 - 124	5 0	- 15			
		BENZIDINE							
		BENZO(A)PYRENE	11	69 - 137	5 0	- 23			
		BENZO(B)FLUORANTHENE	11	50 - 169	5 0	- 29			
		BENZO(G,H,I)PERYLENE	11	15 - 194	5 0	- 58		-	
			- •		5 0				
		BENZO(K)FLUORANTHENE	11	36 - 165		- 47			
		(BENZYLBUTYLPHTHALATE) BUTYLBENZYLPHTHAL Ale	11	42 - 171	5 0	- 16			
		BIS(2-CHLOROETHOXY)METHANE	11	30 - 142	5 0	- 18			
		BIS(2-CHLORDETHYL)ETHER	11	57 - 123	5 0	- 26	,	-	
		BIS(2-CHLOROISOPROPYL)ETHER	11	49 - 142	4 0	- 20			
		BIS(2-ETHYLHEXYL)PHTHALATE	11	0 - 232	5 0	- 35			
		4-BROMOPHENYL-PHENYLETHER	11	77 - 112	5 0	- 26		•	
		4-CHLORO-3-METHYLPHENOL	22		0 0	- 55			
		2-CHLORONAPHTHALENE	10	87 - 103	5 0	- 16			
		2-CHLOROPHENOL	25		0 0	- 45			
		4-CHLORDPHENYL-PHENYLETHER	11	83 - 109	5 0	- 17			
									
		CHRYSENE	11	70 - 122	5 0	= 1			
		DI-N-BUTYLPHTHALATE	12	48 - 136	5 0	- 21			
		DI-N-OCTYLPHTHALATE	11	0 - 239	5 0	- 35			
		DIBENZ(A, H) ANTHRACENE	11	17 - 195	5.0	63			
		1,3-DICHLOROBENZENE (M-DICHLOROBENZENE)	11	36 - 119	5 0	- 88			
		1,2-DICHLOROBENZENE (O-DICHLOROBENZENE)	11	45 - 121	5 0	- 19			
		1,4-DICHLOROBENZENE (P-DICHLOROBENZENE)	18	19 - 119	8 0	- 29			
		3,3'-DICHLOROBENZIDINE	10	13 - 131	5 0	- 45			
		2,4-DICHLOROPHENOL	11	65 - 111	5 0	- 17			
		DIETHYLPHTHALATE	11	79 - 188	5.0.	<u>- 55</u>			
		2,4-DIMETHYLPHENOL	1.1	33 - 92	5 0	- 39			
		DIMETHYLPHTHALATE	11	53 - 151	5 0	- 10			
		4,6-DINITRO-O-CRESOL	11	0 - 160	5 0	- 105		-	
		2,4-D1N1TROPHENOL	11	0 - 170	5 0	- 330		<u></u>	
		2,4-DINITROTOLUENE	55	0 - 175 1	0 0	- 33			
		2,6-DINITROTOLUENE	11	82 - 123	5 0	- 53			
		FLUORANTHENE	11	52 - 124	5 0	- 19			
		FLUORENE	11	76 - 114	5 0	- 15			
		2-FLUOROBIPHENYL	1	89 - 89					
		2-FLUOROPHENOL	i	71 - 71	•				
		HEXACHLOROBENZENE	11	52 - 143	5 0	- 41			
		HEXACHLOROBUTAD1ENE	!!	36 - 125	_	- 47			
		HEXACHLOROCYCLOPENTAD1ENE	11	18 - 142		- 37			
,									

N = Sample count

N = Sample count
REC = Percent Recovery (observed/actual+100)
RPD = Relative Percent Difference (2*(rep1-rep2)/(rep1+rep2)+100)
(qao2.sq1)

EMS HERITAGE LAGORATORIES

Table 5.1

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 THRU 30-SEP-91

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Method	Matrix	Parameter	N	X	RE	С		N	F	PD	Limit	Units	
EPA 625	Non-Spec Water	HEXACHLORDETHANE	11	11	- 1	53		5 () -	37			
	•	INDENO(1,2,3-CD)PYRENE	1.1	28	- 1	73		5 () –	41			
		ISOPHORONE	. 11	69	- 12	21		5 (· •	16			
		N-NITROSO-DIPROPYLAMINE	16	-	- 1			6 (-	9	-		
		N-NITROSODIMETHYLAMINE	1			20		•		•			
		N-NITROSODIPHENYLAHINE	11		- 1			5 () <u>-</u>	29	***************************************		
		NAPHTHALENE	ii		- i			5	•	7			
		NITROBENZENE	11		- 19			5 (-	20			
		NITROBENZENE-D5		68				- '	•				
		2-N1TROPHENOL	11		- "]			5 (-	18			
		4-NITROPHENOL	55		- i				_	109			
		PENTACHLOROPHENOL	50	-	-		•	-	-				
		PHENANTHRENE			- 1			9 (•	13		• • • • • • • • • • • • • • • • • • • •	
		PHENOL	10		- 1				_	19			
			50	-	- 1			7 () –	21			
		PHENOL-D5	. 1	72	-	72		± .	-		·		
		PYRENE	80	29			1	0 () -	18			
		TERPHENYL-D14	1	90		90							
		2,4,6-TRIBROMOPHENOL	1	98		98							
		1,2,4-TRICHLOROBENZENE	55		- 1			•) -	13			
		2,4,6-TRICHLOROPHENDL	11	57	- 1	14		5 () -	27			
	5/5/5	ACENAPHTHENE	4	57	- 1	07		2 (0 -	26			÷ .
		ACENAPHTHYLENE	2	46	- 1	14		1 () -	20			
		ANTHRACENE	2	69	_ (90		1 (-	18			
		BENZ (A) ANTHRACENE	2	38	- j.	40		i () -	27			
		BENZO(A)PYRENE	2	46	- 1	35		1 (· •	23			
		BENZO(B)FLUDRANTHENE	٤	35				1 6	· -	16			
		BENZO(G,H,I)PERYLENE	2		- i				-	40			•
		BENZO(K)FLUORANTHENE	Ē	82				iè	-	Ä			
		(BENZYLBUTYLPHTHALATE) BUTYLBENZYLPHTHAL	ē	69		77		i		3			
		A1E	_	• • • • • • • • • • • • • • • • • • • •		•••				. •		 -	
		BIS(2-CHLORDETHOXY)METHANE	2	37	- (84		1 0) -	18			
		BIS(2-CHLOROETHYL)ETHER	2	· · ·		08		i	-	20			** -
		BIS(2-CHLOROISOPROPYL)ETHER	2	54	- 10	09		1 0	-	16	***************************************		
		BIS(2-ETHYLHEXYL)PHTHALATE	2	61	- 1	82		1 0	-	7			
		4-BROMOPHENYL-PHENYLETHER	ē	40	- 18			1 0	· -	24		-	
		4-CHLORD-3-METHYLPHENOL	2	63 -		92		1 0	–	9			
		2-CHLORONAPHTHALENE	2		- 14			1 0	, <u> </u>	_87			
		2-CHLDRDPHENOL	2		- 11			1 0		18			
		4-CHLOROPHENYL-PHENYLETHER	٤		- i c			1 6	-	A			
		CUDVOENE	-		- ii			1 6		18			
		DI-N-BUTYLPHTHALATE	2	63		79		1 Y		9 6	*******		
		DI-N-DCTYLPHTHALATE	٤	55				1 0		33			
			٤		- 10								
		DIBENZ(A, H)ANTHRACENE	5	•				! .¥	: - -	42	-		
		1,3-DICHLOROBENZENE (M-DICHLOROBENZENE)	2	47 -	- 11	13		1 0	· -	20			

N = Sample count

REC = Percent Recovery (observed/actual+100)

RPD = Relative Percent Difference (2*(rep1-rep2)/(rep1>rep2)*100) (qao2.sq1)

Quality Assurance Objectives for Location ALL PERIOD: 01 APR 91 THRU 30-SEP-91

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Hethod	Matrix	Parameter	N	X REC	N	RPD	Limit	Units
EPA 625	S/S/S	1,2-DICHLOROBENZENE (D-D[CHLOROBENZENE) 1,4-DICHLOROBENZENE (P-D[CHLOROBENZENE)	2	47 - 115 51 - 100	10	- 20 - 34		
		3,3'-DICHLOROBENZIDINE	ž	39 - 55	10			
		2,4-DICHLOROPHENOL	ē	51 - 98	1 0	- 15		
		DIETHYLPHTHALATE	2	78 - 94	10	- '5		
		2,4-DIMETHYLPHENOL	2	46 - 96	1 0	- 11		
		DIMETHYLPHTHALATE	2	82 - 108	i ö	- 6		
		4,6-DINITRO-O-CRESOL	2	0 - 71	1 0	- 9ž		
		2,4-DINITROPHENOL	- ī	25 - 25				
		2,4-DINITROTOLUENE	4	27 - 141	20	- 43		
		2,6-DINITROTOLUENE	2	85 - 93	10	- 2		
		FLUORANTHENE	2	32 - 116	10	- 27		
		FLUORENE	2	88 - 104	1 0	- 14		
		HEXACHLOROBENZENE	2	49 - 122	1 0	- 20		
		HEXACHLOROBUTADIENE	2	50 - 113	1 0	- 18		
		HEXACHLOROCYCLOPENTADIENE	2	5 - 120	10	- 40	1	
		HEXACHLORDETHANE	2	52 - 102	10	- 16		
		INDENO(1,2,3-CD)PYRENE	2	2 - 117	10	- 45		
		ISOPHORONE	2	58 - 87	1 0	- 10		
		N-NITROSO-DIPROPYLAMINE	2	52 - 91	10	- 13		
		N-NITROSODIPHENYLAMINE	2	50 - 105	1 0	17		
		NAPHTHALENE	5	47 - 115	10	- 20		
		NITROBENZENE	2	54 - 101	10.	- 14		
		2-NITROPHENOL	2	65 - 128	10.	?		
		4-NITROPHENOL	5	0 - 179	10	- 76		
		PENTACHLOROPHENOL	2	1 - 85	10.	- 47		
		PHENANTHRENE	2	45 - 126	. ! 0 :	- 14		
		PHENOL	2	51 - 98		- 15		
		PYRENE	4	51 - 116	20			
		1,2,4-TRICHLOROBENZENE	4	49 - 105	50.			
		2,4,6-TRICHLOROPHENDL	5	55 - 94	10-	- 12		
ILMOI	Non-Spec Water	ALUMINUM	40	88 - 107	19 0	- 5		
		ANTIHONY	37	75 - 103		- 7		
		ARSENIC	28	74 - 116	13 0 -	- 11		
		BARIUM	42	87 - 110	190	⁻ 🙎		
		BERYLLIUM	39	81 - 129	17 0 -	- 5		
		CADMIUM	42	71 - 120	20 0 -	_		
		CALCIUM	_ 38	81 - 104	17.0 :			
		CHRONIUM Cobalt	42	82 - 109	20 0 -			
		COPPER	40	84 - 110	19 0 -			
		IRON	39 38	86 - 120 75 - 116	18 0 -		-	
		LEAD	38 62	75 - 116 82 - 111	16 0 -			
		MAGNESIUM	38	82 - 113	17 0 -			
		HANGANESE	40.	87 - 112	18 0 -			
		1777	70	01 115	10 V -			

N : Sample count

REC = Percent Recovery (observed/actual*100)
RFO = Relative Percent Oifference (2*(rep1-rep2)/(rep1+rep2)*100)

(qao2.sql)

THS HERITAGE LABORATORIES

fable 5.1

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 [HRU 30-5EP-91

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Method	Matrix	Parameter	N	X REC	N RPD Limit	Units
ILH01	Non-Spec Water	HERCURY NICKEL POTASSIUM SELENIUM SILVER SODIUM THALLIUM VANADIUM ZINC	28 41 38 26 42 38 65 40 37	76 - 111 82 - 110 87 - 107 75 - 117 85 - 120 84 - 109 70 - 111 82 - 108 77 - 116	14 0 - 7 19 0 - 3 18 0 - 5 13 0 - 15 20 0 - 11 18 0 - 5 30 0 - 11 19 0 - 3 17 0 - 7	
	Oil	ALUMINUM ANTIMONY ARSENIC BARIUM BERYLLIUM CADMIUM CALCIUM CHROMIUM COBALT COPPER IRON LEAD MAGNESIUM MANGANESE NICKEL POTASSIUM SELENIUM SILVER SODIUM THALLIUM VANADIUM ZINC		71 - 79 55 - 63 83 - 104 85 - 102 81 - 115 20 - 144 69 - 77 47 - 133 74 - 74 81 - 89 68 - 110 26 - 142 80 - 80 77 - 85 66 - 66 77 - 85 66 - 66 77 - 85 75 - 90 83 - 91 75 - 75 75 - 83 64 - 64	1 0 - 1 1 0 - 2 1 0 - 5 2 0 - 4 1 0 - 7 2 0 - 3 1 0 - 2 2 0 - 11 1 0 - 0 1 0 - 2 1 0 - 11 3 0 - 4 1 0 - 1 1 0 - 2 1 0 - 1 1 0 - 2 1 0 - 1 1 0 - 2 1 0 - 1 1 0 - 2 1 0 - 1 1 0 - 2 1 0 - 2 1 0 - 2 1 0 - 2 1 0 - 2 1 0 - 2 1 0 - 2 1 0 - 0	
	S/S/S	ANTIMONY ARSENIC BARIUM BERYLLIUM CADMIUM CALCIUM CHRONIUM COBALT COPPER LEAD MAGNESIUM MANGANESE MERCURY	12 40 54 51 45 34 49 51 42 42 47 13	0 - 105 45 - 137 76 - 118 64 - 128 64 - 127 61 - 125 67 - 128 69 - 119 71 - 120 66 - 112 75 - 122 70 - 123	6 0 - 31 19 0 - 28 25 0 - 12 23 0 - 11 22 0 - 16 15 0 - 38 24 0 - 25 25 0 - 11 19 0 - 15 20 0 - 19 23 0 - 19 4 0 - 28 23 0 - 11	

N = Sample count

N = Sample count REC = Percent Recovery (observed/actual+100) RPD = Relative Percent Difference (2*(rep1-rep2)/(rep1+rep2)+100) (qao2.sq1)

idS i i AGE ORA ES I Table > 1

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 THRU 30-SEP-91 Revision ()
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Hethod	Matrix	Parameter	N	% REC	N RPD	Limit	Page 21 of 38 Units
ELMOI	S/3/3	NICKEL	49	70 - 118	25 0 - 19	5	
		POTASSIUM	49	75 - 113	22 0 - 0	3	
		SELENIUM	38	61 - 128	19 0 - 24	,	
		SILVER	48	75 - 120	24 0 - 1:	·	
		SODIUM	52	72 - 120	24 0 -	3	
		THALLIUM	83	60 - 126	36 0 - 11		
		VANADIUM	53	70 - 122	26 0 19	;	
		ZINC	36	70 - 126	18 0 - 24	'	
N105H P4CAM 125	Non-Spec Water	FORMALDEHYDE	Ë	73 - 107	<u>.</u>		
	5/5/5	FORMALDEHYDE			2 0 - 13	3	
SAS	Non-Spec Water	ACIDITY TO PH 11			1 0 - 19	·	
	•	ACRYLAMIDE	2	150 - 155	10-	٠	
		ACRYLIC ACID	25	86 - 124	13 0 - 9	5'	
		BENZOIC ACID	8	110 - 110	10-)	
		BIS-ACRYLAMIDE	2	100 - 113	10:	3	
		DIQUAT	4	59 - 114	20-5	7	
		HYDROQUINDNE	5	85 - 8 9	1 0 -	·	
		4-METHYLPHENOL (P-CRESOL)	2	95 - 95	10	0	
		NITRATE	2	99 - 104	10-	·	
		PARAQUAT	3	31 - 92	20-71	·	
		PHENOL	2	95 - 95	10- ()	
		UREA	5	12 - 177	10-58	2	
	011	LEAD	2	92 - <u> </u> 113	1.0	3	
	9/\$/\$	ACIDITY TO PH 11			40 0 - 16	.	
		ETHYLENE GLYCOL	E	20 - 75	10-25	5	
		HYDROFLUORIC ACID			3 0 - 34		
		ISOFENPHOS	2	99 - 112	10- 2	2	
		4-METHYLPHENOL (P-CRESOL)	2	68 - 136	1 0 - 16	5	
		MOISTURE CONTENT	-		3 0 - 10)	
		NITRIC ACID			4 0 - 17	,	
		PH			10-)	
		PHENOL	2	97 97	10-) 	
		SOLIDS	_		17 0 - 19	,	
		TETRAMETHYLTHIURAM DISULFIDE	2	79 - 129	1 0 - 18		
SM 320A 14TH ED	Drinking Water	CHLORIDE	1	114 - 114	10-5	·	
	Non-Spec Water	CHLORIDE	1	118 - 118	1 0 - 4		
SM 405	\$/\$/\$	BROMIDE		92 - 118	1 9 - 19)	

N : Sample count

RFC = Percent Recovery (observed/actual+100)

RFO : Relative Fercent Oifference (2+(rep1-rep2)/(rep1+rep2)+100)

(qao2.sq1)

TMS HERITAGE LABORATORIES

fable 5.1

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 THRU 39-SEP-91 Section 5.0 Revision 0 Nov. 1, 1991 Page 22 of 38

Hethod	Matrix	Parameter	N	% REC	N	RPD	Limit	Units	
SM 407A	Drinking Water	CHLORIDE	 1	100 - 100			*******		
	Non-Spec Water	CHLORIDE	11	83 - 124	14 0	_ 6			
<u> </u>	S/S/S	CHLORIDE	4	92 - 120	15 0	- 11			
; SM 503B	S/S/S	OIL AND GREASE	2	56 - 124	1 0	- 18			
SM 503E	5/5/5	PETROLEUM HYDROCARBONS	49	76 - 108	29 0	- 22			
J. 2032	u. u. u	TETROLLON IN BROOMS DONG	• •						
SW846-1010	S/S/S	IGNITABILITY			19 0	- 8			
SW846-6010	Non-Spec Water	ALUHINUM	15	72 - 124	10 0	- 35		- ·	
1		ANT I HONY	30	65 - 100	70	- 6			
!		BARIUM	149	82 - 111	85_0	8			
1		BERYLLIUM	28	53 - 154	9 0	- 855	·		
i		BORON	70	B1 - 127	14 0	- 50			
		CADMIUM	137	59 - 124	52 0	- 14		~	
		CALCIUM	37	79 - 104	27 0	- 4			
		CHRONIUM	143	77 - 111	52 0	- 19			
		COBALT	25	77 - 107	7 0	. = 3			
		COPPER	58	84 - 112	16 0	- 290			
:		IRON	111	72 - 114	69 0	- 30			
1		LEAD	17	75 - 111	3 0	- 24			
		LITHIUM	40	89 - 110	19 0	- 6			
!		MAGNESIUM	35	85 - 109	25 0	- 3			
!		HANGANESE	112	82 - 114	57 0	4			
į		MOLYBDENUM	40	85 - 107	19 0	- 4			
		NICKEL	126	79 - 111	57 0	- 27			
		POTASSIUM	25	85 - 109	16 0	6			
1		SILICON			4 0	- 5			
1		SILVER	147	80 - 121	47 0	- 11		·	
		SODIUM	124	81 - 113	73. 0	5			
!		STRONTIUM	40	78 - 109	18 0	- 4			
1		THALLIUM	1	89 - 89					
1		TIN	34	80 - 105	16 0	- 4			
4		TITANIUM	38	86 - 111	18 0	- 4			
1		VANADIUM	28	78 - 103	10 0	- 40			
		ZINC	 53	81 - 106	35_0				
	011	ALUMINUM	4	37 - 132	6 0			<u> </u>	
		BARIUM	 16	. 77 - 110		13.			
		BERYLLIUM	4	40 - 151					
		CADHIUM	13	52 - 108	8 0	- 26			
1		CALCIUM	. 3	42 - 112	7 9	12.		· -	_
1		CHRONIUM	21	37 - 142	18 0	- 16		Child Advant-1-1	

N = Sample count

REC = Percent Recovery (observed/actual+100)

RPD = Relative Percent Difference (2*(rep1-rep2)/(rep1+rep2)*1001

(1pa. Soap)

Table 5 1

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 THRU 30-SEP-91

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Hethod	Matrix	Parameter	N	% REC	N	RPD	Limit	Units
SW846-6010	011	COPPER	2	42 - 136		- 18		
		IRON	1	68 - 68		- 11		
		LEAD	. 13	. 43 - 197.		- 50		
		LITHIUM	2	89 - 97		- 1		
		MAGNESIUM	4	49 - 111	5 0	- 30		
		HOLYBDENUM	2	71 - 79	1 0	- 8		
		NICKEL	4	58 - 100	2 0	- 4		- · · · · · · · · · · · · · · · · · · ·
		S1L1CON			5 0	- 16		
		SILVER	10	63 - 150		= - 53		
		SODIUM	4	54 - 113	6 0	- 61		
		STRONTIUM	8	80 - 96		- 3		
		TIN	1	68 - 68		- 57		
		TITANIUH	5	71 - 92		- 10		
		ZINC	1	124 - 124	7 0	- 13		
	5/5/5	ALUMÍNUM	5	95 - 102	14 0	- 15	1	
		ANTIHONY	5	45 - 87	1 0	- 0		
		BARIUM	74	72 - 121	54 0	- 24		
•		BERYLLIUM	27	52 - 130	9 0	- 1		
		BORON	4	82 - 125	8 0	- 150		
		CADHIUM	64	47 - 127	24 0	- 40		
		CALCIUM	12	67 - 116	21 0	- 18		
		CHRONIUM	82	56 - 132	-	- 36		 _
		COBALT	11	71 - 105		- 5		
		COPPER	52	72 - 121		- 55		
l.		IRON	6	60 - 126		- 149		
		LEAD	82	52 - 128		- 39		
		LITHIUM	53	82 - 112		- 7		
		MAGNESIUM	21	71 - 125		- 26		
		MANGANESE	8	57 - 135	12 0			
		MOLYBDENUM	57	76 - 104	25 0			
		NICKEL	53	65 - 121	52 0			
		POTASSIUM	12	74 - 116		- 6		
		SILICON		_		- i		
		SILVER	64	64 - 123	15 0	- 21		
		SODIUM	20	80 - 110	_	- 9		
		STRONTIUM	53	75 - 112		- 10		
		THALLIUM	13	67 - 94				
		TIN	44	67 - 117	20 0	9		
		TITANIUM	29	0 - 199	13 0	- 39		The state of the s
		VANADIUM	14	66 - 126		- 11		
		ZINC	32	61 - 122		- 149		
SH846-7020	Non-Spec Water	ALUHINUH	13	0 - 258	7 0	- 6		
	5/5/5	ALUHINUH	4	50 - 174	6 0	- 27		

N : Sample count

REC = Percent Recovery (observed/actual+100)

REC = Percent Recovery tobserved/actual-100, RFO = Relative Percent Difference (2*(rep1-rep2)/(rep1+rep2)+100) (qao2.sq1)

LMS HERITAGE LAHURATURIES Fable 5.1

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 FHRU 30-SEP-91 Section 5.0 Revision 0 Nov. 1,1991

Hethod	Matrix	Parameter	N	X REC	N RPD Limit	Page, 24 of 38
SU846-7040	S/S/S	ANTIHONY	4	48 - 136		
SW846-7041	Non-Spec Water	ANT I MONY	5	56 - 125	2 0 - 49	· · · · · · · · · · · · · · · · · · ·
	011	ANTIMONY	4	55 - 114		
	S/S/S	ANTIMONY	5	71 - 97	1 0 - 1	who be to make
SU846-7060	Non-Spec Water	ARSENIC	171	70 - 123	72 0 - 16	
	011	ARSENIC	21	33 - 189	4 0 - 26	
	S/S/S	ARSENIC	158	47 - 139	94 0 - 23	
SU846-7080	Non-Spec Water	BARIUH	ē 9	71 - 131	13 0 - 28'	
	011	BARIUM	2	68 - 118	1 0 - 8	• Mindrey (Mindre)
!	S/\$/S	BARIUM	60	55 - 150	85 0 - 18	
SU846-7090	S/S/S	BERYLLIUM	5	88 - 132		
SH846-7091	Non-Spec Water	BERYLLIUM	4	83 - 95	z o - 9	
	5/5/5	BERYLLIUM	4	44 - 162	3 0 - 20	
SW846-7130	Non-Spec Water	CADMIUM	10	69 - 123	4 0 - 21	
	011	CADMIUN	7	61 - 121	3 0 - 16	
1	S/S/S	CADMIUM	118	69 - 124	69 0 - 20	
SW846-7131	Non-Spec Hater	CADMIUN	ã7	56 - 135	17 0 - 20	
[]	011	CADMIUM	8	56 - 134		
	S/S/S	CADMIUM	5	45 - 136	7 0 - 11	
SU846-7140	Non-Spec Water	CALCIUN	5	99 - 99	5 0 - 17	
SW846-7190	Non-Spec Water	CHRONIUM	19	72 - 113	12.0 - 10	
	011	CHRONIUM	6	52 - 143	2 0 - 10	
:	S/S/S	СНКОНІЦН	178	45 - 140	125 0 - 25	

N = Sample count

REC + Percent Recovery (observed/actual*100)

RPD = Relative Percent Difference (2*(rep1-rep2)/(rep1+rep2)*100)

(1pa. Soap)

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fable 5.1

Quality Assurance Objectives for Location ALL

PERIOD: 01-APR-91 THRU 30-SEP-91

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Method	Matrix	Parameter	.N	% REC	N RPD Limit	Page 25 of 38
SU846-7191	Non-Spec Water	CHRONIUM	12	73 - 117	3 0 - 17	
SN846-7196	Non spec Water	HEXAVALENT CHROMIUM	7	76 - 124	3 0 - 12	
	5/5/5	HEXAVALENT CHROMIUM			1 018	
SU846-7200	\$/\$/\$	COBALT	6	66 - 130	2 0 - 92	and the second
SH846-7210	Non-Spec Water	COPPER	55	80 - 124	9 0 - 6	-
	5/5/5	COPPER	53	54 - 136	36 0 - 21	
SW846-7380	Non-Spec Water	IRON	18	70 - 122	8 0 - 5	
	\$/\$/\$	IRON	6	37 - 179	9 0 - 18	-
SU846-7420	Non-Spec Water	LEAD	13	89 - 113	7 0 - 7	
	011	LEAD	6	65 - 103	6 0 - 12	
\ t 1	5/5/5	LEAD	181	65 - 134	117 0 - 21	
SH846-7421	Non-Spec Water	LEAD	298	72 - 125	107 0 - 16	· -
! 	011	LEAD	В	51 - 145		
	5/5/5	LEAD	6	62 - 124	26 0 - 19	
SW846-7450	Non-Spec Water	MAGNESIUM	s	80 - 80	5 0 - 13	
	S/S/S	MAGNESIUM	1	88 - 88	1 0 - 12	
SU846-7460	Non-Spec Water	MANGANESE	4	57 - 123	20-5	<u> </u>
! :	S/S/S	MANGANESE	4	59 - 152	5 0 - 16	·
SW846-7470	Non-Spec Water	MERCURY	167	65 - 125	55 0 - 13	
1	S/S/S	MERCURY	Ŝİ	65 - 128	10 0 - 10	
SU846-7471	011	MERCURY	7	69 - 120	2 0 - 0	
	S/S/S	MERCURY	112	60 - 129	33 0 - 26	
SU846-7480	S/S/S	MOLYBDENUM	<u>s</u>	74 - 150		

N : Sample count

REC = Percent Recovery (observed/actual+100)

RCO = Relative forcent Oifference (2*(rep1-rep2)/(rep1+rep2)+100)

EMS HERITAGE LAMORATORIES

fable 5.1

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 THRU 30-SEP-91

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-	Hethod	Hatrix	Parameter	N	% REC	N	RPD	Limit	Units	38
	SW846-7520	Non-Spec Water	NICKEL	14	58 - 128	7	0 - 50			
. !		5/5/5	NICKEL	34	61 - 132	50	 0 - 17			
	SU846-7610	Non-Spec Water	POTASSIUM	4	89 - 92	3	0 - 27			
.		5/5/5	POTASSIUM			2	0 - 3			
	SU846-7740	Non-Spec Water	SELENIUM	141	64 - 134	46	0 - 19			-
		011	SELENIUM	9	41 - 117	4	0 - 6			-
		S/S/S	SELENIUM	100	39 - 146	27	0 - 28		-	
!	5W846-7760	Non-Spec Water	SILVER	3	72 - 141	ī				
į		011	SILVER	2	96 - 96	1	0 - 0		·	
		s/s/s	SILVER	51	51 - 135	13	0 - 9		·	
	SU846-7761	Non-Spec Water	SILVER	4	104 - 118	1	0 - 3			
	SU846-7770	Non-Spec Water	SODIUM	5	74 - 148	4	0 - 6			
į		\$/8/8	SODIUM	2	105 - 105	;3	0 - 22	·		
į	SU846-7780	Non-Spec Water	STRONTIUM	17	42 - 150	9	0 - 12			
İ		5/5/5	STRONTIUM	8	68 - 140	4,	9 - 02			
	SU846-7840	\$/\$/\$	THALLIUM	2	74 - 121					
j	SU846-7841	Non-Spec Water	THALLIUM	42	57 - 126	11	0 - 9			
:		011	THALLIUM	. 4	55 - 115					
.		\$/\$/\$	THALLIUM	20	32 - 145	7	0 - 24			
	SW846-7870	Non-Spec Water	TIN							
		0,11	TIN	.1	89 - 89					
	SU846-7910	Non-Spec Water	VANADIUM			1	0 - 3		-	
!		5/5/5	VANADIUM	4	28 - 165	1	0 - 0			

REC = Percent Recovery (observed/actual*100)
RFO = Relative Percent Ofference (2*(rep1+rep2)/(rep1+rep2)*100)

(1pa. Soap)

MS I TAGE +ORA ES Table 5.1

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 [HRU 30-SEP-91

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Method	Matrix	Parameter	N	X REC	N	RPD	Limit	Units	
SW846-7950	Non-Spec Water	ZINC	21	77 - 118	10 0	- 11		<u> </u>	
	5/5/5	ZINC	43	61 - 136	35 0	26			
SU846-8000	Non-Spec Water	CYCLOHEXANONE	Ē	73 - 1 2 3	. 1 0	- 12			
		DIESEL FUEL	2	71 - 134	1 0	- 15			
		1,4-DIOXANE	2	65 - 78	1 0	- 5			
		ETHYLENE GLYCOL	5	0 - 150	1 0	- 5			
		ETHYLENE DXIDE	2	60 - 110	1 0	- 14			
		CASOLINE	В	47 - 133	4 0	- 10			
		ISOBUTANOL	2	110 - 118	1 0	- 2			
		ISOPROPYL ALCOHOL (2-PROPANOL)	2	99 - iis	ió	- 4			
		METHANOL	2	105 - 186	1 0	- 13			
		MINERAL SPIRITS	1	62 - 62					
		N,N-DIMETHYLFORMAMIDE (DMF)	ġ	94 - 180	1 0	- 6	T		
		UNKNOWN HYDROCARBON	_		1 0	- 0			
						•			
	011	BENZENE	2	89 - 94	1 0	- 2	•	•	
		CHLORDBENZENE	ž	88 - 96	1 6	- 2			
		2-METHYLPHENOL (0-CRESOL)	5	93 - 93	1 0	- 0			
		XYLENES (O/M/P-XYLENE)	٤	90 - 95	iõ	- ž			
	0.40.45	A 0.5.7.5.115		(2 127		. 04			
	S/S/S	ACETONE	16	62 - 183	ė į	- 24			
		BENZENE	5	76 - 110	1 0	- 9			
		CHLDROBENZENE	5	57 - 130	1 0	- 18			
		DICHLOROMETHANE (METHYLENE CHLORIDE)	2	69 - 124	1. 0	!3			
		DIESEL FUEL	2	11 - 147	1 0	- 34			
		ETHANOL	7	51 - 142	30	- 25			
		ETHYL ACETATE	8	35 - 119	4 0	- 17		 	
		CASOLINE	2	70 - 78	1 0	- 3			
		ISOAMYL ALCOHOL, PRIMARY	2	87 - 92	1 0	- 1			
		METHANOL	6	41 - 127	3 0	- 30			
		2-HETHYL-1-BUTANOL	2	87 - 92	1 0	- 1			
		2-METHYLPHENOL (O-CRESOL)	2	40 - 163	1 0	- 29			
		3-METHYLPHENOL/4-METHYLPHENOL (M/P-CRESOL)	4	62 - 150	Š 0	- 88			
		MINERAL SPIRITS	1	94 - 94					
		N,N-DIMETHYLFORMANIDE (DMF)	10	73 - 132	5 0	- 18			
		N-AMYL ALCOHOL, PRIMARY	٤	68 - 89	1 0	- 6			
		N-BUTYL ACETATE	2	85 - 90	i ō	- 1			
		N-HEXANE	ē	94 - 95	iŏ	- 11			
		2-NITROPROPANE	5	68 - 115	1 0	- 12			
		OTHER HYDROCARBONS	4	86 - 109	2 1	- 14			
		PYRIDINE	<u>-</u>	44 - 146	<u>-</u> <u>- </u>	- 25			
		1 1 1 4 M 4 17 L	-	77 - 170	, 0	ES			

N = Sample count

(qao2.sq1)

REC * Percent Recovery (observed/actual*100)

RPD = Relative Percent Difference (2*(rep1-rep2)/(rep1+rep2)*100)

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Table 5 1

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 THRU 30-SEP-91

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Method	Matrix	Parameter	N	×	REC	N	RPD	Limit	Units	
SU846-8000	5/5/5	TOLUENE	4	68 -	- 112	2 () - 39	, , , , , , , , , , , , , , , , , , , 		
1		1,1,2-TRICHLOROETHANE	٤		- 125	1 (
•		TRICHLORGETHENE	5		- 103	iò	-			
		TRICHLOROETHYLENE	ī	-	- 110	, ,				
		TRIETHYLAMINE	•		- 20					
			2	59						
<u> </u> 		XYLENES (D/M/P-XYLENE)	ξ.	21.	- 59	1 (y - y			-
SW846-8010	Non-Spec Water	•				1 (·	***************************************	
1		1,2-DICHLORDETHANE					<u>- 27</u>			
Į.		I, I-DICHLOROETHENE				1 (0 - 22			
]		DICHLOROMETHANE (METHYLENE CHLORIDE)				1 (D – 6			
1		TRANS-1,2-DICHLOROETHENE				1 (0 - 10)		
		1,1,1-TRICHLOROETHANE				2	0 - 66		-	
]		TRICHLOROETHENE				2				
		THE OTHER THE NE				• •	-	·		
	S/S/S	1,4-DICHLOROBENZENE (P-DICHLOROBENZENE)			• -	1 (0 - 10			
		1,1-D1CHLQROETHENE	14	46	- 133	7 (0 - 17	·		
		1,1,1-TRICHLOROETHANE	14	52	- 135	7 (0 - 18			
SW846~8015	Non-Spec Water	CYCLOHEXANONE	2	100	- 108	1 (o – a	<u>!</u>		
i	•	ETHANOL	2	81	- 86	1 (0 - 1			
Į.		CASOLINE	5	56		• `				
		ISOBUTANOL	6		- 113	3 (0 - 16			
1		METHANOL	2		- 125) - 10 1 - 7			
		N-BUTYL ALCOHOL	5		- 105	- 1 3				
		N-BOTTE MECONDE	č	31	- 105	, ,	, z			
	011	CYCLOHEXANONE	2	84 -	- 113	1 () - 7			
1		ETHANOL	2	72 -	- 127	1 (- 13			
		ISOBUTANOL	2		- 128	i				
		METHANOL	2		- 116	1 6				
		N-BUTYL ALCOHOL	2		- 112	iò				
		N BOTTE RECONCE	_	00		' '	, 0			
	3/5/5	CYCLOHEXANONE	7	44 -	- 143	3 (- 16			
i		GASOLINE	. 6	92 -	- 92	• • • • • • • • • • • • • • • • • • • •				
1		ISOBUTANOL	9	53 -	- 131	4 0	- 17			
I		ISOPROPYL ALCOHOL (2-PROPANOL)	1		- 97					
†		METHANOL	i 0		- 157	5 0	- 31			
}		N-BUTYL ALCOHOL	9		- 147	4 4	- 10			
		TERTIARY BUTYL ALCOHOL	2			7 0				
		TERTIARI BUTTE ALLUHUL	E	. 40 -	~ >0	1_0	<u> </u>			
SW846-8015 MOD	Non-Spec Water	2-BUTOXYETHANOL				1 0	- 3			
[GASOLINE	1 Q	B0 -	- 127	5 0	- 11		<u></u> .	_
!		HEXANE	é	101 -	- 101	i d	- Ô			
İ		ISOBUTANOL	2	102 -	- 110	1 0	- 2			
1		ISOPROPYL ALCOHOL (2-PROPANOL)				1 0	- 0			
		METHANOL	٤	85 -	- 9ã	i ă	- 1			
I			-			. •				

N = Sample count

REC * Percent Recovery (observed/actual*100)

RPD = Relative Percent Difference (2*(rep1-rep2)/(rep1+rep2)*100) (qao2.sq1)

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Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 THRU 30-SEP-91

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i								148c 17 01 J
Method	Matrix	Parameter	N	X REC	N	RPD	Limit	Units
SW846-8015 MOD	Non-Spec Water	PYRIDINE	5	57 - 65	1 0		 3	
34046-8015 HUD	Mon-Spec water		•	21 - 62				
! 		TERPINEOL (ALPHA)			1 0	-	5	
	S/S/S	ACETONE	2	85 - 101	. — · · ·		4	
	3, 3, 3	DIESEL FUEL	6	79 - 137	4 0	- 2		
1		2-ETHOXYETHANOL	2	106 - 111	1 0	_	·	
i		· · · · · · · · · · · · · · · · · · ·				-	:	··· ·
		GASOLINE	26	72 - 129	13 0	- 3	<u> </u>	
]		HEXANE	6	60 - 155	3 0		·	
		ISOBUTANOL	2	95 - 100	_ 1 0	-	·	
		METHANOL	4	73 - 110	2 0	- 2	3	
1		2-NITROPROPANE	2	93 - 109	1 0	-	4	
,		PYRIDINE	8	61 - 138	4 0	-	7	
011046 0000	N- 0 11 .		_			_	_	
SW846-8020	Non-Spec Water	BENZENE	3	5 - 108	2 0	_		
1		ETHYL BENZENE	3	54 - 1 <u>17</u>	5 Ó	- 3	7,	
i		M/P-XYLENE	3	61 - 118	2 0	-	4 '	
i		METHYL-T-BUTYL ETHER (MTBE)	3	19 - 105				
1		O-XYLENE	3	74 - 113	2 0	- 2	6	
1		TOLUENE	3	12 - 140	2 0	- 5	1	
	5 / 5 / 5	RENZEME		/7 12E	74.0	_	=	
	5/5/5	BENZENE	68	63 - 125	34.0			
!		ETHYL BENZENE	74	59 - 127	36 0	•		
• :		M/P-XYLENE	59	55 - 12 9	30 0	•		
		METHYL-T-BUTYL ETHER (MTBE)	1	138 - 138	1 Q	- 1	!	
		NAPHTHALENE	2	70 - 117	1 0	- 1	2	<u></u>
		O-XYLENE	59	54 - 137	28 0	- 5	1	
		TOLUENE	74	51 - 136	37_0	E	6	
511845 8024	N C N	BENZENE						
SW846-8021	Non-Spec Water	BENZENE	1	92 - 92				
İ		1,1-D1CHLOROETHANE	1	114 - 114				
		DICHLOROMETHANE (METHYLENE CHLORIDE)	1	85 - 8 5				
		ETHYL BENZENE	1	89 - 89				
		TOLUENE	1	90 - 90	<u> </u>			
	S/S/S	BENZENE	22	64 - 122	11 0	- 1	4	
	37 37 3	CARBON TETRACHLORIDE	2	79 - 87		_ '	· ——	
•			2		i 0			
1		1,4-DICHLOROBENZENE (P-DICHLOROBENZENE)	_	82 - 124	• •	•		
ı		1,1-DICHLOROETHANE	4	37 - 162	2 0	- 13	·	
i		1,2-DICHLOROETHANE	2	101 - 117	1 0		!	
		1,1-DICHLOROETHENE	16	59 - 120	8 0	- 2	·	
		ETHYL BENZENE	20	55 - 129	100	- 13	3	
		M/P-XYLENE	2	74 - 90	1 0	!	5	
İ		O-XYLENE	2	83 - 83	i o	- ()	
		TOLUENE	18	60 - 126	8 0	- 1	3	
		1,1,1-TRICHLOROETHANE	20	60 - 136	10 0		, ———	
i		TRICHLOROETHENE	٦	116 - 121	1 0			
		IN SOMEON OF LINEAR	_		. •	!		****

N : Sample count

REC = Percent Recovery (observed/actual+100)

RFO : Relative Parcent Difference (2*(rep1-rep2)/(rep1+rep2)*100)

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Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 FHRU 30-SEP-91

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Meth	od	Hatrix	Parameter	N	l <u>-</u>	X 1	REC	N		RPD	Limit		30 of 3	18	
5484	6-8021	S/S/S	VINYL CHLORIDE	<u> </u>	: 1	43 -	203	i	Ō	- 8					
3984	6-8040	\$/\$/\$	4-CHLDRD-3-METHYLPHENOL	٤	•	40 -	100	1	0	=29	•				
	• • • • • • • • • • • • • • • • • • • •		2,4-DICHLOROPHENOL	. 2		58 -	79	1	ò		,				
1			4,6-DINITRO-2-METHYLPHENOL	- 2	-	32 -	45	i	ŏ	_	· ———				
ŀ			2-NITROPHENOL	2	-	70 -	75	i		_ i					
1			PENTACHLOROPHENOL	5		36 -	57	- ;	7	- 11	· 				
Ì			PHENOL	4	-	90 -		و	ň	- 11	· 				
1			2,4,6-TRICHLOROPHENOL	2	•	88 -	93	ī	0	- 1					
									_						
SW84	6-8080	Drinking Water	ENDRIN	4		64 -			0						
			GAMMA-BHC (LINDANE)	4	}	76 -	85	2	. 0,_	1 ()				. –
1		Non-Spec Water	ALDRIN	59)	24 -	121	29	0	- 34	·				
į			4,4'-DDT	57	,	48 -	135	29	0	- 36	3				
i			DIELDRIN	59)	60 -	120	28	Ō	- 18	1				
			ENDRIN	56	,	59 -	148	27	0	- 30	·				
i			GAMMA-BHC (LINDANE)	59			114	30	0	- 22		-			
[HEPTACHLOR	61		39 -		29	ō	- 34					•
i i			PCB AROCLOR 1232					1	ò	- 12					
1			PCB AROCLOR 1254					1	Ō	- 21					
İ			PCB AROCLOR 1260	88	:	59 -	126	38		- 20					-
ļ		011	ALORIN	. 2	<u>.</u>	66 -	79	1	0	- 13	1				
1		·	4,4'-DDT	E E		16 -	•	1	Õ	- 4	·				
ļ			DIELDRIN	2	-	10 -		i	ō	- 6	·				
			ENDRIN	Ē		33 -		i	Ŏ	- 1	i				
i			CAMMA-BHC (LINDANE)	S.			113	··· · · · i			i ———				
1			HEPTACHLOR	2		02 -		i	ě	_ •	·				
			PCB AROCLOR 1016	_		-		•	•						
			PCB AROCLOR 1221					•		•		· · · · · · · · · · · · · · · · · · ·			
1			PCB AROCLOR 1232												
			PCB AROCLOR 1242					1	٨	- 39					
ł			PCB AROCLOR 1248	24		61 -	119	13	- <u>-</u>						
!			PCB AROCLOR 1254	Ę		50 -	66	Ĩ	-						
i			PCB AROCLOR 1260	154		54 -		74	-	_					
			PCB AROCLOR 1262	,		3 7	167	(7	V				• • •		
į															
!		3/3/3	ALDRIN			. 4	193	3	Q.,	13					
			2-CHLOROBIPHENYL	24		0 -	197	8	0	- 64					
			3-CHLOROBIPHENYL	24			164	8	0	- 61					
1			4-CHLOROBIPHENYL	26		0 -	\$35	2	. 9 :	= 69	·				
			4,4'-DDT	8		0 -	164	4	0	- 36					
			DIELDRIN	6	i	22 -	172	3	0 .	- 50					
1			ENDRIN	15		0 –	362	4	Q:	- 159					-
			GAMMA-BHC (LINDANE)	14		21 -	157	4	0	- 88					

		_		_			
N	2	-S.	3 -	0.1	•	-	 11

RFC = Percent Recovery (observed/actual*100)

RFO : Relative Forcent Oifference (2*(repl-rap2)/(repl+rep2)*100)

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Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 THRU 30-SEP-91

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								. uge 31 0130	
Method	Matrix	Parameter	N	% REC	N	RPD	Limit	Units	
						=====			
SU846-8080	5/5/5	HEPTACHLOR	15	29 - 151	4 0	- 53			
		HEPTACHLOR EPOXIDE	9	0 - 189		- 55			
		METHOXYCHLOR	. 9	18135	20_	- 30			
		PCB AROCLOR 1232	1	59 - 59					
		PCB AROCLOR 1242	7	46 - 106	5 0	- 83			
		PCB AROCLOR 1248	77	35 - 150	12 0_	- 15			
		PCB AROCLOR 1254	18	51 - 154	9 0	- 37			
		PCB AROCLOR 1260	366	41 - 140	124 0	- 53			
		TOXAPHENE	8	67 - 124_	2 _0_	17		······	
SW846-8100	S/S/S	ACENAPHTHENE	7	22 - 151	3.0	- 32			
34040-0100	37373	ACENAPHTHYLENE		25 - 147	4 0	- 30			
		•	7	: : :					
		ANTHRACENE	-	0 - 200	3 0				
		BENZO(A)PYRENE	9	31 - 182	4 0	- 51			
		BENZO(B)FLUDRANTHENE	. 5	51 - 500	<u>-</u>	- 51			
		BENZO(G,H,I)PERYLENE	11	44 - 166	5 0	- 47			
		BENZO(K)FLUORANTHENE	5	19 - 175	2 0	- 13			
		CHRYSENE	7	46 - 167	3 0	- 26			***
		DÍBENZ(A,H)ANTHRACENE	5	1 - 200	ŽÖ	- 25			
		FLUORANTHENE	11	0 - 180	50	- 66			
		FLUORENE	11	21 - 164	5 0	- 26			
		INDENO(1,2,3-CD)PYRENE	5	0 - 231		- 152			
		NAPHTHALENE	11	32 - 136	5 0	- 33			
		PHENANTHRENE	7	0 - 209	7.0	- 38			
		PYRENE	5	74 - 152	. 5 0				
		FIRENE	5	14 - 156	2 0	- 16			
SW846-8141	Non-Spec Water	DISULFOTON	2	100 - 113	1 0	3			
		FAMPHUR	8	103 - 108	1 0	- 1			
		METHYL PARATHION	6	23 - 169	3 0	- 16			
		PARATHION	6	25 - 143	3 0	_ 16			
		PHORATE	2	90 - 116		- 5		· · · · · · · · · · · · · · · · · · ·	
						_			
	S/S/S	MALATHION	2	_ 52 - 149	3_0_	- 35			
SW846-8150	Drinking Water	2.4-DICHLOROPHENOXYACETIC ACID	5	41 - 189	2 0	- 48			
38040-0130	Dilliking water	SILVEX		68 - 137					
1 			5 2	87 - 121	2 0.	- 13 - 8			
		2,4,5-TRICHLOROPHENOXYACETIC ACID (SILVE X)	E	87 - 121	, 0	- 0			
	Non-Spec Water	2,4-DICHLOROPHENOXYACETIC ACID	11	0 - 183	4 0	- 25			
		SILVEX	11	13 - 174	🙎 🗓	=39			
		2,4,5-TRICHLOROPHENDXYACETIC ACID (SILVE X)	6	41 - 129	3 0	- 41			

N : Sample count

REG = Percent Recovery (observed/actual+100)
RFO = Relative Percent Oifference (2*(rep1-rep2)/(rep1+rep2)+100)

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Quality Assurance Objectives for Location ALL

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PERIOD: 01-APR-91 JHRU 30-SEP-91

Method	Matrix	Parameter	N	% REC	N RPD Limit	Page 32 of 38
SU846-8150	5/5/5	2,4-DICHLOROPHENOXYACETIC ACID	23	0 - 194	5 0 - 43	
		SILVEX	21	22 - 160	4 0 - 27	
		2,4,5-TRICHLOROPHENDXYACETIC ACID (SILVE X)	. 2	_54 - 88 _	1 0 - 11	
SU846-8240	Non-Spec Water	BENZENE Bromodichloromethane	27	85 - 184	14 0 - 16	
		4-BROMOFLUOROBENZENE Bromoform Bromomethane	2	8896	10-2	
		CARBON TETRACHLORIDE CHLOROBENZENE CHLOROETHANE CHLOROETHANE	i 9	86 - 117	10 0 - 11	
	.	2-CHLOROETHYLVINYLETHER CHLOROFORM CHLOROMETHANE C1S-1,3-DICHLOROPROPENE				
		DÍBROMOCHLOROMÉTHANE 1,1-DÍCHLOROETHANE 1,2-DÍCHLOROETHANE				
		DICHLOROETHANE-D4 1,1-DICHLOROETHENE 1,2-DICHLOROETHENE (CIS AND TRANS)	2 17	93 - 101 59 - 137	9 0 - 36	
		DICHLOROMETHANE (METHYLENE CHLORIDE) 1,2-DICHLOROPROPANE ETHYL BENZENE FLUDROTRICHLOROMETHANE				**************************************
		METHYL-T-BUTYL ETHER (MTBE) 1,1,2,2-TETRACHLORGETHANE TETRACHLORGETHENE				
		TOLUENE	25	84 - 121	13 0 - 13	
		TOLUENE-D8 Trans-1,3-DICHLOROPROPENE	Ë	90 - 95	2 - 0 1	
		1,1,1-TRICHLORGETHANE 1,1,2-TRICHLORGETHANE		والمستوافية والمستواد		
		TRICHLOROETHENE Vinyl Chloride	16	72 - 124	9 0 - 13	
	oii	BENZENE Chlorobenzene	4	92 - 109 63 - 153	2 0 - 11	
		1,1-DICHLOROETHENE Toluëne Trichloroethene	4	70 - 96 0 - 205 60 - 102	2 0 - 7	
	\$/5/\$	ACETONE		89 - 89		

N = Sample count

REC = Percent Recovery (observed/actual*100)

RPD = Relative Percent Difference (2*(rep1-rep2)/(rep1+rep2)+100)

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Quality Assurance Objectives for Location ALL

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bodieM Matrix X REC Units Parameter RPD Limit SU846-8240 S/S/S 4-BROMOFLUOROBENZENE 70 - 11914 CARBON TETRACHLORIDE15 71 - 129 CHLOROBENZENE 18 69 - 119CHLOROFORM 17 331 - 122 1.2-DICHLOROETHANE 79 - 126DICHLOROETHANE-D4 14 76 - 1231,1-DICHLORDETHENE 39 - 1591,1-DICHLOROETHYLENE 15 81 - 126 METHYL ETHYL KETONE 17 0 - 209 STYRENE TETRACHLOROETHYLENE 15 63 - 118TOLUENE 25 60 - 139TOLUENE-DB 14 80 - 127 3 0 TRICHLOROETHENE 16 66 - 110TRICHLOROETHYLENE 71 - 122VINYL CHLORIDE 15 69 - 122 SW846-8270 Non-Spec Water ACENAPHTHENE 40 - 129 **ACENAPHTHYLENE** ANTHRACENE BENZ(A)ANTHRACENE BENZO(A)PYRENE BENZO(B)FLUORANTHENE BENZO(G, H, I)PERYLENE BENZO(K)FLUORANTHENE (BENZYLBUTYLPHTHALATE) BUTYLBENZYLPHTHAL ATE BIS(2-CHLOROETHOXY) METHANE BIS(2-CHLOROETHYL)ETHER BIS(2-CHLORDISOPROPYL)ETHER 81S(2-ETHYLHEXYL)PHTHALATE 4-BROMOPHENYL-PHENYLETHER 4-CHLORD-3-METHYLPHENOL 2-CHLORONAPHTHALENE 35 - 130 2-CHLOROPHENOL 4-CHLOROPHENYL-PHENYLETHER CHRYSENE DI-N-BUTYLPHTHALATE DI-N-OCTYLPHTHALATE DIBENZ(A, H)ANTHRACENE 1,3-DICHLORDBENZENE (M-DICHLORGBENZENE) 1,2-D1CHLOROBENZENE (O-DICHLOROBENZENE) 1,4-DICHLOROBENZENE (P-DICHLOROBENZENE) 3.3'-DICHLOROBENZIDINE

N : Samole count

REC = Percent Recovery (observed/actual+100)

RPO : Relative Forcent Oifference (2*(repi-rep2)/(repi+rep2)*100)

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Quality Assurance Objectives for Location ALL PER 100: 01-APR-91 THRU 30-SEP-91

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Hethod	Matrix	Parameter	N	X RE	EC	N	RPD	Limit	Units
SU846-8270	Non-Spec Water	2,4-DICHLOROPHENOL			:-:				
		DIETHYLPHTHALATE							
		2,4-DIMETHYLPHENOL		-			· • • • • • • • • • • • • • • • • • • •	- -	
		DIMETHYLPHTHALATE							
		4,6-DINITRO-O-CRESOL							
		2,4-DINITROPHENOL	= = =	== -	= -			·	
		2,4-DINITROTOLUENE	54	38 - 1	148	28	0 - 65		
		2,6-DINITROTOLUENE							
		FLUORANTHENE					· · · - · ·		
		FLUDRENE HEYARIN COORSINES							
		HEXACHLOROBENZENE							
	HEXACHLOROBUTADIENE						 -		
	HEXACHLOROCYCLOPENTADIENE								
		HEXACHLOROETHANE INDENO(1,2,3-CD)PYRENE							
		ISOPHORONE						ı 	
		N-NITROSO-DI-N-PROPYLAHINE	52	32 - 1	149	26	0 - 28		
		N-NITROSODIMETHYLAMINE	36	36	173	20	V - E0		
		N-NITROSODIPHENYLAHINE							
		NAPHTHALENE							
		NITROBENZENE							
		2-NITROPHENOL							** · · · · · · · · · · · · · · · · · ·
		4-NITROPHENOL	57	0 -	70	29	0 - 56		
		PENTACHLOROPHENOL	55	0 - 1	187	28			
		PHENANTHRENE							
		PHENOL	21	0 ~	81	11	0 - 61		
		PYRENE	58	43 - 1	34	27	0 - 19		
		1,2,4-TRICHLOROBENZENE	53	29 -		26			
		2,4,6-TRICHLOROPHENDL							
	Oit	ACENAPHTHENE	3	12 -	73		0 - 7		
		4-CHLORO-3-METHYLPHENOL	3	0 -	88	i	0 - 15		
		2-CHLOROPHENOL	3	14 -	65	1	0 - 17		
	-	1,4-DICHLOROBENZENE (P-DICHLOROBENZENE)	3	18 -	67		0 - 7		
		2,4-DINITROTOLUENE	3	0 -	83	1 -	0 - 7		
		N-NITROSO-DI-N-PROPYLAMINE	3	16 -	77	1 .	Q 8.		
		4-NITROPHENOL	3	0 -	95	1	0 - 21		
		PENTACHLOROPHENDL	3		53	1	0 - 24		
		PHENOL	3	23	. 53	11	Q. <u>-</u> _13		
	-	PYRENE	3	17 -	41	1	D - 11		
		1,2,4-TRICHLOROBENZENE	3	7 -	63	1 (0 - 10		
	5/\$/\$	ACENAPHTHENE	36	35 - 1	32	18	9 - 22		
		ACENAPHTHYLENE	2	69 - 1	_	1			
		ANTHRACENE	2	69 - 1		i.			
		BENZIA)ANTHRACENE	2	50 - 1			- 12		
N r Samula ca									

N = Sample count

REC = Percent Recovery (observed/actual+100)

RPD = Relative Percent Difference (2*(rep1-rep2)/(rep1+rep2)+100)

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Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 THRU 30-SEP-91

Revision 0 Nov. 1, 1991

Manhad	Materia	PERIOD: 01-APR-91 1HRU			n non 1	Page 35 of 38
Method	Hatrix	Parameter	N	X REC	N RPD Li	lmit Units
SV846-8270	8/8/8	BENZO(A)PYRENE		93 - 114	10-5	
		BENZO(B)FLUORANTHENE	2	115 - 120	10-1	
		BENZO(G,H,I)PERYLENE	2	48 - 197	10- 29	
		BENZO(K)FLUORANTHENE	2	79 - 105	10-7	
		(BENZYLBUTYLPHTHALATE) BUTYLBENZYLPHTHAL	2	73 - 112	10 - 10 -	
		ATE				
		B1S(2-CHLORDETHOXY)METHANE	2	89 - 97	10- 2_	
		BIS(2-CHLOROETHYL)ETHER	2	0 - 536	1 0 - 49	
		BIS(2-CHLORDISOPROPYL)ETHER	2	63 - 123	10-15_	
		BIS(2-ETHYLHEXYL)PHTHALATE	2	48 - 137	1 0 - 23	
		4-BROMOPHENYL-PHENYLETHER	2	87 - 100	10-3	
		4-CHLORD-3-METHYLPHENOL	34	13 - 135 1	6 0 - 41	
		2-CHLORONAPHTHALENE	2	84 - 113	10-7_	
		2-CHLOROPHENOL	36	21 - 142 1	8 0 - 29	
		4-CHLOROPHENYL-PHENYLETHER	2	68 - 107	10 - 10	
		CHRYSENE	2	44 - 133	10- 8_	
		DI-N-BUTYLPHTHALATE	2	57 - 125	10-18	
		DI-N-OCTYLPHTHALATE	2	78 - 104	10-7	
		D1BENZ(A,H)ANTHRACENE	2	69 - 166	10- 20	
	_	1.3-DICHLORDBENZENE (M-DICHLOROBENZENE)	2	70 - 112	10-11	
		1,2-DICHLOROBENZENE (D-DICHLOROBENZENE)	5	68 - 118	10-13_	
		1,4-D1CHLOROBENZENE (P-DICHLOROBENZENE)	40	35 - 123 2	0 0 - 21	
		3,3'-DICHLOROBENZIDINE	2	62 - 67	10- 2	
		2,4-DICHLOROPHENOL	2	43 - 140	i 0 - 25	
•		DIETHYLPHTHALATE	2	39 - 133	10-21	
		2,4-DIMETHYLPHENOL	2	44 - 125	1 0 - 22	
		DIHETHYLPHTHALATE	2	85 - 78	1 0 - 3	
		4,6-DINITRO-O-CRESOL	2	21 - 183	1 0 - 35	
		2,4-DINITROPHENOL	2	0 - 302	10- 98	
		2,4-DINITROTOLUENE	36		6 0 - 23	
		2,6-DINITROTOLUENE	1	84 - 84	10-7	
		FLUORANTHENE	è	7 - 172	1 0 - 44	
		FLUORENE	2	42 - 126	0 - 24	
		HEXACHLOROBENZENE	ē	B6 - 102	10-4-	
		HEXACHLOROBUTADIENE	2	93 - 98	10-1	
		HEXACHLOROCYCLOPENTAD1ENE	٤	35 - 142	1 0 - 28 -	
		HEXACHLORDETHANE	Ž	75 - 109	10-9	
		INDENO(1,2,3-CD)PYRENE	2	59 - 182	10-16	
		ISOPHORONE	Ž	70 - 104	1 0 - 9	
		N-NITROSO-DI-N-PROPYLAMINE	30		40- 26	
		N-NITROSODIPHENYLAMINE	2	68 - 128	10-14	
		NAPHTHALENE	<u> </u>	73 - 112	1 0 - 10	
		NITROBENZENE	2	85 - 98	10-3	
		2-NITROPHENOL	2	72 - 119	10-12	
		4-NITROPHENOL	36		8 0 - SE	

N = Sample count

REC = Percent Recovery (observed/actual+100)

RPD = Relative Percent Difference (2*(rep1-rep2)/(rep1+rep2)+100)

EMS HERTTAGE LABORATORIES

Table 5.1

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 THRU 30-SEP-91 Section 3.0 Revision 0 Nov. 1, 1991 Page 36 of 38

% REC Method Parameter Matrix RPD Limit Units SU846-8270 5/5/5 PENTACHLOROPHENOL 0 - 16118 0 ~ 64 PHENANTHRENE 58 - 134 2 10-19 PHENOL 34 0 - 14317 0 PYRENE 37 26 - 124 18 0 34 33 - 14317 0 -1,2,4-TRICHLORGBENZENE 26 71 - 1132,4,6-TRICHLOROPHENOL 1 0 11 SU846-8310 Non-Spec Water **ACENAPHTHENE** 25 ACENAPHTHYLENE 55 - 84 ANTHRACENE 68 - B1 76 - 84 BENZO(A)PYRENE BENZO(B)FLUORANTHENE BENZOIG.H. I)PERYLENE - 125 BENZO(K)FLUORANTHENE 73 - 99 CHRYSENE 71 - 92 DIBENZ(A, H) ANTHRACENE 47 - 100 FLUORANTHENE 63 - 10514 69 - 96 FLUORENE INDENO(1, 2, 3-CD)PYRENE 65 - 100 NAPHTHALENE 23 - 86 1 0 -28 PHENANTHRENE 66 - 106PYRENE 46 - 119 5/5/5 ACENAPHTHENE 202 - 03 0. - . 31 ACENAPHTHYLENE 38 - 194 ANTHRACENE 25 - 136 43 0 - 182 BENZO(A)PYRENE BENZO(B)FLUDRANTHENE - 139 - 233 BENZO(G,H,I)PERYLENE BENZO(K)FLUORANTHENE 41 - 151CHRYSENE 53 - 158 11 DIBENZ(A, H) ANTHRACENE 87 - 122 77 FLUDRANTHENE FLUDRENE - 171 INDENO(1,2,3-CD)PYRENE 28 - 165 13 NAPHTHALENE - 198 PHENANTHRENE 53 - 165 60-28 PYRENE 15 - 16934 SU846-9010 Non-Spec Water CYANIDE 7 55 - 130 1 0 - 16 5/5/5 CYANIDE 31 - 1345 0 - 18 _____ SW846-9012 Non-Spec Water CYANIDE 69 - 11528 0 ~ 26 5/5/5 CYANIDE 60 - 125 10 0 - 47

N : Sample count

RFC = Percent Recovery (observed/actual+100)

RFO : Relative Percent Offference (2*(rep1-rep2)/(rep1+rep2)*100)

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Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 THRU 30-SEP-91

Sel. on L. Revision u Nov. 1, 1991 Page 37 of 38

Method Matrix Parameter Z REC Units SU846-9012 S/S/S CYANIDE, AMENABLE 1 100 - 100 SU846-9020 Drinking Water TOTAL ORGANIC HALOGEN (TOX) 20-26 Non-Spec Water TOTAL ORGANIC HALOGEN (TOX) 68 - 129 136 0 - 34 ____ 6 0 - 33 _____ 011 TOTAL ORGANIC HALDGEN (TOX) S/S/S TOTAL ORGANIC HALDGEN (TOX) 79 - 124 15 0 - 29 _____ SU846-9030 Non-Spec Water SULFIDE 91 - 107 011 SULFIDE 93 - 93 10-90 - 109 ___ 4 0 - 21 ____ 5/5/5 SULFIDE SU846-9038 Non-Spec Water SULFATE 81 - 183 5 0 - 11 5/5/5 SULFATE 88 - 124 SU846-9040 Non-Spec Water S/S/S PH SW846-9041 Non-Spec Water PH 5/5/5 PH 56 0 - 0 _____ SU846-9045 Oil PH 5/5/5 PH 182 0 -SU846-9050 Drinking Water CONDUCTIVITY Non-Spec Water CONDUCTIVITY SW846-9060 Drinking Water TOTAL ORGANIC CARBON (TDC) 2 100 - 100 10-Non-Spec Water TOTAL ORGANIC CARBON (TOC) 84 0 - 21 75 - 125 TOTAL ORGANIC CARBON (TOC) S/S/S , ___ 75 - 135SU846-9065 5/5/5 PHENOL 10 0 - 23 _____ 43 - 125 SU846-9066 <u> 15 68 - 119 5 0 - 12 _____</u> Non-Spec Water PHENOL

N = Sample count

REC = Percent Recovery (observed/actual+100)

RPD = Relative Percent Difference (2*(rep1-rep2)/(rep1+rep2)*100)

EMS HERTTAGE LAGORATORIES

Table 5.1

Quality Assurance Objectives for Location ALL PERIOD: 01-APR-91 THRU 30-SEP-91 Section 5.0 Revision 0 Nov. 1, 1991 -Page 38 of 38 -

Hethod	Matrix	Parameter	N	X	REC		N	RPD	Limit	Units
SU846-9066	\$/\$/\$	PHENOL	12	64	- 12	8	7 0	- 1		
SU846-9067	Non-Spec Water	PHENOL					1, 0	. – 9	5	
SW846-9070	5/5/5	OIL AND GREASE					1 0	- 1		
SU846-9071	5/5/5	OIL AND GREASE					8 0	- 4	<u> </u>	
SU846-9071(MOD)	5/5/5	OIL AND GREASE	3	108	- 11	7	S0	- E		
SH846-9251	Non-Spec Water	CHLORIDE	42	76	- 12	4	30 0	- !	5	
	S/S/S	CHLORIDE	2	115	- i i	5	2 0)	

N = Sample count

REC = Percent Recovery (observed/actual+100)

RPD = Relative Percent Difference (2+(rep1-rep2)/(rep1+rep2)+100)

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1337 rows selected.

PREP	scription	Matrix To detho		
	CLP PESTICIDE/PCB EXTRACTION			
	CLP PCB EXTRACTION	1 CLP SOW 2/	BB CLP POLYCHLORINATED BIPHENYLS (PCBB)	
	CLP SEMI-VOLATILE EXTRACTION	I CLP SON 2/		
	CLP PESTICIDE/PCB EXTRACTION	2 CLP SOW 2/	BBCLP_PESTICIDE/PCB_ORGANICS	
	CLP PCB EXTRACTION	2 CLP SON 2/	BB CLP POLYCHLORINATED BIPHENYLS (PCBB)	
	CLP SENI-VOLATILE EXTRACTION	5 CI P 30M 5/	BB CLP SEMI-VOLATILE ORGANICS	
		S CLP SON Z/	BB CLP SEMI-YOLATILE ORGANICS	
		4 CLP SOW 2/	BB CLP SEMI-VOLATILE ORGANICS	
	CLP PESTICIDE/PCB EXTRACTION	S CLP SON E/	88 CLP PESTICIDE/PCB ORGANICS	
	CLP PCB EXTRACTION	5 CLP SON 2/	BB CLP POLYCHLORINATED BIPHENYLS (PCBS)	
	CLP SENI-VOLATILE EXTRACTION	5 CLP 80W 2/	BB CLP SEMI-VOLATILE ORGANICS	SA
PA 200.0	FAA OR ICP ACID DIGESTION	1 APHA 303A	LITHIUM	AMPLE
		EPA 200.7	ALUMINUM (ICP-SEQ) ALUMINUM ICP	मि
	The second secon	EPA 200.7 EPA 200.7	ANTINONY (ICP-SEQ)	
		EPA 200.7	ANTIHONY ICP	PREPARATIO
		EPA 200.1	BARIUM (ICP-SEQ)	<u>¥</u>
		EPA 200.7	BARIUM ICP	\$
		EPA 200.7	BERYLLIUM (ICP-SEQ)	1 :
•	and the second s	EPA 200.7		g
		EPA 200.7 EPA 200.7	BISMUTH ICP Boron ICP	24
		EPA 200.7	CAOMIUM (ICP-SEQ)	Ħ
	· · · · · · · · · · · · · · · · · · ·	EPA 200.7	CADMIUM ICP	THOD
		EPA 200.7	CALCIUM ICP	Ö
		EPA 200.7	CHRONIUM (ICP-SEQ)	209
	•	EPA 200.7	CHRONIUM ICP	
		EPA 200.7	COBALT (ICP-SEQ)	
		EPA 200.7	COBALT ICP	
		EPA 200.7	COPPER (ICP-SEQ)	
		EPA 200.7	COPPER [CP	
		EPA 200.7	DYSPROSIUM 1CP 1CP SCAN	
		EPA 200.7	IRON (ICP-SEQ)	
		EPA 200.7	IRON ICE	
		EPA 200.7	LEAD (ICP-SEQ)	
		EPA 200.7	LEAD ICP	
		EPA_200,7	LITHIUM ICP	——— ਚ ਖ਼ ਅਰ
		EPA 200.7	HAGNESIUM ICP	י מוֹ כע
		EPA 200.7	MANGANESE (ICP-SEQ)	99 V Y K
		EPA 200.7	MANGANESE ICE	ω -
		EPA 206.7	HISCELLANEOUS (ICP-SEQ)	
				£,

le P	scription	Hatri	x 1c	Method	Description
A 200.0	FAA OR ICP ACID DIGESTION		1. EPA	200.7	HOLYBDENUM 1CP
			EPA	7.003	NICKEL (ICP-8E0)
				200.7	NICKEL ICP
				1.200.7 1.200.7	POTASSIUM ICP SILICON ICP
				200.7	SILVER (ICP-SEO)
					SILVER 1CP
				200.7	SODIUM ICP
				200.7	STRONTIUM ICP
				200.7	THALLIUM (ICP-8EQ)
				1 200.7 1 200.7	THALLIUM ICP Ten icp
				200.7	TITANIUM ICP
				200.7	TUNGSTEN LCP
			EPA	200.7	VANADIUM (ICP-SEQ)
	-		EPA	200.7	VANAOIUM . I CP
			-	200.7	ZINC (ICP-SEQ)
				200.7	ZINC ICP
			. FEG	r sas' ('	ALUMINUM EAA
	GFAA ACID DIGESTION		1 EPA	2.203	ALUMINUM GFAA
	FAA OR ICP ACID DIGESTION		1 EPA	204.1	ANTIHONY FAA
	GFAA ACID DIGESTION		1 EPA	8.4.2	ANTIHONY SEAA TO
				206.2	ARSENIC GFAA
			EPA	206.2	ARSENIC GFAA ('3 POINT HBA)
	FAA OR ICP ACID DIGESTION		I EPA	208.1	BARIUM GAA
	GFAA ACID DIGESTION		I EPA	3.80S	BARTUM GFAA
	FAA OR ICP ACID DIGESTION		1 EPA	210.1	BERYLLIUM FAA
	GFAA ACID DIGESTION		1 EPA	210.2	BERYLLIUM GFAA HE BORON COLORINEIRIG DB
	FAA OR ICP ACID DIGESTION		1 FPA	212.3	BORON COLORINEIRIC OD
	THE OR TO HOTE DISCUSSION			213.1	CADMIUN FAA
	•				
•	GFAA ACID DIGESTION		1 EPA	213.2	CADMIUM GFAA
	#44 an tan tan namatan				
	FAA OR ICP ACID DIGESTION			215.1 _218.1	CALCIUM FAA CHROMIUM FAA
			e E û		CHRONIUM FAA
	GFAA ACID DIGESTION		1 EPA	216.2	CHRONIUM GFAA
			EPA	218.2	CHRONIUM GFAA (3 PQINT MSA)
	744 OR 100 AAIN DIOCATION				
	FAA OR ICP ACID DIGESTION		I EPA	. 219.1	COBALT FAA
	GFAA ACID DIGESTION		1 EPA	219.2	COPPER FAA
	FAA OR ICP ACID DIGESTION		1 EPA	1.055	COPPER FAA
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PREP	scription	 Matrix	16	nethod	Description
EPA 200.0	OSAA AOIN DICEOTION	 	 Epa	226 2	COPPER GEAA
EK 500.0	and won biotofion	 		220.2	COPPER GFAA (3 POINT MBA)
	FAA OR ICP ACID DIGESTION	 1		236.1	IRON FAA
	GFAA ACID DIGESTION	 1		239.2 239.2	LEAD GFAA (3 POINT MBA)
	FAA OR ICP ACID DIGESTION	 1	EPA	242.1 243.1 246.1	MAGNESIUM FAA Manganese faa Molybuenum faa
	GFAA ACID DIGESTION	1	EPA	246.2	HOLYBOENUM GFAA
	FAA OR ICP ACID DIGESTION	1	EPA	249.1	NICKEL FAA
	GFAA ACID DIGESTION	1	EPA	249.Z	NICKEL GFAA
	FAA OR ICP ACID DIGESTION	 i	EPA	258.1	POTASSIUM FAA
	GFAA ACID DIGESTION	1	EPA	2.013	SELENIUM GFAA
	FAA OR ICP ACID DIGESTION	1	EPA	272.1	SILVER FAA
	GFAA ACID DIGESTION	 1	EPA	272.2	SILVER GFAA
	FAA OR ICP ACID DIGESTION	1	EPA	273.1	E AB
	GFAA ACID DIGESTION	1	EPA	£16.2	VANADIUM GFAA
	FAA OR ICP ACID DIGESTION	1	EPA	279.1	THAILIUM FAA
	GFAA ACID DIGESTION	,	EPA	219.2	THALLEUM GFAA
	FAA OR ICP ACID DIGESTION	1	EPA	282.1	TIN FAA
	GFAA ACID DIGESTION	i	EPA	282.2	TIN GFAA
	FAA OR ICP ACID DIGESTION	1	EPA	283.1	TITANIUM FAA
	GFAA ACID DIGESTION	1	EPA	283.2	TETANTUM GFAA
	FAA OR ICP ACID DIGESTION			286.1 289.1	VANADIUM FAA Zing faa
	GFAA ACID DIGESTION	i	EPA	289.2	ZINC GFAA
	FAA OR ICP ACID DIGESTION		EPA EPA	200.7 200.7 200.7 200.7 200.7	LITHIUM ALUMINUM (ICP-SEQ) ALUMINUM ICP ANTIHONY (ICP-SEQ) ANTIHONY ICP Section On the section of the sec
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Page			EPA 200.7	÷			
Page (1941) Page			EPA 200.7	BORON 1CP			
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	FAA OR ICP ACID DIGESTION	3 EPA 258.1	POTASSIUM FAA
	GFAA ACID DIGESTION	3 EPA 270.2	SELENIUM GFAA
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	FAA OR ICP ACID DIGESTION	5 EFA 273.1	SOOIUM FAA
	GFAA ACID DIGESTION	3 EPA 276.2	VANADIUM CFAA
	FAA OR ICP ACID DIGESTION	3 EPA 279.1	THALL TUR FAA
	GFAA ACID DIGESTION	3 EPA 279.2	THAILIUM GFAA
	FAA OR ICP ACID DIGESTION	3 EPA 282.1	TIN FAA
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	FAA OR ICP ACID DIGESTION	3 EFA 283.1	TITANIUM FAA
	GEAA ACID DIGESTION	3 EPA 283.2	TITANIUM GFAA
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	FAA OR ICP ACID DIGESTION	S EFA POS. 1	VANADIUM FAA	S
	GFAA ACID DIGESTION	S EPA 289.2	ZINC CFAA	AMPI
CPA 335.1	CYANIDE ANENABLE DISTILLAFION	1 EPA 335.1 EPA 335.1	CYANIDE AMENABLE TO CHLORINATION	1
		4 EPA 335.1 EPA 335.1	CYANIDE AMENABLE TO CHLORINATION CYANIDE AMENABLE TO CHLORINATION	TABLE EPARA
		S EPA 335.1 EPA 335.1	CYANIDE AMENABLE TO CHIOBINATION	5.2 ATION
CPA 335.2	CYANIDE DISFILLATION	I EPA 335.2 EPA 335.3	CYANIDE, FOTAL (MANUAL)	метн
		E EPA 335.2 EPA 335.3	CYANIDE, IOTAL (MANUAL)	ons
		3 EPA 335.2 EPA 335.3	CYANIDE, TOTAL (MANUAL)	
		1_EPA_315.2 EPA_315.3	CYANIDE, TOTAL (MANUAL)	
		S EPA 315.2 EPA 315.3	CYANIDE, FOTAL (MANUAL).	
EPA 340.1	FLUORIDE DISTILLATION	1 EPA 310.2	FLUORIDE (10N SCIECTIVE ELECTRODE)	
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EPA 35	350. E	O VINOUIA D	· •	ILLATION			-	EPA	350.1	AHHONIA		MITROGEN (AUT	OMATED	(AUIONATED PHENATE	METH		
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FPA 420.1	- -	PHENOL S	PHENOLS DISTILLATION	ATION			-	EPA	420.2	PHEN	HENOLS 4AAP	CAUTOMATED	(ED)		· ·		
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							•	EPA	420.2	PHENOLS	IOLS JAAF	(AUTOMATED)	(E0)			1	
							s	FPA	420.2	PHENOLS	IOLS AAAP	(AUTORATED)	(E.D.)				
EPA 504	•	MICRO-E	XFRACTIO	MICRO-EXTRACTION FOR EDB AND DCB	ANO OCE	<u>a.</u>	-	EPA	204	ETHY	ETHYLENE DIE	DIBRUMIDE AN	AND DIBR	O I BRONOCHI. OPROFAN	ROFAN		
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		PCB EXTRACTION	RACTION				-	EPA 6	80	POL.Y	POLYCHLORINATED	TEO BIPHENYLS		(FCBS)			
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		PCB EXTRACTION	RACTION				N:	EPA	808	F01. Y	POLYCHLORINATED	TEO BIPHENYLS		(PC85)		age	age
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	PESTICIDES OND PCBS EXF	5 EPA 508	ORGANOCHLORINE_FESTICIDES_IN_NATER_BY_GCORGANOCHLORINE PESTICIDES AND PCBS IN WA	
	PCB EXIRACTION	3 EPA 608	POLYCHLORINATED BIPHENYLS (PCBS)	
	ORGANOCHLORINE PESTICIDES AND EGBS EXFRA.	5 EPA 608 FPA 608	ORGANOCHLORINE PESTICIDES AND CGS IN WA	
	PCB EXTRACTION	5 EPA 608	POLYCHLORINATED BIPHENYLS (PCBS)	
FPA 625	SEMI-VOLATILE EXTRACTION (ACID FRACTION)	1 EPA 625	SENT VOLATILE PRIORITY POLLUTANIS (ACID	
	SENI-VOLAFILE EXTRACTION INCURAL/BASE/A	1 EPA 625	SENI-VOLAFILE PRIORITY POLLUFANTS (BASE/	!
	SEMI-VOLATHE EXTRACTION CHASE/NEUTRAL F	1 EPA 625	SEMI-VOLATILE PRIORITY POLLUTANIS (BASE/	
	SENI-VOLAFILE EXFRACTION INCUFRAL/BASEZA	2 EPA 625	SENI-VOLAFILE PRIORITY POLLUTANIS (BASE/	
	SENT-VOLATILE EXTRACTION (ACID FRACTION)	S EPA 625	SEN1-VOLATILE PRIORITY POLLUTANTS (ACID	
	SENI-VOLAFILE EXTRACTION (NEUFRAL/BASE/A	S EPA 625	SEMI-VOLAFILE PRIORITY FOLLUTANTS (BASE/	SAM
	SEMI-VOLATILE EXTRACTION (BASEZNEUTRAL F	S EPA 625	SEMI -VOLATILE PRIORITY POLLUTANTS (BASE/	PLE
FUTURE CLP 0	ONLY CLP PESTICIDE/PCB EXTRACTION	I FUTURE CLF ONLY	Y CLF PESTICIOS/FCB ORGANICS	1
	CLP SEMI-VOLATILE EXTRACTION	1 FUTURE CLP ONLY	Y CIP SENI-VOLATILE ORGANICS	TABI EPAI
	CLP PESFICIDE/PCB EXFRACTION	2 FUTURE CLP ONLY	Y CLP PESFICIOE/PCB ORGANICS	
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		3 FUTURE CLP ONLY	Y CLP SEMI-VOLAFILE ORGANICS	MET
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	CLP PESFICIDE/PCB EXTRACTION	S FUTURE CLP ONLY	Y CLP PESFICION/PCB ORGANICS	DB
	CIP SEMI -VOLATILE EXTRACTION	S FUTURE CLP ONLY	Y CLP SEMI-VOLATILE ORGANICS	:
	FAA OR ICP ACID DIGESTION OF AQUEOUS SAM	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	ALUMINUM FAA (CLP) ALUMINUM 1CE (CLP) ANTIMONY FAA (CLP) ANTIMONY 1CP (CLP)	
	GFAA ACIO DIGESTION OF AQUEOUS SAMPLES (1 [1, 10]	ARSENIC GFAA (3 POINT MSA) (CLP) ARSENIC GFAA (CLP)	
	FAA OR ICP ACID DIGESFION OF AQUEOUS SAN	11.70	BARIUM FAA (CLP) BARIUM ICP (CLP) GERYLLIUM FAA (CLP) BERYLLIUM ICP (CLP)	Section Revisi Nov. 1 Page 1
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	**Cription Matr	trix 1 Method	Description		
1011	FAA OR ICE ACID DIGESTION OF AQUEOUS SAM	1 11.1101	CADMIUM EAA (CLP)		: E
	-	1.1.1 1.1.1	₽ ₹ 5		
:	- Commence of the Commence of	11.801	CHRONIUM FAA (CLP)		1
		ILNO!	201		
:	- Company - Comp	ILMOI	COBALT ICP (CLP)		i
	,	ILM01	FAA ICP		
	CYANIDE DISTILLATION (CLP)	1 11.801	CYANIDE, TOTAL (MANUAL) (CLP)		
	FAA OR ICP ACID DIGESTION OF AQUEOUS SAM	10.70	SKON FAA (CLP)		,
		11.701	FAA	: :	
	GFAA ACID DIGESTION OF AQUEOUS SAMPLES (1 11.801	LEAD GFAA (CLF)		
	FAA OR ICP ACID DIGESTION OF AQUEOUS SAM		LEAD ICP (CLP) MAGNESIUN FAA (CLP) MAGNESIUN ICP (CLP) MANGANESE FAA (CLP) MANGANESE FAA (CLP)	SAMPLE	
:	HERCURY CVAN ACID DIGESTION OF AQUEOUS S	1 (1.80)	HERCURY CVAA (CLP)	1	1
	FAA OR ICP ACID DIGESTION OF AQUEOUS SAM	1.1.1.1.1.1.1.1.1.1.1.1.1.1.1.1.1.1.1.	NICKEL EAA (CLP) NICKEL ICP (CLP) POTASSIUM FAA (CLP) PO(ASSIUM ICP (CLP)	TABLE 5.	
	GFAA ACID DIGESTION OF AQUEOUS SAMPLES (11 11 1101	SELENIUM GFAA (3 POINT MBA) (CLP) Selenium Gfaa (CLP)		
	FAA OR ICP ACID DIGESTION OF AQUEOUS SAM	11.301	SILVER FAA (CI.P.) SILVER ICP (CL.P.) SODIUM FAA (CI.P.) SODIUM ICP (CL.P.)	THODS	
	GFAA ACID DIGESTION OF AQUEOUS SAMPLES (1 11.401	THALLTUM GFAA (CLP)		T.
	FAA ON ICP ACID DIGESTION OF AQUEOUS SAN	1 (LMO)	YANADIUH FAA. (CLE)		
	FAA ON ICP ACID DIGESTION OF 8/3/S SAMPL.	1 11 1101	VANADIUM ICP (CL.P.)		
	FAA OR ICP ACID DIGESTION OF AQUEOUS SAN	1 11.801 11.801	ZINC FAA (CLP) Zinc icp (CLP)		
		7. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1.	ALUMINUM FAA (CLP) ALUMINUM 1CP (CLP) ANTIMONY FAA (CLP) ANTIMONY 1CP (CLP)	Sectio Revisi Nov. 1 Page 1	
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11.001	GFAA ACID DIGESTION OF AQUEOUS SAMPLES (ILMO1 ILMO1	
	FAA OR (CP ACED DIGESTION OF AQUEOUS SAM	2	[LMO]	BARIUN FAA (CLP)
			ILM01 ILM01	BARIUM 1CP (CLP) BERYLLIUM FAA (CLP)
	- · · · · · · · · · · · · · · · · · · ·		ILMOI	BERYLLIUM ICP (CLP)
			[LHO]	CAOMIUN FAA (CLP)
			1LM01	CADMIUM ICE (CLP)
			11.1101	CALCIUM FAA (CLP)
			1 L.HO 1	CALCIUM ICP (CLP)
			11.M01	CHRONIUN FAA (CLE)
			11.M01	CHROMIUM ICP (CLP)
			El.Mot	COBALI FAA (CLP)
			ILHO1	COBALT ICP (CLP)
			[]_MO 1	COPPER FAA (CLP)
			1L.M01	COPPER ICP (CLP)
	CYANIDE DISTILLATION (CLP)		(1,NO1	CYANIDE, TOTAL (MANUAL) (CLP)
	FAA OR ICP ACID DIGESTION OF AQUEOUS SAM	9	[[.110]	IRON EAA (CLP)
	THE OF THE BIGGIIGH OF MACEOUS SAIL		IL.M01	IRON FAA(CLE)
			[LHO]	LEAD FAA (CLP)
••••	GFAA ACID DIGESTION OF AQUEOUS SAMPLES (JLM01	LEAD GFAA (GLP) LEAD ICP (GLP) MAGNESIUM FAA (GLP) MAGNESIUM ICP (GLP) MANGANESE EAA (GLP)
	FAA OR ICP ACID DIGESTION OF AQUEOUS SAN	2	I I.MO 1	LEAD (CP.(CLP)
			IL MO1	MAGNESIUM FAA (CLP)
			ILMOI	MAGNESIUM ICP (CLP)
			ILMOI .	MANGANESE EAA (CLP)
			ILH01	HANGANESE ICP (CLP)
	MERCURY CVAA ACID DIGESTION OF AQUEOUS S	2	11.7101	MERCURY CVAA (CLP)
	FAA OR ICP ACID DIGESTION OF AQUEOUS SAN	2	ELMO I	NICKEL FAA (CLP) NICKEL 1CP_(CLP)
			1LH01	NICKEL ICP (CLP)
	The second of th		[1,001	POTASSIUM FAA (CLP)
	•	1	JEM01	POTASSIUM ICP (CLP)
	GEAA ACID DIGESTION OF AQUEOUS SAMPLES (8 ([1.M0]	SELENIUM GFAA (3 POINT MSA) (CLP)
			ILH01	SELENIUM GFAA (CLP)
•	FAA OR ICP ACID DIGESTION OF AQUEOUS SAM		ILMO1	SILVER FAA (CLP)
			ILM01	SILVER ICP (CLP)
		1	ILM01	SODIUM FAA (CLP)
			ILM01	SODIUM ICP (CLP)
	GFAA ACED DEGESTION OF AQUEOUS SAMPLES (2	LM01	THALLIUM GFAA_(GLP)
	FAA OR 1CP ACID DIGESTION OF AQUEOUS SAM	5 1	ILM01	VANADIUM FAA (CLP)
	FAA OR ICP ACID DIGESTION OF S/S/S SAMPL	2 (LENGT	VANADIUM FAA (CLP) VANAGIUM ICP (CLP) VANAGIUM ICP (CLP)
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PREP	escription	Matrix		Oescription
ILMOI	FAA OR ICP ACID DIGESTION OF AQUEOUS SAM		(LMO)	ZINC FAA (CLP) Zing igp (CLP)
	CYANIDE DISTILLATION (CLP)		[1.80]	CYANIDE, FOTAL (HANUAL) (CLP)
	ACID DIGESTION OF OILSZSQLVEN(\$ (CLP)	4	11-M01 11-M01 11-M01 11-M01 11-M01 11-M01 11-M01	ALUMINUM EAA (CLP) ALUMINUM ICP (CLP) ANTIMONY FAA (CLP) ANTIMONY_ICP (CLP) ARSENIC GFAA (3 POINT MSA) (CLP) ARSENIC GFAA (CLP) BARIUM FAA (CLP) BARIUM ICP (CLP) BERYLLIUM FAA (CLP)
			11 MOT 11 MOT 11 MOT 11 MOT 11 MOT 11 MOT 11 MOT	BERYLLIUM ICP (GLP) CADMIUM FAA (CLP) CADMIUM ICP (GLP) CALCIUM FAA (GLP) CALCIUM ICP (GLP) CHROMIUM FAA (CLP) CHROMIUM ICP (GLP) COBALI FAA (CLP) COBALI ICP (GLP)
			11.M01 11.M01 16 M01	COPPER ICP (CLP)
	CYANIDE DISTILLATION (CLP)	4	[1.M01	CYANIDE, TOTAL (MANUAL) (CLP) REPARAT REPARAT
	ACID DIGESTION OF OILS/SOLVENTS (CLP)		ILM01 ILM01 ILM01 ILM01 ILM01 ILM01 ILM01	IRON ICP (CLP) LEAD FAA (CLP) LEAD ICP (CLP) MAGNESIUM FAA (CLP) MAGNESIUM ICP (CLP)
	•		11.M01 ILM01 ILM01 ILM01 ILM01 ILM01 ILM01	MANGANESE [AA (CLP) MANGANESE ICP (CLP) NICKEL FAA (CLP) NICKEL ICP (CLP) POTASSIUM FAA (CLP) POTASSIUM ICP (CLP) SELENIUM GEAA (3 POINT MSA) (CLP)
			ILMO1 16.M01 64.M01 16.M01 19.M01 16.M01	SELENIUM GFAA (CLP) SILVER FAA (CLP) SILVER ICP (CLP) SODIUM FAA (CLP) SODIUM ICP (CLP) THALLIUM GFAA (CLP)
		(prep	ILMOI IEMOI [LMO] ILMOI DE#SE, \$ql)	VANADIUM FAA (CLP) VANADIUM ICP (CLP) ZINC FAA (CLP) ZINC ICP (CLP)
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	ALUMINUM (CP (CLP)
ICM ICM	1 ANTIMONY CCP (CLP)
GFAA ACID DIGESTION OF S/S/S SAMPLES (CL S ILMOI	1 ARSENIC GFAA (3. POINT MSA)_(CLP) ARSENIC GFAA (CLP)
FAA OR ICP ACID DIGESTION OF S/S/S SAMPL. S ILMO ILMO ILMO	01 BARIUM EAA (CLP) 01 BARIUM ICP (CLP) 01 BERYLLIUM FAA (CLP) 01 BERYLLIUM ICP (CLP)
	CHROMIUM FAA (CLP) CHROMIUM ICP (CLP) COBALT FAA (CLP) COBALT ICP (CLP) COPPER FAA (CLP) COPPER FAA (CLP)
CYANIDE DISTILLATION (CLP)	1 CYANIDE, TOTAL (MANUAL) (CLP)
FAA OR ICP ACID DIGESTION OF S/S/S SANPL S ILNO ILNO ILNO	1 IRON FAA (CLP) 1 IRON ICP (CLP) 1 LEAD FAA (CLP)
GEAA ACID DIGESTION OF SYSTS SAMPLES (CL. 5 11.NO	1 LEAD GFAA (CLP)
FAA OR ICP ACID DIGESTION OF S/S/S SAMPL S ILNO ILNO ILNO ILNO ILNO ILNO ILNO ILNO	01 LEAD ICP (CLP) 01 MAGNESIUN FAA (CLP) 01 MAGANESE FAA (CLP) 01 MANGANESE FAA (CLP)
CID DIGESTION OF S/S/S S	I HERCURY CVAA (CLP)
FAA OR ICP ACID DIGESTION OF S/S/S SANPL S ILM ILM ILM	NICKEL FAA (CLP) NICKEL ICP (CLP) POTASSIUB EAA (CLP) POTASSIUN ICP (CLP)
GFAA ACID DIGESTION OF 8/8/5 SAMPLES (CL. 5 11MG	SELENIUM GFAA (3 POINT MSA) (CLP).
FAA OR ICP ACID DIGESTION OF S/S/S SAMPL S ILNO ILNO ILNO ILNO ILNO ILNO ILNO ILNO	SILVER FAA (CLP) SODIUM FAA (CLP) SOOIUM ICP (CLP)
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FREF		trix le Method	Description
1LM01	GFAA ACID DIGESTION OF S/S/S SAMPLES (CL	S 11.M01	THALLIUM GFAA (CLP)
	FAA OR ICP ACID DIGESTION OF 8/S/S SAMPL		VANADIUM FAA (CLF)
		ILM01	3 2
01.801	CLP PESFICIDE/PCB EXTRACTION	1 OLMO1	CLP PESTICIDE/PCB ORGANICS
	CLP SEMI-VOLATILE EXTRACTION	1 OLM01	CLP SENI-VOLATILE ORGANICS
	CLP FESTICIDE/PCB EXTRACTION	2 OLHOI	CLP PESTICIOE/PC8 ORGANICS
	CLP SEMI-VOLATILE EXTRACTION	2 OLMO1	CLP SEMI-VOLATILE ORGANICS
	CLP PESTICIDE/PCB EXTRACTION	S 01.801	CLP PESSICIDEZPOB ORGANICS
	CLP SEM1-VOLATHE EXTRACTION	5 01.1101	CLP SEMI-VOLATILE ORGANICS
SW 7.3.4.1	TOTAL AVAILABLE SULFIDE EXTRACTION	5 SW846-9030	SULF 10E
SUB46-1310	NEUTRAL WATER LEACHING METHOD (MOD)F)ED	1 SW846-9030	SULF IDE. (LEACHATE)
		2 SN846-9030	SULFIDE (LEACHATE)
		S EPA 340.1 EPA 350.2 EPA 375.4	FLUDRIDE DISTILLATION (LEACHATE) AMMONIA DISTILLATION SULFATE (LEACHATE)
	EXTRACTION PROCEDURE FOXICITY (EPFOX)	5 SW846- 3010	RAT:
	NEUTRAL WATER LEACHING METHOD (HODIFIED	5 SW846-3010	FAA UR ICP ACID DIGESTION OF LEACHATE SA COL
	EXTRACTION PROCEDURE FOXICITY (EPFOX)	S SW846-3020	GFAA ACID DIGESTION OF LEACHATE SAMPLES
	NEUTRAL WATER LEACHING METHOD (MODIFIED	S SUB46-3020	GFAA ACID DIGESTION OF LEACHATE SAMPLES
	EXTRACTION PROCEDURE COXICITY (EPFOX)	5 SUB46-8150	DIAZONETHANE HERBICIDE DERIVATIZATION
	NEUTRAL WATER LEACHING METHOD (MODIFIED	S 54846-7010 SU846-9030	CYANIDE LEACHATE DISTILLATION SULFIDE (LEACHATE)
SU846-1311	10X CHAR LEACHING PROCEDURE (ICLP METALS	4 SUR46-3030 HOD	ACIU DIGESTION.OFOILS
	TOX CHAR LEACHING PROCEDURE (FCLP W/ ORG	4 SUB46-3030 NOD	ACID DICESTION OF OILS
	TOX CHAR LEACHING PROCEDURE (ICLP W/ORG)	4 SUR46-3030 HOD	ACID DIGESTION OF OILS
	FOX CHAR LEACHING PROCEDURE (FCL.P. METALS	5 SW846-3010	FAA OR ICP ACID DIGESTION OF LEACHATE SA
	TOX CHAR LEACHING PROCEDURE CICLP WORGE	S 8U846-3010	Rev Nov
	TOX CHAR LEACHING PROCEDURE (FCLP W/ ORG	0105 -918M3 5	risi 7. l
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GFAA ACID DIGESTION OF LEACHATE SAMPLES	GFAA ACID DIGESTION OF LEACHATE SAMPLES	GFAA ACID DIGESTION OF LEACHATE SANPLES	ACIU DICESTION OF OILS	ACID DIRECTION OF OILS	ACID DIGESTION OF OILS GC SEPARATORY FUNNEL LIQUID-LIQUID EXTRA	GC SEPARATORY FUNNEL LIQUID-LIQUID EXTRA	GC/MS SEPARATORY FUNNEL LIQUID-LIQUID EX	GC/MS SEEARATORY EUNNEL LIQUID:-LIQUID EX	folp GC SEPARATORY FUNNEL LIQUID-LIQUID	TCLP GC SEPARATORY FUNNEL LIQUID-LIQUID	fci.p' GC/MS SEFARATORY FUNNEL LIQUID=LIQU	TCLP GC/MS SEPARATORY FUNNEL L.IQUID-LIQU	GC/HS WASTE DILUTION FOR ORGANICS	CC/NS WASTE DILUTION FOR ORGANICS	TCI.P GC WASTE OII.UTION FOR ORGANICS	TCLF GC WASTE DILUTION FOR ORGANICS	TCLF GCZNS WASTE DILUTION FOR ORGANICS	TCLP GC/MS WASTE BILLUTION FOR ORGANICS	WASTE DILUTION FOR ORGANICS	WASTE DILUTION FOR ORGANICS	ALCOHOLS BY GC:FID	DIAZONETHANE HERBICIDE DERIVATIZATION	ICLP DIAZOMETHANE HERBICIDE DERIVATIZATI	TCL.F DIAZOMETHANE HERBICIDE DERIVATIZATI	
			100	100	10H																				
3020	3020	3020	3030	30.30		3510	. 55 1 0	3510	3510	3510	3510	3510	1580	3580	3580	3580	3580	3580	3580	3580	8015 9015	8150	B150	8150	
SU846-3020	3V846-3020	SW846-3020	SUR46 - 3030	SN846-3030	SUB46-3030 SUB46-3510	SUB46-3510	SV846 5510	3V846-3510	SV846-3510	SUR46 - 3510	SW846-3510	SUB46-3510	SV846- 5580	SUR46-3580	SV846-3580	SUB46 -3580	SU846-3580	SUB46-3580	SW846-3580	sun46-3580	SV846-8015 SV846-8015	SV846-8150	SUB46 - 8150	SNA46-8150	
្ន	i)	S)	: Ø	S.	N SU CO	N S	S S	iV S	S)	i)	S.	Ŋ	S.	Ŋ	ις: C3	N S	S.	Ŋ	₩	S.	r r	S S	. N	ស	
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ME FALS	W/ORE !	W/ ORG	HE TALS	W/ORG+	U/ ORG	W/ORE >	W/ ORG	W/ORF	W/ ORG	WORE	W/ ORG	W/ORG+	W/ ORG	4/086 b	U/ ORG	W/ORG +	W/ ORG	W/ORG+	W/ ORG	W/ORG.		W/ORG+	U/ ORG	W/ORG+	
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								5 T	SUB46-7761	SILVER GEAN	
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	GFAA A	010	GFAA ACID DIGESTION OF LEACHA	ž O	. LEA	نبز	SAMPLES	8	EPA 206.2	ARSENIC GFAA (3 POINT MSA)	
	GFAA A	010	GFAA ACTU DIGESTION OF AQUEOU	Ö	AGU	S	SAMPLES	ន	SW846 7060	ARSENIC GFAA (1 POINT HSA)	
			,			;					
	CFAA A	<u>ء</u>	CFAA ACIU DIGESTION OF LEACHA	ž	الله الله الله	CHATE	SAMPLES	S.	SUCA6 - 7060	STATE OF THE STATE	
								<u> </u>	SN846-7060	NIC CFAA (3 POINT MBA)	
								8	SUB46-7060	ARSENIC GFAA (1 POINT	
								ᄧ	SN846-7060	ENIC CFAB. (3. F	
								<u></u>	SW846 -7131	GFAA (1 POINT	
								ਲ	SW846-7131	(3 POINT MSA)	
								ਲ	SU846 -7131	UM SEAA.	
								2	SU846-7421	GFAA (1 POINT	
								8	SU846-7421	LEAD GFAA (3 POINT MSA)	
								ਲ	SW846- /421	LEAD GFAA	;
								8	~	LEAD GFAA (3	
								S	7	I POINT	
								8	SUB46-7740	CIUM SEAA (3 POINT MSA)	
								3	SU846-7740	SELENIUM GFAA (1 POINT	
				•				ਲ	SUB46-7740	CFAA (3 P	
								<u>8</u>	SV846-7841	GFAA (1 POINT	i
								ซ	SU846 -7841	THALLIUM CFAA (3 POINT MSA)	
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004 04 04 - 040		0.2		7	į				SUB46-6016	ALCENIER CICP-SED	
									SU846:-6010	ICP	
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GFAA ACID DIGESTION OF LEACHATE SAMPLES GFAA ACID DIGESTION OF LEACHATE SAMPLES GFAA ACID DIGESTION OF LEACHATE SAMPLES	1 SW846-7131 1 SW846-7131 3 SW846-7131	CADMICH GFAA (1 POINT MSA)
, 67	12-9	1
EOUS SAMPLES		ADMIUM GFAA (3 POINT CLP CAUMIUM GFAA (1 P
,	SUB46-7191 SUB46-7201 SUB46-7211 SUB46-7421	CHRONIUM GFAA COBALT GFAA COPPER GEAA (SW) LEAD GFAA LEAD GFAA (3 POINT MSA)
ACID DIGESTION OF LFACHATE SAMPLES	SUB46-7421 SUB46-7421 SUB46-7421	LEAD GFAA (3 POINT HSA) TCLP LEAD GFAA (1 POINT HSA) TCLP LEAD GEAA (3 POINT HSA)
ACID DIGESTION OF AQUEOUS SAMPLES	1 SU846-7461 SU846-7521 SU846-7740	NOLYBDENUM GFAA Nickel Gfaa Selenium Gfaa
GFAA ACID DIGESTION OF LEACHATE SAMPLES	5W846-7740 5W846-7740 5W846-7740	SELENIUM-GEAA (3-POINT-MSA) TCLP SELENIUM GFAA (1 POINT MSA) TCLP SELENIUM GFAA (3 POINT MSA)
AQUEOUS SAMPLES	1 57.461 54846-7841 54846-7841	SILVER GFAA THALLIUM GFAA THALLIUM GFAA (3 POINT MSA)
GFAA ACID DIGESTION OF LEACHATE SAMPLES	1 SU846 - 7841	THALLIUM GFAA (3 POINT MBA)
AQUEOUS SARPLES	1 SU8467911	VANADIUM GFAA
	Subade - 1021 Subade - 7041 Subade - 7041 Subade - 7091 Subade - 7131 Subade - 7131 Subade - 7201 Subade - 7201 Subade - 7201 Subade - 7201 Subade - 7201 Subade - 7201	ALUMINUM GFAA ANTIMONY GFAA ANSENIC GFAA BERYLLIUM GFAA CADMIUM GFAA COBALT GFAA COPPER GFAA (SW) LFAD GFAA NICKEL GFAA

HAGNESIUM FAA Manganese faa Mercury Cvaa

LEAD GFAA

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SW846=7380 SW846-7420 SW846-7421 SW846:7450

ACID_DIGESTION_OF_OILS

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MOLYBOENUM CF

MOLYBDENUM F.

MERCURY_CYAA

POTASSIUM FAA Selenium gfaa

SILVER CFAA

SW846-7761

SILVER FAA

NICKEL GFAA

NICKEL FAA

SUB46-7460 SUB46-7470 SUB46-7471 SUB46-7480 SUB46-7520 SUB46-7521 SUB46-7521 SUB46-7521

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CADMIUM (ICP_SEQ)	CARCIUM (ICP-SED)	1 CP-3	2	COPPER (177-SEQ)) - Z	-	TRON TOP	 LITHIUM ICP	10L	TAREBURGE (CT - ORE)	HOLYBDENUM ICE	KICKEL (ICP-SEG)	NICKEL ICF Potassium ICP	LICON IC	ILVER (ICP-SEQ)	SILVER ICP	TIUN ICP	HALLIUR (10P-SEQ)	MALLIUM ICP	TITANIUM ICE	VARADION (ICT-SES)	_	ZINC (ICP-SEQ)	ALUMINUM FAA	LUNINUM GFAA	NATUROXY FAA	RSENIC GFAA	ROUTIC GTAA (M POINT MSA)	ERYLLIUM FAA	ERYLLIUM GFAA	CADMIUM EAA	-	CHRONIUM FAA	CHRONIUM GFAA	COBALT GEAA	COPPER FAA			
846-6010	6-6010	846-6010	0-6010	6-6010	6-6010	0109-91	16-6010	6466010	16-6010		46-6010	16-6010		0109-91	846-6010		0.09.918	946-6010	57.04-6-100 50.04-6-6010	010991	0104-6-10 V	16-6010	16-6010	17 0109-978789 14 0807-97978	1501-91	SU846:7040 AN	V 0902-91	16-7060		1601-91	50.017-9180 50.017-9180	16-7140	16-7190	SUB46-7191 CT	846-7201	846-7210			
ACID DIGESTION OF OILS																																							
SN846-3030 HOD ACE																																							

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	scription		latrix Ti Method	Description	
W846-3050	FAA OR ICP ACID DISESTION OF			SILVER ICP. (1 POINT MSA)	
			SW846-6010	SILVER ICP (3 POINT HBA)	
			SW846-6010	SILVER ICP-SEQ (1 POINT HSA)	
			SW846-6010	SODIUM ICP (1 POINT MSA)	
			SUB46-6010 SUB46-6010	STRONTIUM ICP	
			SW846-6010	1HALLIUM_(1CP-SEQ)	
	- · · · ·		SW8466010	THALLIUM ICP	
			SU846-6010	TIN ICP	
			SW846-6010	TIFANIUM ICP	
			SU846-6010	VANADIUM (ICP-SEQ)	
			5846-6010	VANADIUM ICP	
			SW846-6010	YANADIUM ICP (1_POINT MSA)	
•			5W846~6010 SW846-6010	VANADIUM ICP-SEQ (1 POINT HSA) YTTRIUM ICP	
			SW846-6010	ZINC (ICP-SEQ)	
			SU846-6010	ZINC 1CP	
			SW8466010	ZINC ICP (1 POINT MSA)	
			SUB46-6010	ZINC ICP (3 POINT USA)	
	•		SU846-6010	ZINC ICP-SEQ (1 POINT MSA)	
			SW846-7020	ALUMINUM FAA	
	GFAA ACID DIGESTION OF S/S/S	SAMPLES	5 SW846-7021	ALUHINUM GFAA	IAMB
	FAA OR ICP ACID DIGESTION OF	S/S/S SAMPL	5 SH846-7040	ANTIHONY FAA	in in
	GFAA ACID DIGESTION OF S/S/S	SAMPLES	5 SW8467041	ANTINONY GFAA	PRI
		4000 223	54846-7060	ARSENIC CFAA	TAB EPA
			SW846-7060	ARSENIC GEAA (3 POINT MBA)	TABLE EPARAT
	FAA OR ICP ACID DIGESTION OF	3/3/3 SAMPL	5 SU846-7080	BARIUM FAA	A T H 5
	W 124 C		SN846-7080	BARIUM FAA (1 POINT MSA)	. 2 ON
			SU846-7080	BARIUM FAA (3 POINT MSA)	
			SN846-7090	BERYLLIUM FAA	Ã
			SU846-7090	BERYLLIUM FAA (1 POINT MSA)	METH
			SN846-7090	BERYLLIUM FAA (3 POINT MSA)	MODS:
	GFAA ACID DIGESTION OF 8/3/3	SAMPLES	5 SU846-7091	BFRYLLIUM GFAA	Ø
	FAA OR ICP ACID DIGESTION OF	S/S/S SAMPL	5 SW846-7130	CADMIUM FAA	
			SUB46-7130	CADMIUM FAA (1 POINT MSA)	
			SW816-7130	CADHIUM FAA (3 POINT MSA)	
	GFAA ACID DIGESTION OF S/S/S		5 SW846-7131	CADMIUM GFAA	
			SW846-7131	CADMIUM GFAA (1 POINT MBA)	
			SW846-7131	CADMIUM GFAA (3 POINT MSA)	
	FAA OR ICP ACID DIGESTION OF	S/S/S SAMPL	5 SW846-7140	CALCIUM FAA	
			04046 3464	CHRONIUM FAA	
			SU846-7190	CHROMIUM FAA (1 POINT MSA)	
			SU846-7190	CHRONIUM FAA (3 POINT MSA)	Sec Rev Nov
	GFAA ACID DIGESTION OF S/S/S	SAMPLES	5 SN846-7191	CHRONIUM GFAA	ecticlevis
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			SUB46-6010 SUB46-6010	GARIUM (CP-SEQ) Barium icp	
			SW846-6010	BARIUM ICE (1 POINT BSA)	1
			SUB46-6010	RIUM ICP (3 POINT MSA)	
			SU846-6010	BARIUM ICP-SEG (1 POINT NSA)	
			SUB46-6010	BEAYLLIUM (1CP-SED)	
			SW846-6010		
			0109-07010	-	
			SUB46 - 6010	BORDA JCP (1 POINT MSA)	!
			16-6010	UM (ICP-SEQ)	
			46-6010	20	:
			46-6010	ICF (I POINT	
			16-6010	CADMIUM ICP (3 POINT MSA)	
			SU846-6010	CALCIUM ICP	
			0109-91		
			9199-919		
				THIS TO BOTH	
			SUR46-6010	1CP-SED (1 PO	
			SU646-6810	(CPSEQ)	s
			SUB46 - 6010	-	Μ
			SV846-6010	COPPER (ICP-SEQ)	PL
			SUB16-6010	106	Ε
•			SU646-6010	H K	P
			SUB46-6010	IRON (ICP-SEQ)	
			SV8466010	LEON ICP	i
			SUB46-6010	ICP (1 POINT MSA)	BL AR
			SW846-6010	CP-8	
			SUB46-6010	-	5 L(
				TEND TO A BOTHT MEAN	
				INION EV SUI	
			0100-010NG	CO TO CHALSTI	Œ'
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			SUB46-6010		lOI
			AUG 46-4616	RACKES CO (1 POINT REA)) DS
	•		SU846-6010	CANESE (ICP-SEQ)	
			SU646-6010	10P	
			0109-91	ICP (1 POINT MSA	
			0.09-95	TARGARESE ICT-SEC (1 POINT MAX)	
			SUB46-6010 SUB44-4010	TOLITHORNEY LON (1 POLET MAX)	
			9109-91	CP - SEQ)	
			16-6010	ZICKEL ICP	,
			0109-91	CKEL ICP (1 POINT	
				RICKEL ICE - FOIR BOX	
	1	1	SUS-6-5010	SION ICP	
			16-6010	10.	Re No
			16-6010	CP8EQ)	ev:
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		Matrix f Method	06.30.10(10)
		. ;	
SN846-3050	FAA OR ICP ACID DIGESTION OF 525/5 SAMPL	S_SW846:: [840	THALL TUN FAA LL POINT HBAI
	GFAA ACID DIGESTION OF S/S/S SAMPLES	S SUB46-7841	THALLIUM GFAA
	FAA OR ICP ACID DIGESTION OF S/S/S SAMPL.	S SEG46-7641 SEG46-7810 SEG46-7810	THALLIUM GFAA (3 FOINT MSA) TIN FAA VANADIUM FAA VANADIUM FAA (1 POINT MSA)
	GFAA ACID DIGESTION OF S/8/S SANPLES		GFAA
	FAA OR ICP ACID DIGESTION OF 3/3/3 SAMPL	S SUB46-7950 SUG46-7950 SUB46-7950	ZINC FAA ZINC FAA (1 FOINT MBA) ZINC FAA (3 FOINT MBA)
SU8163510	GC SEPARATORY FUNNEL LIQUID-LIQUID EXFRA	2012 2012 2012 2012 2012 2012 2012 2012	ORGANOCHLORINE PESTICIDES BY GC:ECD. ORGANOCHLORINE PESTICIDES BY GC:ECD (19 ORGANOCHLORINE PESTICIDES BY GC:ECD (4 P DRGANOCHLORINE PESTICIDES BY GC:ECD (7 P PCB/PESTICIDE SCAN GC:ECD
	TCLP GC SEPARATORY FUNNEL LIGUID-LIGUID	1 SW846-8080	TCLF URGANOCHLORINE PESTICIDES BY GC:ECO
	GC/HS SFPARATORY FUNNEL_LIQUID-LIQUID EX	1 \$4846-8270 \$4846-8270	APPENDIX IX SEMIVOLATILE ORGANICS
	GC/HS SEPARATORY FUNNEL LIQUID-LIQUID (A	1 SW846-8270	SEMI-VOLATILE ORGANICS (ACID FRACTION)
	GC/NS SEFARATORY FUNNEL LIQUID-LIQUID (B	1 SW846-6270	SEMI-VOLATILE ORGANICS (BASE/NEUTRAL FRA
	GC/HS SEPARATORY FUNNEL LIQUID-LIQUID EX	1 SWR46-R270 SW846:8R70	SEMI-VOLATILE ORGANICS (BASE/NEUTRAL/ACI
	TELP CC/MS SEPARATORY FUNNEL LIQUID-LIQU	1 SUR46-8270	TCLP SEMIVOLATILE ORGANICS (LANDBAN)
	GC/MS SEPARATORY FUNNEL LIQUID-LIQUID EX	1 SW846-8270	foll SEMIVOLATILE ORGANICA (TOXICITY CHA
	TELP GC/HS SEPAKATORY FUNNEL LIQUID-LIQU	1 SUB46-8270	ICLP SEMIVOLATILE ORGANICS (TOXICITY CHA
	GC SEPARATONY FUNNEL L.IQUID-L.IQUID EXFRA	52 52 52 52 52 52 52 52 52 52 52 52 52 5	ORGANICHLORINE PESTICIOE(S) BY GC:ECO ORGANOCHLORINE PESTICIDES BY GC:ECO (19 ORGANOCHLORINE PESTICIDES BY GC:ECO (4 P ORGANOCHLORINE PESTICIDES BY GC:ECO (7 P PCB/PESTICIDE SCAN GC:ECO
	GC/MS SEFARATORY FUNNEL LIQUID-LIQUID EX	2 SUB46-B270 SUB46-B270	AFPENDIX IX SEMIVOLAFILE ORGANICS 1DEM SEMI-VOLATILE TARGET COMPOUND LIST
	GC/MS SEPARATORY FUNNEL LIQUID-LIQUID (A	2 SU846-8270	SEMI-VOLAFILE ORGANICS (ACID FRACTION)
	GC/HS SEPARATORY FUNNEL LIQUID-1.1QUID (B	2 SUR46-8270	SEMI VOLATILE ORGANICS (BASE/NEUTRAL FRA
		(preprest.sql)	

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Uescription	COBALT FAA (1 POINT MSA)	COBAL F GFAA	COPPER FAA (1 POINT MBA) COPPER FAA (3 POINT MBA)	COPPER GFAA (SU)	IRON FAA (I POINT MSA) LEAD FAA (I POINT MSA) LEAD FAA (I POINT MSA) LEAD FAA (3 POINT MSA)	LEAD GFAA (1 POINT MGA)	LEAD GFAA (3 FOINT MSA) HAGNESIUM FAA (1 POINT MSA) HAMMENHSSF FAA	MANGANERE FAA (1 POINT MBA) MOLYBDENUM FAA	HOLYBOENUM GFAA	NICKEL FAA (1 POINT MSA) NICKEL FAA (3 POINT MSA)	NICKEL GFAA	POTASSIUM ENA	SELENIUM GFAA	SELENIUM GFAA (1 POINT MSA) SELENEM GFAA (3 POINT MSA) SILVER FAA (1 POINT MSA) SILVER FAA (1 POINT MSA)	SILVER GFAA	SODIUM FAA (1 POINT MSA) STRONTIUM FAA THALLIUM FAA (1 POINT MSA)	
icrix f methou	4 SUB16:7208	5 SW646-7201	S. SKR46-7219 SU846-7219 SU846-7210	5 SW846-7211	0271-97875 0271-97075 0271-97075 0271-97075 0271-97075 0271-97075 0271-97075 0271-97075 0271-97075	5 SW846-7421 SW846-7421	1311-91035 0511-91035 0511-91035		5 SH846-1481	SUBAC-1550 81646-7529 81646-7529	5 SUB46-7521	5_SUB46-7610	S SUB46-7740	SUBAL-1740 SUBAL-1740 SUBAL-1760 SUBAL-1760 SUBAL-1760	1921-91903 S	5 52 54 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	(preplest.sql)
₹	OF 8/8/8 SAMPL	S/S/S SAMPLES	OF 5/3/3 SAMPL	S/S SAMPLES	OF 373/3 SAMPL	S/8/S SAMPLES	OF 8/5/5 SANPL		S/S/S SANPLES	OF \$7575 SAMPL	S/S SAMPLES	OF \$2578 SAMPL.	S/S SAMPLES	OF 5/5/5 SAMPL	S/S/S SAMPLES	OF 5/3/5 SAMPL	-
uo.	ICP ACID DIGESTION	GFAA ACID DIGESTION OF SA	FAA OR ICP ACID DIGESTION OF S73/S SAMPL	GFAA ACID DIGESTION OF 8/8/S SAMPLES	FAA OR ICP ACID DIGESTION OF	GFAA ACID DIGESTION OF S/	CP ACID DIGESTION OF		GFAA ACID DIGESTION OF S/	CP ACID DIGESTION OF	GFAA ACID DIGESTION OF 8/S/S SAMPLES	FAR OR ICP ACID DICESTION OF	GFAA ACID DIGESTION OF 8/8/8	CP ACID DICESTION	GFAA ACID DIGESTION OF S/	CP ACID DIGESTION OF	
scription	FAA OR	GFAA ACT	FAA OR 1	GFAA ACE	FAA OK 1	GFAA ACI	FAA OR 1CP		GFAA ACI	FAA OR ICP	GFAA ACI	FAA OR I	GFAA ACI	FAA OR 1CP	GFAA ACI	FAA OR 1CP	_
fact	SN646-3050	:				: :				,	: - <u>-</u>			·			

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SV846- 3580	WASTE DILUTION FOR ORGANICS	S 57546-6060	O GIFTENTLS (TCBS)
			CREATORING THROUGHT TO BE TO THE DESCRIPTION OF THE TRUE
	Confirmation of the Confir	3000 Broke	STOLINITES BY CO-FFC A
		2020: 548UA	PERTICIPES BY CC:ECD
		SW846 -8080	SCAN GC : ECD
	PCB OIL EXTRACTION	S SW846-8080	POLYCHLORINATED BIPHENYLS (PCBS)
	TELP GE WASTE DILUTION FOR ORGANICS	S SUN46-8080	TCLP ORCANOCHLORINE PESTICIDES BY GC:ECD
	GC/HS WASTE DILUTION FOR ORGANICS	5 SW846-8270 SW846-8270 SW846-8270 SW846-8270	APPENDIX IX SEBIVOLATILE ORGANICS
	TCLP GC/NS WASTE BILUTION FOR ORGANICS	S SW846-8270	TCLP SEMIVOLAFILE ORGANICS (LANOBAN)
	CC/MS WASTE DILUTION FOR ORGANICS	5 SUR46-8270	TCLP SEMIVOLATILE ORGANICS (TOXICITY CHA
1	TCLP GC/HS WASTE DILUTION FOR ORGANICS	5 SW846-8270	FCLP SEMIVOLATILE ORGANICS (FOXICITY CHA
SUB46-5030	PURGE AND TRAP METHOD FOR GREANIC ANALYT	5 EPA 601	PURCEABLE HALOCARBONS BY GC/ELCD Haiocemated and abomatic volatiff orcant
		602 16-8010 16-8020 16-8021	PURCEABLE AROMATICS BY GC/PID ELCD HALOGENATED VOLATILE ORGANICS BY G PURCEABLE AROMATICS BY GC/PID. HALOGENATED AND AROMATIC VOLATILE
SUB46-1410	MERCURY CVĄA ACID DIGESTION OF LEACHAIE	1 SUR46-7470 SU846-7470	TCLP MERCURY CVAA (3 FOINT MSA)
		S SUB46-7470 SUB46-7470	TCLP MERCURY CVAA (1 POINT HSA)
3UR46 - 8150	DIAZONETHANE HERBICIDE DERIVATIZATION	3UR46-8150 SUR46-8150 SUR46-8150	CHLORINATED HEBBICIDES (2,4 D. AND SILVEX CHLORINATED HERBICIDES (2,4 D, SILVEX AN TCLP CHLORINATED HERBICIDES (2,4-D AND S
	ICLP DIAZOMETHANE HERBICIDE DERIVATIZATI	1 SUB46-8150	TCLP CHIORINATED HERBICIDES (2,4-D AND S
	DIAZONETHANE HEKBICIDE DERIVALIZATION	2 SUR46-8150 SU846-8150	CHLORINATED HERBIGIDES (2,4 D. AND SILVEX CHLORINATED HERBIGIDES (2,4 D, SILVEX AN
	:	SUR46-0150 SUR46-0150	CHLORINATED HERBICIDES (2,4 D. AND SILVEX CHLORINATED HERBICIDES (2,4 D, SILVEX AN
		5 5W846-8150 5W846-8150 5W846-8150	CHIORINATED HERBICIDES (2,4 D AND BILVEX CHIORINATEO HERBICIDES (2,4-0 AND S D
	ICLP DIAZOMETHANE HERBICIDE DERIVAFIZAFI	5 SW846-8150	ICLP CHLORINATED HERBICIDES (2,4-0 AND S
		(preplest.sql)	8 of

	crt.	c :	cri, n		Ė	*:	.leth	14.22	ption		1		
SV846-3510	6c/NS SE	PARATORY F	GC/MS SEPARATORY FUNNEL LIQUID-LIQUID	וחנט-ד נפחנ	D EX	878 878 878 878	846-8270 846-8270 846-8270	SENI-V TCLP 8	SEMIVOLATILE SEMIVOLATILE	NICS 18A	SE/NEUTRAL, (LANDBAN) (TOXICITY	/AC1	
	GC SEPAR	ATORY FUM	SEPARATORY FUNNEL LIQUID-LIQUID EXTRA	1-LIAUID E	XTRA	SUS E	146-2020	ORGANO	ORINE	PESTICIDES BY	۵	L T	
	36 C/NS SE	PARATORY	GC/HS SEPARATORY FUNNEL L. (QUID-1, IQUID EX	10 f D-1, [QU [1 X	97.5	SW846-8270 SW846-8270 SW846-8270	AFFENDIX IDEN SENI SENI-VOLA	NDIX IX SEMINOL. SEMI-VOLATILE -VOLATILE_ORGAN	ATTLE ORG TARGET CO	ANICS MPOUND L.	LIST	
	GC SFPAR	ATORY FUN	SEPARATORY FUNNEL LIQUID-LIQUID EXTRA	-t.19UJD E	XIRA	shs s	846 -B080	ORGANOCHL	CHLORINE PESTI	C1 DE (S)	BY GC:ECD	0	
	ICLP GC	SEPARATOR	ICLP GC SEPARATORY FUNNEL LIQUID-LIQUID	.10010-1.19	010	S SUB	846-8080	TCLP 0	DRCANOCHLORINE	PESTICIO	ES BY GC	: ECO	
	GC/NS SE	PARATORY.	GC/HS SEPARATORY_FUNMEL_LIQUID-LJQUID_EX	1U10-1,1QU1	D EX	Pars S	846-B270	161.8	SEMIVOLATILE	ORGANICS	(TAMBBAN)		
	TCLP GC/	HS SEPANA	TCLP GC/AS SEPARATORY FUNNCI. LIQUID-LIQU	L1991D-	L. 10U	S 5W84	846-8270	TCLP S	EMIVOLAFILE	83	(LANOBAN)		
	GC/NS SE	PARATORY	GC/HS SEPARATORY FUNNEL LIQUID-LIQUID EX	NUID-LIQUI	. C	5 8084	846-8270	TCLP S	SENI VOLATILE	ORGANICS (1	TOXICITY	CHA	
	TCLP GC/	HS SEPAKA	TCLP GC/HS SEPANAFORY FUNNEL LIGUID-LIGU	-0166F1 T	L. 19U	ens s	846-8270	TCLP S	SEMIVOLATILE	ORGANICS (1	(TOXICITY	CHA	SAM
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	MATRIX 1 - NON-SPECIFIC WATER								
	MATRIX 2 - GROUND WATER								
	MATRIX 3 - DRINKING WATER			-					
	MATRIX 4 - OIL								
	MATRIX 5 - SOIL/SEDIMENT/SLUDGES, O	THER						···	
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	CYANIDE AMENABLE DISTILLATION	1 50848-9012	CYANIDE AMENABLE TO CHLORINATION (AUTORA
	CYANIDE DISTILLATION	1 SW646-9012 SUB46-9012	CYANIDE LEACHATE (AUTOMATED) CYANIDE, TOTAL (AUTOMATED)
		E 11.001 SW846-9010	CYANIDE, FOTAL (AUTOHATED) (CLP) CYANIDE, FOTAL (MANUAL)
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	CYANIDE AMENABLE DISTILLATION	4 SW846-9010	CYAN) DE. AMENABLE
	CYANIDE DISTILLATION	4 SUS46-9010	CYANIDE, TOTAL (MANUAL)
	CYANIDE AMENABLE DISTILLATION	4 SW846-9012	CYANIDE AMENABLE TO CHLORINATION (AUTOMA
	CYANIDE DISTILLATION	4 SW816-9012	CYANIDE, TOTAL (AUTONAFED)
		5 11.801	CYANIDE, FOTAL (AUTONATED) (CLP)
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	CYANIDE DISTILLATION	5 57846-9910 57846-9910	CYANIDE, LEACHAFE (MANUAL) CYANIDE, FOTAL (MANUAL)
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	CYANIDE DISTILLATION	S 54646-3618	CYANIDE LEACHATE (AUTOMATED) CYANIDE, TOTAL (AUTOMATED)
SN816-916S	PHENOLS DISTILLATION	1 SW846-9066	PHENOLS JAAP (AUTOMATED)
		2 SUB46-9066	PHENOLS 4AAP (AUTOMATED)
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		4 SW646-9866	PHENOLS 4AAF (AUTOMATED)
		3 SUBA6-9066	PHENOLS AAP (AUTOMATED)
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6.0 Sampling Procedures

6.1 Introduction

During normal operations, no sampling will be performed by EMS Heritage. The sampling procedures for specific projects will be addressed on a case by case basis by the project manager.

In the event that sampling is to be performed, procedures in the EPA Region IV SOP and QA Manual, dated April 1, 1986 will be used and referenced.

6.2 Sample Containers

EMS Heritage does provide sample containers and preservatives to clients. All sample containers will be for one-time use only, therefore no sample bottle cleaning will be performed at EMS Heritage. Sample containers are prepared as "kits" which make up all the containers needed for a sample with multiple tests to be performed, including additional containers required for QC sample analysis. Clients requiring coolers are shipped kits in a styrofoam cooler packed inside of a cardboard box. Those clients using their own coolers for delivery are shipped containers in a cardboard box. All shipments are sealed such that any tampering or attempt to open the shipping container is evident. There are basically two general types of container categories: 1) new pre-cleaned and 2) new, as-is from the manufacturer.

6.2.1 Pre-Cleaned Containers

The pre-cleaned containers will be used when the client's Data Quality Objectives (DQO's) demand them or when the client or QA Unit requests pre-cleaned containers. These pre-cleaned containers will primarily be used for clients performing superfund type investigations (EPA, CLP methodology) or any other sensitive sampling event. The cleaning procedures used are broken down into protocols for different test requirements (see Table 6.1, Sample Container Preparation Protocols). These pre-cleaned containers may be purchased either with or without a certificate of analysis (QA analyzed) pertaining to the container lot from the vendor. If QA analyzed containers are not available from the vendor, EMS Heritage will perform the analysis from randomly selected containers of the lot shipped when the client or QA Unit requires containers that have been QA analyzed. The use

Table 6.1

Sample Container Preparation Protocols

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-Protocol A-

Amber Glass Bottles
 Clear Glass Bottles
 Wide Mouth Amber Glass Jars
 Wide Mouth Clear Glass Jars

Cleaning Protocol A Specifications:

- 1. Wash containers, closures and teflon liners in hot tap water with laboratory grade non-phosphate detergent.
- 5. Rinse with pesticide grade methylene chloride.

2. Rinse three times with tap water.

6. Oven dry at 125°C. Allow to cool.

3. Rinse with 1:1 nitric acid.

7. Remove containers, closures, and teflon liners from oven.

4. Rinse three times with deionized water.

8. Place teflon liners in closures and place closures on container. Attendant to wear gloves and containers are not to be removed from preparation room until sealed.

Sample Containers Prepared According to Protocol A Are For Use In The Analysis Of:

Acidity/Alkalinity	Hardness
Sulfate	Chloride
Color	Oil & Grease
Phenols	рН
Phosphate	Turbidity

Settleable Residue/Suspended Solids	Mercury
BOD (Biological Oxygen Demand)	Metals
COD (Chemical Oxygen Demand)	Sulfide
Sulfite	Conductivity
TOC (Total Organic Carbon)	Cyanide
Extractable Organics	•

Table 6.1

Sample Container Preparation Protocols

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-Protocol B-

◆40 ml Borosilicate Clear and Amber Glass Vials ◆10 ml Polyethylene Vials
 ◆40 ml and 20 ml Flint Glass Vials ◆8 oz. and 4 oz. Septum Bottles
 ◆Wide Mouth Clear Glass Jars*
 and High Density Polyethylene Bottles (allow extra time for delivery)
 *Wide Mouth Clear Glass Jars prepared according to Protocol B are supplied with Teflon-lines polypropylene closures attached

Cleaning Protocol B Specifications:

- 1. Wash containers, septa, and closures in hot water 4. Oven dry containers, septa and closures at 125°C. with laboratory grade non-phosphate detergent.
- 2. Rinse three times with tap water.

- 5. Remove containers, septa, and closures, at 125°C.
- 3. Rinse three times with deionized water.
- 6. Place liners in closures, teflon side down, and place on containers. Attendant to wear gloves and containers are not to be removed from preparation room until sealed.

Sample Containers Prepared According to Protocol B Are For Use In The Analysis Of:

COD (Chemical Oxygen Demand) Purgeable (Volatile) Organics
Trihalomethanes Nitrate-Nitrite TOC (Total Organic Carbon)

Note: Protocol B cleaning is performed in an environmentally controlled facility, free of volatile organic vapors.

Table 6.1

Sample Container Preparation Protocols

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-Protocol C-

High Density Polyethylene Bottles and jars
 Polyethylene Cubitainers
 High Density Polyethylene Wide Mouth Jars
 High Density Polyethylene Products

Cleaning Protocol C Specifications:

- 1. Wash containers, closures, and teflon liners in hot 4. Rinse three times with deionized water. tap water with laboratory grade non-phosphate detergent.
- 2. Rinse three times with tap water.

5. Air dry in contaminant-free-environment.

3. Rinse with 1:1 nitric acid.

6. Place liners in closures and place closures on containers. Attendant to wear gloves and containers are not to be removed from preparation room until sealed.

Sample Containers Prepared According to Protocol C Are For Use In The Analysis Of:

All analysis noted for Protocol A except: Phenols; Oils & Grease; TOC (Total Organic Carbon) of QA analyzed containers may not be required in cases where the client will submit field blanks.

Containers that have been pre-cleaned or pre-cleaned and QA analyzed will be shipped to EMS Heritage with a custody seal. The certificate of analysis (when purchased) will be shipped inside the shipping container (box) and will remain there until the box is opened (custody seal is broken). Some client needs will not necessitate that the shipping container be opened in all instances. If this is the case, the certificate of analysis for bottle cleanliness must be retained by the client receiving the containers. In instances where preservatives must be added prior to transfer to the client and/or the original custody seal is broken on the container box, the certificate of analysis will be removed and filed for future reference. The containers will be resealed to establish a proper chain of custody.

All sample containers will be stored in their original packing containers. When packages of un-capped sample containers or closures are opened, they will be stored in plastic bags and sealed to prevent contamination during storage. Only EMS Heritage employees will have unsupervised access to the storage area. All other persons must be accompanied by an EMS Heritage employee while in the container storage area.

6.2.2 New, As-Is Containers

Each month EMS Heritage will use as-is sample containers to prepare blind performance evaluation samples. Any significant contamination will show up in these monthly reports, which is a check on container cleanliness. Any target analytes present at equal to or greater than the laboratory reporting detection limits will require that corrective actions be taken, including but not limited to, obtaining an alternate source of containers.

New containers used directly from the manufacturer without precleaning will have reagent water blanks analyzed for the tests which are conducted in that container type not covered in the monthly blind P.E. samples. The frequency of these analyses will be once per lot of the container type received or once per month when lot numbers are not available. These container blanks containing reagent water and preservative will be analyzed using the most sensitive techniques

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routinely performed from those containers. The blank samples will be entered into the LIMS system for analysis and reporting. The QA Unit will review this final report and retain the report for cleanliness of container documentation.

6.3 Preservation

A basic requirement for preservation of samples for many types of analyses includes sample shipping and storage at 4°C immediately after collection. Wet ice is the only acceptable means of reducing temperature for all environmental samples with the exception of any biological tissue samples (i.e. fish). Biological tissue samples may be shipped with dry ice.

Chemical preservatives are to be added on-site at the time of sample collection. As a convenience to our clients and an assurance that chemical preservation is not omitted, EMS Heritage supplies containers which already contain the required preservative in the proper amount and concentration. All preservatives consist of reagent grade chemicals. Lot numbers of all chemical used are tracked so that any questions relating to the reagent (preservatives) used are traceable to the lot of reagent used for preservation.

The chemical preservatives used for preserving samples are listed in Table 6.2, Aqueous Sample Preservatives.

Table 6.3, <u>Sample Requirements</u>, includes all of the parameters listed in Table 5.1 (QA Targets for Precision, Accuracy and Method Detection Limits) and specifies:

- ♦ Container type
- ♦ Required sample amounts for analysis
- ♦ Preservation requirements
- ♦ Holding times
- ♦ Methods

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Table 6.2
Aqueous Sample Preservatives

Reagent	Concentration	Pre-Preserved ¹ Container Amount	Required pH	Test Categories
Hydrochloric Acid	Concentrated	0.1 mL/40mL	<2	Volatile Organics
Hydrochloric Acid	0.1M	See Footnote ²	4 to 5	Acrolein, Acrylonitrile
Hydrochloric Acid	1:1	5mL/L	<2	Oil & Grease, TPH
Phosphoric Acid	Concentrated	4mL/L	<4	Phenois
Nitric Acid	Concentrated	5mL/Liter	<2	Metals, Alpha, Beta, Radium
Sulfuric acid	Concentrated	1mL/Liter	<2	Phenols, T-Phosphorus, Ammonia, COD, TKN, Nitrate-Nitrite, Oil & Grease, TOC
Sodium Hydroxide	Pellets	2 pellets/Liter	>9	Sulfide
Sodium Hydroxide	Pellets	2 pellets/Liter	>12	Cyanide
Zinc Acetate	Solid	0.1g/Liter	NA	Sulfide
Sodium Thiosulfate	Solid	80mg/Liter	NA	Coliforms, Organics
Ascorbic Acid	Solid	25mg/40mL	NA	SDWA Volatile Organics, Cyanide
Mercuric Chloride ³	10mg/mL	1mL/Liter	NA	SDWA Herbicides, SDWA N&P Containing Pesticides
Monochloroacetic Acid Buffer	See Footnote 4	1.2 mL/40mL	3	Carbamates

¹ Preservative added per sample amount

To be adjusted in the field

Alternative preservative preferred (HCl)

¹⁵⁶ mL of 2.5M Monochloroacetic acid and 100mL of 2.5M potassium acetate

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Test	Method	Reference	Container!	Water Sample	Required Sa	mple Quantity	Maximum	Maximum
		Politica de la composición del composición de la composición de la composición de la composición de la composición de la composición del composición de la composición de la composición de la composición de la composición de la composición de la composición de la composición de la composición de la composición de la composición de la composición de la composición de la composición de la		Preservation ^{2.3}	Water	S/S/S	Holding Time Water ⁴	Holding Time S/S/S
General Chemistry				1001.00 i 120 150.00 i 120 150.00 i 120				
Acidity	Titrimetric	EPA 305.1	P,G	Cool 4°C	200mL	50g	14 days	
Alkalinity	Titrimetric	EPA 310.1	P,G	Cool, 4°C	200ml.	50g	14 days	
Alkalinity	Titrimetric	EPA 310.2	P,G	Cool, 4°C	200ml.	50g	14 days	
Ammonia Nitrogen		EPA 350.1	P,G	Cool, 4°C H ₂ SO ₄ to pH <2	500ml.	100g	28 days	28 days
Ammonia Nitrogen		EPA 350.2	P,G	Cool, 4°C H ₂ SO ₄ to pH <2	500ml.	100g	28 days	28 days
Ammonia Nitrogen	Potentiometric	EPA 350.3	P,G	Cool, 4°C H ₂ SO ₄ to pH <2	IL	Sg	28 days	28 days
Asbestos		N105H 239					NA	NA
Biochemical Oxygen Demand	Potentiometric	EPA 405.1	P,G	Cool 4°C	1L	NA	48 hours	NA
Boron	Colorimetric	EPA 212.3	P	Cool, 4°C H ₂ SO ₄ to pH <2	200mL	NA	28 days	NA
Bromide	Colorimetric	SM405	P,G	None	50mL	NA	28 days	NA
Bromids	Titrimetric	EPA320.1	P,G	None	50mL	NA	28 days	NA
Chemical Oxygen Demand	Орев Кеймх	SM508A	P,G	Cool, 4°C H ₂ SO ₄ to pH <2	100mL	100g	28 days	28 days
Chemical Oxygen Demand	Mid-Level Titrimetric	EPA410.1	P,G	Cool, 4°C H ₂ SO ₄ to pH <2	100ml	100g	28 days	28 days
Chemical Oxygen Demand	Low Level	EPA410.2	P,O	Cool, 4°C H ₂ SO ₄ to pH <2	100ml,	100g	28 days	28 days
Chemical Oxygen Demand	Block Digester, Colorimetric	EPA 410.4	P,G	Cool, 4°C H ₂ SO ₄ to pH <2	100mL	100g	28 days	28 days
Chloride	Colorimetric, Automated	EPA 325.1	P,G i	None	100mL	5g	28 days	28 days

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Test	Method	Reference	Container ¹	Water Sample	Required Sa	mple Quantity	Maximum	Maximum
				Preservation ^{2.3}	Water	S/S/S	Holding Time Water ⁴	Holding Time S/S/S
Cynnide, Total	Antomated	SW 846-9012	P,G	Cool, 4°C NaOH to pH > 10	500ml.	100g	14 days ⁶	14 days
Cynnide, Total Available		SW 7.3.3.2	P,G	Cool, 4°C NaOH to pH > 10	1L	100g	14 days ⁶	14 days
Dissolved Oxygen	Membrane Blectrode	EPA 360.1	P,G	None	300mL	NA	Analyze Immediately	NA
Dissolved Oxygen	Winkler	EPA 360.2	P,G	Pix on site and store in dark	300mL	NA	8 hours	NA
Dissolved Solids, Total	Gravimetric	EPA 160.1	P,G	Cool, 4°C	500mL	NA	7 days	NA
Phioride	Ion Selective Electrode	EPA 340.2	P,G	Cool, 4°C	500mL	50g	28 days	28 days
Fluoride		EPA 340.1	P,G	Cool, 4°C	500mL	50g	28 days	28 days
Forms ide hyde		AOAC 20.047	o	Cool, 4°C	100mL	50g	28 days	28 days
Gross Alpha		EPA 900.0	P	HNO, to pH < 2	1 galion	NA	6 months	NA
Gross Beta		EPA 900.0	P	HNO, to pH < 2	1 galloa	NA	6 months	NA
Hardness	Calculation	SM 314A	P,G	HNO, to pH < 2	200mL	NA	6 months	NA
Hardness	EDTA, Titrametric	EPA 130.2	P,G	HNO, to pH < 2	100mL	NA	6 months	NA
Hezavalent Chromina	Colorimetric	SM 312B	P,G	None	300mL	100g	24 hours	7 days
Hexavalent Chromium	Colorimetric, Massal	SW 846-7196	P,G	None	300mL	100g	24 hours	7 days
Iodine	Leuco Crystal Violet	16th Ed SM 415A	P,G	Cool 4°C	500mL	100g	Analyze Immediately	
Kjeldahl Nitrogea, Total	Automated	EPA 351.2	P.G	Cool, 4°C H ₂ SO ₄ to pH <2	500ml.	10g	28 days	28 duys
Kjeldahl Nitrogea, Total	Potentiometric	EPA 351.3	P,G	Cool, 4°C H ₂ SO ₄ to pH <2	100mL	50g	28 days	28 days
Nitrogen, Total Organic	Calculation	EPA 351.3	P,G	Cool, 4°C H ₂ SO ₄ to pH <2	100ml.	50g	28 days	28 days

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Test	Method	Reference	Container!	Water Sample	Required Sa	mple Quantity	Maximum	Maximum
				Prescryation ^{2,3}	Water	S/S/S	Holding Time Water ⁴ 28 days 28 days 28 days 28 days 28 days NA 6 hours 48 hours 14 days ⁴	Holding Time S/S/S
Chloride	Colorimetric, Automated	EPA 325.2	P,G	None	500mL	Sg	28 days	28 days
Chloride	Titrimetric, Mercuric Nitrate	EPA 325.3	P,G	Nose	500mL	5g	28 days	28 days
Chloride	Ion Selective Electrode	14th Ed SM 320A	P,G	None	500mi.	Sg	28 days	28 days
Chloride		SM 407A	P,G	None	500mE.	Sg	28 days	28 days
Chlorine Demand	Starch-Iodine Titration	SAS	P,G	None	1L	NA	28 days	NA
Chlorine, Total		ASTM D-806	P,G	None			NA	
Coliform Bacteria, Fecal	Microbiological	SM 909C	P,G	0.008% Na ₂ S ₂ O ₃ ² Cool, 4°C	100mL	NA	6 hours	NA
Coliform Bacteria, Total	Microbiological	\$M 909A	P,G-Sterile	0.006% Na ₂ S ₂ O ₃ ³ Cool, 4°C	100mL	NA	6 hours	NA
Color Determination	HACH	EPA 110.3	P,G	Cool, 4°C	100mL	NA	48 hours	NA
Cyunide, Amenable		BPA 335.1	P,G	Cool, 4°C NaOH to pH > 10	1L	100g	14 days ⁶	14 days
Cynnide, Amenable	Colorimetric, Automated	EPA 335.2	P,G	Cool, 4°C NaOH to pH > 10	11.	100g	14 days ⁴	14 days
Cyanide, Total	Momel	EPA 335.2	P,G	Cool, 4°C NaOH to pH > 10	500mL	100g	14 deys ⁶	14 days
Cynuide, Total	Automated	EPA 335.3	P,G	Cool, 4°C NaOff to pH > 10	500mL	100g	14 days ⁶	14 days
Cyanide, Amenable	Mesusi	SW846-9010	P,G	Cool, 4°C NaOH to pH > 10	1L	100g	14 dayu ⁶	14 days
Cynnide, Amenable	Automated	SW 846-9012	P,G	Cool, 4°C NaOH to pH > 10	ıL	100g	14 days ⁶	14 days
Cynuide, Total	Masual	SW 846-9010	P,G	Cool, 4°C NaOH to pH > 10	500mL	100g	14 days ⁴	14 days

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Teit	Method	Reference	Container ¹	Water Sample	Required San	ple Quantity	Maximum	Maximum
				Preservation ^{2.3}	Water	S/S/S	Holding Time Water ⁴	Holding Time S/S/S
pH	Solids	SW 846-9045	P,G	Cool, 4°C	NA	50g	Analyze Immediately	Analyze Immediately
Phenole	Masual, 4AAP	EPA 420.1	O	Cool, 4°C H,PO,	500mL	100g	28 days	28 days
Phenols	Automated, 4AAP	EPA 420.2	G	Cool, 4°C H,PO ₄	500mL	50g	28 days	28 days
Phenois	Manual, 4AAP	SW 846-9065	o	Cool, 4°C H,PO4	500mL	50g	28 days	28 days
Phenois	Manual Distillation w/Automated, 4AAP	SW 846-9066	G	Cool, 4°C H,PO₁	500m1.	SO _B	28 days	28 days
Phenols	Spectrophotometric MBTH w/Distillation	SW846-9067	G	Cool, 4°C			28 days	28 days
Phosphorus, Ortho	Automated	EPA 365.1	P,G	Cool, 4°C	500m1.	50g	48 hours	
Phosphorus, Ortho	Colorimetric, Manual	EPA 365.2	P,G	Cool, 4°C	500mL	50g	48 hours	
Phosphorus, Total	Colorimetric, Manual	EPA 365.4	P,G	Cool, 4°C H ₂ SO ₄ to pH <2	500mL	50g	28 days	28 days
Radium		EPA 903.0	P,G	HNO ₃ to pH <2	1 Gallon	NA	6 months	NA
Radium 226		EPA 903.1	P,G	HNO, to pH <2	1 Gallon	NA	6 months	NA
Residual Chlorine		EPA 330.5	P,G	None	100mL	25g	6 hours	
Saturation Index (Calculated from pH, Alkalinity, TDS, Calcium, Temperature)	Langlier	SM203	P,G	Cool, 4°C	1000mL	NA	NA	NA
Settleable Matter	Imhoff Cone	EPA 160.5	P,G	Cool, 4°C	1000mL	NA	48 hours	NA
Silica, Total		EPA 370.1	P	Cool, 4°C	200mL	NA	28 days	NA
Specific Conductance	Conductivity Bridge	EPA 120.1	P,G	Cool, 4°C	100mL	NA	28 days	NA
Specific Conductance	Conductivity Bridge	SW 846-9050	P,G	Cool 4°C	100mL	NA	28 days	NA
Su life te	Turbidimetric	EPA 375.4	P,G	Cool 4°C	100mL	25g	28 days	
Sulfate	Turbidimetric	SW 846-9038	P,G	Cool, 4°C	100mL	25 ₈	28 days	

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Tost	Method	Reference	Container ¹	Water Sample	Required Sa	mple Quantity	Maximum	Maximum
				Preservation ^{2.3}	Water	S/S/S	Holding Time Water	Holding Time S/S/S
Nitrogen, Total Organic	Calculation	EPA 351.4	P,G	Cool, 4°C H ₃ SO ₄ to pH <2	500mL	100g	28 days	28 days
Nitrogen, Nitrate	Colorimetric, Bracine	EPA 352.1	P,G	Cool, 4°C H ₂ \$O ₄ to pH <2	100mL	50g	48 hours	
Nitrogea, Nitrate	Colorimetric, Automated - Cd Red.	EPA 353.2	P,G	Cool, 4°C H ₂ 5O ₄ to pH <2	100mL	50g	48 hours	
Nitrogea, Nitrite	Colorimetric, Automated	EPA 353.2	P,G	Cool, 4°C H ₂ SO ₄ to pH <2	100mL	50g	48 hours	
Nitrogea, Nitrate	Colorimetric, Automated-Hydrazine	EPA 353.1	P,G	Cool, 4°C H ₂ SO ₄ to pH <2	100mL	50g	48 hours	
Nitrogea, Nitrate-Nitrite	Colorimetric, Automated-Hydrazine	EPA 353.1	P,G	Cool, 4°C H ₂ SO ₄ to pH <2	100mL	50g	28 days	
Nitrogen, Nitrite	Spectrophotometric	EPA 354.1	P,G	Cool, 4°C	100ml.	50g	48 hours	
Nuisance Dust	Analytical Micro- Balance	NIOSH 500	Air Filter	NA	NA	NA	NA	NA
Oil & Grease	Gravimetric, Sep. Pessot	BPA 413.1	a	Cool, 4°C H ₂ SO ₄ to pH <2	2x1L	NA	.28 days	NA
Oil & Grease	Sozhlet, Gravimetric	SW 846-9071	G	NA	NA	50g	NA	28 days
Paint Pilter Test	Paint Filter	SW 846-9095	G	NA	NA	500g	NA	NA
Percent Solids	Gravimetric	SM 209A	P,G	Cool 4°C	250mL	50g	7 days	7 days
Percent Solids		ASTM D-2042	,					
Percent Solids	Contriluge	ASTM D-96						
pH	Electrode	EPA 150.1	P,G	Cool, 4°C	50mL	NA	Analyze Immediately	NA
pH	Blectrode	SW 846-9040	P,G	Cool, 4°C	SOun.L.	NA	Analyze Immediately	NA
pH	Paper	SW 846-9041	P,G	Cool, 4°C	50mL	NA	Analyze Immediately	NA

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Test	Method	Reference	Container!	Water Sample	Required San	mple Quantity	Meximum	Maximum
				Preservation ^{2.5}	Water	S/S/S	Holding Time Water ⁴	Holding Time S/S/S
Total Solids	Gravimetric	EPA 160.3	P,G	Cool 4°C	250mL	50g	7 days	7 days
Total Suspended Solids	Gravimetric	EPA 160.2	P,G	Cool, 4°C	500mL	NA	7 days	NA
Total Volatile Solida	550C Oven	EPA 160.4	P,G	Cool 4°C	250mL	10 g	7 days	7 days
Total Volatile Solids	Weight Percent	ASTM 2369	P,G	Cool, 4°C	250mL	10g	7 days	7 days
Turbidity	Nephelometric	EPA 180.1	P,G	Cool 4°C	100m].	NA	48 hours	NA
Volatile Acids	Distillation	SM 504B	G	Cool 4°C	1L	NA	_	NA
Volstile Suspended Solida	550C Oven	EPA 160.4	P,G	Cool, 4°C	250mL	NA	7 days	NA
Miscellaneous Physical Teets								
API (Specific Gravity)	Hydrometer	ASTM D-287	P,G	NA	NA	Oil: 100mL	NA	NA
Boiling Point	Distillation	ASTM D-86	P,G	NA	NA	Oil: 200mL	NA	NA
Cation Exchange Capacity		AOAC 57-3	P,G	Cool, 4°C	NA	100g	NA	14 days
Conting Density		ASTM 1475	P	NA	NA	NA	NA	NA
Density at 60P		ASTM D-70	P,G					
Plash Point	Pensky-Martens	A5TM D-93	o	None	120mL	NA		NA
Ignitability Potential	Solids Assessment	ASTM D-49623	P,G	NA	NA	200g	NA	
Plash Point		SW 846-1010	G	None	120mL	NA		NA
Mineral Oil Flash Point	Cleveland Open Cup	ASTM D-92	P,G	NA	NA	200g	NA	
Flash Point	Cleveland Open Cup	ASTM D-92	P,G	NA	NA	200g	NA	
Percent Bottom Sediment & Water	Cestrifage	ASTM D-96	P,G	Cool, 4°C	NA	Oif: 200mL	NA	

^{*}Oil matrix includes solvents, fuels and other petroleum products.

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Teat	Method	Reference	Container ¹	Water Sample	Required Sa	mple Quantity	Maximum	Maximum
				Preservation ^{2.3}	Water	S/S/S	Holding Time Water ⁴	Holding Time S/S/S
Sulfide	Colorimetric, Methylene Blue	EPA 376.2	P,G	Cool, 4°C NaOH + ZaC ₄ H ₄ O ₄ to pH >9	iL.	100g	7 days	7 days
Sulfido	Titrimetric, Iodine	EPA 376.1	P,G	Cool, 4°C NaOH + ZaC _i H ₂ O _i to pH >9	1L	100g	7 daya	7 days
Sulfide	Titrimetric, Iodine	SW 846-9030	P,G	Cool, 4°C NeOH + ZeC _i H ₂ O ₄ to pH >9	1L	100g	7 days	7 days
Sulfine	Titrimetric	EPA 377.1	P,G	None	500mL	NA	Analyze Immediately	NA
Suifier	Bomb	ASTM D-129	P,G	NA	NA	Oil:2g	NA	NA
Surfactuats	MBAS	EPA 425.1	P,G	Cool 4°C	500mL	NA	48 hours	NA
Temperature		EPA 170.1	P,G	None	100mL	-	Analyze Immediately	-
Thiographic		SM 412L	P,G	Cool, 4°C See Method	500mL	100g		
Sulfide, Total Available Extraction	Maguel	SW 7.3.4.1	P,G	Cool, 4°C NaOH + ZaC ₁ H ₄ O ₄ to pH >9	200mL	50g	7 days	7 days
Total Halogen		ASTM D-806	P,G	None	NA	10g	NA	
Total Organic Carbon	Combustion-IR	EPA 415.1	P,G	Cool, 4°C HCl or H ₂ SO ₄ to pH <2	250mL		28 days	28 days
Total Organic Carbon	Combustion-IR	SW 846-9060	P,G	Cool, 4°C HCl or H ₂ SO ₄ to pH <2	250mL		28 days	28 days
Total Organic Halides	Micro-Coulometric	SW 846-9020	GW/Teffon	Cool, 4°C	500mL	10g	7 days	7 days
Total Recoverable Oil & Grease	Gravimetric	EPA 160.3	G ·	Cool, 4°C HCl to pH <2	1L	50g	28 days	28 days

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Test	Method	Reference	Container	Water Sample	Required Sa	mple Quantity	Maximum	Maximum
				Preservation ^{2,3}	Water	S/S/S	Holding Time Water ⁴	Holding Time S/S/S
Al, Sb, Ba, Bc, Cd, Ci	a, Cr, Co, Ca, Pe, Ph, Mg, I	da, Mo, Ni, K, Ag, Na, Ti	, Sa, Ti, V, Za (23)					
Metals: Listed Below	FAA	SW 846-7000 Series	P,G	HNO, to pH <2	500mL	100g	180 days	180 days
Al, Sb, Ba, Ba, Cd, Ca	a, Cr, Co, Ca, Pe, Ph, Mg, 3	da, Mo, Ni, K, Ag, Na, Si	, Tl, Sa, V, Za (23)					
Metals: Listed Below	GFAA	BPA 200 Series	P,G	HNO, to pH <2	500mL	100g	180 days	180 days
Al, Sb, Aa, Ba, Be, Cd	l, Cr, Co, Ca, Pb, Ma, Ni, S	e, Ag. V, Tl. Sa. Ti. Za (19)					
Metala: Listed Below	GFAA	SW 846-7000 Series	P,G	HNO, to pH <2	500mt,	100g	180 days	180 days
AL AA Be, CA Cr, Co	a, Pb, Ma, Se, Ag, Tl, V (12)						
Metala: Listed Below	ICP (200.7 CLP-M)	CLP SOW 7/88	P,O	Cool, 4°C HNO, to pH <2	500m[.	100g	180 days	150 days
Al, Sb, Ba, Be, Cd, Ca	, Cr, Co, Ca, Fe, Mg, Ma, I	Ni, K, Ag, Na, Za, V						
Metals: Listed Below	ICP (200.7 CLP-M)	CLP SOW ILMOLO	P,G	Cool, 4°C HNO ₃ to pH <2	500mL	100g	180 days	180 days
Al, Sb, Ba, Be, Cd, Ca	, Cr, Co, Ca, Fe, Mg, Ma, I	NI, K, Ag, Na, Za, V						
Metals: Listed Below	GPAA (CLP·M)	CLP SOW 7/88	P,G	Cool, 4°C HNO, to pH <2	500ml.	100 g	180 daya	180 days
As (206.2 CLP-M), Se	(270.2 CLP-M), Pb (239.2	CLP-M), TI (279.2 CLP-I	4)					
Metals: Listed Below	GFAA (CLP-M)	CLP SOW ILMOL1	P,G	Cool, 4°C HNO ₃ to pH <2	500mL	100g	180 daya	180 days
As (206.2 CLP-M), Se	(270.2 CLP-M), Pb (239.2	CLP-M), TI (279.2 CLP-I	d)					
Метонгу	Magual CVAA (245.1 CLP-M)	CLP SOW 7/88	P,G	Cool, 4°C	SOOmL,	100g	26 days	NA
Mercusy	Magnal CVAA (245.1 CLP-M)	CLP SOW ILM01.0	P,G	Cool, 4°C	500mL	10g	26 days	NA
Mercury	Masual CVAA (245.5 CLP-M)	CLP SOW 7/88	P,G ı	Soil/Sediment Cool, 4°C	NA	10g	NA	26 days

l GCE	Method	Reference	Container ¹	Water Sample	Required Sau	nple Quantity	Maximum	Maximum
				Preservation ^{2.5}	Water S/S/S Holding Time Water NA	Holding Time S/S/S		
Percent Oil	Solubility	ASTM D-2042	P,G	NA	NA	Oil*: 200mL	NA	
Percent Oil	Contribuge	ASTM D-96	P,G	NA	NA	Oil*: 200mL	NA	
Percent Water	Xylene Trap	ASTM D-95						
Percent Water	Karl Fischer Titration	ASTM 1744	P,G Airtight	None	100ml.	Oil*: 100mL	NA	NA
Pounds per Cabic Poot		ASTM D-696						
Pour Point		ASTM D-97	P,G	NA	NA	Oil*: 100mL	NA	NA
Reid Vapor Pressure	Reid	ASTM D-323	Airtight	NA	NA	Oil*: 1L	NA	NA
Residue, Percent Ash		ASTM D-482	P,G	NA	NA.	Oil*: 100g	NA	NA
Viscosity	Saybolt Purol	ASTM D-244	P,G	NA	NA	Oil*: 200mL	NA	
Viscosity	Saybolt Universal	ASTM D-445	P,G	NA	NA	Oil*: 200mL	NA	
Specific Gravity		ASTM D-244	P,G	NA	100g	100g	NA	NA
Specific Gravity	(Solids)	SM 213B	P,G	NA	100g	100g	NA	NA
Heat of Combustion	BTU/LB	ASTM D-240	P,G	NA	50mL	20g	NA	NA
Weight Per Gallon		ASTM D-244	P,G	NA	1		NA	NA
Metals;								
Metals: Listed Below	ICP	EPA 200.7	P,G	HNO ₃ to pH <2	500mL	100g	180 days	180 days
Al, As, Sb, Ba, Be, Bi,	B, Cd, Ca, Cr, Co, Ca, Fe,	Pb, Li, Mg, Mn, Mo, Ni, I	K, Si, Sc, Ag, Na, S	r, TL Sa, Ti, V, Za, Y (3	1)			
Metals: Listed Below	ICP	SW 846-6010	P,G	HNO ₃ to pH <2	S00mL	100g	180 days	180 days
Al, As, Sb, Ba, Be, Bi,	B, Cd, Ca, Cr, Co, Cu, Fe,	Pb, Li, Mg, Mn, Mo, Ni, I	K, Si, Sc, Ag, Na, S	r, Tl. Sa, Ti, V, Za, Y (3	1)			
Metals: Listed Below	FAA	EPA 200 Series	P,G	HNO, to pH <2	500mL	100g	180 days	180 days

^{*}Oil matrix includes solvents, fuels and other petroleum products.

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Test	Method	Reference	Container ¹	Water Sample	Required Se	mple Quantity	Maximum	Maximum
				Preservation ^{2.5}	Water	S/S/S	Holding Time Water ⁴	Holding Time S/S/S
Nitrosamines ^{11, 27}	GC/NPD	EPA 607	o	Cool. 4°C, 0.008% Na ₁ S ₂ O ₁ ³		NA	7 days to extraction, 40 days to snalysis	NA
Organochlorine Pesticides/ PCBs ¹⁴	GC/BCD-	EPA 408	a	Cool, 4°C, pH 5-9 ¹⁸	21.	100g	7 days to extraction, 40 days to analysis	7 days to extraction, 40 days to analysis
Volatile Priority Pollutants	GC/MS	EPA 624	G - Zero Headspace	Cool, 4°C, HCl to pH2, 0.008% Na ₃ S ₂ O ₃ S ₃₂	2x40mL	40g	14 days	14 days
Semi-Volatile Priority Pollutants (BNA) ¹⁴	GC/MS	EPA 625	G	Cool, 4°C, 0.008% Na ₃ S ₂ O ₃₁ ³	21.	100g	7 days to extraction, 40 days to analysis ¹⁶	7 days to extraction, 40 days to analysis ¹⁶
Hydrocarbon Scan	GC/FID	5W 846-8000	G	Cool #C	2x40mL	100g	14 days	14 days
RCRA Listed Solvents	GC/FID	SW 846-8000	G	Cool, 4°C	2x40mL	40g	14 days	14 days
Halogeunted Volatile Organics	GC/BLCD	SW 846-8010	G	Cool, 4°C, 0.008% Na ₂ S ₂ O ₃	2x40mL	40g	14 days	14 days
Alcohols	GC/FID	SW 846-8015	o	Cool 4°C	2x40mL	40g	14 days	14 days
Petroleum Puels	OC/FID	SW 846-8015 Modified	G	Cool, 4°C	2x40mL	100g	14 days	14 days
Total Petroleum Hydrocarbons	FTIR	EPA 418.1	0	Cool, 4°C HCl to pH 2	21.	NA	28 days	NA
Total Petroleum Hydrocarbons	FTIR	SM 503B	G	Cool, 4°C	NA	100g	NA	25 days
Purgeable Aromatics	GC/PID	SW 846-8020	G	Cool, 4°C, 0.008% Na ₂ S ₂ O ₃ HCl to pH2 ^{S,12}	2x40mL	40g	14 days	14 days
Halogeanted and Aromatic Volatile Organica	GC/PID/ ELCD	SW 846-8021	O	Cool, 4°C HCl to pH2, 0.008% Na ₂ S ₂ O ₃ ^{5,13}	2x40mL	40g		
Acroleia, Acrytonitrile, Acetonitrile	OC/FID	SW 846-8030	G 1	Cool, 4°C, 0.008% Na ₃ S ₂ O ₃ ² Adjust pH to 4-5 ¹³	2x40m L	40g		

Test	Method	Reference	Container ¹	Water Sample	Required Sa	mple Quantity	Maximum	Maximum
			je Pala, sept. Pala sept. Nasara a a a	Preservation ^{2,3}	Water	\$/\$/\$	Holding Time Water	Holding Time S/S/S
Mercury	Manual CVAA (245.5 CLP-M)	CLP SOW ILM01.0	P,G	Soil/Sediment Cool, 4°C	NA	10g	NA	26 days
Mercury	CVAA - Massal	EPA 245.1	P,G	HNO ₃ to pH <2	500mL	NA	13 days ⁷	NA
Метомгу	CVAA - Maqual	SW 846-7470	P,G	HNO, to pH <2	500ml.	NA	13 days ⁷	NA
Mercury	CVAA - Manual, \$/\$/\$	SW 846-7471	P,G	Nose	NA	10g	28 days	28 days
Organic Tests ^{(4) 11}								
2,3,7,8 TCDD ¹⁴	GC/MS	EPA 613	o	Cool, 4°C	21.	NA	7 days to extract, 40 days to analysis	NA
2,3,7,8 TCDD ¹⁴	OC/MS/MS	CLP SOW 8/88	o	Cool, 4°C	2L.	100g	NA	NA
Organochlorine Pesticides/PCBs	GC/ECD	CLF SOW 2/88	a	Cool, 4°C	2L.	100g	Extraction 5 days	Extraction 10 days
Organochlorine Pesticides/PCBs	OC/ECD	CLP SOW OLM01.5	o	Cool, 4°C	2L.	100g	from VTSR ^{**} , 40 days to analysis	from VTSR**, 40 days to analysis
Semi-Volatilo Organics	GC/MS	CLP SOW 2/88	O	Cool, 4°C	21.	100g		
Semi-Volatile Organica	GC/M\$	CLP SOW OLM01.5	G	Cool, 4°C	21.	100g		
Volatile Organics	GC/MS	CLP SOW 2/88	O · Zero Headspace	HCl to pH 2	2x40mL	40g	10 days from VTSR	10 days <u>from</u> VTSR
Volatile Organics	GC/MS	CLP SOW OLM01.2	G - Zero Headepace	HCl to pH2	2x40m].	40g		
Purgeable Halocarbons	GC/ELCD	EPA 601	O - Zero Headspace	Cool, 4°C, 0.006% Na ₂ S ₂ O ₃ ³	2x40mL	40g	14 days	14 days
Halogenated and Aromatic Volatile Organics	GC/PID/ELCD	EPA 601/602 Modified	G - Zero Headspacs	Cool, 4°C, 0.006% Na ₂ S ₂ O ₃ ³	2x40mL	40g	14 days	14 days
Purgeable Aromatics	GC/PID	EPA 602	G - Zero Headspace	Cool, 4°C, 0.006% Na ₂ S ₂ O ₂ ³ HCl to pH2 ¹³	2x40mL	40 g	14 days	14 days

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[&]quot;VTSR-Validated Time of Sample Receipt, EPA contracts only.

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Test	Method	Reference	Container	Water Sample	Required Sar	nple Quantity	Maximum	Maximum
				Preservation ^{2.3}	Water S/S/S		Holding Time Water ⁴	Holding Time S/S/S
Nitrogen & Phosphorus Containing Pesticides	OCMPD	EPA 507	O	Cool, 4°C, HgCl ₂ to cosc. of	2L.	NA	28 days after extraction	NA
Volatile Organic Compounds in Drinking Water	GCA4S	EPA 524.2	G	1089/L Cool, 4°C, 0.008% Na ₂ S ₂ O ₃ ⁵ , HCl to pH2 ⁵	3x40mL	NA	14 days	NA
Semi-Volatile Organic Compounds in Drinking Water	GC/MS	EPA 525	a	Cool, 4°C, 0.008% Na ₂ S ₂ O ₃ ⁵ , HCl to pH2 ⁵	2xil.	NA	7 days to extract, 40 days to analysis	NA
SDWA Carbamates	нис	EPA 531.1	G	Mosochloro-scetic Acid Buffer to pH3, Cool, 4°C	120mL	NA	28 days	NA
SDWA Herbicides	GC/BCD	BPA 515.1	G	Cool, 4°C, HgCl ₂ to 10 mg/L conc.***	ıL	NA	7 days to extract, 40 days to analysis	NA
Estraction Procedure Toxicity - Metals Only	Leachate Generation	SW 846-1310	P,G	Nonc	100g @ 100% So >5,000g @ ≥.5%		NA	NA
Indians-Neutral Leaching Method	Leachate Generation	329LAC	G	None	100g @ 100% So >10kg @ ≥.5%		NA	NA
Extraction Procedure Toxicity- Metals, Pesticides, Herbicides	Leachate Generation	SW 846-1310	G	None	100g @ 100% So >10kg @ ≥.5%		NA	NA
Toxicity Characteristic Leaching Procedure (TCLP)-Metals Only	Leachate Generation	SW 846-1311	P,G	None	105g @ 100% Solida >5kg @ ≥.5% Solids		180 days to TCLP extraction 180 days to Analysia, except Hg - 28 days	
Toxicity Characteristic Leaching Procedure (TCLP)-All Regulated	Leachate Generation	SW 846-1311	G	None	500g @ 100% Solids 20kg @ ≥.5% Solids		See individual analyte	Bronbiata
Toxicity Characteristic Leaching Procedure (TCLP)-Regulated Non-Volatile Organics	Leachate Generation	SW 846-1311	G	None	305g @ 100% Solide 15kg @ ≥.5% Solide		14 days to TCLP extra TCLP extraction to pe 40 days from preparat analysis	eparative extraction.

^{***}Alternate preservative preferred

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	Method	Reference	Container ¹	Water Sample	Required Sa	mple Quantity	Maximum	mumicaM
		All Bylt His Miles Miles		Preservation ^{2.5}	Water S/S/S		Holding Time Water ⁴	Holding Time S/S/S
Organochlorine Pesticides & PCBs ¹⁴	GC/BCD	SW 846-8080	o	Cool, 4°C pH 5-918	2L	100g	7 days to extract, 40 days to analysis	14 days to extract 40 days to analysi
PCBi ^M	GC/BCD	SW 846-8080	G	Cool, 4°C	2L	100g		
Polymedest Aromatic Hydrocarbous ¹⁴	GC/PID	SW 846-8100	a	Cool, 4°C, 0.008% Na ₃ S ₂ O ₃ ³	2L	100g		
Chlorinated Hydrocarbons	GC/ELCD	SW 846-8120	G	Cool, 4°C	2x40mL	40g	14 days	14 days
Organophosphorus Pesticides ¹⁴	OC/NPD	SW 846-8141	o	Cool, 4°C, pH 5-918	2L	100g	7 days to extract, 40	14 days to extract
Chlorinated Herbicides	GC/BCD	3W 846-8150	o	Cool, 4°C	1L	100g	days to aunlysis	40 days to analysis
Volntile Organics	GC/MS	SW 846-8240	G	Cool, 4°C, HCl to pH 2 ¹²	2x40mL	40g	14 days	14 days
Volatile Organics	GC/MS	SW 846-8260	G	Cool, 4°C, HCl to pH 2 ¹²	2x40mL	40g		
Appendix IX Volatile Organics	GC/MS	SW 846-8240	G	Cool, 4°C, HCl to pH 2 ¹²	2x40mL	40g		
Semi-Volatile Organics ¹⁴	GC/MS	SW 846-8270	a	Cool, 4°C, 0.008% Na ₂ S ₂ O ₃ ³	2L	100g	7 days to extract, 40 days to manhysis	14 days to extract 40 days to sasiys
Appendix [X Somi-Volntile Organics ¹⁴	GC/MS	SW 846-8270	G	Cool, 4°C, 0.006% Na ₂ \$ ₂ O ₃ ³	2L	100g		
Polychlorinated Dibenzo-P- Dioxina and Dibenzolurans ^H	GC/MS	SW 846-8280	G	Cool, 4°C, 0.008% Na ₂ S ₂ O ₃ ³	2L	100g		
Polymeclear Aromatic Hydrocarbons	HPLC	SW 846-8310	G	Cool, 4°C, 0.008% Na ₂ S ₂ O ₃ ³	21.	100g		
Volatile Organic Compounds in Drinking Water	GC/PID/ELCD	EPA 502.2	G	Cool, 4°C, 0.008% Na ₂ S ₂ O ₃ , HCl to pH 2 ⁵	3x40mL	NA	14 days	NA
SDWA EDB & DBCP	Micro-Extraction, OC/BCD	EPA 504	o ,	Cool, 4°C	3x40m1.	NA	7 days to extract, 40 days to analysis	NA

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- 5. Should only be used in the presence of residual chlorine. May use ascorbic acid or sodium bisulfate as allowed per regulation.
- 6. Maximum holding time is 24 hours when sulfide is present. Optionally all samples may be tested with lead acetate paper before pH adjustments in order to determine if sulfide is present. If sulfide is present, it can be removed by the addition of cadmium nitrate powder until a negative spot test is obtained. The sample is filtered and then NaOH is added to pH 12.
- 7. 13 days in plastic, 28 days in glass.
- 8. All organics containers have teflon lined caps. Required Sample quantity is listed to include MS or back-up sample.
- 9. Samples should be filtered immediately on-site before adding preservative for dissolved metals.
- 10. Guidance applies to samples to be analyzed by GC, LC, or GC/MS for specific compounds.
- 11. All drinking water samples must be preserved, regardless of holding time.
- 12. Sample receiving no pH adjustment must be analyzed within 7 days of sampling.
- 13. The pH adjustment is not required if acrolein will not be measured. Samples for acrolein receiving no pH adjustment must be analyzed within 3 days of sampling.
- 14. When the extractable analytes of concern fall within a single chemical category, the specified preservative and maximum holding times should be observed for optimum safeguard of sample integrity. When the analytes of concern fall within two or more chemical categories, the sample may be preserved by cooling to 4°C, reducing residual chlorine with 0.008% sodium thiosulfate, storing in the dark, and adjusting the pH to 6-9; samples preserved in this manner may be held for seven days before extraction and for forty days after extraction. Exceptions to this optional preservation and holding time procedure are noted in footnote 5 (re: the requirement for thiosulfate reduction of residual chlorine), and footnotes 15 and 16 (re: the analysis of benzidine).

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Ter	Method	Reference	Container!	Water Sample Preservation ^{2,5}	Required Sam	ple Quantity S/S/S	Maximum Holding Time Water ⁴	Maximum Holding Time S/S/S
Toxicity Characteristic Leaching Procedure (TCLP)-Regulated Volatiles	Leachate Generation	SW 846-1311	G	None	125g @ ≤ 100% 5 1200g @ <5% So		14 days to TCLP extr. TCLP extraction to a	•

- 1. Polyethylene (P) or Glass (G). All organics containers will have a teflon-lined cap.
- 2. Sample preservation should be performed immediately upon sample collection. for composite chemical samples each aliquot should be preserved at the time of collection. When use of an automated sampler makes it impossible to preserve each aliquot, then chemical samples may be preserved by maintaining at 4°C until compositing and sample splitting is completed.
- 3. When any sample is to be shipped by common carrier or sent through the United States Mails, it must comply with the Department of Transportation Hazardous Materials Regulations (49 CFR Part 172). The person offering such material for transportation is responsible for ensuring such compliance. For the preservation requirements of Table II, the Office of Hazardous Materials, Materials Transportation Bureau, Department of Transportation has determined that the Hazardous Materials Regulations do not apply to the following materials: Hydrochloric acid (HCl) in water solutions at concentrations of 0.04% by weight or less (pH about 1.96 or greater); Nitric acid (HNO₃) in water solutions at concentrations of 0.15% by weight or less (pH about 1.62 or greater); Sulfuric acid (H₂SO₄) in water solutions at concentrations of 0.080% by weight or less (pH about 1.230 or less).
- 4. Samples should be analyzed as soon as possible after collection. The times listed are the maximum times that samples may be held before analysis and still be considered valid. Samples may be held for longer periods only if the permittee, or monitoring laboratory, has data on file to show that the specific types of samples under study are stable for the longer time, and has received a variance from the Regional Administrator under § 136.3(e). Some samples may not be stable for the maximum time period given in the table. A permittee, or monitoring laboratory, is obligated to hold the sample for a shorter time if knowledge exists to show that this is necessary to maintain sample stability. See § 136.3(e) for details.

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- 15. If 1,2-Diphenylhydrazine is likely to be present, adjust the pH of the sample to 4.0 ± 0.2 to prevent rearrangement to benzidine.
- 16. Extracts for benzidines may be stored up to 7 days before analysis if storage is conducted under an inert (oxidant-free) atmosphere.
- 17. For the analysis of diphenylnitrosamine, add 0.008% Na₂S₂O₃ and adjust pH to 7-10 with NaOH within 24 hours of sampling.
- 18. The pH adjustment may be performed upon receipt at the laboratory and may be omitted if the samples are extracted within 72 hours of collection. For the analysis of aldrin, add 0.008% Na₂S₂O₃.

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7.0 Sample Custody

7.1 <u>Introduction</u>

The maintenance of sample chain of custody is part of the laboratory QA program. Where samples and analytical results may be needed for legal purposes, it is imperative that the laboratory, client and everyone else involved in such proceedings be able to unequivocally demonstrate that analytical results were obtained from analysis of specified samples.

A sample is in someone's "custody" if:

- 1. It is in their actual physical possession, or
- 2. It is in their view, after being in their physical possession, or
- 3. It is in their physical possession and locked up so that no one could tamper with it, or
- 4. It is kept in a secured area restricted to authorized personnel only.

7.2 Facilities and Sample Security

EMS Heritage laboratory facilities are protected with an access control system. In this way, access to all samples at all times remains controlled and documented.

Visitors to the facilities are prohibited from entering the cooler and laboratory areas unescorted. Special viewing areas have been provided to facilitate laboratory "tours".

Rules regarding custody have been established and must be followed as outlined below:

- 1. Samples must never be left unattended in the delivery/receiving area. If the receiving area is to be left unattended, the sample receiving door must be closed and locked.
- 2. All samples must be returned to their storage area/cooler at the end of the final shift.
- 3. All sample extracts and digestates must be placed in their respective laboratories.

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7.3 Sample Custody Records

Sample custody will be supported by records which trace a sample from its point of origin to include container storage and shipment through disposal after analysis. These records will include, but are not limited to:

- 1. Field notebooks;
- 2. Field sample I.D. tags, labels;
- 3. Laboratory (transmittal forms), Task Orders, Request Forms;
- 4. Chain of Custody forms;
- 5. Sample extraction/preparation logs or worksheets;
- 6. Analytical (instrument) logs or worksheets;
- 7. Calibration and quality control data associated with a sample set;
- 8. Instrument maintenance logs;
- 9. Sample disposition logs; and,
- 10. Final reports.

Item numbers 1, 3 and 4 above are generally the responsibility of the client unless EMS Heritage performs the sampling.

7.4 <u>Legal Chain of Custody</u>

The laboratory will only be responsible for maintaining the records of the activities it performs. Sampling documentation for samples not collected or sub-contracted for collection by EMS Heritage will be the responsibility of the sample collector.

EMS Heritage will follow special procedures when strict Legal Chain of Custody is required. Legal Chain of Custody is a special type of sample custody in which <u>all</u> events associated with a specific sample must be documented in writing. In addition to the records described above, chain of custody records must include the following:

- 1. Sample transmittal forms or tags that have adequate space for the dated, original signatures of all individuals who handle the sample (or empty containers) from time of collection (or container receipt) through laboratory delivery.
- 2. Custody seals.
- 3. Laboratory sample storage logs that identify date, time, and individuals who remove samples from storage.
- 4. Secure, limited access storage areas.
- 5. Errors in all documentation are deleted by drawing one line through the error, writing the correction, placing the initials of the person

making the change, the date of the change and the reason for the change. All documentation is recorded in indelible black ink (no other colors allowed).

7.5 Sample Labels, Seals and Chain of Custody

All sample containers will have a sample container label attached. An example of this label is shown in Figure 7.1. Those samples requiring Legal Chain of Custody will have a custody seal placed on the closure mechanism on the sample such that any attempt to remove the seal will be obvious. An example of this seal is shown in Figure 7.2. All samples will be accompanied by a Chain of Custody as shown in Figure 7.3 when provided by EMS Heritage. Clients may substitute Chain of Custody forms of their own when desired. Labels and seals may utilize alternate forms containing identical information.

7.6 Field Sampling/Custody

When Legal Chain of Custody is required, the custody documentation will begin with the preparation and/or shipment of sample containers to the sampler(s). Sampling performed by EMS Heritage will utilize the following procedure for identifying and labeling samples in the field:

Sample Prefixes

RW = Residential Well
MW = Monitoring Well
WA = Water, Non-Specific

SW = Surface Water

SD = Sediment (lake, ditch, river)

SL = Sludge LE = Leachate SS = Soil Sample

Sample Suffixes

DP = Duplicate
FB = Field Blank
TB = Trip Blank

EB = Equipment Blank

BG = Background

When samples are already sufficiently identified by type and exact location without the need of assigning an arbitrary identification, that pre-established

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Figure 7.1

CONTAINER LABEL

DATE SAMPLED COMPANY NAME (TO	TIME SAMPLED D BE SHOWN ON REPORT)
SAMPLE DESCRIPTION	
LAB USE ONLY	PRESERVATIVE
EMS SAMPLE#	

CUSTODY SEAL

00203



CUSTODY SEAL

EMSIN 7901 WEST MORRIS STREET INDIANAPOLIS, INDIANA 46231 Section 7.0 Revision 0 Nov. 1, 1991 Page 1 of 1

Figure 7.3

CHAIN OF CUSTODY



TO ENSURE PROPER HANDLING OF SAMPLES PLEASE COMPLETE THIS ENTIRE FORM

EMS HERITAGE LABORATORIES, INC.

7901 West Morris Street Indianapolis, Indiana 46231 (317) 243-0811 Fax (317) 243-0360

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Figure 7.4

EMS HERITAGE LABORATORIES, INC.

SAMPLING PROJECT SUMMARY

Submitter:					
Site Contact:Site Identification:					
Project Number:			7 Numbers:		
Samplers:			_Z Numbers.		
Sampling Date:					
Weather Conditions:	 ·				
Sample Iced? YES					
Preservative Used? YES		If Yes, Prese	rvative Used	Lot #	
Sample Types (circle):	Lagod Leach Sludge	ate	Ash Soil Solvent	Indus Waste Waste Pile Solid	Ditch Creek Liquid
	Oil Res. V	Well	Sand other	Drummed Waste	Truck
Sample Plan Review: Information on equ					
Decontamination Procedu	res Use	ed:			
Equipment is (dedicated/d	leconta	minated)	Source of de	econ. water	
Source of blank reagent w		,			
Miscellaneous: Photos taken?	YES	NO			
Environmental Pro	gram:	RCRA NPDES DISPOSAL OTHER	CERCLA IWP SDWA	SOLID WASTE SURFACE WATE CWA	UST R

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identification will be used. Samples requiring arbitrary field number assignments will be given consecutive numbers in the following manner:

Examples:

Soil Sample #1	=	SS0001
Background Soil Sample #1	=	SS0001BG
Surface Water Sample #3	=	SW0003
Trip Blank Sample #1	=	0001TB
Surface Water Sample #3 Duplicate	=	SW0003DP

EMS Heritage field samplers will use the appropriate field data sheets as given in Figures 7.4 through 7.8 for sampling. Field records will be stored, filed and maintained by submitter (client) and project number at EMS Heritage.

7.7 Sample Dispatch

Samples to be shipped by common carrier will be properly preserved and placed in a secure shipping container. All transmittal documents including the chain of custody form will be placed inside the shipping container in water-tight plastic bags, taped to the inside lid. The container will be sealed with custody tape such that any tampering between shipment and receipt at the laboratory will be obvious to receiving personnel.

When samples are delivered by common carrier it is only necessary to sign the documents that are required by the carrier service at the time of delivery, i.e., airbills, shipping invoices, etc. After the shipping papers have been signed, the packages are entered into the Parcels Received log book. This log book contains:

- 1. Date package received
- 2. Carrier Service delivering
- 3. Client/Company shipped from
- 4. Destination
- 5. Type of package (i.e. samples, stock, instruments)

The next steps in laboratory custody will be the same regardless of whether samples arrive via common carrier, or are hand delivered.

EMS HERITAGE LABORATORIES, INC.

SAMPLE DETAIL

Sample ID					
Field Test		Sample Type	es (circle all	applicable)	
Performed	Result	Mon. Well		Ash	Indus. Waste
		Res. Well		Soil	Waste Pile
		Creek	Oil	Sludge	Solid
		Ditch	Solvent	Sand	4
		Truck		other	
		Blank (Equi Background		Duplicate	(of)
		Dackground			
Sample Date:	Time	e:: AM/I	PM		
Containers	#	Prese	ervatives Add	ed	
1 L plastic		H ₂ SC			
1 L glass		HNC	•		
500 ml glass		NaO	Ĥ	**************************************	
40 ml vial			cetate		
		HCl			
		•	ole Iced		
		•	reservatives vaqueous sam		
Additional Sample	e Location Inf		•	, , , , , , , , , , , , , , , , , , , ,	
Additional Sample density, suspended					odor, size, clarity,
Deviations From	Sampling Plan	l:			
Sampling Equipm	ent Used:				
			Sampler Si	gnature	
			-	Date	

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Figure 7.6

	EMS HERITAGE L	ABORATORIES, II	NC. Page of
	Monitoring Well Sam	oling - Project Sump	nary
Site Contact:		County:	
Project Number:		Z Numbers: _	
Sampling Date: Weather Condition			
Preservative Used		Environmental RCRA C UST S OTHER	ERCLA SDWA OLID WASTE
	Pumping	Information	
Pump Type: make: Bailer: (PVC	cy / consultant / EMS) per	size tubing	terial
diam	eterle	ngth c	apacity
	Filtration	Information	
Filtered by: (facil Filtration Method:	Filtered:	ersonnel sure); Devise Type_	
	Field Instr	uments Used	
pH Meter S.C. Meter Thermometer			

Figure 7.7

SAMPLE DETAIL

Sample I.D.	_							
Field Tests Test Result Temperature°C pHumhos	Sample type Mon. well Res. well Duplicate of Split	Preservatives Added Hcl HNO ₃ H ₂ SO ₄ NAOH						
	Blank (field/trip/equip.)	Zn-Acetate						
Containers Used # 1 liter plastic 1 liter glass 40 ml vials 250 ml glass 500 ml glass	1 liter plastic Well I.D. 1 liter glass Reference point on (steel / plastic) casing top 40 ml vials Casing Stick-up 250 ml glass Depth to Water							
Were METALS filtered prior to Color of water subsequent Deviations from Sampling Plan_	t to filtration							
Other Observations Water Appearance (clear slightly / very turbid) (color: gray / brown) Well purged (less / greater) than (1 2 4 6 12 24 48) hours prior to sampling Purged approx. (to dryness) / (1 2 3 4 5 8 10 G.T. 10 well volumes)								
Reaction upon addition of preser	rvatives? (YES / NO) (exp	lain)						

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Figure 7.8

Equipment Decontamination Procedures

Specific Procedures Utilized to Decontaminate Equipment for this Event:

7.8 Laboratory Custody/Receiving

The following steps are taken when samples are received.

- 1. Examination of the shipping container. Note the presence or absence of custody seals, locks, evidence tape, etc.
- 2. Generating any log books applicable (i.e. recording all information into a central case file, if client requests such).
- 3. Removal of samples from shipping containers and inspection of any traffic reports, chain of custody documents, etc. If no chain of custody is received the lab creates a chain of custody from that point forward.
- 4. If a chain of custody is received, it is inspected to determine the accuracy of the document (i.e. relinquishing signatures, date and time sampled, sample locations, etc.). If any discrepancies exist, they are noted on the chain of custody and in the case or client file if applicable.
- 5. Inspect sample containers for any damage. If damage has occurred it is to be noted directly on the chain of custody or a sample integrity form attached to the chain of custody, and in the case file if applicable. Notify the project manager immediately to determine corrective action required. All corrective actions are recorded on the chain of custody and in the case file.
- 6. Examine sample bottles for their agreement with the chain of custody.
- 7. Sample containers are inspected for proper containers and preservation. Preservation is checked according to the following protocol:
 - a. If EMS Heritage containers were supplied and used, the pH is checked on one out of 10 samples by using a disposable pipet and litmus paper.
 - b. All samples that are not received in EMS-supplied containers are checked for pH, with the exception of volatile organic sample vials. VOC vials are not opened until the time of analysis, but are checked for the presence of air bubbles. Samples are also checked for proper preservation at the time of analysis.

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- c. If an analysis requires a specific type of preserved container and the container was not received, it is noted on the chain of custody or sample integrity form, and in the case file if applicable.
- d. All preservations, if needed, should be performed under a fume hood and are performed one at a time to avoid cross contamination.
- 8. The analyses requested are compared to the quote for the project, the containers provided and the preservation. If the analyses to be performed are not fully specified or if the condition of the test substance does not conform to the description provided or is different from what was expected, the Project Manager is immediately notified. The Project Manager will contact the client for further instructions before proceeding.
- 9. The laboratory will establish whether the test substance has received all necessary preparation, or whether the client requires preparation to be undertaken or arranged by EMS Heritage.

7.9 Sample Log-In

Samples are logged into the LIMS system with the following information:

- 1. A computer assigned EMS Heritage sample number. If another sample is copied or a specific quote "Z sample" is used to log in a sample, this template sample number is recorded in addition to the unique sample number;
- 2. <u>Project number</u>. This project number defines a discreet project for a given set of samples;
- 3. Quote number. An EMS Heritage quotation;
- 4. <u>Purchase Order number</u>. A client provided billing reference number;
- 5. <u>Submitter number</u>. This is computer shorthand for the EMS Heritage client:

- 6. Regulation. This field describes the environmental program area under which the sample is submitted;*
- 7. Matrix of the sample;*
- 8. Received identification. This entry identifies the staff receiving the sample;
- 9. <u>Lab Days</u>. This entry states the number of days targeted for analysis;
- 10. <u>Due Date</u>. This entry gives the report due date;
- 11. Status. This describes the status of the sample with regard to analysis and reporting;
- 12. Entered. This entry identifies the staff entering the sample into LIMS;*
- 13. <u>Project Manager</u>. This is the assigned staff person for this sample/project;
- 14. Modified. This entry shows the last person to modify a sample record;
- 15. Complete. This field shows the date that all tests and tasks show completed in the computer system;
- 16. Date sampled. Taken from the chain of custody;
- 17. <u>Time sampled</u>. Taken from the chain of custody;
- 18. Sample received within holding time? yes or no;
- 19. <u>Sample properly preserved?</u> yes or no;
- 20. Sample chilled to 4°C? yes or no;
- 21. <u>Sample received in proper container?</u> yes or no;
- 22. <u>Chain of custody received?</u> yes or no;

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- 23. Chain of custody number, if applicable;
- 24. Method of disposal;
- 25. Client sample description;
- 26. Client reporting address;
- 27. <u>Sample comments to the laboratory</u>. This field displays any information pertinent to the sample;
- 28. Sample comments to appear on final report;
- 29. <u>Tasks assigned</u>. Tasks are any activity or service performed not inherent in the act of sample analysis, i.e. a special reporting package, etc.;
- 30. <u>Protocol.</u> A pre-defined grouping of tests routinely requested together;
- 31. Tests assigned. All tests are assigned in the order they are to appear on the final report. Those tests requiring a preparation step are linked together indicating such. The location where analysis will be performed, the priority of analysis and the final discounted price for the analysis also appear here. All test documentation includes the names of the analyst, reviewer, data entry person, all associated dates, run numbers, QC and results;
- 32. All information relating to each test assigned is contained in the next two blocks of the LIMS system, including the actual results;
- 33. Any comments relating to the individual test performed are contained in the next two blocks of the LIMS system.

The key point to the EMS Heritage Lab ID number is linking that number to the client's sample description. After the computer has assigned the Lab ID number, that number is to be written on the cap and the label of all containers.

7.10 Sample Storage

Samples are to be placed in their appropriate storage area which is specific to the analysis to be performed. Analysts will be notified immediately of

samples received with extremely short or sensitive holding times, so that analysis may be performed immediately. All sample storage areas are accessible only to EMS Heritage staff. Sample digestates and extracts will be stored separately from samples or standards and are accessible only to EMS Heritage staff. These digestates and extracts will also be signed in and out when Legal Chain of Custody is required (see Section 7.4). Standards are stored separately from all samples and VOC samples are segregated from other samples.

7.11 <u>Log-In Accuracy</u>

As a cross check for data entry, all paperwork is reviewed against the information entered into the LIMS system either by an assisting data entry operator, their supervisor or a Project Manager. This paperwork is retained on file. Project paperwork is forwarded to the project manager for the samples received as a further check on accuracy and as a means for project management. A log audit report from the LIMS system is checked daily by the appropriate Project Manager.

7.12 Project Schedules

Incoming samples are scheduled by sales and/or project management staff. The sample quotes, requests to send containers, "Z" template samples, etc. are forwarded to the sample shipping/receiving area, where they are matched with incoming samples/projects. Once samples are logged into the LIMS system the work assignments will appear on an analyst's worklist according to a pre-assigned list of tests per work group. These work lists are generated and distributed daily. Analysts receive worklists as needed (bi-weekly at a minimum) which lists the samples due, sorted by their maximum allowable holding time from the date sampled. This helps to insure that all samples are analyzed within holding time. Analysis within holding time takes precedence over RUSH analysis requests. As work assignments are completed, they are sent to data entry and given the status of complete in LIMS upon completion of data entry.

7.13 Inter and Intra-Lab Shipping

When it is necessary or desirable to sub-contract to another laboratory, it is crucial that the sub-contract laboratory be given complete instructions and information regarding the sample and analysis. The information required includes the list of analytes and the methods required for analysis. Sample shipments between EMS Heritage laboratory divisions will present no problem in this area, since all share a common data base which describes the

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information for all samples. Chain of custody procedures will be used for all sample shipments. All shipments of samples are recorded in a logbook containing:

- 1. shipping date;
- 2. EMS sample ID numbers;
- 3. field sample ID numbers
- 4. collection date and time
- 5. requested analyses
- 6. date of sample preparation (if any)
- 7. location from which the samples were sent;
- 5. the carrier ID and shipping paper ID.

Subcontracting is permitted only when certifications, client Data Quality Objectives and/or contractual requirements are not prohibitive (e.g. CLP). Any sub-contract laboratory must be approved by the QA Officer or the QA Unit.

7.14 <u>Documentation of Sample Disposition</u>

All samples and laboratory waste are disposed of in adherence with all environmental regulations. Sample disposition is documented in the LIMS system by the sample custodian or his/her designee. Sample disposition is not routinely documented as to the precise fate but only for the fact that EMS Heritage no longer has the sample. Clients requiring more elaborate documentation may request this service on a case by case basis. Refer to Section 8.0, Analytical Procedures for further details of Sample and Waste Disposal.

7.15 Electronic Data Records (LIMS)

EMS Heritage Laboratories utilizes a computerized LIMS to track samples, temporarily store data for the purpose of integrated and/or electronic reporting, and to perform calculations as part of data reduction.

This LIMS system (hardware and software) is maintained and operated by a professional staff of computer programmers, analysts and operators under the direction of a data processing manager. All LIMS policies, procedures and operations are approved by laboratory management.

7.15.1 Security System

LIMS security is maintained on three levels; system resource access, information access and information manipulation access. The security

system is maintained by the data base administrator (DBA). All security policies and procedures are approved by Management. Access to information is given on a need to know basis only, and is granted according to job function. Inactive or terminated users are immediately removed from the system. All system users receive training appropriate to their job function prior to gaining access to the information.

Access to computer system resources is restricted through the assignment of a user specific login id and a confidential password to each user. No system resources are available to a user without the appropriate login id and password. This information is kept confidential and periodically modified.

Users are classified according to job function. This limits the users' access to information not specific or required for the performance of their duties. Once a user gains access to system resources as described above, the user is presented with a restricted view of all of the information available according to his/her user classification.

Data manipulation is also controlled by the user classification. For specific job functions, users may have the ability to view only, create, modify, and/or delete specific information. Users may have multiple capabilities custom tailored to their job function. As with system access, information access and manipulation is maintained by the DBA.

Once all tests on a sample are completed, the sample status becomes "released". This "released" status will not allow any data manipulation except by laboratory QA staff.

7.15.2 Analytical Records Management

All raw analytical data is stored as the original hard copy. Hard copy information includes:

- 1. Chain of Custody and/or Sample Submission Forms
- 2. Analytical Bench Sheets and Log Books
- 3. Instrument Printouts and Chromatograms

All hard copy information is uniquely identified for control purposes. Sample preparation procedures and final results are documented on forms containing all pertinent information and comments, including

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analyst and reviewer initials. Those forms are specific to the test being performed. Tests which are directly down-loaded into the LIMS system from the instrument are reviewed and approved before release to LIMS; the hardcopy instrument readout is filed for reference.

Analytical records maintained in the LIMS are only temporarily stored for the purpose of integrated or electronic reporting. Information may be entered into LIMS in two ways, direct computer to computer transfer and manual data entry. Both data entry procedures are quality assured by checking the computerized information against the original hard copy or the electronic output from the other computer.

7.15.3 Electronic Data Transfer

Electronic Data Transfer (EDT) may be utilized to facilitate the integration of the laboratory information with a client's own data management system. All hard copy information (i.e. Certificate of Analysis, QAQC report) is distributed to the client to provide source documentation for validation of information electronically transferred. A data transfer log is maintained of all electronic transfers indicating the client, date and time of transfer and the file id and contents description. Electronic (floppy disk) copies of all information transferred are also maintained on file.

7.15.4 Software Documentation

LIMS software is documented as necessary for the proper development and maintenance of the programs. Software modifications, problems and/or failures are documented using a System Support Request (SSR) form submitted to the data processing department. All SSR's are managed by the programming staff on a priority basis. Original program documentation and modifications are documented within the program logic or data structure itself. Summaries of modifications are communicated to the users using Software Maintenance Bulletins (SMB).

7.15.5 Data Entry Validation

Data entry is validated by manually comparing hard copy documentation (Chain of Custody, Lab Bench Sheets, etc.) to LIMS reports of the information entered. At least 5% of all data entry is validated routinely in this manner. Errors found during review are corrected and re-reported if necessary. A log book and print-out of

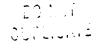
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data entry validation is kept indicating the run, date and auditor of the information and a notation of any records found in error.

7.15.6 Software/Hardware Validation

All software is functionally validated before use by the programming staff. Once software is put into production, it has essentially passed all preliminary validation criteria. Validation criteria include manual validation of calculations and manual validation of translation from data storage to report. All software is re-validated following modification.

Hardware validation is accomplished using the hardware vendors own internal checking mechanisms. No additional hardware validations are performed.



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8.0 Analytical Procedures

8.1 Introduction

EMS Heritage uses EPA approved analytical methods appropriate for the regulatory program, sample type and DQO's. The analytical methods used and their analyte lists are given in Table 5.1, QA Targets for Precision, Accuracy and Method Detection Limits, and Table 6.3, Sample Requirements, Tables 8.1 and 8.2 list the approved methods and references for the various types of matrices commonly encountered. If a client requests a method not listed in these tables, a full method validation package must be prepared, including Method Detection Limit (MDL) studies, before any analyses are reported for that method to be in compliance with this QAP. Instrument Detection Limit studies may replace the MDL studies in some situations if matrix spike/matrix spike duplicates are analyzed and approval is obtained from the Corporate QA Officer.

If applicable, alternative, modified or proposed new methods covered under this QAP must be summarized and a detailed description with method validation package must be submitted as an appendix to this QA Plan.

Many of the USEPA Contract Lab Program (CLP) methods are very similar if not identical to RCRA methods with the exception of the reporting forms and the QA protocols required. Commercial clients requesting a particular CLP SOW may opt upon consultation with EMS Heritage to accept different analytical reporting and QA protocols which do not compromise the data produced, but may not include all of the CLP forms, documentation and QA protocols. Samples analyzed under contract will be fully compliant with all contract requirements.

8.2 Standard Operating Procedures

In order to obtain reliable results, adherence to prescribed analytical methodology is required at all EMS Heritage laboratory divisions. In any operation that is performed on a repetitive basis, reproducibility is best accomplished through the use of Standard Operating Procedures (SOPs). An SOP is a written document which provides directions for the step-by-step execution of an operation, analysis or action which is commonly accepted as the method for performing certain routing or repetitive tasks. It is necessary that every aspect of an analysis or study every step in the sample trail) be linked by one or more SOPs. EMS Heritage has adopted a formal written SOP program for all procedures. Every SOP is to be written according to the

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EPA APPROVED ANALYTICAL METHODS AND REFERENCES FOR WATER ANALYSES

DRINKING WATER

- 1. 40 CFR Part 141, Subpart C and Subpart E Monitoring and Analytical Requirements.
- 2. "Methods for the Determination of Organic Compounds in Drinking Water", EPA/600/4-88/039, December 1988.
- 3. "Methods for Chemical Analysis of Water and Wastes", EPA-600/4-79-020, Revised March 1983.
- 4. Standard Methods for the Examination of Water and Wastewater, 16th Edition, 1985.
- 5. "Manual for Certification of Laboratories Analyzing Drinking Water", EPA 570/9-90/008, April 1990.
- 6. "Prescribed Procedures for Measurement of Radioactivity in Drinking Water", EPA-600/4-80/032, August 1980.

Note: "500" series methods shall be used for drinking water related analyses unless client requirements dictate low detection limits achieved by those methods and analyses require no dilutions. Regulatory approval may be required for use in other matrices. Application of drinking water QC procedures may be inappropriate for some sample matrices.

SURFACE WATER, GROUNDWATER AND WASTEWATER EFFLUENTS

- 1. 40 CFR Part 136, Tables IA, IB, IC, ID, and IE, July 1989.
- 2. Methods for Chemical Analysis of Water and Wastes, revised March 1983.
- 3. "Test Methods for Evaluating Solid Waste, Physical/Chemical Methods", Third Edition (EPA SW-846), 1986 and Revision I dated December 1987.*

[&]quot;USEPA notes that, for guidance purposes, the Third Edition and its Revision I supersede the Second Edition and its Updates I and II. However, for regulatory purposes, the Second Edition and Updates I and II remain in effect together with 47 methods of the Third Edition and its Revision I cited above.

- 4. "Test Methods for Evaluating Solid Waste, Physical/Chemical Methods", Second Edition (EPA SW-846), 1982 as amended by Update I (April 1984), and Update II (April 1985).
- 5. Methods listed in 40 CFR Part 261, Appendix II, Appendix III and Appendix X, 1990.

WATER SOURCES (SURFACE WATER AND GROUNDWATER) ANALYZED PURSUANT TO 40 CFR PART 261 (Resource Conservation and Recovery Act) and the DER Chapter 17-700 Series Rules

- 1. "Test Methods for Evaluating Solid Waste, Physical/Chemical Methods", Third Edition (EPA SW-846), 1986 and its Revision I dated December 1987.¹
- 2. Methods listed in 40 CFR Part 261, Appendix II, Appendix III and Appendix X, 1990.
- 3. "Test Methods for Evaluating Solid Waste, Physical/Chemical Methods", Second Edition (EPA SW-846), 1982 as amended by Update I (April 1984), and Update II (April 1985).
- 4. "USEPA Contract Laboratory Program Statement of Work for Inorganic Analyses", Multi-Media, Multi-Concentration EPA SOW 7/88, July 1988.
- 5. "USEPA Contract Laboratory Program Statement of Work for Organic Analyses", Multi-Media, Multi-Concentration EPA SOW 2/88, February 1988.
- 6. "Prescribed Procedures for Measurement of Radioactivity in Drinking Water", EPA-600/4-80/032, August 1980.
- 7. "USEPA Contract Laboratory Program Statement of Work for Inorganic Analysis, Multi-Media, Multi-Concentration, Document Number ILM01."
- 8. "USEPA Contract Laboratory Program Statement of Work for Organic Analysis, Multi-Media, Multi-Concentration, Document Number OLM01".

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EPA APPROVED ANALYTICAL METHODS AND REFERENCES FOR SEDIMENTS, SOILS, RESIDUALS AND SOLID AND HAZARDOUS WASTES

- 1. "Test Methods for Evaluating Solid Waste, Physical/Chemical Methods", Third Edition (EPA SW-846), 1986 and its Revision I dated December 1987.
- 2. "Test Methods for Evaluating Solid Waste, Physical/Chemical Methods", Second Edition (EPA SW-846), 1982 as amended by Update I (April 1984), and Update II (April 1985).
- 3. "USEPA Contract Laboratory Program Statement of Work for Inorganic Analyses", Multi-Media, Multi-Concentration, EPA SOW 7/88, July 1988.
- 4. "USEPA Contract Laboratory Program Statement of Work for Organics Analyses", Multi-Media, Multi-Concentration, EPA SOW 2/88, February 1988.
- 5. "USEPA Contract Laboratory Program Statement of Work for Inorganic Analysis, Multi-Media, Multi-Concentration, Document Number ILM01."
- 6. "USEPA Contract Laboratory Program Statement of Work for Organic Analysis, Multi-Media, Multi-Concentration, Document Number OLM01."
- 7. Methods listed in 40 CFR Part 261, Appendix II, Appendix III and Appendix X, 1990.
- 8. American Society for Testing and Materials (ASTM), 1916 Race Street, Philadelphia, PA 19103.

USEPA notes that, for guidance purposes, the Third Edition and its Revision I supersede the Second Edition and its Updates I and II. However, for regulatory purposes, the Second Edition and Update I and II remain in effect together with 47 methods of the Third Edition and its Revision I cited above.

guidelines given in the EMS Heritage "Standard Operating Procedure for the Development and Maintenance of Standard Operating Procedures".

All SOPs must be approved by management and distribution is strictly controlled, which precludes the use of outdated or inappropriate SOPs. Copies of all SOPs which have been replaced and retired are collected upon the retirement. Copies of all SOPs will be maintained at each division; a master copy will be maintained at the Indianapolis Division.

This procedure will assure standardization among all locations. As new SOPs are developed, they must be coordinated through the Vice President of operations to avoid overlap or duplication of existing SOPs.

SOPs prepared by EMS Heritage will be clear, comprehensive, up-to-date, and sufficiently detailed to permit duplication of results by qualified analysts. In addition, all SOPs will be:

- 1. Consistent with current EPA regulations, guidelines and contract requirements;
- 2. Consistent with instrument manufacturer's specific instruction manuals;
- 3. Available to clients and regulatory personnel during On-Site Laboratory Evaluations. A complete set of SOPs shall be bound together and available for inspection at such evaluations;
- 4. Capable of providing for the development of documentation that is sufficiently complete to record the performance of all tasks required by the protocol;
- 5. Capable of demonstrating the validity of data reported;
- 6. Capable of describing the corrective measures and feedback mechanism utilized when analytical results do not meet protocol requirements;
- 7. Reviewed regularly and updated as necessary when contracts, facility, or procedural modifications are made;
- 8. Archived for future reference in usability or evidentiary situations; and,
- 9. Available at specific work stations as appropriate.

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8.2.1 <u>Documented Sampling Techniques</u>

SOP's must be in place and followed for sampling techniques and methodology as well as for sample control (including chain of custody).

8.2.2 <u>Documented Analytical Methods</u>

Routine analytical methods must be documented in an Analytical Methods Manual. Reference can be made to published methods, but are not a substitute for methods written specific to EMS Heritage.

Modifications to methods or non-standard methods must be approved by the QA Officer, Laboratory Director, QA Unit or Chief Chemist before use. Analytical methods in use must have a method detection limit established, determined and documented. All concerned staff will be notified of any method modifications or non-standard methods.

8.2.3 Equipment Calibration

Calibration or USEPA standards and their verification standards shall be traceable to NIST wherever possible or to another reliable source which may be determined by the QA Unit.

Calibration must be documented and this documentation will include:

- 1. Date(s) of calibration
- 2. Identification of standards used.
- 3. Results of calibration (raw data and summary statistics).
- 4. Corrective actions taken.

8.2.4 Training and Training Documentation

EMS Heritage must have an SOP for training and training documentation. Training shall continue, according to written procedures, to keep all personnel aware of new methods and technologies. Attendance at symposia, training courses, and further formal education is encouraged. Training which has been completed must be documented for each employee and kept as a part of the training records. Refer to Appendix H, <u>Training Forms</u> for "example" training documentation forms.

8.3 Laboratory Glassware Cleaning

8.3.1 Metals Analyses

Glassware may be segregated in the metals section according to the following categories as required to avoid cross contamination:

- 1. ICP/FAA water samples
- 2. GFAA water samples
- 3. Solid samples and/or non-water matrices
- 4. Mercury samples

Acid baths may also be segregated as required to avoid cross contamination according to the above categories. Only designated acid baths will be used. All glassware is air-dried on adsorbent covered laboratory bench tops and stored in labelled drawers adjacent to the preparation areas.

The glassware cleaning procedure for metals analysis is summarized below.

- 1. After preps are completed, glassware is to be rinsed with tap water and placed in a soak bath of mildly acidic water.
- 2. Wash with warm tap water and micro-soap. Scrub each piece thoroughly and inspect for cracks or chips while rinsing.
- 3. Rinse with warm tap water to remove all soap residue.
- 4. Rinse twice with deionized (DI) water.
- 5. Rinse once with 1+1 Nitric Acid.
- 6. Rinse three times with DI water.
- 7. Soak in an acid bath (approximately 30 percent Nitric Acid) for a minimum of 2 hours. Glassware must be completely submerged with no air bubbles.
- 8. Remove from acid bath and rinse 3 times with DI water.
- 9. Check pH of random pieces with pH paper. Re-rinse if pH<2 on any piece.

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- 10. Allow to air dry on adsorbent covered lab bench.
- 11. Place in appropriate adsorbent lined lab drawer.
- 12. Final rinse of all GFAA waters glassware should be done with Milli-Q (or equivalent) water. Analysts must rinse glassware with 1+1 Nitric Acid and Milli-Q (or equivalent) DI water (three times) just prior to using for preps.

Mercury Bottles

- 1. Cold water rinse when test is complete.
- 2. Rinse three times with hot tap water.
- 3. Rinse 3 times with DI water.
- 4. Rinse 2 times with concentrated Nitric Acid.
- 5. Rinse 3 times with Milli-Q (or equivalent)(DI) water.
- 6. Check pH of random pieces with pH paper. Re-rinse if pH <2 on any piece.

EPTOX, TCLP and Other Large Glassware

Large pieces of glassware (filtering flask, etc.) do not need to be soaked in an acid bath. Large glassware must be thoroughly rinsed with 1+1 Nitric Acid and rinsed two times with DI water, followed by another rinse with 1+1 Nitric Acid and rinsing 3 times with Milli-Q (or equivalent)(DI) water.

8.3.2 Extractable Organics

- 1. Wash with hot soapy water, taking care to not scratch w/brush.
- 2. Sonicate small pieces if necessary to remove all residues.
- 3. Rinse with hot tap water.
- 4. Rinse three times with DI water.
- 5. Air dry, or rinse with ACS Reagent Grade Acetone.
- 6. Rinse with Pesticide grade Methylene Chloride.
- 7. Store with clean glassware in a dust free environment or cap with aluminum foil.

8. Rinse with the solvent used for analysis immediately before use.

Note: Keep all plastics away from glassware.

8.3.3 Volatile/Purgeable Organics

Purge vessels are washed in hot soapy water and rinsed with hot tap water and DI water after each use unless there are <u>no</u> detectable analytes. Purge vessels are replaced in the same positions after cleaning, therefore carryover from high concentration samples can be easily detected.

Volumetric glassware, syringes, etc. (i.e. used in standard preparation) are cleaned according to the following protocol:

- 1. Rinsed three times with tap water
- 2. Rinse three times with Milli-Q (or equivalent)(DI) water.
- 3. Syringes rinsed five times with methanol.
- 4. Placed in storage at room temperature until use.

8.3.4 Classics/Nutrients (General Chemistry)

- 1. Wash with a hot micro-soap solution, taking care not to scratch with the brush.
- 2. Rinse three times with hot tap water.
- 3. If glassware is to be used for TKN or Total Phosphorus, rinse with 1+1 HCl
- 4. Rinse three times with DI water.
- 5. Air dry.
- 6. Store with clean glassware in a dust free environment or cap with aluminum foil.
- 7. Rinse with the solvent used for analysis, if applicable.

8. Refer to specific SOPs for special decontamination procedures for distillation apparatuses, etc.

8.4 Reagent and Standard Storage

All reagents and analytical standards received into the laboratory will be labelled with the date received and the expiration date. The label will also contain a space for "first opened". All reagents and analytical standards will be delivered to the appropriate group, who will store them as appropriate.

Some bulk reagents such as acids and solvents will be delivered in large cases which may be stored until use in the storeroom. The laboratory Receiving Department will be responsible for labelling the date of receipt on the outside of the un-opened case of bulk reagents. Any containers that have been opened may no longer remain in the store room on open shelves, and must be moved to the designated storage as identified in Table 8.3, Reagent and Standard Storage. All chemicals will be stored under temperature and humidity controlled conditions. Compatibility of chemicals will be considered for storage.

8.5 Sample and Waste Disposal

It is the policy of EMS Heritage to comply with all regulations concerning proper disposal and safe handling of laboratory samples, chemicals and waste products. Documentation will consist of manifesting of all sample and chemical analysis waste products.

- 8.5.1 Sample Disposal: It is the policy of EMS Heritage Labs to hold samples for fifteen days after analysis has been complete. EMS may store samples for an extended amount of time upon request, but an additional storage fee may apply. Once a sample is ready for disposal, it may be disposed of in two ways: Internal Sample Disposal or Return to Generator.
 - 8.5.1.1 Return to Generator: A generator may request that a sample be returned after analysis is complete. Samples will be returned by common carrier or picked up by the generator.

REAGENT AND STANDARD STORAGE

CHEMICAL	METHOD OF STORAGE
Acetic Acid	Stored in original containers in cabinets designated for liquid reagent acid storage.
pH Standards	see above
Acetone	Stored in original containers in cabinets designated for
Pyridine	flammable chemicals.
Hexane	
Methanol	
Acetonitrile	
Carbon disulfide	
Methyltertbutyl ether	
Sodium hydroxide	Stored in original containers in cabinet designated for base storage.
Freon 113 Dimethyl sulfoxide Dimethyl formamide	Stored in original containers in cabinet designated for solvent storage.
Phenylarsine oxide	Stored in original containers in cabinet designated for
Methylene chloride	reagents.
Sodium sulfate anhydrous	Stored in original containers in a cabinet designated for standard and reagent storage.
Ether (anhydrous)	Stored in original containers inside a vented flammable solvent storage cabinet.
Cadmium metal	Stored in original container in a cabinet designated for
Mercuric thiocyanate	standard and reagent storage.
Ammonium chloride	
Sulfanilimide	
Ferric nitrate	

CHEMICAL	METHOD OF STORAGE
EDTA	Stored in original containers on benchtop.
Sodium sulfide	
Potassium Bi-Iodate	
Iodine	
Nitric acid	Stored in original containers in cabinet designated for
Sulfuric acid	acid storage.
Hydrochloric acid	
Hydrogen peroxide	
Phosphoric acid	
Hypophosphorus acid	
85% O-phosphoric acid	
Sodium hydroxide	Stored in original containers in cabinets designated for
Hydroxylamine hydrochloride	dry chemical storage.
Stannous chloride	
Potassium persulfate	
Barbituric acid	
Sodium phosphate dibasic	
Potassium ferricyanide	
4-Aminoantipyrine	
Brij 35	
Sulfamic acid	
Calcium hypochlorite	
Boric acid	
Potassium chloride	
Chloramine-T	
L-Ascorbic acid	
Magnesium chloride	
Potassium cyanide	
Potassium permanganate	Stored in original containers in a cabinet designated for standard and reagent storage.

- 8.5.1.2 Internal Sample Disposal (solids/others): After the proper holding time beyond analysis completion, samples may be disposed of internally. Samples ready for disposal are taken from their storage area and segregated by type of waste. The type of waste is determined by examining information available on the sample such as client descriptions and analytical results. Once the type of waste is determined, the sample bottle is clearly marked and placed in the Sample Disposal Storage Area. Samples are stored until a contracted company removes the waste, generally on a monthly or as-needed basis. The contractor packs the waste which is then sent to the Heritage Treatment Center's Lab-Depacking Program or, depending on the type of waste, sent to an incinerator.
- 8.5.1.3 Internal Sample Disposal (waters): After the proper holding time beyond analysis completion, water samples are disposed of internally. Samples ready for disposal are taken from their storage area and reviewed for their method of disposal. The method of disposal is determined by examining information available on the sample such as client descriptions and analytical results. All waters that are nonhazardous can be poured down the drain. If a water sample has any contaminants which would preclude disposal to a POTW (e.g. Dioxins, PCBs) it is segregated in the same fashion as solids/others. After determining the type of waste, these samples are placed in the appropriate area in Sample Disposal Storage. The samples are stored until a contracted company removes the waste according to the waste type.
- 8.5.1.4 Internal Sample Disposal (waste approval samples): Waste approval samples are received and entered into the computer by the QA/QC Laboratory of Heritage Treatment Center. Therefore, after a waste approval sample has been completed for at least fifteen days, the samples are returned to QA/QC lab for disposal.
- 8.5.2 Lab Waste Disposal: There are 7 different types of lab generated waste Lab Prep Waste, Auto Sampler Vials, Extraction Waste, Sulfuric Acid Waste, Nitric Acid Waste, and Waste Solvents. This

waste is accumulated on a daily basis and is handled according to it's type of waste.

- 8.5.2.1 Lab Prep Waste: Wastes generated by sample preparation and spill clean up are placed in this category. This would include used pipets, absorbent materials and gloves or towels that have come in contact with hazardous materials. The waste is segregated by determining the type of spill, or the laboratory that generated the waste. For example, if the waste is from a nitric acid spill, the absorbent used to clean the spill will be collected and placed in the Nitric Acid Section in the Sample Disposal Area, whereas waste pipets from the prep lab for PCB's would be collected and placed in the Disposal Area for incineration along with all other PCB waste.
- 8.5.2.2 Auto-Sampler Vials: Vials are generated by the organics labs containing solvents from sample extracts. The vials are accumulated in storage containers in the laboratory then are placed in the Sample Disposal Area for incineration.
- 8.5.2.3 Extraction Waste: Extraction waste is generated from the extraction of samples containing of a mixture of soils, solids, and oils in solvents. These containers are accumulated in the laboratory, then are placed in the Sample Storage Area for incineration. Note that the area is designated for flammables because of the waste.
- 8.5.2.4 Sulfuric Acid Waste: Sulfuric acid waste is generated in all areas of the lab and in the General Laboratory from COD analysis. This waste acid is accumulated in Carboys and stored in acid cabinets until full. The waste acid is then placed in the sulfuric acid section of the Sample Disposal Area. It is stored in the disposal area until several Carboys are accumulated, then sent to Heritage Treatment Center (as a Small Quantity Generator) who manifests and picks up the waste upon request.
- 8.5.2.5 Nitric Acid Waste: The inorganics (metals) section of the laboratory generates nitric acid waste which is stored

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in Carboys in acid cabinets until full. The Carboys are then placed in the nitric acid section of the Sample Disposal Area. It is stored until several Carboys are accumulated then sent to Heritage Treatment Center (as a Small Quantity Generator) who manifests it along with the sulfuric waste.

8.5.2.6 Waste Solvents: Solvents used to rinse glassware, solvents reclaimed during sample extractions, or expired solvents are accumulated in the laboratory. Bottles containing waste solvents are clearly marked and are placed in the flammable section of the Sample Disposal Area for incineration.

Samples and sample derived waste products are not routinely tracked with regard to their precise fate. If clients desire this precise tracking of the fate of their sample, this can be accommodated on a project or client specific basis with adequate advance notice and agreement on procedures.

9.0 Calibration Procedures and Frequency

9.1 <u>Laboratory Instrumentation</u>

A list of laboratory instrumentation is found in Table 9.1, Instrument List.

9.2 Standards Receipt and Traceability

The Material Safety Data Sheet (MSDS) must be on file or sent with a substance before acceptance into the lab. MSDS's included with the shipping container should be in an external envelope so that it can be read and all precautions followed before opening the container.

All chemicals delivered to the lab will be accepted in the shipping and receiving area. Bulk reagents such as acids and solvents will be delivered in large cases which may be stored in the store room until use. The shipping and receiving department will label the date of receipt on the outside of the unopened case and of bulk reagents. All other reagents and analytical standards will be delivered (unopened) immediately to the appropriate Study Manager or Group Leader. The containers will be inspected to check for damage, leakage, etc. If the reagents, standards and bulk reagents taken from the store room are acceptable they will be entered into a logbook with the following information:

- 1. Identification of compound (CAS number, if possible)
- 2. Source of Standard or reagent
- 3. Date of receipt
- 4. Date first opened
- 5. Purity
- 6. Batch or Lot number
- 7. Stability expiration date
- 8. Approximate amount

For ease and/or confidentiality, chemicals or reagents can be assigned (and all bottles labelled with) a sequential laboratory compound (project specific) identification number based on the protocol or contract number. This information is logged in the project specific logbook. Solutions and dilutions of this compound will be tracked through this number.

All manufacturer's certification and traceability statements are maintained on file. The statements will be marked with the date received and the identification of the logbook where they are referenced. When logbooks are

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Table 9.1 INSTRUMENT LIST

Division	Instr. Code	Name	Manufacturer	Model
Indianapolis	16	Auto Analyzer II	Technicon	CN, N-NO ₃ , Phenol
	7	FTIR 1	Biorad	FTS-7
	5	GC/ECD #5 Channel 1	Hewlett Packard	5880A
	9	GC/ECD #9 Channel 1	Hewlett Packard	5880A
	24	GC/ECD #24 Channel 1	Hewlett Packard	5890
	25	GC/ECD #25 Channel 1	Hewiett Packard	5890 Series II
	5	GC/ECD #5 Channel 2	Hewlett Packard	5880A
	9	GC/ECD #9 Channel 2	Hewlett Packard	5880A
	24	GC/ECD #24 Channel 2	Hewlett Packard	5890
	25	GC/ECD #25 Channel 2	Hewlett Packard	5890 Series II
	41	GC/FID/NPD	Hewlett Packard	5890
	106	GC/FID (Dual Channel)	Hewlett Packard	5890 Series II
	104	GC/FID/PID/ELCD w/Purge & Trap	Varian	3400
	35	GC/MS 1 w/Purge & Trap	Hewlett Packard	5985 w/5840 GC
	36	GC/MS 2	Hewlett Packard	5987A w/5880 GC
	37	GC/MS 3 w/Heated Purge & Trap	Hewlett Packard	5970 w/5890 GC
	38	GC/MS 4 w/Purge & Trap	Hewiett Packard	5970 w/5890 GC
	39	GC/MS 5 w/Purge & Trap	Hewiett Packard	5970 w/5890 Series II
	40	GC/MS 6	Hewlett Packard	5970 w/5890 GC
	107	GC/MS 7 w/Purge & Trap	Hewlett Packard	5890 w/5890 GC
	160	GC/MS 8	Hewlett Packard	5971 MSD
	19	GC/PID/ELCD w/Purge & Trap	Hewlett Packard	5890
		GC/PID/FID - Portable	HNu	
	101	Gas Flow Proportional Counter	Canberra	2405
	96	HPLC 1 w/Photodiode Array Detector and Fluorescence Detector	Waters	600E
	100	HPLC 2 w/Multi-Wave Length Detector, Scanning Fluorescence Detector and Pickering Post-Column Derivitization	Waters	600E
	32	ICP, Simultaneous	Thermo Jarrell Ash	61
	29	FAA	Instrumentation Laboratories	S12

Table 9.1 INSTRUMENT LIST

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Division	Instr. Code	Name	Name Manufacturer Mod	
Indianapolis	46	ISE Meter 1 (Gen)	Beckman	РНІ 34
	47	ISE Meter 2 (Metals)	Beckman	PHI 34
		Digital Ion Analyzer (Gen)	Orion	601A
	98	Auto Analyzer (Lachat)	Lachat Instr.	2300
	33	FAA (CVAA, PE2380)	Perkin Elmer	2380
	34	GFAA (PE3030)	Perkin Elmer	3030/Zeeman
	30	GFAA (PE5100-A)	Perkin Elmer	5100/Zeeman
	31	GFAA (PE5100-B)	Perkin Elmer	5100/Zeeman
	45	Spectrophotometer	Sequoia-Turner	690
	110	TOC (Combustion IR)	Rosemount Analytical/ Dohrman	DC-190
	26	1 хот	Envirotech/ Dohrman	MC-1
	27	тох іі	Xertex/Dohrman	MC-1(DX-20)
	44	Turbidimeter	Hach	2100A
	63	GC/ECD 3 (1 of 2 CH)	Hewlett Packard	5890
	63	GC/FID 1 (2 of 2 CH)	Hewlett Packard	5890
	114	Conductivity Meter	Yellow Springs Instruments	35
	125	Dissolved Oxygen meter	Yellow Springs Instruments	59
	126	Balance 1 (to 0.0001g)	Sartorius	1601 MP8
	127	Balance 2 (to 0.0001g	American Scientific	SP180
	128	Balance 3 (Top Loading to 0.001g)	Ohaus	B300D
	129	Balance 4 (Top Loading to 0.01g)	Fisher	7240DA
	130	Balance 5 (Top Loading to 0.01g)	Mettler	PM4000
	131	Balance 6 (Top Loading to 0.01g)	Ohaus	300
	132	Balance 7 (Top Loading to 0.01g)	Fisher	S-400
	133	Balance 8 (to 0.0001g)	Mettler	AE200
	134	Balance 9 (Top Loading to 0.001g)	Ohaus	E400D
	135	Balance 10 (Top Loading to 0.001g)	Ohaus	E400D
	136	Balance 11 (Top Loading to 0.01g)	Fisher	LX500
	137	Balance 12 (microBalance to 0.00001)	Mettler	М3

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Table 9.1 INSTRUMENT LIST

Division	Instr. Code	Name	Manufacturer	Model
Charlotte	92	FAA (CVAA, PE360)	Perkin Elmer	360
	91	FAA	Perkin Elmer	3030
	90	GC/NPD/FPD	Hewlett Packard	5890 Series II
	87	GC/ECD	Hewlett Packard	5890
	89	GC/FID/PID/ELCD	Hewlett Packard	5890
	88	GC/FID/ECD	Hewlett Packard	5730
	85	GC/MS 1	Hewlett Packard	5970 w/5890 GC
	86	GC/MS 2	Hewlett Packard	5970 w/5890 GC
	103	GC/MS 3	Hewlett Packard	5970 w/5890 GC
	108	GC/PID/FID	Varian	3400
	93	GFAA 1	Perkin Elmer	3030/Zeeman
	94	GFAA 2	Perkin Elmer	3030/Zeeman
	84	Auto Analyzer, Traacs	Bran & Luebbe	800 AA
	138	Conductivity Meter	Myron	EP
	139	Spectrophotometer	Milton Roy	21DV
	140	Dissolved Oxygen Meter	Yellow Springs Instruments	S4A
	141	Bomb Calorimeter	Parr	1241
	142	ISE Meter #2	Orion	920A
	144	ISE Meter #1	Cole Parmer	5986.50
	145	Turbidimeter	Hach	2100A
	149	Analytical Balance, Top Loading	Ohaus	B300D
	150	Analytical Balance	Mettler	Н30
Kansas City	77	Bomb Calorimeter	Parr	1241
	76	ISE 1	Fisher Scientific	Accummet 925
i	72	Balance 1 (to 0.0001g)	Mettler	AE620
	73	Balance 2 (Top Loading to 0.001g)	American Scientific Products	
	74	Balance 3 (to 0.0001g)	American Scientific Products	SP 120
		Balance 4 (Top Loading to 0.001)	Denver Instrument Company	XE Model 3000D
		Balance 5 (Triple Beam to 0.1g)	Fisher	711

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Table 9.1 INSTRUMENT LIST

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Division	Instr. Code	Name	Manufacturer	Model
Kansas City		Digital Thermometer	Fisher	Serial #L064041
	65	GC/ECD 1 (1 of 2 CH)	Hewlett Packard	5890
	65	GC/ECD 2 (2 of 2 CH)	Hewiett Packard	5890
	115	GC/MS w/Purge & Trap	Hewlett Packard	5970 w/5890 GC
	116	GC/MS	Hewiett Packard	5970 w/5890 GC
	117	IR Spectrophotometer	Perkin Elmer	1310
	70	FAA (PE2380) w/CVAA	Perkin Elmer	2380
	71	GFAA (PE3030)	Perkin Elmer	3030/Zeeman
	28	TOC 1	Beckman	915B Tocamaster
	118	Dissolved Oxygen Meter	Yellow Springs Instruments	54A
	119	Conductivity Meter	Orion	124
Romeoville	120	Balance 5 (to 0.01g)	Mettler	BB300
(Chicago)	121	Balance 6 (to 0.001g)	Ohaus	E400
	60	Balance 2 (to 0.0001g)	Mettler	AE260-S
	99	Balance 3 (to 0.0001g)	Mettler	AF100
	61	Bomb Calorimeter	Рагт	1241 Ea
	54	Conductivity meter	Cole/Parmer	1484-20
	50	GC/ECD/FID	Hewlett Packard	5890A
	51	GC/ECD/FID	Hewlett Packard	5890A
	102	GC/MS 1	Hewlett Packard	5970 w/5890 GC
	49	GC/MS w/Heated Purge & Trap	Hewlett Packard	5970 w/5890 GC
	20	FAA, CVAA, GFAA	Instrumentation Laboratories	V12E
	58	Balance 4, Top Loading	Ohaus	E300D
	57	Dissolved Oxygen Meter	Yellow Springs Instruments	54ABP
	97	FAA (CVAA, PE2380)	Perkin Elmer	2380
	105	ICP, Sequential	Perkin Elmer	P40
	22	GFAA (PE5100)	Perkin Elmer	5100/Zeeman
	55	ISE Meter 1	American Scientific Products	
	56	ISE Meter 2	Orioa	SA520

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Table 9.1 INSTRUMENT LIST

Division	Instr. Code	Name	Name Manufacturer M	
Romeoville	59	Analytical Balance	American Scientific	B1240-2
(Chicago)	159	pH Meter	Orioa	520A
95		ISE Meter 3	Orion	520
	52	Spectrophotometer 1	Milton Roy	SPEC 21
	53	Spectrophotometer 2	Milton Roy	SPEC 21
	122	GC/MS A w/Purge and Trap	Hewlett Packard	5970 w/5890 GC
	123	GC/MS B	Hewlett Packard	5970 w/5890 GC
124		FTIR	Matheson	Galaxy 2020
	161	Spectrophotometer 3	Milton Roy	SPEC 21

Abbreviations/Acronyms:

GC	=	Gas Chromatograph	MS	**	Mass Spectrometer
ECD		Electron Capture Detector	NPD	-	Nitrogen-Phosphorus Detector
FTIR	=	Fourier Transform Infra-Red	PID	=	Photoionization Detector
ELCD	-	Electrolytic Conductivity Detector	ICP	•	Inductively Coupled Argon-Plasma
FAA	=	Flame Atomic Absorption	CVAA	=	Cold Vapor Atomic Absorption
GFAA	-	Graphite Furnace Atomic Absorption	TOC	=	Total Organic Carbon
TOX	=	Total Organic Halides	FPD	=	Flame Photometric Detector
ISE	-	Ion Selective Electrode	CH	-	Channel(s)
IR	=	Infra-Red, Dispersive	FID	=	Flame Ionization Detector

archived, the associated certification file documentation is archived with the logbook.

Only standard sources approved by the QA Unit may be used. Standards will be traceable to NIST when possible.

9.3 Standard Sources and Preparation

Stock solutions are prepared by the Group Leader or their designee for the purpose of making standards and performing analyses. There must be no handling of the reagents directly from the reagent bottles. Instead, minimal amounts are to be placed in interim holding containers. Unused stock solutions must not be poured back into original containers. The possible toxicity of the reagents must be noted and respected. Stock solutions are to be made using analytical reagent grade or other appropriate grade solvents. Storage precautions must be noted and observed with regard to heat and light sensitive compounds. Solutions must be kept in the proper container tightly sealed (amber glass where necessary) with Teflon lined lids in an appropriate refrigerator. No mouth pipetting is permitted.

When preparing stock solutions, at least 0.1000 grams or 1.00 ml of reagent is to be used, when possible or practical. Direct delivery into a volumetric flask is required. Stock solution preparation noting lot numbers of reagents and solvents in addition to date and initials of person preparing the solutions must be noted and recorded on the label and in the appropriate standards logs.

Stock solutions under proper storage (refrigerated if necessary) may be kept for a period not to exceed the maximum recommended storage time. Standard solutions are prepared from the stocks by the analyst on a daily, weekly, or monthly basis as necessary.

Standard sources and preparation protocols are listed in Table 9.2, STANDARD SOURCES AND PREPARATION. This table includes the following information:

- 1. Instrument group
- 2. Source and traceability of primary standards
- 3. Frequency of preparation
- 4. How primary and working standards are stored.

Table 9.2 STANDARD SOURCES AND PREPARATION

Instrument Group	Standard Sources/ Traceability	Concentrations Received	Source Storage	Preparation from Source ¹	Lab Stock Storage	Prep Frequency
GC/MS - Volatile						
Methods 624, 824Ø	Supelco or equivalent	200ppm	frozen	1:10 dilution in methanol ¹	Refrig.	Weekly or as needed
Method 524	Supelco or equivalent	2000ррт	frozen	Stock solution 1:20 in methanol ¹ Working solution 1:10 in methanol ¹	Refrig.	Quarterly, Every 2 weeks or as needed
CLP OLMØ1	Ultra or equivalent	100ppm	frozen	1:5 dilution in methanol ¹	Refrig.	Weekly or as needed
Appendix IX	AccuStandard or equivalent	Available 100-500 ppm	frozen	Dilute to 20ppm ¹	Refrig.	As needed
GC/MS - Semi-Volatile						
Methods 625, 8270	Supelco, Restek or equivalent	2000ррт	frozen	Make dilution to 160 ppm ¹	Refrig.	Every 2 weeks
Method 525	AccuStandard or equivalent	100ррт	frozen	Stock solution - 1:10 in methanol ¹	Refrig.	As needed
CLP OLMØ1	Ultra or equivalent	2000ррт	frozen	Stock solution - 160ppm ¹	Refrig.	As needed
Appendix IX	AccuStandard or equivalent	available from 100- 250ppm	frozen	Run as received or dilute to 160ppm ¹	Refrig.	As needed

¹Daily calibration standards are made from stock solutions at method specified concentrations.

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Table 9.2 STANDARD SOURCES AND PREPARATION

Instrument Group	Standard Sources/ Traceability	Concentrations Received	Source Storage	Preparation from Source ¹	Lab Stock Storage	Prep Frequency
HPLC	Supelco Polynuclear Aromatic Hydrocarbon Mix Supelcopreme-HC or equivalent	Solutions of 2000µg/mL	Refrig.	Working stocks prepared from source stock	NA	Spiking solution extract with each sample set
	Supelco Polynuclear Aromatic Hydrocarbons Mixture 610-M or equivalent	Solutions of 100 to 2000 g/ml µ	Refrig.	Working stocks prepared from source stock	Refrig.	Semi-annual
	NIST-1647a PNA's	Solutions of 0.771 to 19.9 µg/mL	Refrig.	Working solutions prepared from source	NA	Per sequence
	AccuStandard PAH Additions to Method 610-M or equivalent	Solution of 0.1 mg/mL in CH ₂ Cl ₂	Refrig.	Working solution prepared from source	Refrig.	Semi-annual
	USEPA QC Sample PNA I WP 485	Solution of 5 to 100 µg/mL	Refrig. if seal is broken	Working solution prepared from source	Refrig.	2 months

Table 9.2 STANDARD SOURCES AND PREPARATION

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Instrument Group	Standard Sources/ Traceability	Concentrations Received	Source Storage	Preparation from Source ¹	Lab Stock Storage	Prep Frequency
HPLC	AccuStandard N-Methyl Carbamoyl Oximes and N- Methyl Carbamates in Drinking Water or equivalent	0.1 mg/mL in ACN	Refrig.	Working solution prepared from source	Room temp. protect from light	2 months
	Aldrich 4-Bromo-3,5- Dimethyl-Phenyl N-Methyl- Carbamate (BDMC) ISTD or equivalent	98% neat	Room temp.	Primary stocks ~ 1000 mg/L Intermediate stock ~ 100 mg/L Working - 5µL of Int. → 50 mL	Room Temp. protect from light	2 months 2 months Working-Daily
	AccuStandard Methiocarb (for LCS) or equivalent	Neat	Room temp.	Primary stock ~ 1000 mg/L Intermediate stock ~ 2 mg/L Working solution ~ 2PPB	Room temp. protect from light	2 months 2 months Working - Daily
	AccuStandard Aldicarb-Sulfoxide (for LCS) or equivalent	Neat	Room temp.	Primary stock ~ 1000 mg/L Intermediate stock ~ 100 mg/L Working solution ~ 100PPB	Room temp. protect from light	2 months 2 months Working - Daily

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Table 9.2 STANDARD SOURCES AND PREPARATION

Instrument Group	Standard Sources/ Traceability	Concentrations Received	Source Storage	Preparation from Source ¹	Lab Stock Storage	Prep Frequency
GC/PID/ELCD	AccuStandard or equivalent	Stock solution 2000 µg/mL	Freezer 0°C	Working solutions 20μg/mL	Freezer (with minimal headspace)	4 weeks, Gaseous stds. 1 week unless validity proven
	Supelco or equivalent	Stock solution 200 - 2000 µg/mL	Freezer 0°C	Working solutions 20μg/mL	Freezer (with minimal headspace)	4 weeks.Gaseous stds. 1 week unless validity proven
	Chem Service or equivalent	Neat compounds	Refrig.	Stock solutions ~ 10,000 µg/mL Working solution ~ 20 µg/mL	Freezer (minimal headspace	As needed for special analyte lists. 4 weeks unless validity proven.
FTTR	USEPA - API reference oil, #2 fuel oil	Neat compound	Refrig.	Stock solutions: Calibration Std. ~ 7500 Spiking soln ~ 20,000	Refrig.	6 month exp.
				Working calibration std. from stock soln. L-1 ~ 8 mg/l L-2 ~ 15 L-3 ~ 77 L-4 ~ 380 L-5 ~ 770 L-6 ~ 2300	NA	with every calibration
				Daily verification std. L-3 ~ 77	NA	Daily

Table 9.2 STANDARD SOURCES AND PREPARATION

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Instrument Group	Standard Sources/ Traceability	Concentrations Received	Source Storage	Preparation from Source ¹	Lab Stock Storage	Prep Frequency
ICP (LCS, ICV)	3 solutions SPEX multi-element or equivalent	50-10,000ppm multi-element mixes	Room temp.	Dil. 1:10 from source for LCS or as required	1%HNO,	Bi-weekly or as needed
ICP (Cal. Std's)	3 solutions - SPEX multi- element or equiv.	100-5,000ppm Multi-element mixes	Room temp.	Dil. 1:50 from source or as required	5% HCI 1%HNO ₃	Weekly
ICP	Perkin Elmer CLP Calibration Stds. or equiv.	1-1,000ррт	Room temp.	Intermediate solutions prepared from stock	1% HNO ₃	Weekly
ICP	EPA ref.	1-5ppm	Room temp.	Working solution prepared from source	1% HNO ₃	As needed
FAA (Cal. Stds.)	Fisher, RICCA or equiv.	1,000ррт	Room temp.	Working stock prepared from source 0.1 to 50 ppm	1% HNO ₃	Monthly or as needed
FAA (LCS, ICV)	Spex multi- element or equiv.	50-2000ppm mix	Room temp.	1:500 dil. for LCS or as needed	1% HNO ₃	Biweekly or as needed
FAA (Cal. Std's)	JM Specpure or equiv.	1000ррт	Room temp.	Ba, Pb-1:10 from source. Others - 1:10 intermed. from source then: 0.1-5ppm working from intermediate	2.5% HNO ₃	Weekly or as needed
FAA	EPA ref.	1-5ppm	Room temp.	Working solution prepared from source	1% HNO ₃	As needed

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Table 9.2 STANDARD SOURCES AND PREPARATION

Instrument Group	Standard Sources/ Traceability	Concentrations Received	Source Storage	Preparation from Source ¹	Lab Stock Storage	Prep Frequency
ICP, FAA, GFAA (LCS)	Conostan S-21 or equiv.	Metals in oil matrix As, Se = 100 mg/L Al, Ca, Fe, K, Mg, Na = 1,000 mg/L All other metals = 400 mg/L	Room temp.	Working solution prepared from source	NA	As needed
CVAA (LCS, ICV)	NIST 1641B	1.52ppm	Room temp.		0.2% KMnO ₄	Daily
CVAA (Cal. Std's)	Fisher, RICCA or equiv.	1000ррш	Room temp.	250ml → 25mL from source = 10ppm intermediate std. 1mL → 100mL = 0.1ppm working std.	5% HNO ₃ 0.2% KMnO ₄	Daily
CVAA	EPA ref. materials	As provided	Room temp.	Working solution prepared from source	1% HNO ₃	As needed
GFAA (Cal. Stds.)	Fisher, RICCA, SPEX, Specpure or equiv.	1,000ррш	Room temp.	Working solution prepared from intermediate	1% HNO ₃ & 5% HNO ₃	Weekly or as needed
GFAA (ICV, LCS)	EPA ref. materials	1-5ppm, As provided	Room temp.	Working solution prepared from source	1% HNO ₃	As needed
GFAA (LCS)	PE pure Inorganic Ventures or equiv.	1,000ррш	Room temp.	Stock prepared from source	5% HNO,	Semi-annually or as needed

Table 9.2 STANDARD SOURCES AND PREPARATION

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Instrument Group		Standard Sources/ Traceability	Concentrations Received	Source Storage	Preparation from Source ¹	Lab Stock Storage	Prep Frequency
Conductivity Meters	CAL	Fisher or equiv.	Potassium hydrogen phthalate (KHP) granular	Room temp.	0.3g/1000mls DI (168µmhos)	.Refrig. 4°C	6 mos. or as needed
	ICV	Fisher or equiv.	Potassium chloride granules	Room temp.	0.7456g/1000ml D1 (1413µmhos)	.Refrig. 4°C	6 mos. or as needed
ISE Meters	(PH)	Commercial Lab Supplies	pH, 7, 4 & 10	Room temp.	None	Room temp.	NA
Chloride	CAL	Fisher or equiv.	NaCl granules	Room temp.	1.6480g/1000mls (500ppm)(CAL)	Room temp.	6 mos. or as needed, exp. date on bottle
	ICV	Ricca or equiv.	Chloride Standard (3540ppm)	Room temp.	premade (3540ppm)(ICV)		
Fluoride	CAL	Mallinckrodt or equiv.	NaF granules (CAL)	Room temp.	0.221g/11 DI (100ppm)(CAL)	Room temp.	6 mos. or as needed, exp. date on bottle
	ICV	Ricca or equiv.	Fluoride 100ppm (ICV)	Room temp.	premade (ICV)		
Ammonia	CAL	Mallinckrodt or equiv.	Ammonium chloride granules (CAL)	Room temp.	3.819g/1c DI (1000ppm)(CAL)	Room temp.	exp. date on bottle
	ICV	Ricca or equiv.	Ammonia (ICV)	Room temp.	premade (ICV)]	
Spectrometers Phenois	CAL	Chem Service or equiv.	Phenol crystals (CAL)	Room temp.	0.1g/100mls of boiled DI (CAL)	Room temp.	6 mos or as needed, exp.
	ICV	RICCA/Mallinc- krodt or equiv.	Phenol 1000ppm (ICV)	Room temp.	Premade (ICV)		date on bottle

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Table 9.2 STANDARD SOURCES AND PREPARATION

Instrument Group		Standard Sources/ Traceability	Concentrations Received	Source Storage	Preparation from Source ¹	Lab Stock Storage	Prep Frequency
Nitrite C	CAL	Fisher or equiv.	KNO ₂ granules (CAL)	Room temp.	3.351g/500mls w/2mls CHCl ₃ (CAL)		3 mos. or as needed, exp.
	ICV	Red Bird or equiv.	premade (ICV)	Room temp.	premade (100ppm) (ICV)		date on bottle
Nitrate	ICV	Red Bird or equiv.	Nitrate 1000ppm (ICV)	Room temp.	premade (100ppm) (ICV)	Room temp.	exp. date on bottle, semi-
	CAL	Fisher or equiv.	KNO ₃ granules (CAL)	Room temp.	3.609g/500mls DI (CAL)		annually
Sulfate ICV CAL	ICV	Fisher or equiv.	NaSO ₄ granules	Room temp.	1.479g/1c DI 100ppm (ICV)	Room temp.	6 mos. or as needed
	CAL	Red Bird or equiv.	1000 ppm SO ₄	Room temp.	Premade 1000 ppm (CAL)		
Cyanide	ICV	Mallinckrodt or equiv.	KCN granules (ICV)	Room temp.	.251g/100, 1000ppm (ICV)	1	6 mos. or as needed
	CAL	Kodak or equiv.	KCN granules (CAL)	Room temp.	.251g/100mls, 1000ppm (CAL)		
PO ₄	ICV	RICCA or equiv.	Phosphorus 1000ppm (ICV)	Room temp.	premade	Room temp.	6 mos. or as needed
	CAL	Fisher or equiv.	KH ₂ PO ₄ granules (CAL)	Room temp.	0.1099g/500mls		

Instrument	Group	Standard Sources/ Traceability	Concentrations Received	Source Storage	Preparation from Source ¹	Lab Stock Storage	Prep Frequency
COD Reactor	CAL	Fisher or equiv.	K ₂ Cr ₂ O ₇ crystals (CAL)	Refrig. 4°C	12.259g/L	Refrig. 4°C	every other month
	ICV	Fisher or equiv.	KHP (ICV)	Refrig. 4°C	0.4250g/L (500ppm)		
Hexavalent Chromium	CAL	Fisher or equiv.	K ₂ Cr ₂ O ₇ crystals 50ppm (CAL)	Room temp.	0.1414g/1000mls 50ppm	nee	6 mos. or as needed, exp.
	ICV	Hach or equiv.	premade (ICV)	Room temp.	premade 50ppm		date on bottle
Alkalinity	ICV	Hach or equiv.	Sodium Bicarbonate	Room temp.	0.673g/1000mls	.Refrig. 4°C	6 mos.
TDS		Fisher or equiv.	KHP granules	Room temp.	0.3g/1000mls	.Refrig. 4°C	3 mos.
Bomb Calorimeter	Soils	Fisher or equiv.	Ethylene glycol	Flammable cabinet premade	N.A. premade	Flammable cabinet (room temp.)	as needed
	Liquids	Parr	Benzoic Acid	premade	.25 (1 tablet)	Room temp.	purchased as new
TKN Digestor	ICV	Hach	glycine p-toluenesulfonate granules	Room temp.	1.7652g/L 100ppm	Room temp.	6 mos. or as needed
Oxygen Meter	CAL	Fisher (CAL) or equiv.	glutamic acid and dextrose	Room temp.	.15g/L	Room temp.	once a month or as needed
	ICV	Fisher (ICV)	КНР	Room temp.	.15g/L		

Daily calibration standards are made from stock solutions at method specified concentrations. Exact dilutions and final concentrations may vary depending on the source available and the final concentrations required.

Table 9.3
STANDARDIZATION OF TITRATING SOLUTIONS

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Solution (N)	Source of Primary Standard	Frequency of Standardization	Methods Utilized
EDTA (0.2N)	CaCO ₃ NIST	Daily	130.2
NaOH (0.02N)	KHC ₈ H ₄ O ₄ NIST	As Needed (Each new soln.)	305.2
H ₂ SO ₄ (0.1N)	Na ₂ CO ₃ NIST	As Needed (Each new soln.)	375.4, 310.1, 350.2
HCl (0.1N)	Na ₂ CO ₃ NIST	As Needed (Each new soln.)	310.1
Phenylarsine Oxide (0.0375N)	KH(IO ₃) ₂ Fisher	Daily	345.1, 320.1
Na ₂ S ₂ O ₃ (0.75N)	KH(IO ₃) ₂ Fisher	Daily	320.1
Hg(NO ₃) ₂ (0.141N)	NaCl NIST	Daily	325.3
$Hg(NO_3)_2$ (0.025N)	NaCl NIST	Daily	325.3
$Hg(NO_3)_2$ (0.0141N)	NaCl NIST	Daily	325.3
Phenylarsine Oxide (0.00564N)	KH(IO ₃) ₂ Fisher	Daily	330.2
KI (0.1N)	PAO Red Bird Service	Daily	330.2
KI (0.0282N)	PAO Red Bird Service	Daily	330.2
Fe(NH ₄) ₂ (SO ₄) ₂ (N)	K ₂ Cr ₂ O ₇ NIST	Monthly	330.4
KCN (0.0192N)	AgNO ₃ Red Bird Service	Monthly	335.2
Na ₂ S ₂ O ₃ (0.0375N)	KH(IO ₃) ₂ Fisher	Daily	345.1
KI (0.0225N)	PAO, Na ₂ S ₂ O ₃	Daily	376.1
Methylene Blue	KI	Daily	376.2
$Fe(NH_4)_2(SO_4)_2(0.25N)$	K ₂ Cr ₂ O ₇ NIST	Daily	410.1, 440.3
Fe(NH ₄) ₂ (SO ₄) ₂ (0.025N)	K ₂ Cr ₂ O ₇ NIST	Daily	410.2
I ₂	Na ₂ S ₂ O ₃	Bi-Weekly	

All stock and standard solutions are to be labeled with:

- Contents and level of concentration
- Discrete identifying ID
- Solvent
- Date of preparation
- Manufacturer's Lot number
- Initials of preparer
- Expiration date
- Other information which may be appropriate
- This information must also be entered into the group's standards log book

9.4 Standardization of Titrating Solutions

The laboratory uses purchased pre-standardized reagents when possible for all those solutions and methods requiring standardization. Refer to Table 9.3, Standardization Of Titrating Solutions for solutions that require standardization, the source of the primary standards used to standardize, and the frequency of standardization. Standardization must be documented in some manner.

9.5 Calibration of Thermometers

All thermometers are labelled and calibrated with a NIST certified thermometer. A logbook documents this calibration. Thermometers are calibrated every two years or more frequently as needed.

9.6 Calibration of Balances

All analytical balances except top loading balances will be checked daily (or each day of use) using a "class S" certified weight. The QA Unit will perform a weekly 5 point check of all balances in the laboratory. Logbooks will be used to document this calibration.

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9.7 <u>Calibration of Eppendorfs/Pipettors</u>

All pipettors required to deliver precise volumes are calibrated weekly. This calibration is documented in a logbook which identifies the specific pipettor.

9.8 Instrument Performance Parameters - Start-up QC

Performance parameters are checked at the beginning of each analytical run and are used as a check to determine acceptable instrument performance. The operating conditions are documented in a bound log book.

9.8.1 Metals Analysis Instrument Performance Parameters

Performance parameters are checked at the beginning of each analytical run. The operating conditions are documented in a bound instrument log book, one for each instrument. The following information is recorded:

GFAA, FAA: Date, element, analyst, standard concentration,

standard absorbance, comments.

CVAA: Date, element, analyst, standard concentration,

standard absorbance, blank absorbance,

comments.

ICP: Date, analyst, emission "counts" for calibration

standard and blank for Ba, Cd and Sb, Ar flow,

Hg (or Cd or Mn) profile, comments.

Hydride AA: Date, element, analyst, standard concentration,

standard absorbance, blank absorbance,

comments.

The following minimum criteria must be met before proceeding with the analytical run, unless approval is obtained from the appropriate supervisor.

ICP: Emission count criteria for Mn is 30,000; data is

being accumulated for other elements. The Hg

(or Cd) profile must be ± 0.20 .

CVAA: The 3 ug/L calibration standard must be greater

than or equal to 0.080 abs. The blank must have

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an absorbance of less than 0.005. The calibration correlation coefficient must be at least 0.995.

FAA:

The highest calibration standard (designated in log book) must have an absorbance of at least 80% of the instrument manual value (Ba and Cr require 70%). This value is found on the chart by the instrument and in the instrument manual. The correlation coefficient must be at least 0.995.

GFAA:

A mid-range standard (designated in the log book; at the level of the analytical spike) must have an absorbance of at least the value in Table 9.4, GFAA Minimum Absorbance Criteria. The correlation coefficient must be at least 0.995.

Hydride AA:

The 3 ug/L calibration standard must be greater than or equal to 0.080 abs. The blank must have an absorbance of less than 0.005. The calibration correlation coefficient must be at least 0.995.

9.8.2 Mass Spectrometry Performance Parameters

Each system used for the analysis of volatile compounds must be tuned to meet the abundance criteria listed below for BFB.

	Method 624/8240	Method 524
Mass	Target Value	Target Value
50	15%-40%	15%-40%
75	30%-60%	30%-80%
95	100%	100%
96	5%-9%	5%-9%
173	< 2% of 174	< 2% of 174
174	> 50%	> 50%
175	5%-9% of 174	5%-9% of 174
176	95%-101% of 174	95%-101% of 174
177	5%-9% of 176	5%-9% of 176

Table 9.4
GFAA MINIMUM ABSORBANCE CRITERIA

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Element	Standard (ug/L)	Min. Abs.
As	10	0.035
Be	2.0	0.025
Cd	2.0	0.050
Cu	10	0.050
Cr	10	0.100
Pb	10	0.050
Ni	10	0.035
Sb	10	0.035
Ag	2.0	0.025
Se	10	0.025
TI	10	0.040

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These criteria cannot (routinely) be met using the AUTOTUNE file. It may be necessary to adjust the various lenses. The ratios of 69/219/502 should be around 100%/40%/4% for the MS's and 100%/25%-30%/2% on the MSD's. Run mass axis and width calibration and look at a profile scan before storing the tune and exiting the program.

The mass spectrometers are considered to be in tuning compliance for 12 hours after the injection (and passing the criteria) of the tuning compound for SW846 methods and for 8 hours for EPA 524. EPA method 624 has no time limitation on the tune or continuing calibration check, but sequences are not run for greater than 24 hours. Any analytical run which lasts longer than the method allows must include a tune check at or before the time period expires. Background correction may only be used to eliminate column bleed or instrument background ions. A hardcopy report of each tune check (spectrum and mass listing) is kept in the run folder. Background correction technique must be recorded.

Each system used for the analysis of semi-volatile compounds must be tuned to meet the abundance criteria listed below for DFTPP.

-	Method 625/8270	Method 525
Mass	Target Value	Target Value
51	30%-60% of 198	10%-80% of 198
68	< 2% of 69	< 2% of 69
70	< 2% of 69	< 2% of 69
127	40%-60% of 198	10%-80% of 198
197	< 1% of 198	< 2% of 198
198	100%, Base Peak Relative Abund.	100% or > 50% of 442
199	5%-9% of 198	5%-9% of 198
275	10%-30% of 198	10%-60% of 198
365	> 1% of 198	> 1% of 198
441	present and < 443	present and < 443
442	> 40% of 198	100% or > 50% of 442
443	17%-23% of 442	15%-24% of 442

These criteria cannot (routinely) be met using the AUTOTUNE file. It may be necessary to adjust the various lenses. Run mass axis and width calibration and look at a profile scan before storing the tune and exiting the program.

The mass spectrometers are considered to be in tuning compliance for 12 hours after the injection (and passing the criteria) of the tuning compound for SW846 methods and for 8 hours for EPA 525. EPA method 625 has no time limitation on the tune or continuing calibration check, but sequences are not run for greater than 24 hours (sample capacity limitation). Any analytical run which lasts longer than the method allows from the injection of DFTPP, must include a tune check at or before the time expires. Background correction may only be used to eliminate column bleed or instrument background ions. A hardcopy report of each tune check (spectrum and mass listing) is kept in the run folder. Background correction technique must be recorded in the log book.

9.8.3 Gas Chromatography and Liquid Chromatography (HPLC) Performance Parameters

Gas Chromatography (GC) and High Performance Liquid Chromatography (HPLC) techniques require three basic checks used to evaluate the system upon start-up.

- 1. Column efficiency
- 2. Resolution (resolution and column eff.)
- 3. Detector efficiency
- 9.8.3.1 Column efficiency is measured by an artificial concept called theoretical plates. In order to compare column efficiencies one must utilize identical conditions for the following:
 - 1. Solvent
 - 2. Solute
 - 3. Temperature

4. Flow Rate

5. Sample size

Broadening of the peaks (under identical instrument and method conditions, see above) indicates a decrease in the number of theoretical plates and therefore less desirable chromatography. A representative compound for the method being utilized will be chosen and a theoretical number of plates will be calculated from the calibration standard(s) run at the beginning of each sequence. The number of theoretical plates will be documented in a log book and will be used to assist in the determination of column acceptability.

Tangents are drawn to the peak at the points of inflection. The number of theoretical plates n, is given by:

$$\frac{n}{cm} = \frac{5.545 \left(\frac{RT}{W_h}\right)^2}{L}$$

Where:

RT = distance from injection to peak maximum in centimeters

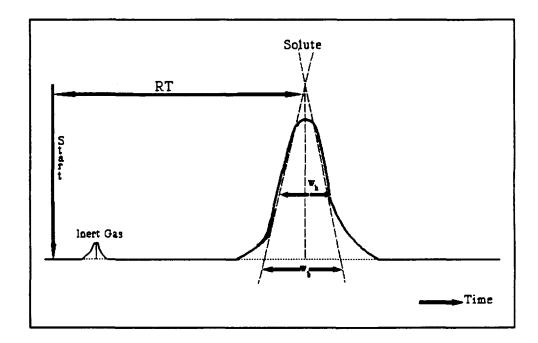
W_h = peak width at half height in centimeters

L = column length in centimeters

See Figure 9.1, Calculation of Theoretical Plates.

The peak gaussian factor (PGF) will be calculated when required as per the method specified analyte and must meet the method specified criteria. The PGF will be documented in a log book. The PGF calculation is required by some drinking water methods.

CALCULATION OF THEORETICAL PLATES



$$\frac{n}{cm} = \frac{5.545 \left(\frac{RT}{Wh}\right)^2}{I}$$

n = # theoretical plates

RT = distance from injection to peak maximum (retention time in centimeters)

W_h = peak width at half height L - column length in centimeters

or;

$$n = \frac{16\left(\frac{RT}{W_b}\right)^2}{L}$$

Where: $W_b = Width of peak at base$

$$PGF = \frac{1.83(W_k)}{W_{1/10}}$$

Where:

 W_h = peak width at half height $W_{1/10}$ = peak width at one-tenth height

9.8.3.2 Resolution is a measure of both the column and solvent efficiencies. It indicates both the separation of peak maxima (retention times) and the narrowness of peaks. Representative or method specified compounds will be utilized to calculate resolution at the beginning of the analytical run. The formula for resolution (R) is:

$$R = \frac{2d}{W_1 + W_2}$$

Where:

d = difference in elution times between two peaks

 W_1 = peak width for analyte 1. W_2 = peak width for analyte 2.

See Figure 9.2, Calculation of Resolution.

Retention time windows will be established for all GC methods each time a new column is installed. The concentration of each analyte will be such that the response is approximately half scale. The standard deviation (SD) from 3 injections shall be used to establish the retention time windows. For multiresponse analytes, one major peak will be chosen for the calculation.

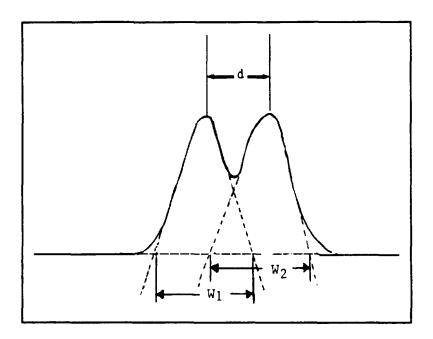
The performance criterion is:

Retention Time: $\pm 3(SD)$

For multi-response analytes the analyst should utilize the retention time window, but will primarily rely on pattern recognition.

9.8.3.3 Detector efficiency will be monitored from the analysis of method specified and/or representative compounds

CALCULATION OF RESOLUTION



$$R = \frac{2d}{W_1 + W_2}$$

Where

d = difference in elution times between two peaks.

W₁ = peak width for analyte 1. W₂ = peak width for analyte 2. Section 9.0 Revision 0 Nov. 1, 1991 Page 10 of 11

chosen such that their response will indicate the optimum minimum acceptable response from the detector. Many methods (i.e.GC/MS) specify compounds called System Performance Check Compounds (SPCC's). SPCC's must meet the minimum RF required by the method for both initial and continuing calibrations. Specifically, USEPA CLP lists minimum RF requirements for each analyte which must be met for both initial and continuing calibrations.

Methods that do not list minimum RF's (detectors other than mass spectrometers) will monitor a chosen analyte or analytes and compare that response to a mean RF (RF).

Minimum RF: ± 20% Difference of the established mean

9.8.4 Miscellaneous equipment and instrumentation performance requirements are listed below:

Analytical Balance : $\pm 0.5\%$ of class "S" weight true value

Spectrophotometer : \pm 5% of NIST or other standard glass filters

intensity

Ovens : \pm 3°C of set point

Furnaces : ± 20°C of set point

Conductivity Meter: ± 1% of cell constant's established mean

Refrigerators/Coolers: \pm 2°C of set point

Freezers : \pm 3°C of set point

Incubators : ± 1°C of set point

D.I. Water System : Conductivity $\leq 1.0 \,\mu\text{mhos/cm}$ (or 1.0 M Ω ·cm)

Zero Headspace Extractors: Maintain 50 PSI for 1 hour

EP, TCLP Extractors: ± 2 RPMs of specified RPM

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9.9 <u>Instrument Calibration Criteria</u>

Instrument Calibration frequency and criteria are detailed in Table 9.5, Instrument Calibration. All linear calibration curves will be fitted by the least squares linear regression technique. This procedure will also be used for calibration curves that require log transforms or similar transforms. The formula for calculation of the regression line is given in Section 12.0, Data Reduction, Validation and Reporting.

Each item of equipment shall, when appropriate, be labeled, marked or otherwise identified to indicate its calibration status, except for those items of equipment which are calibrated more frequently than every 6 months.

Table 9.5 INSTRUMENT CALIBRATION

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Instrument	Method Reference	Standards Initial Calibration	Acceptance/Rejection Criteria - Initial Calibration	Frequency of Calibration	Prequency of Initial Calibration Verification ¹	Acceptance/ Rejection Criteria - Initial Calibration Verification	Frequency of Continuing Calibration Verification ¹	Acceptance/ Rejection Criteria - Continuing Calibration Verification
FAA	SW-846	4	Correlation coefficient must be	At least	Every	90-110%R	Every 10	90-110%R
	EPA600/4-79/080	4	≥ 0.995	daily, or as required	calibration	90-110%R	analytical samples	90-110%R
	CLP	4	-	(when CCV fails		90-110%R		90-110%R
CVAA	SW-846	4		acceptance criteria)		80-1 20% R		80-120%R
	EPA600/4-79/080	4		,		80-120%R	_	80-120%R
	CLP	4]			80-120%R		80-120%R
ICP	SW-846	1				90-110%R		90-110%R
	EPA600/4-79/080	1			,	90-110%R		90-110%R
	CLP	1				90-110%R		90-110%R
GFAA	SW-846	4				85-115%R	i	85-115%R
	EPA600/4-79/080	4				85-115%R		85-115%R
	CLP	4				90-110%R		90-110%R
pH Meter	SW-846	3	±0.1 STD units of true value			± 0.1 STD units of true value		± 0.1 STD unit
ļ	CLP	3						

Table 9.5 INSTRUMENT CALIBRATION

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Instrument	Method Reference	Standards Initial Calibration	Acceptance/ Rejection Criteria - Initial Calibration	Frequency of Calibration	Frequency of Initial Calibration Verification ¹	Acceptance/ Rejection Criteria - Initial Calibration Verification	Frequency of Continuing Calibration Verification ¹	Acceptance/ Rejection Criteria - Continuing Calibration Verification
GC/MS- volatiles	SW-846 (8240,8260)	5	%RSD<30% (CCC) 1,1-dichloroethene; chloroform 1,2-dichloropropene; toluene ethyl benzene; vinyl chloride RF>0.30(SPCC) chloromethane; 1,1-dichloroethane; bromoform (0.25); 1,1,2,2-tetrachloroethene; chlorobenzene	As needed	As needed	± 20%	daily 12 hr.	CCC %D <25% same SPCC criteria as initial calibration
	40CFR136, 624	5	all cmpds %RSD <35% or use calibration curve	As needed	As needed	± 20%R	daily 24 hr.	Compare w/Table 9.5 "Q" (attached)
	CLP SOW 2/88	5	same as SW846	As needed	As needed, usually w/PE's	± 20%R	daily 12 hr.	same as SW-846

Table 9.5 INSTRUMENT CALIBRATION

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Instrument	Method Reference	Standards Initial Calibration	Acceptance/ Rejection Criteria - Initial Calibration	Prequency of Calibration	Frequency of Initial Calibration Verification ¹	Acceptance/ Rejection Criteria - Initial Calibration Verification	Frequency of Continuing Calibration Verification ¹	Acceptance/ Rejection Criteria - Continuing Calibration Verification
GC/MS-volatiles	CLP SOW OLMO1.5	5	min RF Bromoform 0.10 Vinyl Chloride 0.10 1,1-dichloroethene 0.20 Chloroform 0.20 1,2-dichloroethane 0.10 1,1,1-trichloroethane 0.10 1,1,1-trichloroethane 0.10 carbon tetrachloride 0.10 bromodichloromethane 0.20 cis-1,3-dichloropropene 0.20 trichloroethene 0.30 dibromochloromethane 0.10 1,1,2-trichloroethane 0.10 benzene 0.50 trans-1,3-dichloropropene 0.10 bromoform 0.10 betrachloroethene 0.20 1,1,2,2-tetrachloroethane 0.50 toluene 0.40 chlorobenzene 0.50 ethylbenzene 0.50 ethylbenzene 0.10 styrene 0.30 xylene (total) 0.30 bromofluorobenzene 0.20 all % RSD <20.5 Other target compounds must meet minimum RF of 0.10 No %RSD criteria	As needed	As needed, usually w/PE's	± 20%R	Daily every 12 hours	RF criteria same as initial cal. %D <25.0
	EPA 524.2	5	% RSD <20% or use cal curve - all target compounds	As needed	As needed	± 20%R	Daily, every 8 hours	All compounds RF%D <30% ISTD areas >30%, <150% of initial cal.

Table 9.5 INSTRUMENT CALIBRATION

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Instrument	Method Reference	# Standards Initial Calibration	Acceptance/ Rejection Criteria - Initial Calibration	Frequency of Calibration	Frequency of Initial Calibration Verification ¹	Acceptance/ Rejection Criteria - Initial Calibration Verification	Frequency of Continuing Calibration Verification ¹	Acceptance/ Rejection Criteria - Continuing Calibration Verification
GC/MS - semi- volatiles	SW846-8270	5	%RSD < 30% (CCC) acenaphthene 1,4-dichlorobenzene hexachlorobenzene hexachlorobutadiene N-nitroso-diphenylamine di-octylphthalate fluoranthene benzo(a)pyrene 4-chloro-3-methylphenol 2,4-dichlorophenol phenol pentachlorophenol 2,4,6-trichlorophenol RF > 0.05(SPCC) N-nitrosodipropylamine hexachlorocyclopentadiene 2,4-dinitrophenol 4-nitrophenol	As needed	As needed	± 20%R	Daily, every 12 hours	CCC % D < 25% same SPCC criteria as initial cal.
	40CFR136 625	5	%RSD < 35% or cal. curve all compounds	As needed	As needed	± 20%R	Daily every 24 hours	% D < 20%
	CLP SOW 2/88	5	Same as SW846-8270	As needed	As needed w/PE's	± 20%R	Daily every 12 hours	Same as SW846- 8270

Table 9.5 INSTRUMENT CALIBRATION

Instrument	Method Reférence	# Standards Initial Calibration	Acceptance/ Rejection Criteria - Initial Calibration	Frequency of Calibration	Frequency of Initial Calibration Verification ¹	Acceptance/ Rejection Criteria - Initial Calibration Verification	Frequency of Continuing Calibration Verification ¹	Acceptance/ Rejection Criteria - Continuing Calibration Verification
GC/MS - semi-volatiles	CLP SOW OLMO1.5	5	min. RF phenol 0.80 bis(2-chloroethyl)ether 0.70 2-chlorophenol 0.80 1,3-dichlorobenzene 0.60 1,4-dichlorobenzene 0.50 1,2-dichlorobenzene 0.40 2-methylphenol 0.70 4-methylphenol 0.60 N-nitrosodipropylamine 0.50 hexachloroethane 0.30 nitrobenzene 0.20 isophorone 0.40 2-nitrophenol 0.10 2,4-dimethylphenol 0.20 bis(2-chloroethoxy)methane0.30 2,4-dichlorophenol 0.20 1,2,4-trichlorobenzene 0.20 naphthalene 0.70 4-chloro-3-methylphenol 0.20 2,4,5-trichlorophenol 0.20 2,4,5-trichlorophenol 0.20 2,4,5-trichlorophenol 0.20 2,-thoronaphthalene 0.80 acenaphthylene 1.30 2,6-dinitrotoluene 0.20 acenaphthene 0.80 dibenzofuran 0.80 dibenzofuran 0.80 2,4-dinitrotoluene 0.20 4-chlorophenylphenylether 0.40 fluorene 0.90 4-bromophenylphenylether 0.10 hexachlorobenzene 0.80					%D < 25 RF criteria same as initial calibration

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Table 9.5
INSTRUMENT CALIBRATION

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Instrument	Method Reference	Fishedards Initial Calibration	Acceptance/ Rejection Crit - Initial Calibration	keria	Prequency of Calibration	Frequency of Initial Calibration Verification ¹	Acceptance/ Rejection Criteria - Initial Calibration Verification	Prequency of Continuing Calibration Verification ¹	Acceptance/ Rejection Criteria - Continuing Calibration Verification
GC/MS - semi- volatiles	CLP SOW OLMO1.5		phenanthrene anthracene fluoranthene pyrene benz(a)anthracene chrysene benzo(b)fluoranthene benzo(k)fluoranthene benzo(a)pyrene indeno(1,2,3,cd)pyrene dibenz(a,h)anthracene benzo(ghi)perylene nitrobenzene d5 2-fluorobiphenyl terphenyl-d ₁₄ phenol-d ₃ 2-fluorophenol 2-chlorophenol-d ₄						
	EPA525	6	%RSD < 30% all compoun Chromatographic separation isomers		As needed	As needed	± 20%R	daily, every eight hours	RF %D < 30% 1STD areas > 30% < 150% from initial cal.

Table 9.5 INSTRUMENT CALIBRATION

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Instrument	Method Reference	# Standards Initial Calibration	Acceptance/Rejection Criteria - Initial Calibration	Frequency of Calibration	Frequency of Initial Calibration Verification ¹	Acceptance/ Rejection Criteria - Initial Calibration Verification	Frequency of Continuing Calibration Verification ¹	Acceptance/ Rejection Criteria - Continuing Calibration Verification
Pesticides E	N-P containing pesticides EPA 507	3	RF < 20% RSD or single point (single point must be within 20% of sample concentration)	As needed when CCV > 20% diff., upon detection of	quarterly	20%D	2 times daily, beginning and end of day	20%D
	Organophosphorus pesticides SW-846 8141	5	RF < 20% RSD or cal. curve	analyte after running low level single point to demonstrate detectability ²	quarterly	15%D	Daily	15%D
	Simetryn & Terbutryn EPA 619	3	RF < 10% RSD or cal. curve	Daily	As needed and with the prep of new std.	10%D	Each working shift	10%D
<u>!</u>	Nitrosamines EPA 607	3	RF < 10% RSD or cal. curve	Daily	As needed and with the prep of new std.	15%D	Each working day	15%D
GC/FID	SW-846 8015	5	RF < 20% RSD or cal. curve	As needed, when CCV > 15%D	Quarterly	15%D	Daily	15%D
	SW-846 8100	5	RF < 20% RSD or cal. curve	With each analytical sequence	As needed, with prep of new std.	15%D	Daily	15%D
	SW-846 8030	5	RF < 20% RSD or cal. curve	As needed when CCV > 15% D	As needed with prep of new standard	15%D	Daily, 10%, ending	15%D

Table 9.5 INSTRUMENT CALIBRATION

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Instrument	Method Reference	# Standards Initial Calibration	Acceptance/ Rejection Criteria - Initial Calibration	Prequency of Calibration	Prequency of Initial Calibration Verification	Acceptance/ Rejection Criteria - Initial Calibration Verification	Prequency of Continuing Calibration Verification ¹	Acceptance/ Rejection Criteria - Continuing Calibration Verification
HPLC	EPA 531.1	3-5	RF < 20% RSD or single point or calibration curve	As needed, when CCV > 20%D	Quarterly	20 % D	Min. of 2 1 beg, 1 end	20%D
	SW-846 8310	5	RF < 20% RSD or cal. curve	As needed, when CCV > 15%D or every 6 months	As needed, with prep of new std.	15%D	Daily, 10%	15%D
	EPA 610	3	RF < 10% RSD or cal. curve	When CCV > 15%D	As needed, with prep of new std.	15%D CCV vs. cal. curve	Daily 10%	15%D
GC-PID/ ELCD	EPA 502.2	3-5	RF < 10% RSD or cal. curve or single point cal.	When CCV > 20%D	As needed, with prep of new std. or quarterly	20%D	Daily	20%D
	EPA 601	3	RF < 10% RSD or cal. curve	As needed, when ICV or CCV > Table 2 criteria	As needed, with prep of new std.	See method 601 Table 2 criteria ~ 30%D (Q Value)	Daily Note: ICV = CCV in this case (different source than calibration stds.)	For % Rec. see method 601 Table 2 (Q Value)
	EPA 602	3	RF < 10% RSD or cal. curve	As needed, when ICV or CCV > Table 2 criteria		See method 602 Table 2 Criteria ~ 25%D (Q Value)		For % Rec. see method Table 2 (Q Value)
	SW-846 8010	5	RF < 20% RSD or cal. curve	As needed,	As needed,	15%D	Daily 10%,	15%D
	SW-846 8020			when CCV > 15%D	with prep of new std.	15%D	ending	15%D

Instrument	Method Reference	# Standards Initial Calibration	Acceptance/ Rejection Criteria - Initial Calibration	Frequency of Calibration	Frequency of Initial Calibration Verification ¹	Acceptance/ Rejection Criteria - Initial Calibration Verification	Frequency of Continuing Calibration Verification ¹	Acceptance/ Rejection Criteria - Continuing Calibration Verification
GC-PID/ ELCD	SW-846 8021	5	RF < 20% RSD or cal. curve	As needed, when CCV > 15%D	As needed with prep of new std.	15%D	Daily 10%, ending	15%D
FΠR	EPA 418.1	5	20%D Correlation Coeff. (r) ≥ 0.995	When CCV is > 20%D	As needed, with prep of new std.	20%D	Beg. and end of each sequence	20%D
	Standard Methods 503	5	20%D Correlation Coeff. (r) ≥ 0.995	When CCV is > 20%D	As needed, with prep of new std.	20%D	Beg. and end of each sequence	20%D
GC-ECD	EPA 548.1 (Endothall)	3	Linearity < 20% RSD	Each Run	As needed with each new std. quarterly at a minimum	80-110%	Every lifth injection	Primary column %D <15. Conf. column %D <20. R.T. Shift, Capp. columns <0.3%. RT Shift Mega-Bore Columns <1.5%
	CLP-SOW 2/88	3	Linearity <20% RSD Generate calibration curve for all single analytes detected in samples where the % RSD ≥ 10% Retention time windows: Wide Bore capp. column: ± 0.75% Narrow Bore Capp. column: ± 0.15%	Each run or every 72 hours	As needed with each new std. quarterly at a minimum	80-110%	Every fifth injection	Primary column %D <15. Conf. column %D <20. R.T. Shift, Capp. columns <0.3%. RT Shift Mega-Bore Columns <1.5% Breakdown criteria: DDT <20% Endrin <20% Combined <30%

Table 9.5 INSTRUMENT CALIBRATION

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Instrument	Method Reference	Standards Initial Calibration	Acceptance/ Rejection Criteria - Initial Calibration	Prequency of Calibration	Frequency of Initial Calibration Verification	Acceptance/ Rejection Criteria - Initial Calibration Verification	Frequency of Continuing Calibration Verification ¹	Acceptance/ Rejection Criteria - Continuing Calibration Verification
GC-ECD	EPA 508	3	Linearity <20% RSD	Each Run	As needed. With each new std. Quarterly at a minimum.	80-110 % R	Every fifth injection	Primary column %D <15. Conf. column %D <20. R.T. Shift, Capp. columns <0.3%. RT Shift Mega- Bore Columns <1.5% Breakdown criteria: DDT <20% Endrin <20%
	EPA 504	5	Linearity <20% RSD	Each Run	As needed. With each new std. Quarterly at a minimum	80-110%R	Every lifth injection	Primary column %D <15. Conf. column %D <20. R.T. Shift, Capp. columns <0.3%. RT Shift Mega- Bore Columns <1.5%
	APHA 509A (Standard Methods)	3	Linearity <20% RSD	Each Run	As needed. With each new std. Quarterly at a minimum	80-110%R	Every fifth injection	Primary column %D <15. Conf. column %D <20. R.T. Shift, Capp. columns <0.3%. RT Shift Mega-Bore Columns <1.5% Breakdown criteria: DDT <20% Endrin <20% Combined <30%

Table 9.5 INSTRUMENT CALIBRATION

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Instrument	Method Reference	Standards Initial Calibration	Acceptance/ Rejection Criteria - Initial Calibration	Frequency of Calibration	Prequency of Initial Calibration Verification ¹	Acceptance/ Rejection Criteria - Initial Calibration Verification	Prequency of Continuing Calibration Verification ¹	Acceptance/ Rejection Criteria - Continuing Calibration Verification
GC-ECD	EPA 608	3	Linearity <20% RSD	Each Run	As needed. With each new std. Quarterly at a minimum	80-110%R	Every fifth injection	Primary column %D <15. Conf. column %D <20. R.T. Shift, Capp. columns <0.3%. RT Shift Mega- Bore Columns <1.5% Breakdown criteria: DDT <20% Endrin <20% Combined <30%
	SW-846 8080 SW-846 8150	5	Linearity <20% RSD	Each Run	As needed. With each new std. Quarterly at a minimum	80-110%R	Every fifth injection	Primary column %D <15. Conf. column %D <20. R.T. Shift, Capp. columns <0.3%. RT Shift Mega-Bore Columns <1.5% Breakdown criteria: DDT <20% Endrin <20% Combined <30%

Table 9.5 INSTRUMENT CALIBRATION

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Instrument	Method Reference	Standards Initial Calibration	Acceptance/ Rejection Criteria - Initial Calibration	Frequency of Calibration	Frequency of Initial Calibration Verification ¹	Acceptance/ Rejection Criteria - Initial Calibration Verification	Frequency of Continuing Calibration Verification ¹	Acceptance/ Rejection Criteria - Continuing Calibration Verification
GC-ECD	EPA 515.1	3	Linearity <20% RSD	Each Run	As needed. With each new std. Quarterly at a minimum	80-110%R	Every fifth injection and beginning and end of run.	Primary column %D <15. Conf. column %D <20. R.T. Shift, Capp. columns <0.3%. RT Shift Mega- Bore Columns <1.5%
	EPA OLM01.3	3+Instr. Blank Multi-Comp. Targets Calib. as single point	All peaks 100% resolved. Performance evaluation mixtures (PEMs) ≤ 25.0 RPD. 1 Chromatogram from each of 2 indiv. A&B must yield peak highs of 50-100% of full scale. Resolution of midpoint std. mixes A&B ≥ 90% linearity ≤ 20% RSD except: Surrogates ≤ 30% Any 2 targets ≤ 30% Resolution check mix ≥ 60% Breakdown of DDT & Endrin ≤20%, Combined < 30%	Each Run	As needed. With each new std. Quarterly at a minimum	80-110%R	Every 12 hours (PEM or indiv. A&B)	PEMs and Indiv. A&B within RT windows of init. calibration. PEMs RPD ≤ 25.0. Resolution of PEM must be 100%. Resolution of indiv. A&B ≥ 90% Breakdown of DDT & Endrin ≤ 20% Combined ≤ 30%

Number of Standards Run is 1, unless noted otherwise

Only when an unusually large analyte list requires analysis of more than one standard mix for injection by GC/NPD.

Table 9.5 - Attachment GC/MS - Volatiles Continuing Calibration Check - EPA Method 624

	Range for "Q" in ug/L
Benzene	12.8-27.2
Bromoform	14.2-25.8
Carbon tetrachloride	14.6-25.4
Chlorobenzene	13.2-26.8
Chloroethane	7.6-32.4
2-Chloroethylvinyl-ether	D-44.8
Chloroform	13.5-26.5
Dibromochloromethane	13.5-26.5
Bromodichloromethane	13.1-26.9
1,4-Dichlorobenzene	12.6-27.4
1,1-Dichloroethane	14.5-25.5
1,2-Dichloroethane	13.6-26.4
1,1-Dichloroethene	10.1-29.9
1,2-Dichloropropane	6.8-33.2
trans-1,3-Dichloropropene	10.0-30.0
Ethylbenzene	11.8-28.2
Bromomethane	2.8-37.2
Chloromethane	D-40.8
Methylene Chloride	12.1-27.9
1,1,2,2-Tetrachloroethane	12.1-27.9
Tetrachloroethene	14.7-25.3
Toluene	14.9-25.1
trans-1,2-Dichloroethene	13.9-26.1
1,1,1-Trichloroethane	15.0-25.0
1,1,2-Trichloroethane	14.2-25.8
Trichloroethene	13.3-26.7
Trichlorofluoromethane	9.6-30.4
Vinyl Chloride	0.8-39.2



10.0 Preventive Maintenance

10.1 Wherever practical, maintenance agreements have been executed with instrument manufacturers. Such is the case with the following major lab instrumentation:

GC Equipment

GC/MS Equipment

HPLC Equipment

FTIR Equipment

Atomic Spectroscopy Equipment

Lab Data Management Systems

For those instruments for which no maintenance agreements are available, an attempt has been made to maintain at least 2 functional instruments (eg. TOX, pH Meters, EP Extractors, DO Meters, Conductivity Meters, Balances, etc.). If adequate in-house expertise exists, maintenance agreements may be cancelled upon approval of the Lab Director.

- 10.2 A log for each instrument is maintained by the appropriate group leader and/or analyst. Information maintained in this log must include, but is not necessarily limited to:
 - 1. Name of the item or equipment
 - 2. Manufacturer's name, type identification, and serial number or other unique identification.
 - 3. Date received and date placed in service.
 - 4. Current location, where appropriate.
 - 5. Condition when received (e.g., new, used, reconditioned).
 - 6. A daily "response check" or "sensitivity check" (curve verification data).
 - 7. A record of problems and their solutions.

- 8. A record of maintenance and preventive maintenance service calls noting what was done each time.
- 9. Initials of analyst and dates.
- 10.3 Group leaders are <u>required</u> to maintain a supply of items critical to the performance of their instruments (eg. electron multipliers for GC/MS, D₂ lamps for AA, key circuit boards). Blanket purchase authority is given to group leaders for any item valued at less than \$1,000.00 for each item to a maximum of \$5,000.00 per group; the Lab Director must approve more costly items and/or higher maximums per group.
- 10.4 Logbooks are also maintained for all equipment/instrumentation monitoring activities, such as:
 - 1. Balances accuracy checks
 - 2. Ovens temperature monitoring
 - 3. Refrigerators temperature monitoring
 - 4. Incubators temperature monitoring
 - 5. Deionized Water Systems resistivity conductivity checks
 - 6. Zero Headspace Extractors leak tests
 - 7. Hot Plates surface temperature monitoring
 - 8. Fume Hoods air flow
 - 9. Extractors (EP, TCLP) RPM checks
 - 10. Gas Tanks changes
- 10.5 Table 10.1, Routine Preventive Maintenance provides the following information:
 - 1. Instrument/Equipment type
 - 2. Specific activity
 - 3. Frequency

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All routine maintenance is documented in a logbook specific to the instrument or equipment. Service calls and all repairs will also be documented in this same logbook.

To minimize the impact of major equipment failures, sufficient equipment redundancy has been established within each laboratory and throughout the network. Where there is no such redundancy, samples are shipped to other EMS locations for analysis (certifications and contracts permitting). Flame atomic absorption may be a suitable replacement technique for ICP.

Preventive maintenance procedures to be employed are described in detail in instrument manuals.

Table 10.1 ROUTINE PREVENTIVE MAINTENANCE

Instrument/Equipment Type	Specific Activity	Frequency
Gas Chromatography/Mass Spectrometry	Changing transfer lines	As needed
(Volatiles)	Changing septa	Every 20-30 injections
	Clean source	As needed
	Change trap (purge & trap)	As needed
	Change column	Every 3-4 months or as needed
	Flush sample lines with methanol (P & Trap)	As needed
	Replace gas filters and traps	Every other carrier gas cylinder
Gas Chromatography/Mass Spectrometry	Change injection liner	Daily or each day of use
(Semi-Volatiles)	Change septum	Daily or each day of use
	Trim head of column	Daily or each day of use
	Bake the source	When response to DFTPP initial standards decrease below acceptable levels
	Clean the source	When baking fails
	Replace gas filters and traps	Every other carrier gas cylinder
	Check syringe/replace	Daily or each day of use
Gas Chromatography/ECD	Change port liners	Twice per week or after 3 days use
	Clean detectors	As needed
	Remove front 6" of column	As needed
Gas Chromatography/ECD	Change septa	Twice per week or after 3 days use

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^{&#}x27;or when criteria can't be met

Table 10.1
ROUTINE PREVENTIVE MAINTENANCE

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Instrument/Equipment Type	Specific Activity	Frequency
Gas Chromatography/ECD	Replace gas filters and traps	Every other carrier gas cylinder
	Check syringe/replace	Daily (minimum) or each day of use
	ECD leak test (swipe)	Every 6 months
Gas Chromatography/PID, ELCD, NPD,	Change port liners	As needed
FPD, FID	Clean detectors	As needed
	Remove front 6" of column	As needed
	Change septa	As needed
	Replace gas filters and traps	Every other carrier gas cylinder
	Check syringe/replace	As needed
	Change/Add Solvent (ELCD)	As needed or as per Mfg'r Guidelines
	Clean window (PID, FPD)	As needed
	Change resin (ELCD)	As needed or as per Mfg'rs Guidelines
	Change trap (purge and trap)	As needed
High Performance Liquid Chromatograph	Change seals	As needed
(HPLC)	Replace pre-column	As needed
	Change lamps	As needed
Fourier Transform-Infra-Red (FTIR)	Purge System w/Nitrogen	Each use

Table 10.1 ROUTINE PREVENTIVE MAINTENANCE

Instrument/Equipment Type	Specific Activity	Frequency
Flame Atomic Absorption	Check gas supply pressure	Daily or each day of use
	Empty drain container	Daily or each day of use
	Inspect capillary tube	Daily or each day of use
	Clean burner heads (acid soak)	Daily or each day of use
	Aspirate DI water for 15 minutes to flush nebulizer and premix chamber	Daily or each day of use
	Check nebulizer uptake rate	Weekly
	Clean premix chamber	Weekly
	Inspect windows	Weekly
	Inspect glass bead	Monthly
	Inspect O-rings	Monthly
	Inspect nebulizer	Monthly
	Inspect frangible diaphragm (where applicable)	Monthly
Graphite Furnace Atomic Absorption	Check Argon supply pressure	Daily or each day of use
	Inspect furnace windows	Daily or each day of use
	Inspect graphite tube and platform	Daily or each day of use
	Check rinse solution and drain container	Daily or each day of use
	Inspect autosampler tip alignment and condition	Daily or each day of use
	Inspect contact cylinders	Weekly

Table 10.1 ROUTINE PREVENTIVE MAINTENANCE

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Instrument/Equipment Type	Specific Activity	Frequency
	Clean and refill ∞oling water recirculators	Quarterly
Graphite Furnace Atomic Absorption	Check exhaust system	Weekly
	Clean computer cooling fans	Yearly
Inductively Coupled Plasma (ICP)	Check argon supply pressure	Daily or each day of use
	Clean nebulizer	Daily or each day of use
	Change peristaltic pump tubing	Weekly
	Clean entrance slit	Monthly
	Change torch	Quarterly
	Inspect focusing mirror (simultaneous ICP)	Quarterly
	Check oil in vacuum pump (simultaneous ICP)	Twice per year
	Clean filters on instrument	Yearly
Total Organic Carbon (TOC)	Change catalyst	Bi-monthly or as needed
	Re-charge acid	Weekly
Total Organic Halogens (TOX)	Clean inlet tube	Weekly
	Clean filtration cell and electrodes	Each use
	Recondition filtration cell reference sidearm	Monthly
	Clean pyrolysis and quartz exit tubes	Each use
Balances	Check with an external class "S" weight	Each use or daily
	Change desiccants	Weekly

Table 10.1 ROUTINE PREVENTIVE MAINTENANCE

Instrument/Equipment Type	Specific Activity	Frequency
Balances	Clean pans and compartments	Each use
	Check alignment (leveling)	Each use
	Service Engineer visit	Every 6 months
Auto Analyzer	Change tubing	Quarterly
	Clean tubing w/10% HCl	Monthly
	Oil pump	Monthly
	Change Cadmium reduction column	Quarterly
Gas Flow Proportional Counter (Gross Alpha	Change gases	As needed
Beta)	Change gas line filters	Every other carrier gas cylinder
Spectrophotometer	Check wavelength with standard filter	Quarterly
Turbidimer	Clean optics	As needed
Conductivity Meter	Clean electrodes	As needed
Dissolved Oxygen Probe	Replace membrane	As needed
Recorder	Dust and clean	As needed
	Clean the carriage rod for smooth operation	As needed
	Lubricate the servomotor with two drops of light oil	Quarterly
Deionized Water System	Conductivity Check, in-line meter	Daily
	Ion exchange beds replaced	As needed
	Conductivity check by lab S.C. meter	Weekly (Daily if no in-line meter)

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Table 10.1 ROUTINE PREVENTIVE MAINTENANCE

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Instrument/Equipment Type	Specific Activity	Frequency
Milli-Q D.I. Water System	Conductivity Check, in-line meter	Daily
	Replace cartridges	As needed
	Conductivity check with lab S.C. meter	Weekly
Oven	Check temperature	Each use
	Oil Motor	Quarterly
Refrigerators/Coolers/Freezers	check temperature	Daily
	Clean cooling coils	Annually
Vacuum pumps & air compressors	Lubricate	As needed
	Check belts, etc.	As needed
Zero Headspace Extractors	Leak test	Each use
Rotary Extractors (TCLP, EP)	RPM check	Daily or each day of use
Hot Plates	Surface temperature checks	Semi-annually
Incubators	Temperature monitoring	Daily or each day of use
Fume Hoods	Check air flow	Monthly
	Replace charcoal filters if applicable	As needed

- 11.0 Quality Control Checks, Routines to Assess Precision and Accuracy and Calculation of Method Detection Limits
 - 11.1 Quality Control Checks Field QC

The laboratory has little to no control over what type of Field QC is collected and submitted for analysis by clients, unless the sampling is performed by EMS Heritage and the client agrees to the increased costs associated with Field QC sample analysis. The implementation of proposed (January 23, 1989) mandatory field and laboratory QC requirements by EPA in the RCRA program will further the EPA goal of providing more reliable analytical data and help assure that this Quality Assurance Plan (QAP) is followed for Field QC sample collection and submission. The proposed mandatory Field QC samples are as follows:

- 1. Field Duplicate
- 2. Field Blank
- 3. Equipment Blank
- 4. Trip Blank

The following types of QC checks and the stated frequency are the basic minimum requirements of this QAP. Deviations from these minimum requirements (i.e. less than 5 samples in a set, weekly or daily monitoring, etc.) are the responsibility and decision of the client and shall be proposed in a site specific QC plan (if required) with appropriate justification. See Appendix D for definition of all terms and QC types.

- 11.1.1 Sampling events with 10 or more samples of a similar matrix will require the following Field QC:
 - 1. Equipment Blanks: At lease one equipment blank on clean sampling equipment will be submitted and analyzed for every 20 samples in each analyte group for water matrices. This blank will be prepared in the field before sampling begins by filling or rinsing the precleaned equipment with analyte free water, filling the appropriate container(s) and preserving and documenting the sample in the same manner as the other collected samples. Suitable blanks for analyte groups of interest

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will be collected and analyzed for each type of equipment set (i.e., bailer, filtration system; or cover, stainless steel pan and spatula, etc.) to be used in sampling. EMS Heritage does not believe it is appropriate to use an aqueous equipment blank for non-aqueous (solid) matrices because different analytical procedures are utilized which have different detection levels at unnecessary expense to the client. The USEPA does not generally require collection of field blanks for solid matrices, however, if their collection is required the use of acid washed beach sand is a recommended blank matrix in place of the rinsing with analyte free water cited above. The use of this solid matrix will be interference free, will generally allow the use of identical methods and will provide similar if not lower detection limits.

If equipment is cleaned on site, then additional equipment blanks will be collected and analyzed for each equipment group at a rate of one blank or 5% of the equipment sets that must be cleaned, whichever is greater. These blanks will be collected and analyzed as described in the preceding paragraph.

If no equipment is involved in sampling (i.e., grab samples using sample container) or passage of blank matrix through the sampling device is not possible (i.e., dedicated well pumps) then a Field Blank will be collected in place of an Equipment Blank.

- 2. <u>Trip Blanks</u>: At least one trip blank for each proposed volatile organic method (601, 624, 8020, 8021, 8240, etc.) shall be prepared and analyzed for each cooler to be used for storage and transport of volatile samples. This type of blank will require the submission of two or more 40mL vials per blank. Trip blanks will generally only be collected for water samples. Trip blanks are prepared in the laboratory, placed in the cooler before shipment to the field and are returned unopened to the lab.
- 3. <u>Field Duplicate</u>: During each independent sampling event, at least one sample or 10% of the samples, whichever is greater, shall be collected in duplicate for analysis. This requirement shall apply to each and all parameter groups and matrices that are sampled.

- 11.1.2 Sampling events involving 5 to 10 samples of a similar matrix will require the following Field QC:
 - 1. Equipment Blanks: If equipment is cleaned in the field, one equipment blank for each parameter group shall be collected and analyzed on the field decontaminated equipment. If no equipment is cleaned, then one equipment blank that is prepared on-site on the precleaned equipment shall be collected and analyzed for each parameter group. If no equipment is involved in sampling (i.e. grab sample using sample container) or passage of blank matrix through the sampling device is not possible (i.e., dedicated well pumps) then a Field Blank will be collected in place of an Equipment Blank.
 - 2. <u>Field Duplicates</u>: One field duplicate shall be collected and analyzed for all parameter groups and matrices.
- 11.1.3 Sampling events involving less than 5 samples of a similar matrix will require the following field QC:
 - 1. One equipment blank on either precleaned or field decontaminated equipment (see above) will be collected and analyzed for each parameter group.
- 11.1.4 Additional QC types and/or submittal frequency of field quality control blanks may be required and are dependent on the methods of analysis and the Data Quality Objectives of a specific project and/or mandatory QC requirements. An example would be the collection and analysis of field blanks (see Appendix D for definition) in addition to the minimum requirements specified above.
- 11.1.5 QC checks on field measurements will require at least one duplicate sample analysis for every 10 field measurements. Field QC samples are treated as any other sample for analysis.

11.2 Quality Control Checks - Laboratory QC

EMS Heritage will follow the minimum quality control requirements specified by each method. Table 11.5, <u>Analytical Run Requirement/Frequency</u> mandates the specific requirements that will be used. Those minimum requirements may be exceeded for specific methods. Please refer to

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Appendix D, <u>Definitions and Acronyms</u> for definitions and acronyms used by EMS Heritage for QC types.

- 11.2.1 If no quality control requirements are listed in the method, or if the method quality control requirements are less stringent that those listed below, the laboratory will follow the general guidelines listed below:
 - 1. Method Reagent Blank, or Prep Blank (BLA02) Prepared and analyzed at a rate of one per sample prep set (see definitions in Appendix D for sample set, etc.)
 - 2. <u>Matrix Spike (SPI02)</u> Prepared and analyzed at a minimum of one sample in a sample set (or 5%, whichever is greater) with similar matrices for a specified method. If a set contains samples of different matrices, matrix spikes should be prepared and analyzed for each matrix type by the specified method.
 - 3. Reagent Water/Reagent Matrix Spikes or Laboratory Control Sample (LCS) A control sample of known analyte concentration and source which has been processed through the entire method (digested/extracted, etc.); may be used as an additional QC check to monitor the effectiveness of the method (not subject to matrix interferences). This LCS (Laboratory Control Sample) should be of the same matrix as samples but must be spiked using a standard from a different source than the calibration standards (EPA or NBS traceable when possible). If used, these must be analyzed at a frequency of one sample in a sample set, or 5%, whichever is greater, for a specified method.
 - 4. Quality Control Check Standards or Performance Evaluation Standards (PE) At a minimum these blind check samples shall be analyzed in duplicate semiannually. Results of all blind performance evaluation (PE) samples will be summarized in the QA report to management and forwarded to clients upon request. EMS Heritage participates in many state certification programs requiring blind sample (PE) analysis in addition to the EPA WP (Water Pollution) and WS (Water Supply) programs and a commercial service providing monthly blind samples (Analytical Products Group). In addition, EMS

Heritage-Indianapolis participates in the USEPA Contract Laboratory Program's Quarterly Blind (QB) program.

- 5. Ouality Control Check Standards or Initial and Continuing Calibration Check Standards (ICV01, CCV) An Initial Calibration Verification (ICV01) standard shall be analyzed at the beginning of each run to verify the standard curve (separate source than calibration standards, traceable to EPA or NBS if possible) and/or a Continuing Calibration Verification (CCV) standard shall be analyzed at a continuing frequency of 10% of the analytical samples in the analytical set (i.e. one every 10 analytical samples in a run).
- 6. Duplicate Samples (DUP02) or Matrix Spike Duplicates (DPS02) At least one or 5%, whichever is greater, of all samples in a sample set with a similar matrix shall be selected and analyzed in duplicate. If a sample set contains samples from different matrices (e.g., effluent and drinking water), then duplicates or matrix spike duplicates should be analyzed for each matrix.
- 7. Additional OC Checks may be included and will be used if specified by the approved method:
 - a. Reagent purity checks
 - b. Internal standards
 - c. Surrogate spikes
 - d. Method of Standard Additions (MSA, 3 Point or 1 Point)

11.3 Determination of Frequency of OC Measurements

The required frequency of analysis of certain QC sample types stated in this QAP may be given as a percentage (i.e., 5%) or translated into a numerical ration (i.e., 1 in 20). The term "analytical sample" is defined in Appendix D, <u>Definitions and Acronyms</u>. As the term is used, analytical sample includes all field samples, including Performance Evaluation samples, received from an external source, but it also includes all required QA/QC samples (matrix

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spikes, analytical/post digestion spikes (1-point MSA), duplicates, serial dilutions, LCS, Interference Check Samples (ICS), CDL standards, preparation blanks (BLA02) and linear range analyses) except those directly related to instrument calibration or calibration verification. A "frequency of 10%" means once every 10 analytical samples.

The following Tables and Figures specify the QC Types and their frequency of analysis for different methods/regulatory programs.

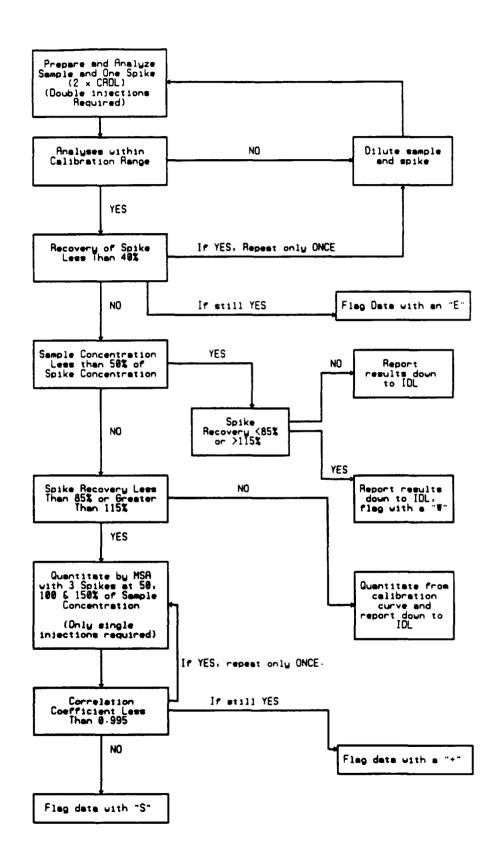
Table # or Figure #	# Pages	Subject						
Table 11.1	1	ICP Run Sequence						
Table 11.2	1	FAA Run Sequence						
Table 11.3	1	GFAA Run Sequence						
Table 11.4	1	CLP-GFAA Run Sequence						
Table 11.5	1	Analytical Run Requirements/Frequency						
Figure 11.1	1	Metals Analysis Scheme For Spiking (SPI 01)						
Figure 11.2	1	CLP-GFAA Analysis Scheme						

11.4 Routine Methods Used to Assess Precision and Accuracy

The targets listed in Section 5.0 were derived where possible from data entered into the QCTS (Quality Control Tracking System). All sample analysis runs and all QC data points are entered into the QCTS. Certain of the QC types for a particular method are not applicable and/or not used to arrive at precision and accuracy statistics; those types can be identified from Table 11.5, Analytical Run Requirements/Frequency for a particular method.

The criteria listed in Table 5.1, Section 5.0 are derived from matrix specific QC. Control limits for Instrumental criteria (calibration) and method performance (LCS, ICV02) are found elsewhere in this document. EMS Heritage defines most QC Types (see Definitions) with either an "01" or "02" suffix. The "01" suffix denotes that the procedure did not include a preparation step and/or a preparation step is not possibly separated from the analytical procedures. An "02" suffix denotes that the procedure was taken

FIGURE 11.2 CLP-GFAA ANALYSIS SCHEME



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through the entire procedure, which includes a discreet or separate preparation (extraction, digestion, distillation, etc.) step.

All Table 5.1 control limits are computed from sample matrices processed through all analytical steps including any preparatory procedure ("02") if applicable.

11.4.1 Accuracy

Acceptance limits are calculated as follows, where x_i represents the individual values, \overline{x} is the mean, n is the number of values and S is the Standard Deviation:

Upper Control Limit (UCL) =
$$\bar{x} + 3S$$

Upper Warning Limit (UWL) = $\bar{x} + 2S$
Center Line (\bar{x}) = $\frac{\sum x_i}{n}$
Lower Warning Limit (LWL) = $\bar{x} - 2S$
Lower Control Limit (LCL) = $\bar{x} - 3S$

11.4.2 Precision

Acceptance control limits are calculated as follows, where x_i represents the individual values, \overline{x} is the mean, n is the number of values and S is the Standard Deviation:

Upper Control Limit (UCL) =
$$\bar{x} + 3S$$

Upper Warning Limit (UWL) = $\bar{x} + 2S$
Center Line (\bar{x}) = $\frac{\sum x_i}{n}$

11.4.3 Identification of Outliers

Outliers are excluded from the data used to calculate acceptance (control) limits. The procedure is described in ASTM E178-80, Section 4.

Outliers may be determined by any of the procedures referenced in ASTM E178-80 applicable, however, as a general rule the inorganic precision limits will be tested for outliers using the coefficient of kurtosis test at the 1 Percent Significance Levels (this is equivalent to a test for normality). The kurtosis formula is recommended for "two-

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sided" tests (changes in level to higher and lower values) and also for changes in scale (variance). Outliers in the data sets for determining accuracy control limits in all tests and precision control limits in organic testing (from MS/MSD determinations) may use the simpler test criterion, T_n at the 1 percent significance level as follows:

$$T_n = (x_n - \bar{x})/S$$

Where:

 \overline{x} = arithmetic average of all n values, and

S = estimate of the population standard deviation based on the sample data

 $x_n =$ the doubtful value

11.5 Methods Used to Generate Precision and Accuracy Targets

The concentration levels used to make control limit determinations are as follows:

- 1. <u>Low level</u> defined as concentrations from the minimum detection limit to a level 5 times the MDL or PQL.
- 2. <u>Mid level</u> defined as the mean level between the minimum detection level and the upper end of the linear range.
- 3. <u>High level</u> defined as the concentration at the upper end of the linear range.

Refer to Table 11.6, Methods Used to Generate Precision and Accuracy Targets, for further information.

11.5.1 The following procedures/equations are to be used to assess data:

1. Accuracy

%Recovery of Standard =
$$\frac{A}{B}$$
 x 100%

%Recovery of Spike =
$$\frac{(SSR - SR)}{SA}$$
 x 100%

Where:

SSR = Spiked Sample Result

SR = Sample Result, 0.0 if < MDL

SA = Spike Added

2. <u>Precision</u> - Replicability expressed as Relative Percent Difference (RPD)

The following formula will be used when sample precision is evaluated utilizing matrix spikes (MS) and matrix spike duplicates (MSD):

$$RPD = \frac{2|MSR - MSDR|}{(MSR + MSDR)} \times 100$$

Where:

MSR = Matrix Spike Recovery (%R)

MSDR = Matrix Spike Duplicate Recovery (%R)

The following formula will be used when sample precision is evaluated from duplicate (un-spiked) sample results:

$$RPD = \frac{2|S - D|}{(S + D)} \times 100$$

Where:

S = Sample Original Measured Value

D = Duplicate Sample Measured Value

3. Standard Deviation (S)

The following formula will be used for calculations by computer programs:

$$S = \sqrt{\frac{\sum_{i=1}^{n} (C_i - \overline{C})^2}{n-1}}$$

Where:

 C_i = Each individual data point C = Average value for all readings

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The following form may be used when S is calculated with a calculator:

$$S = \sqrt{\frac{\sum C_i^2 - \frac{(\sum C_i)^2}{n}}{n-1}}$$

4. Relative Standard Deviation (RSD)

Relative Standard Deviation is the ratio of the standard deviation S of a set of numbers to their mean (X) expressed as percent. It relates standard deviation (or precision) of a set of data to the size of the numbers; sometimes referred to as the coefficient of variation (CV).

$$CV = RSD(percent) = \frac{S}{\overline{X}} \times 100\%$$

5. Percent Difference (%D)

Instrument calibration methods require this calculation to compare initial and continuing calibration response factors.

%Difference =
$$\frac{\overline{RRF_i} - RRF_c}{\overline{RRF_i}} \times 100$$

Where: RRF_i = Average relative response factor

from initial calibration

RRF_c = Relative response factor from continuing calibration standard

11.6 Significant Figures

The following rule will be used for the number of significant figures to use in calculations: For all mathematical operations (addition, subtraction, multiplication, division and exponentiation) retain the equivalent of two more places of figures than present in the single observed value. No rounding of calculated values or intermediate results should be done until the final values are obtained. No rule can be given for the number of significant places to be

: 1

retained in final values, however, most tests will be reported with two significant figures. Percent recovery (%R) data will be rounded and reported to include the first decimal place. Relative percent difference (RPD) calculations will be performed using the above rounded %R result when hand calculated, or will use all digits carried by a computer when calculated by computer. The RPD result will be reported to 2 significant figures. The rules to be used for significant figure reporting of %R and RPD are as follows:

% Recovery

#	Significant Figures Reported	Magnitude of Value (x)
	4	x ≥ 100
	3	$100 > x \ge 10$
	2	$10 > x \ge 1$
	1	x < 1

RPD

Significant Figures Reported Magnitude of Value (x)
$$\begin{array}{ccc}
2 & x \ge 1 \\
1 & x < 1
\end{array}$$

11.6.1 The rounding procedure to be followed is:

- 1. If the figure following those to be retained is less than 5, the figure is dropped, and the retained figures are kept unchanged. As an example, 11.444 is rounded off to 11.44.
- 2. If the figure following those to be retained is greater than 5, the figure is dropped and the last retained figure is raised by 1. As an example, 11.446 is rounded off to 11.45.
- 3. If the figure following those to be retained is 5, and if there are no figures other than zeroes beyond the five, the figure 5 is dropped, and the last-place figure retained is increased by one if it is an odd number or it is kept unchanged if an even number. As an example, 11.435 is rounded off to 11.44, while 11.425 is rounded off to 11.42.
- 11.6.2 Calculations using Percent Recovery (%R of spikes, etc.) or observed result from the RPD calculation will use the rounded (reported) value in the calculation, so that reported %R and final results used in calculations (vs. using un-rounded numbers not found on final reports) will arrive at the same result.

Organic testing (GC, GC/MS, etc.) requires reporting of %RSD and Percent Difference (%D) to at least one decimal place.

11.7 Control Charts

In addition to the Quality Control Tracking System (QCTS) which establishes control limits quarterly for all QC types (in a tabular format), certain QC types will be evaluated using graphical quality control charts at the bench level.

Control charts are graphical methods for monitoring and improving analytical quality over time. They are process-control tools that can be applied to spiked sample or reference material recoveries (ICVO1, LCS) or other QC test results. They can be used to document data quality, detect the existence of quality problems (special causes, motivate better performance and improve the analytical process.

Graphical quality control charts will include at a minimum:

- 1. ICV01
- 2. LCS
- 3. BLA02

Any other graphical control charting required by method or contract will be performed as required.

Graphical quality control charts and their associated control and warning limits will be updated with each analytical run at the bench.

Refer to Section 11.4 and Appendix A, Selected References, for additional guidance.

11.8 Instrument Detection Limits (IDLs) and Method Detection Limits (MDLs)

Instrument Detection Limits (IDLs) and Method Detection Limits (MDLs) are a quantitative measure of capability. Determination of the IDLs and/or MDLs are mandatory at EMS Heritage Laboratories.

All operating instruments in the laboratory have documented IDLs on file with the QA unit. All newly acquired instruments will have IDLs determined and reported to the QA Unit for approval, before any samples are analyzed. Any significant manipulations/modification to any instrument will require a new determination of the IDLs, which shall be submitted to the QA Unit for approval, before any samples are analyzed. The frequency for determination of IDLs, in the absence of any major instrument manipulations/modifications, shall be determined by the QA unit. Refer to Appendix D, Definitions, under Detection Limits, for the definition of an IDL. See Figure 11.3 for the

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IDL/MDL reporting form. Copies of the IDL/MDL form will be sent to the Corporate Quality Assurance Officer for formal test code location approval.

11.8.1 All analytical methods used in the laboratory will have MDLs determined and reported to the QA Unit (not required when an MDL is equivalent to an IDL) for approval, before any samples are analyzed and reported. At a minimum, MDLs may be determined on only 1 instrument of several equivalent instruments having essentially equivalent IDLs, however, they must be determined on the instrument of least sensitivity (highest IDLs). Methods requiring individual analysts' demonstration of ability may be an exception to this rule. At a minimum, the MDLs will be determined in a reagent water matrix and will represent the best achievable detection limit. These reagent water MDLs will be determined by using the procedure outlined in 40 CFR Part 136, Appendix B. This procedure is designed for applicability to a wide variety of sample types ranging from reagent (blank) water containing analyte to wastewater containing analyte. Client or regulatory program specific requirements may call for the determination of client matrix specific MDLs, in which case the lab may opt to use the MDL procedure specified in the Third Edition of SW-846 for all non Clean Water Act (CWA) testing programs (those not requiring 40 CFR Part 136).

11.8.2 MDL Determination as Outlined in 40 CFR Part 136 App. B

A summary of the procedure, as used by EMS Heritage, follows:

- 1. Make an estimate of the detection limit using one of the following:
 - a. The concentration value that corresponds to an instrument signal/noise in the range of 2.5 to 5.
 - b. The concentration equivalent of three times the standard deviation of replicate instrumental measurements of the analyte in reagent water.

Figure 11.3 DETECTION LIMIT STUDY

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			DATE:
NSTRUMENT:		_ ANALYST:	
NALYTE:			d:
METHOD:		_	
REP METHOD:		_	
AMPLE MATRIX:			
	CURVE	DEVELOPMENT	
Conc.	<u>Units</u>		
•		_	
		-	
	·	-	
·			
		— Ohmi —	%R=
. 10.4011.4			
Cons	<u> </u>	D CONCENTRATION	
Conc.	<u>Units</u>		
•		-	
0		-	
1.			
2.		_	
3.		-	
4		 -	
5. ICV01TV=_		_Obsv.=	_%R=
6. BLA01	_		
Mean (%) of #'s 8 thr	ough 14 values	*	_
tandard Deviation (SD)		=	_ x 3.143 =
	IDL/MDL	=	_
5% LCL = IDL/MDL	x 0.64	=	_
5% UCL = IDL/MDL	x 2.20	*	

- c. That region of the standard curve where there is a significant change in sensitivity, i.e. a break in the slope of the standard curve.
- d. Instrumental limitations.
- 2. Prepare reagent (blank) water that is as free of analyte as possible. This reagent water must not contain analyte or interferant at concentrations at or above the MDL.
- 3. a. MDLs determined in reagent water will contain analyte concentrations which are at least equal to or in the same concentration range as the estimated MDL. The standard concentration will not exceed 10 times the actual MDL. EMS Heritage will set the standard concentration between 1 and 5 times the estimated detection limit.
 - b. MDLs required to be determined in sample matrices must first be analyzed for the background concentration. Depending on the level of analyte present in the sample a known amount of analyte may be added to bring the level of analyte between one and five times the estimated detection limit. If the level of analyte is greater than the estimated detection limit, then the option of choosing another sample with a lower analyte level or the option using that sample (if analyte does not exceed 10 times the reagent water MDL) may be chosen.
- 4. a. Seven aliquots of the sample will be used to calculate the MDL. All samples will be processed through the entire analytical method. If a blank measurement is required to calculate the measured level of analyte, obtain a separate blank measurement for each sample aliquot analyzed. The average blank measurement is subtracted from the respective sample measurements.
 - b. EMS Heritage may opt to take two aliquots of the sample to be used to calculate the MDL and process each entirely. If these measurements indicate that the sample is in the desirable range, then five additional aliquots will be analyzed, combined with the two aliquots

for a total of 7 measurements. If the sample is not in the correct range, the MDL will be re-estimated and the process repeated.

- 5. The variance (S²) and standard deviation (S) of the replicate measurements will be calculated as given in the referenced procedure. The standard deviation of the 7 replicate measurements will be multiplied by 3.143 and rounded to three significant figures.
- 6. The 95% confidence interval estimates for the MDL will be calculated as follows:

LCL = 0.64 MDL

UCL = 2.20 MDL

7. EMS Heritage will generally not use the optional iterative procedure for MDL determinations.

Where EMS Heritage opts to use the SW 846 procedure for determining the MDL, the same procedure as 40 CFR, Part 136, Appendix B will be used except for the following points:

- a. Triplicate analyses will be utilized instead of seven analyses.
- b. The standard deviation from the triplicate analyses will be multiplied by 7 to obtain the MDL.
- 11.8.3 EMS Heritage will report a practical quantitation limit (PQL) in place of the actual MDL where allowed. Exceptions will be made to Practical Quantitation Limit (PQL) reporting where regulatory or client DQO's push the limits of reporting close to the MDL. In no case will EMS Heritage report data lower than the MDL without the "estimated" qualifier on the result.

The PQL will be 10 times the standard deviation (S_{pooled}) that is derived from the procedures used to determine the MDL. It will be calculated to two significant figures in most cases. Alternatives to the PQL calculated in this manner will be action levels or other criteria used as the reporting detection limit as long as the reported limit is

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equal to or greater than the actual MDL. The QA Officer will make the final determination for all reporting detection limits.

EMS Heritage recognizes the definition of PQL as stated in John Taylor's book <u>Principles of Quality Assurance of Chemical Measurements</u>, 1987, p. 79-82.

TABLE 11.1 ICP Run Sequence

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TYPE	DESCRIPTION/FREQUENCY
BLA00	Instrument Blank
CAL01	Calibration Standards (Single Point Calibration)*
ICV01	Initial Calibration Verification (EPA QC or equivalent; Mid-Range Std.)*
BLA01	Calibration Blank (Beginning, Initial Calibration Blank)*
CDL01"	Detection Limit Standard (2x CRDL, beginning, 2 times/8 hrs.)
ICS	Interference Check Sample(s) (Beginning, 2 times/8 hrs.)
LRA	Linear Range Analysis (Upper limit of linear range - quarterly)
LCS	Laboratory Control Sample (ICV02 or EPA supplied LCS, 1 per 20 or prep batch)
BLA02	Preparation Blank (1 per 20 or prep batch, whichever is more frequent)
SAMPLES	6 samples, including required SPI02's, DPS02's, DUP02's and SPI01's (MSA)***
CCV	Continuing Calibration Verification (May be same as ICV, 10% frequency or every 2 hours; mid-range std.)*
BLA01	Continuing Calibration Blank (10% frequency during run)*
SDA**	Serial Dilution Analysis (each matrix and/or each case)
SAMPLES	9 samples, including required SPI02's, DUP02's and SPI01's (MSA)***
CCV	Continuing Calibration Verification (May be same as ICV, 10% frequency or every 2 hours; Mid-range Std.)*
BLA01	Continuing Calibration Blank (end of run)*
	GO TO: SAMPLES OR CDL01
ICS	Interference Check Sample (End, 2 times/8 hrs)
CDL01**	Detection Limit Standard (2x CRDL, end, 2 times/8 hours)

NOTE: Each full (3 point) MSA counts as two "analytical samples".

Does not count as an "analytical sample" for the 10% frequency requirement.

Performed only for specific projects as required (contracts, etc.)

Single addition post digestion spike (1 point MSA) analysis is counted as an "analytical sample" for ICP; required immediately following unspiked analysis of that sample.

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TABLE 11.2 FAA Run Sequence

TYPE	DESCRIPTION/FREQUENCY
BLA00	Instrument Blank
CAL01	Calibration Standards (4+BLA00 Point Calibration)*
ICV01	Initial Calibration Verification (EPA QC or equivalent; Mid-Range Std.)*
BLA01	Calibration Blank (Beginning, Initial Calibration Blank)*
CDL01"	Detection Limit Standard (2x CRDL, beginning, 2 times/8 hrs.)
ICS	Interference Check Sample(s) (Beginning, 2 times/8 hrs.)
LCS	Laboratory Control Sample (ICV02 or EPA supplied LCS, 1 per 20 or prep batch)
BLA02	Preparation Blank (1 per 20 or prep batch, whichever is more frequent)
SAMPLES	7 samples, including required SPI02's, DUP02's and SPI01's (MSA)***
CCV	Continuing Calibration Verification (May be same as ICV, 10% frequency or every 2 hours; mid-range std.)*
BLA01	Continuing Calibration Blank (end of run)*
SAMPLES	10 samples, including required SPI02's, DPS02's, DUP02's and SPI01's (MSA)***
CCV	Continuing Calibration Verification (May be same as ICV, 10% frequency or every 2 hours; Mid-range Std.)*
BLA01	Continuing Calibration Blank (end of run)*
	GO TO: SAMPLES OR CDL01
→ CDL01**	Detection Limit Standard (2x CRDL, end, 2 times/8 hrs.)

NOTE: Each full (3 point) MSA counts as two "analytical samples".

Does not count as an "analytical sample" for the 10% frequency requirement.

Performed only for specific projects as required (contracts, etc.)

Single addition post digestion spike (1 point MSA) analysis is counted as an "analytical sample" for FAA; required immediately following unspiked analysis of that sample.

# INJECTIONS	TYPE	DESCRIPTION/FREQUENCY
1	BLA00	Instrument Blank
5	CAL01	Calibration Standards (4+Blank Point Calibration)*
1	ICV01	Initial Calibration Verification (EPA QC or equiv.; Mid-range Std.)*
1	BLA01	Calibration Blank (Beginning, Initial Calibration Blank)*
1	CDL01***	Detection Limit Standard (2x CRDL, beginning, 2 times/8 hrs.)
1	LCS	Laboratory Control Sample (ICV02 or EPA supplied LCS, 1 per 20 or prep batch)
1	BLA02	Preparation Blank (1 per 20 or prep batch, whichever is more frequent)
7	SAMPLES	7 samples, including required SPI02's, DPS02's, DUP02's
7	SAMPLES- SP*	7 sample spikes (2x CRDL, post digestion spike, one per analytical sample except SPI02's, DPS02's)**
1	CCV	Continuing Calibration Verification (May be same as ICV, 10% frequency or every 2 hours; Mid-range Std.)*
1	BLA01	Continuing Calibration Blank (10% frequency during run)*
10	→ SAMPLES	10 samples, including required SPI02's, DPS02's, DUP02's
10	SAMPLES- SP*	10 sample spikes (2x CRDL, post digestion spike, one per analytical sample except SPI02's, DPS02's)**
1	CCV	Continuing Calibration Verfication (May be same as ICV, 10% frequency or every 2 hours; Mid-range Std.)*
1	BLA01	Continuing Calibration Blank (end of run)*
-		GO TO: SAMPLES OR CDL01
1	→ CDL01***	Detection Limit Standard (2x CRDL), end, 2 times/8 hrs.)

NOTE: Duplicate injections (burns) not performed; maximum of 20 injections between CCVs. Each full MSA (3-point) counts as 2 "analytical samples".

Does not count as an "analytical sample" for the 10% frequency requirement.

Single addition post digestion spike (1 point MSA) required immediately following unspiked analysis of that sample.

Performed only for specific projects as required (contracts, etc.)

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TABLE 11.4 CLP - GFAA Run Sequence

# INJECTIONS	TYPE	DESCRIPTION/FREQUENCY _
1	BLA00	Instrument Blank
5	CAL01	Calibration Standards (4+Blank Point Calibration)*
2	ICV01	Initial Calibration Verification (EPA QC or equiv.; Mid-range Std.)*
2	BLA01	Calibration Blank (Beginning, Initial Calibration Blank)*
2	CDL01	Detection Limit Standard (2x CRDL, beginning, 2 times/8 hrs.)
2	LCS	Laboratory Control Sample (ICV02 or EPA supplied LCS, 1 per - 20 or prep batch)
2	LCS-SP	Laboratory Control Sample Spike (2x CRDL, post digestion spike, one per LCS)
2	BLA02	Preparation Blank (1 per 20 or prep batch, whichever is more frequent)
2	BLA02-SP	Preparation Blank (2x CRDL, post digestion spike, one per Prep Blank)
4	SAMPLES	2 samples, including required SPI02's, DPS02's, DUP02's -
4	SAMPLES-SP	2 sample spikes (2x CRDL, post digestion spike, one per analytical sample except SPI02's, DPS02's)
2	CCV	Continuing Calibration Verification (May be same as ICV, 10% frequency or every 2 hours; Mid-range Std.)*
2	BLA01	Continuing Calibration Blank (10% frequency during run)*
10	→SAMPLES	5 samples, including required SPI02's, DPS02's, DUP02's
10	SAMPLES-SP*	5 sample spikes (2x CRDL, post digestion spike, one per analytical sample except SPI02's, DPS02's)**
2	CCV	Continuing Calibration Verfication (May be same as ICV, 10% frequency or every 2 hours; Mid-range Std.)*
2	BLA01	Continuing Calibration Blank (end of run)*
-		GO TO: SAMPLES OR CDL01
2 _	→CDL01	Detection Limit Standard (2x CRDL), end, 2 times/8 hrs.)

NOTE: Duplicate injections (burns) are performed; maximum of 20 injections between CCVs. Each full MSA (3 poin' counts as 2 "analytical samples"; for analytical runs containing only MSAs, single injections can be used for Q samples during that run.

Does not count as an "analytical sample" for the 10% frequency requirement.

Single addition post digestion spike (1 point MSA) required immediately following unspiked analysis of that samp! (counts as an analytical sample).

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Test	Method			ail a sa	Instrumen	al QC			Prep. QC								
	Reference	ICV01	BLAOI	CCV2	CDL01	DPS01 ³	DUP01 ³	SPI01 (or MSA)	BLA02	(CLP) BLA02 SP101	ICV02 or LCS	(CLP) LCS/ SP101	SP102 ¹	DPS02 ³	DUP023	DLCS	
Acidity	EPA	R	5%	B,E 10%	•	•	R, 10%	٠	NA		R	•	NA	NA	NA	-	
Alkalinity	EPA	NA	5%	B,E 10%	•	10%	R, 10%	10%	NA	-	R	5%	NA	NA	NA		
Ammonia - N	ЕРА	С	B,E 10%	B,E 10%	•	R, 10%	R, 10%	R, 10%	P		P	•	P, 10%	P, 10%	P. 10%		
Asbestos	NIOSH	R	R	NA	•_	•	R, 10%	•	NA	NA	R, 10%		NA	NA	NA		
BOD	EPA/SM	R	R				R, 10%	-	NA		R, 5%		NA	NA		-	
Boron Colorimetric	EPA	С	B,E 10%	B,E 10%	•	R, 10%	R, 10%	R, 10%	NA	NA	R, 5%	•	NA	NA	NA		
Bromide	EPA/SM	R	R, 5%	•	-	R, 10%	R, 10%	R, 10%	NA	NA	R, 5%		NA	NA	NA		
COD	EPA/SM	С	R	R, 10%	·	•	R, 10%	•	NA	NA	R, 5%	•	NA	NA	NA		
Chloride	EPA/SM	С	R, 5%	B,E 10%		R, 10%	R, 10%	R, 10%	NA	NA	R, 5%	-	NA	NA	NA		
Chlorine, Total	ASTM	R	R	NA	В	R, 10%	R, 10%	R, 10%	NA	NA	R, 5%	-	NA	NA	NA		
Coliform, Fecal	SM	NA	10%	NA	-		R, 10%		NA	NA		•	NA	NA	NA	·	
Coliform, Total	SM	NA	10%	NA		•	R, 10%	•	NA	NA	•	•	NA	NA	NA	_ ·_	
Color	ЕРА	N	NA	B,E 10%	NA	•	R, 10%	NA	NA	NA	NA	•	NA	NA	NA		
Cyanide, Amen.	EPA/SW	С	B,E 10%	B,E 10%	-	•		•	P, 5%		P, 5%	-	P, 10%	P, 10%	P, 10%		
Cyanide, Total	EPA/SW	С	B,E 10%	B,E 10%		•		٠	P, 5%	·	P, 5%	•	P, 10%	P, 10%	P, 10%	-	
Cyanide, Total	CLP	С	B,E 10%	B,E 10%	B,E			-, N	P, 5%		P, 5%	-	R, 10%		R, 10%	-	
Cyanide, T. Avail.	sw	С	B,E 10%	B,E 10%	B,E	P, 10%	P, 10%	P, 10%	NA	NA	R, 5%	-	NA	NA	NA		
Dissolved Oxygen	EPA								NA	NA		-	NA	NA	NA		

	,, ,	Instrumental QC								Prep. QC								
Test	Method Reference	ICV01	BLAOI	ocv₁	CDL01	DPS013	DUP01 ³	SPI01 (or MSA)	BLA02	(CLP) BLA02 SP101	iCV02 or LCS ⁴	(CLP) LCS/ SP101	SP102 ³	DPS02 ⁵	DUP023	DLCS		
Dissolved Solids	EPA	R, 10%	R, 5%	NA	NA	R, 10%	R, 10%	R, 10%	NA	NA	R, 5%	NA	NA	NA	5%	·		
ISE Fluoride	EPA	С	B,E 10%	B,E 10%	В	P, 5%	P, 5%	P, 5%	P, 5%		P, 5%	-	P, 5%	P, 5%	P, 5%			
Formaldehyde	AOAC						R, 10%				, 5%							
Gross Alpha, Beta (Radioactivity)	EPA	NA	B,E 20%	В		•	R, 10%	-	NA	NA	-	-	NA	NA	NA	-		
Titrimetric Hardness	EPA	R	B,E 10%	NA	٠	٠	R, 10%	•	NA	NA	R, 5%	-	NA	NA	NA	-		
Hexavalent Chromium	SM	R	B,E 10%	B,E 10%	В	R, 10%	R, 10%	R, 10%	P, 10%	•	R, 5%	-	R, 10%	R, 10%	R, 10%			
Hexavalent Cr	sw	R	B,E 10%	B,E 10%	В	R, 10%	R, 10%	R, 10%	P, 10%	-	R, 5%		R, 10%	R, 10%	R, 10%			
Iodine	SM	С	B,E 10%	B,E 10%		R, 10%	R, 10%	R, 10%	NA	NA	R, 5%		NA	NA	NA			
TKN	EPA	С	B,E 10%	B,E 10%		R, 10%	R, 10%	R, 10%	NA	NA	R, 5%		NA	NA	NA			
Nitrate-N (Brucine)	EPA	С	B,E 10%	B,E 10%		R, 10%	R, 10%	R, 10%	NA	NA	R, 5%	_	NA	NA	NA			
Nitrate-N (Colorimetric)	EPA	С	B,E 10%	B,E 10%	•	R, 10%	R, 10%	R, 10%	NA	NA	R, 5%	-	NA	NA	NA			
Nitrite-N	EPA	С	B,E 10%	B,E 10%	•	R, 10%	R, 10%	R, 10%	NA	NA	R, 5%		NA	NA	NA	-		
Nitrate-Nitrite-N	EPA	С	B,E 10%	B,E 10%	•	R, 10%	R, 10%	R, 10%	NA	NA	R, 5%		NA	NA	NA	-		
Nuisance Dust	NIOSH	R	R	B,E 10%	NA	NA	R, 10%	NA	NA	NA	NA	NA	NA	NA	NA			
Oil & Grease	EPA413.1	R	R	NA			R, 10%		NA	NA	•	•	NA	NA	NA	-		
Oil & Grease	SW 9071	R	R	NA		-	R, 10%	_	NA	NA	R, 5%		NA	NA	NA			

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Table 11.5 ANALYTICAL RUN REQUIREMENT/FREQUENCY

Test					Instrumen	ml QC				Prep. QC							
	Method Reference	ICV01	BLAGI	ccv*	CDL01	DPS013	DCP01 ³	SP101 (or MSA)	BLA02	(CLP) BLA02 SP101	er IC24	(CLP) LCS/ SP101	SP102 ¹	DP9023	DUP023	DLCS	
Paint Filter	sw	NA	NA	NA	NA	NA	R, 10%	NA	NA	NA	•	NA	NA	NA	NA		
Percent Solids	SM	R	R	NR%	NA	_ NA	R, 10%	NA	NA	NA	R, 5%	NA	NA	NA	NA	-	
Percent Solids	ASTM	R	R	NA	NA	NA	R, 10%	NA	NA	NA	R, 5%	NA	NA	NA	NA	<u>.</u>	
рН	EPA/SW	С	NA	B,E 10%	NA	NA	R, 10%	NA	-	•	R, 5%	-	NA	NA	NA		
Phenois	EPA/SW	С	B,E 10%	B,E 10%	В	R, 10%	R, 10%	R, 10%	P	•	P, 5%	-	P, 10%	P, 10%	P, 10%	-	
Phosphorus	EPA	С	B,E 10%	B,E 10%	-	R, 10%	R, 10%	R, 10%	P	•	P, 5%	-	P, 10%	P, 10%	P, 10%		
Radium	EPA	NA	B,E 5%	В	-	•	R, 10%	•	NA	NA		-	NA	NA	NA		
Radium 226	EPA	NA	B,E 5%	В	•		R, 10%	•	NA	NA	•	_ •	NA	NA	NA		
Residual Chlorine	EPA	R	B,E 10%	B,E 10%	•	R, 10%	R, 10%	R, 10%	NA	•	R, 5%		NA	NA	NA		
Saturation Index							See ind	ividual method	s for this cal	culation							
Settleable Matter	EPA		NA	NA		R, 10%	R, 10%	NA	NA	•	R, 5%		NA	NA	NA		
Silica, Total	EPA	R	R	10%	-	R, 10%	R, 10%	•	NA	•	R, 5%		NA	NA	NA		
S. Conductance	EPA/SW	R	R	B,E 10%	•	•	R, 10%	•	NA	NA	R, 5%	NA	NA	NA	NA		
Sulfate	EPA/SW	R	R	B,E 10%	•	R, 10%	R, 10%	R, 10%	NA	NA	R, 5%	-	NA	NA	NA	•	
Sulfide	EPA/SW	R		B,E 10%	·	R, 10%	R, 10%	R, 10%	NA	NA	R, 5%		NA	NA	NA		
Sulfur	ASTM	R	R	10%		R, 10%	R, 10%	R, 10%	NA	NA	R, 5%		NA	NA	NA		
Surfactanta	EPA	R	R	10%		R, 10%	R, 10%	R, 10%	NA	NA	R, 5%		NA	NA	NA		
Temperature	EPA	И			NA	NA	R, 10%	NA	NA	NA	•	NA	NA	NA	NA		
Thiocyanate	ЕРА	R	R	B,E 10%	·	R, 10%	R, 10%	R, 10%			R, 5%						

-	111				Instrumen	ul QC						Prep.	QC	1		
Test	Method Reference	ICV01	BLAOI	ocv*	CDL01	DPS01*	DUP01	SPI01 (or MSA)	BLA02	(CLP) BLA02 SPI01	1CV02 or LCS ⁴	(CLP) LCS/ SP101	SPI02 ⁵	DPS02 ³	DUP023	DLCS
Sulfide, T. Available	sw	R	R	B,E 10%	•	P, 10%	P, 10%	•	P, 10%	-	P, 5%	<u>.</u>	P, 10%	P, 10%	P, 10%	-
Sulfite	EPA	R	R	10%	-	R, 10%	R, 10%	R, 10%	NA	NA	R, 5%	-	NA	NA	NA	
Total Halogen	ASTM	R	B,E 10%	NA	-	R, 10%	R, 10%	R, 10%	NA	NA	R, 5%	-	NA	NA	NA	
тос	EPA/SW	R	B,E 10%	B,E 10%	В	R, 10%	R, 10%	R, 10%	NA	NA	R, 5%	-	NA	NA	NA	-
T. Volatile Residue	EPA/SM /ASTM	•	R	•	-	NA	R, 10%	NA	NA	NA	R, 5%		NA	NA	NA	
тох	sw	R	B,E 10%	B,E 10%	В	R, 10%	R, 10%	R, 10%	NA	NA	R, 5%	-	NA	NA	NA	•
Total Recoverable Oil & Gresse	sw	R	R	NA		•	R, 10%	-	NA	NA	R, 5%	-	NA	NA	NA	-
Total Solids	EPA	R	R, 5%	NA	•	•	R, 10%		NA	NA	R, 5%	-	NA	NA	NA	
Total Suspended Solids	ЕРА	R	R, 5%	NA	•	•	R, 10%	•	NA	NA	R, 5%		NA	NA	NA	
Turbidity	EPA	R	R	B,E 10%	•	NA	R, 10%	NA	NA	NA	R, 5%	-	NA	NA	NA	
Volatile Susp. Solids	ЕРА		R, 5%		٠	NA	R, 10%	NA	NA	NA	R, 5%	-	NA	NA	NA	
Metals-ICP	EPA	R	B,E 10%	B,E 10%	B,E,T	10%*	10%*	10%*	P, 10%	•	P, 10%	•	P, 10%	P, 10%	P, 10%	
Metais-ICP	sw	R	B,E 10%	B,E 10%	B,E,T	NA	NA	NA	P, 10%	•	P, 10%		P, 10%	P, 10%	P, 10%	
Metale-ICP	CLP	R	B,E 10%	B,E 10%	B,E,T	NA	NA	NA	P, 10%	-	P, 10%	•	P, 5%		P, 5%	

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^{*}Only applies in limited situations (samples not digested).

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		Instrumental QC							Prep. QC							
	Mathod Reference	ICV01	BLA01	ocv*	CDL01	DPS01 ³	DUP01 ³	SPIO1 (or MSA)	BLA02	(CLP) BLA02 SP101	or ICS4	(CLP) LCS/ SPI01	SP102 ³	DP\$023	DUP029	DLCS
Metale-FAA	EPA	R	B,E 10%	B,E 10%	B,E,T	NA	NA	NA	P, 10%	•	P, 10%		P, 10%	P, 10%	P, 10%	-
Metale-FAA	sw	R	B,E 10%	B, E 10%	B,E,T	NA	NA	NA	P, 10%	-	P, 10%		.P, 10%	P, 10%	P, 10%	
Metale-FAA	CLP	R	B,E 10%	B,E 10%	B,E,T	NA	NA	NA	P, 10%	•	P, 10%	-	P, 5%		P, 5%	
Metals-GFAA	ЕРА	R	B,E 10%	B,E 10%	B,E,T	NA	NA	100%**	P, 10%		P, 10%		P, 10%	P, 10%	P, 10%	
Metals-GFAA	sw	R	B,E 10%	B,E 10%	B,E,T	NA	NA	100%**	P, 10%	-	P, 10%	•	P, 10%	P, 10%	P, 10%	
Metals-GFAA	CLP	R	B,E 10%	B,E 10%	B,E,T	NA	NA	100%**	P, 10%	P, 5%	P, 10%	P, 5%	P, 5%	-	P, 5%	
Mercury, CVAA	EPA	R	B,E 10%	B,E 10%	B,E,T	NA	NA	NA	P, 10%	-	P, 10%		P, 10%	P, 10%	P, 10%	
Mercury, CVAA	sw	R	B,E 10%	B,E 10%	B,E,T	NA	NA	NA	P, 10%	-	P, 10%		P, 10%	P, 10%	P, 10%	
Mercury, CVAA	crs	R	B,E 10%	B,E 10%	B,E,T	NA	NA	NA	P, 10%		P, 10%		P, 5%		P, 5%	-
2,3,7,8-TCDD	613 EPA	N		R,D		NA	NA	NA	P. 5%	•	10%		P, 10%		-	
2,3,7,8-TCDD	CLP	N	•	12		NA	NA	NA	P, 5%		10%	-	P, 10%	P, 10%		
PCDD's & PCDF's	8280 SW	N	•	12	•	NA	NA	NA	P, 5%		10%	-	P, 5%	P. 5%	-	
Organochlorine Pesta/PCBs	2/88 CLP	N		72 Hr. Seq.	•	NA	NA	NA	P, 5%	•	P, 5%	•	SDG, 5%	SDG, 5%	-	
Organochlorine Pesta/PCBs	OLM01 CLP	N	12	12		NA	NA	NA	P, 5%	•	P, 5%		SDG, 5%	SDG, 5%		

[&]quot;Single Point MSA

Test	Method	Instrumental QC							Prep. QC							
	Reference	ICV01	BLADI	CCV3	CDL01	DPS01'	DUP013	SPIO1 (or MSA)	BLA02	(CLP) BLA02 SPI01	ICV02 or LCS	(CLP) LCS/ SPI01	SP102 ³	DPS02 ³	DUP023	DLCS
Organochlorine Pesta/PCBs	8080 SW	N	•	10%, 12	•	NA	NA	NA	P, 5%	-	P, 10%	•	P, 5%	P, 5%	-	-
Semi-volatiles GC/MS	2/88 CLP	N	•	12	-	NA	NA	NA	P, 5%	-	-	-	SDG, 5%	SDG, 5%	•	-
Semi-volatiles GC/MS	OLM01 CLP	N	•	12	•	NA	NA	NA	P, 5%	•	•	•	SDG, 5%	SDG, 5%	-	-
Semi-volatiles	625 EPA	N	•	R,D	-	NA	NA	•	P, 5%		R, 5%	-	P, 5%	P, 5%	-	N
Semi-volatiles	8270 SW	Ŋ		12	•	NA	NA	NA	P, 5%		N	-	P, 5%	P, 5%		N
Volatiles	2/88 CLP	2	12	12	•	SDG, 5%	•	SDG, 5%	NA	NA	-	-	NA	NA	NA	-
Volatiles	OLM01 CLP	N	12	12	-	SDG, 5%	•	SDG, 5%	NA	NA	-	-	NA	NA	NA	•
Volatiles	624 EPA	N	R,D	R,D	•	R, 5%	•	R, 5%	NA	NA	R, 5%	-	NA	NA	NA	_
Volatiles	8240/8260 SW	И	R, 12	12	-	R, 5%	-	R, 5%	NA	NA	R, 5%	•	NA	NA	NA	
Organochlorine Pests/PCBS	608 EPA	N	•	R,D	•	NA	NA	NA	P, 5%	•	P, 10%	•	P, 10%	P, 10%		-
Purgeable Halocarbons	601 EPA	NA	R,D	NA	•	R, 10%	•	R, 10%	NA	NA	R, 10%	-	NA	NA	NA	
Purgeable Halocarbons (Halogenated Volatiles)	8010 SW	N	R,D	10% R,D	•	R, 5%	-	R, 5%	NA	NA	R, 10%	•	NA	NA	NA	-
Purgeable Aromatics	602 EPA	NA	R,D	NA	-	R, 10%	-	R, 10%	NA	NA	R, 10%	•	NA	NA	NA	-
Purgeable Aromatics	8020 SW	N	R,D	10% R,D		R, 5%	•	R, 5%	NA	NA	R, 10%	•	NA	NA	NA	-
Halogenated & Aromatic Volatile Organics	601/602 EPA	NA	R,D	NA	-	R, 10%	•	R, 10%	NA	NA	R, 10%	-	NA	NA	NA	-

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Test	Method	Instrumental QC							Prep. QC								
	Reference	ICVOI	MLABI	8	CDL01	DP301,	DUP01 ³	SP101 (or MSA)	BLA02	(CLP) BLA02 SP101	or I'C2,	(CLP) LCS/ SP101	SP102 ⁵	DPS02 ³	DUP029	DLCS	
Halogenated & Aromatic Volatile Organics	8021 SW	N	R,D	10% R,D	•	R, 5%	•	R, 5%	NA	NA	R,10%	•	NA	NA	NA	•	
Acrolein, Acrylonitrile, Acetonitrile	8030 SW	N	R,D	10%, R,D	•	R, 5%	•	R, 5%	NA	NA	R, 10%	•	NA	NA	NA	•	
Hydrocarbon Scan (FID)	8000 SW 8015-Mod	N	R,D	10% R,D	•	R, 5%	•	R, 5%	NA	NA	R, 10%	•	NA	NA	NA	•	
RCRA Listed Solvents (FID)	8000 SW	N	R,D	10% R,D	-	R, 5%		R, 5%	NA	NA	R, 10%	•	NA	NA	NA	-	
Alcohols (FID)	8015 SW	N,Q	R,D	10% R,D	•	R, 5%	•	R, 5%	NA	NA	R, 10%	•	NA	NA	NA	•	
PCBs (ECD)	8080 SW	N	•	10% 12	-	NA	NA	NA	P, 10%	_	P, 10%	•	P, 5%	P, 5%	-	-	
PNAs (FID)	8100 SW	N	-	R,D	•	NA	NA	NA	P, 10%	•	P, 10%	•	P, 5%	P, 5%	-	•	
Chlorinated Hydrocarbons	8120 SW	И	-	10% R,D	-	NA	NA	NA	P, 10%	·	10%	•	P, 5%	P, 5%	-	•	
Organophos- phorus Pesticides	8141 SW	N,Q	-	10% R,D	•	NA	NA	NA	P, 10%	-	P, 10%	•	P, 5%	P, 5%	-	•	
Chlorinated Herbicides	8150 SW	И	•	10% R,D	٠	NA	NA	NA	P, 10%	•	P, 10%	,	P, 5%	P, 5%	-	-	
PNAs (HPLC)	8310 SW	N	•	10% R,D	-	NA	NA	NA	P, 10%	-	P, 10%	-	P, 5%	P, 5%		•	
T. Petroleum Hydrocarbons	418.1 EPA 503E SM	N	D	B,E 5%	-	R, 10%	-	R, 10%	NA	-	P, 10%	-	NA	NA	-	-	
Volatiles (PID/ELCD)	502.2 EPA	N	R,D	R,D		•	R, 10%	<u>-</u>	NA	NA	R, 5%	-	NA	NA	NA	Q	
EDB, DBCP	504 EPA	Д,И	R,D	NA	Weekly	R, 5%	R, 5%	R, 5%	NA	NA	R, 10%		NA	NA	NA	Q	
N & P Pesticides	507 EPA	И,Q	R,D	10% R,D	D	NA	NA	NA	P, 10%	-	P. 5%	-	P, 5%	P, 5%	-	Q	

	1 1 1 1 1	Instrumental QC							Prep. QC							
Test	Method Reference	ICV01	BLADI	CCV*	CDL01	DPS01*	DUP013	SPI01 (or MSA)	BLA02	(CLP) BLA02 SPI01	ICV02 or LCS	(CLP) LCS/ SP101	SP102 ³	DPS02 ⁸	DUP02 ⁵	DLCS
Volatiles (GC/MS)	524.2 EPA	N	R, D	D, 8	•	•	R, 10%	•	NA	NA	R, 10%		NA	NA	NA	Q
PNAs (HPLC)	610 EPA	N	-	10% R,D		NA	NA	NA	P, 10%	-	P, 10%	•	P, 5%	P, 5%	-	
Carbamates (HPLC)	531.1 EPA	Q,N	R,D	D,B,E		10%		10%	NA	NA	10%	•	NA	NA	NA	Q
Herbicides (ECD)	515.1 EPA	N		D,B,E 10%	N	NA	NA	NA	P, 10%		P, 10%	-	P, 10%	P, 10%	-	
TCLP Regulated Analytes	sw	N	R,D	10% R,D	N	10%	10%	N	P, 10%	-	10%	-	10%	10%	10%	-
Semi-Volatiles GC/MS	525 EPA	N	NA	В	NA	NA	NA	NA	P	-	P, 5%	-	P, 5%	P, 5%	-	Q

Table Abbreviations:

В	_	Beginning of run	T	-	Twice per 8 hour shift	NA	=	Not applicable
C	=	Each calibration curve (ICV01)	SW		SW-846	-	=	Not routinely performed
D	=	Daily	N	=	As needed	SM	-	Standard Methods
E	_	End of run	SDG	=	Each sample delivery group	EPA	=	Methods for Chemical Analysis of Water and Wastes
P	-	Each prep batch	12	=	Once per 12 hours	AOAC	=	Association of Official Analytical Chemists
Q	=	Quarterly	8	=	Once per 8 hours	ASTM	**	American Society for Testing Materials

Q = Quarterly
R = Each run

S = 2 times per 8 hour shift

- 1. Number of Standards Run is 1, unless noted otherwise.
- 2. CCV may be the same as ICV.
- 3. Where preps are optional and not performed then "01" QC is used. Either MS/MSD or Spike and Dup used, not both. "01" QC will be used only when "02" QC is not applicable. An analytical spike (MSA) is not required on the predigestion spike samle (SPI02).
- LCS can be "01" or "02".

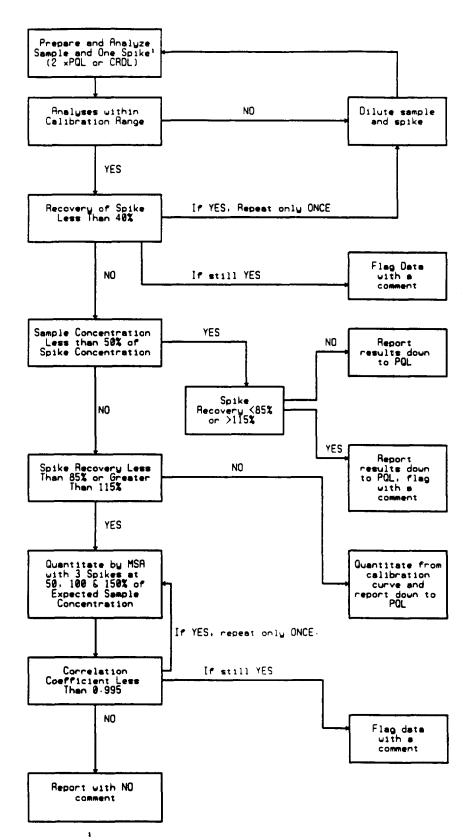
Table 11.6 METHODS USED TO GENERATE PRECISION AND ACCURACY TARGETS

QC Type (Method)	Purpose	Concentration Level	Method References					
Deplicate Matrix Spikes (DPS01, DPS02)	Precision and Accuracy	Low Level	All EPA "500"Series (Drinking Water) Organic Methods EPA "600" Series Organic Methods: 601, 602, 624 EPA: 340.2, 350.1, 350.3, 351.2, 351.3, 375.4, 410.2, 415.1, 418.1 All Furnace Metals Methods SW-846: 9020, 9038, 9060; SM: 407A					
		Mid Level	EPA "600" Series Organic Methods: 607, 608, 625 All SW-846 Organic Methods ("8000" Series) All EPA SOW (CLP) Organic Methods EPA: 325.1, 325.2, 335.1, 335.2, 365.2, 376.1, 376.2, 410.4, 420.1, 420.2 All ICP & FAA metals Methods SW-846: 7196, 9010, 9012, 9030, 9065, 9066; SM: 3128, 503E					
Matrix Spikes (SPI01, SPI02, DPS01, DPS02)	Accuracy	Low Level	All EPA "500" Series (Drinking Water) Organic Methods EPA "600" Series Organic Methods: 601, 602, 624 EPA: 340.2, 350.1, 350.3, 351.2, 351.3, 375.4, 410.2, 415.1, 418.1 All Furnace Metals Methods SW-846: 9020, 9038, 9060; SM: 407A					
		Mid Level	EPA "600" Series Organic Methods: 607, 608, 625 All SW-846 Organic Methods ("8000" Series) All EPA SOW (CLP) Organic Methods EPA: 325.1, 325.2, 335.1, 335.2, 365.2, 376.1, 376.2, 410.4, 420.1, 420.2 All ICP and FAA Metals Methods SW-846: 7196, 9010, 9012, 9030, 9065, 9066; SM: 312B, 503E					
Sample Duplicates (DUP01, DUP02)	Precision	Mid Level	All Methods					
		Low Level	All Methode (Maximum RPD=20)					
Sample Duplicates (DPS01, DPS02)	Precision	Mid Level	See "Daplicate Matrix Spikes"					
		Low Level	See "Deplicate Matrix Spikes"					

See Table 11.5

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FIGURE 11.1 METALS ANALYSIS SCHEME FOR SPIKING (SPI01)



Spike required for all GFAA, optional for FAA/ICP except EPTOX or TCLP analyses.

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12.0 Data Reduction, Validation and Reporting

12.1 Introduction

The laboratory analyst holds the key to the successful completion of valid and documentable analytical testing. Analysts must be provided with proper training, equipment and supervision so that they know and follow proper analytical procedures. It is management's responsibility to make available at the bench level, this QA Plan and it is the analysts' responsibility to know and to follow its' requirements. In addition to the QA Plan, pertinent EMS Heritage SOPs, EMS Heritage written methods and published (EPA, etc.) methods, the analysts must be fully aware and have easy access to any specific contract requirements and DQO's. Data validation will never result in acceptable data if the proper QC types are not analyzed as dictated by the specific project or contract.

12.2 Data Reduction

This section outlines how the raw data are reduced (calculated) into reportable values. Refer to Appendix E, <u>Job Descriptions</u>, for the responsibilities and specific duties of the analyst.

- 12.2.1 All lab analysts share some portion of the responsibility for maintaining documentation of the testing being performed. In addition to the various bound notebooks which are maintained documenting standard preparations, instrument maintenance, instrument performance, logbook of logbooks, etc., analysts must maintain some combination of bound notebooks and/or original raw data printouts with bench sheets indicating the analytical run contents, sequence and results. Some documentation requirements under various testing programs may make the use of loose bench sheets in lieu of bound sequentially numbered notebooks an unacceptable procedure, in which case bound notebooks will be utilized exclusively. Exclusive use of bound notebooks may necessitate the use of loose sheets for data entry of results and associated information into the LIMS system.
- 12.2.2 All analytical run notebooks must contain the following information at a minimum:
 - 1. Analyst identification
 - 2. Date of analysis

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- 3. Instrument identification
- 4. Analytical method (Edition or Version) used and any deviations
- 5. Date of initial calibration
- 6. Computer file name if applicable
- 7. Concentration and source identification of all standards and spikes
- 8. EMS sample I.D. number
- 9. Reference to prep performed
- 10. Any comments pertinent to the analysis or sample conditions (homogeneity)
- 11. Any dilutions, concentrations, sample manipulations (initial and final weights/volumes, etc.)

Calculation of results from raw data if not contained in the bound lab notebook, will be maintained and filed along with the original hard copy of instrument printouts.

- 12.2.3 When data entry of results into LIMS is required (direct data file transfer not available), it will be necessary to summarize the final results and all information needed for data entry. A combination of the analytical run notebook, the instrument printouts/chromatograms, calculation work sheets, prep bench sheets/notebooks and summary of final results will be sufficient to re-construct the sample analysis. This simple final result summary will include at a minimum:
 - 1. Analyst identification
 - 2. Date of analysis
 - 3. Instrument identification
 - 4. Analytical method used
 - 5. Computer file names

- 6. EMS sample I.D. number
- 7. Comment to appear on the final report
- 8. Any dilutions or concentrations involved
- 9. Final results with proper units and detection limits
- 10. Reviewer initials and date of review
- 11. Data entry operator and date of data entry
- 12.2.4 All instrument printouts/chromatograms must contain the following at a minimum:
 - 1. Analysts' initials and date of review of hard copy
 - 2. Sample or standard identification (and conc.)
 - 3. Computer file numbers
 - 4. Dilutions or concentrations
 - 5. Date and time of injection or analysis
 - 6. Units (μ g/L, mg/L, mg/kg, etc.)
 - 7. Identification of samples that are spiked or duplicated (in addition to computer assigned "Q" numbers).
 - 8. Peaks chosen for quantitation and their identification.
- 12.2.5 Some testing will generate only bench sheets when no instrument printout is possible or practical and no bound notebook is maintained. The following tests are <u>directly read</u> from instruments:
 - 1. pH and ISE methods
 - 2. Manual spectrophotometric methods
 - 3. Gravimetric methods
 - 4. Coliform bacteria

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- 5. Heat of combustion
- 6. Total organic halides
- 7. Temperature tests (i.e. Flash Point, Boiling Point)
- 8. Titrations
- 9. Turbidity
- 10. Biochemical oxygen demand (BOD)
- 11. Specific conductance

In those cases this bench sheet must contain all information required in the analytical run notebook previously discussed <u>and</u> the following at a minimum:

- 1. Reviewer initials and date of review
- 2. Data entry operator initials and date of entry
- 3. Page numbers
- 4. Calculated final results, %R of spikes, RPD of duplicates, etc.
- 5. Any raw data used to arrive at the final result
- 6. Results of the calibration curve (linear regression, log transforms, etc.)
- 12.2.6 All calibration curves conforming to a linear equation will be fitted by the least squares linear regression technique. This procedure will also be used for calibration curves that require log transforms or similar transforms. The formula for calculation of the regression line is as follows:

Where:

Y = mX + b

Y = Point on the y axis

m = Slope of line

X = Point on the x axis

b = y intercept

The least squares estimate of b and m uses the following equations:

$$m = \frac{\sum XY - \frac{\sum X\sum Y}{n}}{\sum X^2 - \frac{(\sum X)^2}{n}}$$

$$b = \overline{Y} - m\overline{X}$$

Other scientifically sound procedures/data reduction techniques may be used as permitted by the Chief Chemist and the QA Officer.

12.2.7 When using the <u>external standard</u> calibration technique the average calibration factor (CF) can be used in place of a calibration curve (non-linear) when the %RSD is less than 20% over the working range and linearity through the origin can be assumed. The calibration factor (CF) is calculated as follows:

or
$$RF(Response\ Factor) = \frac{Mass\ injected\ (in\ nanograms)}{Total\ Area\ of\ Peak*}$$

*For multi-response pesticides/PCBs, gasoline, diesel, etc., use the total area of all peaks used for quantitation.

The concentration of each analyte in the sample may be determined by calculating the amount of standard purged or injected, from the peak response, using the calibration curve or the calibration factor (CF) above. The concentration of a specific analyte is calculated as follows:

Aqueous samples:

Concentration (ug/L) =
$$[(A_x)(A)(V_i)(D)]/[(A_s)(V_i)(V_s)]$$

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where:

 A_x = Response for the analyte in the sample, units may be in the area counts or peak height.

A = Amount of standard injected or purged, ng.

 A_x = Response for the external standard, units same as for A_x .

 V_i = Volume of extract injected, μL . For purge-and-trap analysis, V_i is not applicable and therefore = 1.

D = Dilution factor, if dilution was made on the sample prior to analysis. If no dilution was made, D = 1, dimensionless.

 V_t = Volume of total extract, μL . For purge-and-trap analysis, V_t is not applicable and therefore = 1.

 V_s = Volume of sample extracted or purged, mL.

By inspection of the above formula one observes that the concentration in the extract is:

$$(A_x)(RF)$$

where:

$$RF(ResponseFactor)^* = \frac{A}{A_s}$$

*RF is the inverse of CF.

Nonaqueous samples:

Concentration
$$(ng/g) = [(A_x)(A)(V_i)(D)]/[(A_s)(V_i)(W)]$$

where:

W = Weight of sample extracted or purged, g. The wet weight or dry weight may be used, depending upon the specific applications of the data.

 A_x , A_s , A, V_t , D, and V_i have the same definition as for aqueous samples.

12.2.8 When using the internal standard technique (all GC/MS analyses) the average response factor (RF) can be used in place of a calibration curve (non-linear) when the %RSD is less than 20% over the working range and linearity through the origin can be assumed. The response factor (RF)* is calculated as follows:

$$RF = (A_sC_{is})/(A_{is}C_s)$$

where:

 A_s = Response for the analyte to be measured.

 A_{is} = Response for the internal standard.

 C_{is} = Concentration of the internal standard, $\mu g/L$.

 C_s = Concentration of the analyte to be measured, $\mu g/L$.

*Also known as "Relative Response Factor" or RRF

The concentration of a specific analyte is calculated as follows:

Aqueous samples:

Concentration
$$(\mu g/L) = [(A_x)(C_{is})(D)]/[(A_{is})(RF)(V_s)]$$

where:

 A_x = Response of the analyte being measured, units may be in area counts or peak height.

 C_{is} = Amount of internal standard added to extract or volume purged, ng.

D = Dilution factor, if a dilution was made on the sample prior to analysis. If no dilution was made, D = 1, dimensionless.

 A_{is} = Response of the internal standard, units same as A_{x} .

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RF = Response factor for analyte.

 $V_s = Volume of water extracted or purged, mL.$

Nonaqueous samples:

Concentration $(\mu g/kg) = [(A_s)(C_{is})(D)]/[(A_{is})(RF)(W_s)]$

where:

W_s = Weight of sample extracted, g. Either a dry weight or wet weight may be used, depending upon the specific application of the data.

 A_{i} , C_{ii} , D, A_{ii} , and RF have the same definition as for aqueous samples.

The RF from the daily standard (CCV) analysis is used to calculate the concentration in the sample unless specific methods dictate otherwise.

12.2.9 Temperature variations for pH are negated by simply allowing solutions to warm-up to room temperature or using an automatic temperature compensation probe. When testing for specific conductance, the operator may use the standard solution and the table to check accuracy of the cell's constant or to determine an unknown constant. The formula is:

$$K = k_1 + k_2$$

 $\frac{1}{R}$

where,

K = cell constant in c.g.s. metric units (cm⁻¹)

k1 = conductivity from Table 12.6

k2 = blank conductivity

1/R = measured conductance of KC1 standard on meter

TABLE 12.6 TEMPERATURE vs CONDUCTIVITY 0.01N KC1 SOLUTION

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TEMPERATURE (°C)	CONDUCTIVITY (ABSOLUTE MICROMHOS/CM)
15	1141.5
16	1167.5
17	1193.6
18	1219.9
19	1246.4
20	1273.0
21	1299.7
22	1326.6
23	1353.6
24	1380.8
25	1408.1
26	1435.6
27	1463.2
28	1490.9
29	1518.7
30	1546.7

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1/R, k1 and k2 must either be determined at the same temperature, or corrected to the same temperature to make the equation valid.

To calculate sample value, use the following equation:

Conductivity of Sample x Cell Constant
$$1 + 0.0191 (t-25)$$

Table 12.7 lists denominator values (1+0.0191 [t-25]) for temperatures ranging from 17°C to 28°C.

12.2.10Manual data entry is the responsibility of the Data Entry Supervisor who will assure that data entry staff have knowledge of the system in order that complete analytical information is entered accurately, in the correct format and labeled consistently. Data entry staff will not perform any calculations; all data entry items will be summarized and submitted on standard forms which have been reviewed by the responsible group leader for acceptability, accuracy and completeness.

All analyses in the data system are based on the "analytical run" concept. All data (prep or analytical, is grouped in the computer, exactly in the order analyzed (as it also appears on a benchsheet). All QC is included in the analytical sequence.

Automated data transfer is the responsibility of each respective Group Leader.

12.3 <u>Data Validation - During Collection</u>

The principal criteria to be used to validate data integrity during collection will be the following:

- 1. Calibration Criteria (Linearity, RF etc.)
- 2. Internal Standard Areas (organics)
- 3. Calibration Blank Value (CAL01)
- 4. Method Prep Blank Value (BLA02)
- 5. Reagent Blank Value (BLA01)

TABLE 12.7

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TEMPERATURE	Km (C) (Denominator of Sample equ.)
17°C	0.8472
18°C	0.8663
19°C	0.8854
20°C	0.9045
21°C	0.9236
22°C	0.9427
23°C	0.9618
24°C	0.9809
25°C	1.00
26°C	1.0191
27°C	1.0382
28°C	1.0573

- 6. Result of calibration verification standard used to verify calibration (ICV01).
- 7. Results of (matrix) spiked non QC samples (DPS01, SPI01, SPI02, DPS02).
- 8. Results of surrogate recoveries (organics).
- 9. Results of replicate measurements (DUP01, DUP02, DPS02).
- 10. Results of Continuing Calibration Verification standard (CCV).
- 11. Results of Laboratory Fortified Blanks and Laboratory Control Samples or QC Check Samples (LCS, DLCS, ICV02).
- 12.3.1 These measurements are to be made by the analyst who, using specific acceptance criteria, will either proceed with analyses or take corrective action. All QC data must be reviewed by the Group Leader to insure that all QC types have been completed and meet acceptance criteria. Group Leaders must also:
 - 1. Check raw data entries and calculations.
 - 2. Check extraction logs and benchsheets.
 - 3. Check Calibration integrity.
 - 4. Check Instrument/analytical logbooks.

The Group Leaders are responsible for assuring that all bench sheets submitted for data entry are accurate and adequate to allow for the rapid and error free recording of data into LIMS.

- 12.3.2 Group Leaders and the Sample Custodian are also responsible for checking and assuring that all internal chain of custody requirements and documentation of such are met.
- 12.3.3 In summary, Group Leaders will review all bench sheets or release computer files to LIMS after review. In addition 20% of all raw data will be reviewed by the Group Leaders to insure that the method was in-control during the analytical run. Group Leaders are responsible for initialing and submitting data for data entry upon completion of

their review. This completes the first step in review of the data integrity and data validity.

12.4 <u>Data Validation - Post Completion</u>

Subsequent to data entry, the Quality Assurance Unit is responsible for comparing bench sheets with actual data entered into LIMS. Such review will occur for 2-5% of all data entered. This activity will be documented and retained on file.

Refer to Figure 12.1, <u>Data Review Protocol</u> for the procedures used at EMS Heritage for data review and validation. The QA Officer and/or the QA Unit must also:

- 1. Confirm that the goals for precision and accuracy on duplicate and spiked compounds are sufficient to achieve the goal of 99% completeness.
- 2. Confirm that all analytical QC precision and accuracy (the measurement system, not sample (matrix) related) is 100% acceptable or that the run has been rejected and corrective action taken.
- 3. Confirm that all client specific or contract specific QC requirements have been met.
- 4. Examine at least 5% of the raw data (e.g. chromatograms, AAS recorder outputs, burn records) on a periodic basis to verify adequacy of documentation, confirm peak shape and resolution, and to assure that the instrument system is sufficiently sensitive and responsive.

12.4.1 Internal and External Assessments

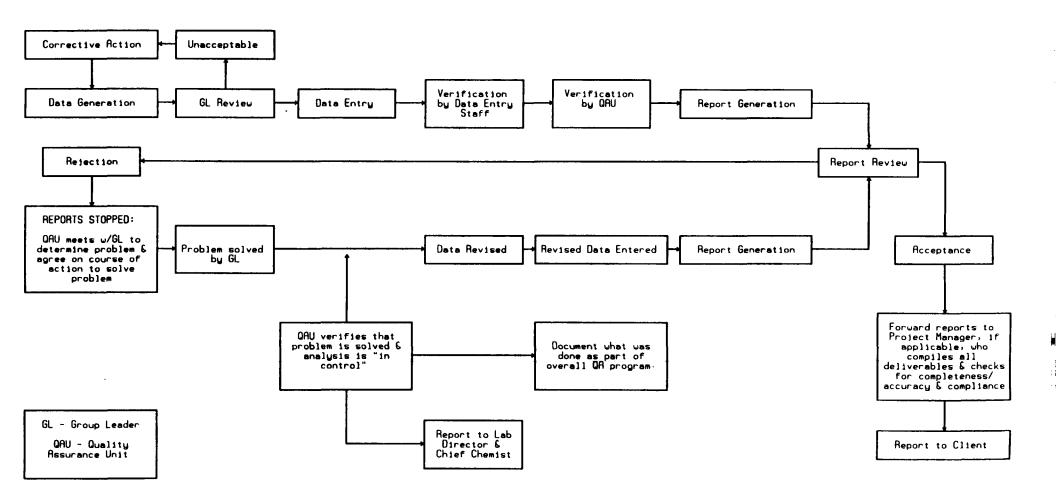
External assessment of the adherence to the QA/QC policy and procedures is the responsibility of the Quality Assurance Officer. Audit forms have been developed in the QAP and are used at all EMS Heritage Laboratories to evaluate their adherence to QA/QC Procedures. External assessments are also accomplished via third party audits.

Internal assessment of the compliance to company QA/QC policies and procedures is conducted by the Laboratory Director and the Quality Assurance Unit or their designee. Review of documentation, use of blind duplicates and standards and monthly reference materials shall be used to evaluate the performance of each laboratory. Refer

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FIGURE 12.1 EMS LABORATORIES

DATA REVIEW PROTOCOL



to Section 14.0, <u>Performance and Systems Audits</u> for the details of internal and external assessments.

12.4.2 Quality Control Policy and Procedures

The EMS Heritage Laboratories, Inc. QAP is mandatory at all laboratories. Specific procedures are required and regular quality control audits are performed to assure compliance to company policy. Quarterly quality control reports are required from the QA Units submitted through the QA Officer. The overall quality of analysis shall be compiled and summarized by the QA Officer, who will report this summary to the President of EMS Heritage. Acceptable levels of performance are presented and monitored by the QA Officer. Refer to Section 15.0, Quality Assurance Reports for further details.

12.5 <u>Laboratory Quality Control Checks</u>

The types and frequency of laboratory QC Checks are presented in Table 11.5, Analytical Run Requirements/Frequency. This table presents basically three categories of QC:

- 1. Instrument Calibration QC
- 2. Instrumental Method QC (tests w/no prep. or analyzed w/o prep)
- 3. Analytical samples QC processed through all steps (prepped QC).

Instrument Calibration criteria must be met before any analyses are allowed to proceed; those criteria are given in Table 9.5, <u>Instrument Calibration</u>. Internal QC checks not related to calibration are discussed below.

12.5.1 Preparation or Method Blanks

Blanks are identified at EMS Heritage as either BLA01 or BLA02 (calibration blanks are treated as a calibration data point, CAL01). Those methods that do not differentiate a prep from the analysis, i.e. purge and trap volatile analyses, are labeled as BLA01, which is equivalent to a BLA02 for tests that do have a separable prep and analysis. The following control limits apply for all method blanks (BLA02) and reagent blanks (BLA01):

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If sample results are greater than 10 times the method blank value or the sample results are below the detectable level, the analysis is acceptable. Otherwise, the following criteria will apply.

1. Organic Analytes: Less than the Reporting Detection level (PQL)

Exceptions to the rule will be for common lab solvents not to exceed 5 times the PQL, including but not limited to the following:

- 1. Acetone
- 2. Methylene Chloride
- 3. Toluene
- 4. Methyl Ethyl Ketone (MEK)
- 5. Methyl Isobutyl Ketone (MIBK)
- 6. Freon(s)
- 7. Phthalates
- 2. Inorganic Analytes: Less than the Reporting Detection Level (PQL)

Exceptions to the rule will be for the following metals - not to exceed 3 times the PQL:

- 1. Iron
- 2. Nickel
- 3. Zinc
- 4. Copper
- 12.5.2 Any target compounds in the blank will be reported. Any sample containing a similar amount of the compound outside the acceptable range will be re-analyzed when the source of the contamination is identified and removed.

Target compounds found to be BDL (Below Detection Limit) in the blank will be entered into the data base at 0.4 times the MDL.

12.5.3 Matrix Spikes

Matrix spikes are included as a routine protocol according to the frequency given Table 11.5, Analytical Run Requirements/Frequency. Matrix spikes serve only to show that an individual sample does or does not exhibit matrix interferences using the prescribed method; they do not definitively demonstrate that a given analytical method is out of control. Other QC types will be utilized to demonstrate that an analytical method is in control (QC check sample, LCS, etc.).

- 12.5.3.1 There are basically two types of spikes utilized and they are as follows:
 - 1. Pre-digestion/extraction spikes
 - 2. Post digestion spikes

Pre-digestion/extraction spikes are performed for all applicable analyses at the required (Table 11.5) frequency. The goals for accuracy are listed in Table 5.1, QA Targets for Precision, Accuracy and Method Detection Limits. Those control limits are determined from actual data entered into the QCTS when a sufficient number of observations (minimum of 7) are available to determine meaningful statistical limits.

Accuracy will be a function of the spiking level chosen. Many methods and testing programs specify the spiking levels to be used; EMS Heritage will use those spiking levels where specified. As a general rule, spiking will be performed at approximately 10 to 20 times the MDL unless other factors preclude that choice.

Several methods have established multi-laboratory performance based control limits for spike recovery of specific matrices (water and soil) through the USEPA. While Table 5.1 contains actual EMS Heritage developed control limits, the control limits used will be those determined by the USEPA for those equivalent analyses (assumes similar spiking levels used). Surrogate

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matrix spikes will be treated in the same manner for organic analyses. Refer to the specific methods for further explanation.

- 12.5.3.2 The spike (%R) recovery <u>OC Acceptance Criteria</u> for those organic methods providing specific limits are as found in Table 12.1, <u>Organic Spike Recovery Criteria</u>:
- 12.5.3.3 The spike recovery (%R) QC Acceptance Criteria for those inorganic methods providing specific limits are given in Table 12.2, Inorganic Spiking Levels and Control Limits. Those criteria are the minimum acceptable criteria, stricter limits may be set by contractual requirements and/or QA Units. Analytes not listed in Table 12.2 will utilize the control limits established in Section 5.0, Table 5.1 where a sufficient number of observations exist, otherwise the control limit of 75-125% Matrix Spike Recovery will apply.
- 12.5.3.4 Post-digestion spikes (SPI01) are primarily used in metals analyses and are also referred to as the Method of Standard Additions (MSA). Two versions of this MSA are referenced in SW-846, Third Edition. The full method of standard additions generates a 4 point curve from which final results can be calculated. Another version of MSA is the Single-Point MSA, which essentially is a sample and matrix specific spike recovery. Section 11.0, Figures 11.1, Metals Analysis Scheme For Spiking and 11.2, CLP-GFAA Analysis Scheme define the criteria used to determine the application of those techniques.

The spike recovery criteria for a post-digestion spike is as follows:

SPI01 Criteria: 85-115% Recovery

Those analytical samples not meeting the above criteria will follow the procedures outlined in either Figure 11.1 and/or Figure 11.2.

TABLE 12.1 ORGANIC SPIKE RECOVERY CRITERIA

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Volatile Methods: SW-846-8240, CLP-VOA			
Parameter	Water, Range %R	Soil, Range %R	Spike Amount
Benzene	76-127	66-142	50 μg/L
Chlorobenzene	75-130	60-133	50 μg/L
1,1-Dichloroethene	61-145	59-172	50 μg/L
Toluene	76-125	59-139	50 μg/L
Trichloroethene	71-120	62-137	50 μg/L

Volatile Methods: SW-846 8240*, EPA 624		
Parameter	Water, Range %R	
Benzene*	37-151	
Bromodichloromethane	35-155	
Bromoform	45-169	
Bromomethane	D-242	
Carbon tetrachloride	70-140	
Chlorobenzene*	37-160	
Chloroethane	14-230	
2-Chloroethylvinyl ether	D-305	
Chloroform	51-138	
Chloromethane	D-273	
Dibromochloromethane	53-149	
1,2-Dichlorobenzene	18-190	
1,3-Dichlorobenzene	59-156	
1,4-Dichlorobenzene	18-190	
1,1-Dichloroethane	59-155	
1,2-Dichloroethane	49-155	
1,1-Dichloroethene*	D-234	
trans-1,2-Dichloroethene	54-156	
1,2-Dichloropropane	D-210	

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TABLE 12.1 ORGANIC SPIKE RECOVERY CRITERIA

Volatile Methods: SW-846 8240*, EPA 624		
cis-1,3-Dichloropropene	D-227	
trans-1,3-Dichloropropene	17-183	
Ethyl Benzene	37-162	
Methylene chloride	D-221	
1,1,2,2-Tetrachloroethane	46-157	
Tetrachloroethene	64-148	
Toluene*	47-150	
1,1,1-Trichloroethane	52-162	
1,1,2-Trichloroethane	52-150	
Trichloroethene*	71-157	
Trichlorofluoromethane	17-181	
Vinyl chloride	D-251	

D = Detected

*These analytes for method 8240 and CLP will use tighter limits established by RCRA and CLP: see previous limits.

Volatile Methods: EPA 601, 602; SW-846 8010, 8020		
Parameter Water, Range %R		
Bromodichloromethane	42-172	
Bromoform	13-159	
Bromomethane	D-144	
Carbon tetrachloride	43-143	
Chlorobenzene	38-150	
Chloroethane	46-137	
2-Chloroethylvinyl ether	14-186	
Chloroform	49-133	
Chloromethane	D-193	
Dibromochloromethane	24-191	
1,2-Dichlorobenzene	D-208	

TABLE 12.1 ORGANIC SPIKE RECOVERY CRITERIA

Volatile Methods: EPA 601, 602; SW-846 8010, 8020		
Parameter	Water, Range %R	
1,3-Dichlorobenzene	7-187	
1,4-Dichlorobenzene	42-143	
1,1-Dichloroethane	47-132	
1,2-Dichloroethane	51-147	
1,1-Dichloroethene	26-167	
trans-1,2-Dichloroethene	38-155	
1,2-Dichloropropane	44-156	
cis-1,3-Dichloropropene	22-178	
trans-1,3-Dichloropropene	22-178	
Methylene chloride	25-162	
1,1,2,2-Tetrachloroethane	8-184	
Tetrachloroethene	26-162	
1,1,1-Trichloroethane	41-138	
1,1,2-Trichloroethane	39-136	
Trichloroethene	35-146	
Trichlorofluoromethane	21-156	
Vinyl chloride	28-163	
Benzene	39-150	
Ethylbenzene	32-160	
Toluene	46-148	
Chlorobenzene	55-136	
1,2-Dichlorobenzene	37-154	
1,3-Dichlorobenzene	50-141	
1,4-Dichlorobenzene	42-143	

D = Detected

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TABLE 12.1 ORGANIC SPIKE RECOVERY CRITERIA

Semi-Volatile Methods: SW-846 8270, CLP-SVOA			
Parameter	Water, Range %R	Soil, Range %R	
Phenol	12-89	26-90	
2-Chlorophenol	27-123	25-102	
1,4-Dichlorobenzene	36-97	28-104	
N-nitroso-di-n-propylamine	41-116	41-126	
1,2,4-Trichlorobenzene	39-98	38-107	
4-Chloro-3-methylphenol	23-97	26-103	
Acenaphthene	46-118	31-137	
4-Nitrophenol	10-80	11-114	
2,4-Dinitrotoluene	24-96	28-89	
Pentachlorophenol	9-103	17-109	
Pyrene	26-127	35-142	

Semi-Volatile Methods: SW-846 8270*, 625		
Parameter	Water, Range %R	
Acenaphthene	47-145*	
Acenaphthylene	33-145	
Aldrin	D-166	
Anthracene	27-133	
Benzo(a)anthracene	33-143	
Benzo(b)fluoranthene	24-159	
Benzo(k)fluoranthene	11-162	
Benzo(a)pyrene	17-163	
Benzo(ghi)perylene	D-219	
Benzyl butyl phthalate	D-152	
δ-ВНC	24-149	
у-ВНС	D-110	
Bis(2-chloroethyl)ether	12-158	

TABLE 12.1 ORGANIC SPIKE RECOVERY CRITERIA

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Semi-Volatile Methods: SW-846 8270*, 625		
Parameter	Water, Range %R	
Bis(2-chloroethoxy)methane	33-184	
Bis(2-chloroisopropyl)ether	36-166	
Bis(2-ethylhexyl)phthalate	8-158	
4-Bromophenyl phenyl ether	53-127	
2-Chloronaphthalene	60-118	
4-Chlorophenyl phenyl ether	25-158	
Chrysene	17-168	
4,4'-DDD	D-145	
4,4'-DDE	4-136	
4,4'-DDT	D-203	
Dibenzo(a,h)anthracene	D-227	
Di-n-butyl phthalate	1-118	
1,2-Dichlorobenzene	32-129	
1,3-Dichlorobenzene	D-172	
1,4,-Dichlorobenzene	20-124*	
3,3'-Dichlorobenzidine	D-262	
Dieldrin	29-136	
Diethyl phthalate	D-114	
Dimethyl phthalate	D-112	
2,4-Dinitrotoluene	39-139	
2,6-Dinitrotoluene	50-158	
Di-n-octylphthalate	4-146	
Endosulfan sulfate	D-107	
Endrin aldehyde	D-209	
Fluoranthene	26-137	
Fluorene	59-121	
Heptachlor	D-192	

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TABLE 12.1 ORGANIC SPIKE RECOVERY CRITERIA

Semi-Volatile Methods: SW-846 8270*, 625		
Parameter	Water, Range %R	
Heptachlor epoxide	26-155	
Hexachlorobenzene	D-152	
Hexachlorobutadiene	24-116	
Hexachloroethane	40-113	
Indeno(1,2,3-cd)pyrene	D-171	
Isophorone	21-196	
Naphthalene	21-133	
Nitrobenzene	35-180	
N-nitrosodi-n-propylamine	D-230*	
PCB-1260	D-164	
Phenanthrene	54-120	
Pyrene	52-115*	
1,2,4-Trichlorobenzene	44-142*	
4-Chloro-3-methylphenol	22-147*	
2-Chlorophenol	23-134*	
2,4-Dichlorophenol	39-135	
2,4-Dimethylphenol	32-119	
2,4-Dinitrophenol	D-191*	
2-Methyl-4,6-dinitrophenol	D-181	
2-Nitrophenol	29-182	
4-Nitrophenol	D-132*	
Pentachlorophenol	14-176*	
Phenol	5-112*	
2,4,6-Trichlorophenol	37-144	

D = Detected

^{*}These analytes for method 8240 & CLP will use tighter limits established by RCRA, CLP-see previous list limits

TABLE 12.1 ORGANIC SPIKE RECOVERY CRITERIA

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Organochlorine Pesticides/PCBs: SW-846 8080, CLP-Pest			
Parameter	Water, Range %R	Soil, Range %R	
gamma-BHC (Lindane)	56-123	46-127	
Heptachlor	40-131	35-130	
Aldrin	40-120	34-132	
Dieldrin	52-126	31-134	
Endrin	56-121	42-139	
4,4'-DDT	38-127	23-134	

Organochlorine Pesticides/Po	CBs: EPA 608; SW-846 8080
Parameter	Water, Range %R
Aldrin	42-122*
α-ВНС	37-134
<i>β</i> -ВНС	17-147
δ-ВНС	19-140
у-ВНС	32-127*
Chlordane	45-119
4,4'-DDD	31-141
4,4'-DDE	30-145
4,4'-DDT	25-160*
Dieldrin	36-146*
Endosulfan I	45-153
Endosulfan II	D-202
Endosulfan Sulfate	26-144
Endrin	30-147*
Heptachlor	34-111*
Heptachlor epoxide	37-142
Toxaphene	41-126

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TABLE 12.1 ORGANIC SPIKE RECOVERY CRITERIA

Organochlorine Pesticides/PCBs: EPA 608; SW-846 8080				
Parameter	Water, Range %R			
PCB-1016	50-114			
PCB-122	15-178			
PCB-1232	10-215			
PCB-1242	39-150			
PCB-1248	38-158			
PCB-1254	29-131			
PCB-1260	8-127			

D = Detected

^{*}These analytes for method 8240 & CLP will use tighter limits established by RCRA, CLP-see previous list limits

TABLE 12.2 INORGANIC SPIKING LEVELS AND CONTROL LIMITS

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Element		For I	CP/AA			For F	urnace AA		Other ⁽¹⁾	Control Limits
	Water (ug/L)	Soil ⁽²⁾ (mg/kg)	Soil ⁽³⁾ (mg/kg)	Soil ⁽⁴⁾ (mg/kg)	Water (ug/L)	Soil ⁽²⁾ (mg/kg)	Soil ⁽³⁾ (mg/kg)	Soil ⁽⁴⁾ (mg/kg)	(ug/L)	% Recovery
Aluminum	2,000	400*	400*	200*						75-125%R
Antimony ³	500	100	100	50	100	20	20	10		75-125%R
Arsenic	2,000	400	400	200	40	8	8	4		75-125%R
Barium	2,000	400	400	200						75-125%R
Beryllium	50	10	10	5						75-125%R
Cadmium	50	10	10	5	2	0.40	0.40	0.20		75-125%R
Calcium	100,000*	20,000*	20,000*	10,000*						75-125%R
Chromium	200	40	40	20	20					75-125%R
Cobalt	500	100	100	50						75-125%R
Copper	250	50	50	25	20					75-125%R
Iron	1,000	200*	200*	100						75-125%R
Lead	500	100	100	50	20	4	4	2		75-125%R
Magnesium	100,000*	20,000*	20,000*	10,000*						75-125%R
Manganese	500	100	100	50						75-125%R
Mercury									1/s/s/s=0.25	75-125%R
Nickel	500	100	100	50						75-125%R
Potassium	100,000*	20,000*	20,000*	10,000*						75-125%R
Selenium	2,000	400	400	200	10	2	2	1		75-125%R
Silver	50	10	10	5						75-125%R
Sodium	100,000*	20,000*	20,000*	10,000*						75-125%R

TABLE 12.2 INORGANIC SPIKING LEVELS AND CONTROL LIMITS

Element For ICP/AA		For Furnace AA			Other ⁽¹⁾	Control Limits				
	Water (ug/L)	Soil ⁽²⁾ (mg/kg)	Soil ⁽³⁾ (mg/kg)	Soil ⁽⁴⁾ (mg/kg)	Water (ug/L)	Soil ⁽²⁾ (mg/kg)	Soil ⁽³⁾ (mg/kg)	Soil ⁽⁴⁾ (mg/kg)	(ug/L)	% Recovery
Thallium	3,000	600	600	300	50	10	10	5		75-125%R
Vanadium	500	100	100	50						75-125%R
Zinc	500	100	100	50						75-125%R
Cyanide									100 ⁽⁵⁾	

^{*}No spike required. NOTE: Elements without spike levels and not designated with an asterisk, must be spiked at appropriate levels.

³Used for USEPA CLP SOW 7/88 or equivalent. The levels shown indicate concentrations in the final digestate of the spiked sample (100 mL for mercury and 200 mL for all other metals) when the wet weight of 1 gram (for ICP, Furnace and Flame AA), or 0.40 grams (for mercury) of sample is taken for analysis.

⁴Used for non-CLP analyses. The levels shown indicate concentrations in the final digestate of the spiked sample (100 mL for mercury and 100 mL for all other metals) when the wet weight of 1 gram (for ICP, Furnace, and Flame AA), or 0.40 grams (for mercury) of samples is taken for analysis.

⁵The level shown indicates the amount of cyanide that must be added to the original (undistilled) sample. for instance, 100 ug must be added per each Liter of aqueous sample. If the sample volume is 500 mL, then 50 ug of cyanide must be added. If the volume if 50 mL, then 5 ug of cyanide must be added.

For soil samples, 25 ug of cyanide must be added per each gram of solid sample taken for analysis. The spiking level is dependent on the weight of the sample taken and the final distillate volume. If one gram of sample is taken for analysis, and the final distillate volume is 250 mL, then the distillate must contain cyanide at a concentration of 100 ug/L. If five grams of sample are taken, then the distillate must contain cyanide at a concentration of 500 ug/L. Assuming a sample of one gram, the manual and semi-automated colorimetric methods call for a cyanide concentration of 25 ug per the 500 mL mixture of the sample, reagents and water before distillation. The final distillate, in this case, contains cyanide at a concentration of 100 ut/L. Fo the midi-distillation method, a cyanide concentration of 25 ug must be added into the 50 mL mixture of sample reagents, and water before distillation. This yields a cyanide concentration of 500 ug/L in the final distillate of 50 mL.

¹Cyanide spiking level reported is for both water and soil/sediment matrices.

²Used for USEPA CLP ILM01. The levels shown indicate concentrations in the final digestate of the spiked sample (100 mL for mercury and 200 mL for all other metals when the wet weight of 1 gram (for ICP, Furnace, and Flame AA), or 0.40 grams (for mercury) of sample is taken for analysis. Adjustment must be made to maintain these spiking levels when the weight of sample taken deviates by more than 10% of these values. Appropriate adjustment must be made for microwave digestion procedure where 0.5 grams of sample or 50.0 mL (45.0 mL of sample plus 5.0 mL of acid) of aqueous sample are required for analysis.

12.5.4 Duplicate Matrix Spikes (DPS01, DPS02)

Duplicate matrix spikes measure both accuracy (see previous section for matrix spikes) and precision. As with matrix spikes, USEPA has established control limits for precision of specified methods for waters and soils. Those control limits are given as Relative Percent Difference (RPD) and will be used as control limits by EMS Heritage when applicable. When the methods do not specify RPD control limits, Table 5.1 of this QAP will be utilized when a sufficient number of observations allow for usable statistical limits. Any method not falling into the above two categories will be allowed a maximum of 50 percent RPD for organic testing and 20 percent RPD for inorganic testing. RPD's exceeding these limits may be considered as out-of-control for Duplicate Matrix Spike analyses.

Methods Without Established Control Criteria:

RPD Inorganic Criteria: 0-20 RPD Organic Criteria: 0-50

12.5.4.1 The Precision Control Limits for those organic methods providing specific limits are given in Table 12.3, Organic Precision Control Limits.

12.5.5 Reagent Water or Reagent Matrix Spikes (LCS, DLCS)

Reagent water or reagent matrix spikes will be used as additional QC checks to monitor the effectiveness of the method. These QC types will utilize either reagent water spiked w/analyte (also known as Laboratory Fortified Blank) or a blank matrix (water, soil, sand, oil) which is known to exhibit minimal or no matrix interferences. EMS Heritage uses the acronyms of LCS (Laboratory Control Sample), DLCS (Duplicate Laboratory Control Sample) and/or ICV02 (Initial Calibration Verification - Prepped Standard) to identify the variations of these check samples. Analysis of this QC type will demonstrate that the method is in control when the Matrix Spike/Matrix Spike Duplicate (MS/MSD) recoveries are out-of-control. These QC types must be extracted/digested (if applicable) in the same preparation batch as the samples failing to meet QC criteria. Analysis may be optional for some organic tests when QC criteria for MS/MSD samples are met.

12.5.5.1 The reagent water or reagent matrix spikes (Laboratory Control Sample) are entered into the QCTS in LIMS as are all QC types. In addition, these QC samples are kept on control charts at the bench level. The minimum

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TABLE 12.3 ORGANIC PRECISION CONTROL LIMITS

Volatil	e Methods: SW-846 8240, CL	P-VOA				
Parameter RPD, Water RPD, Soil						
Benzene	11	21				
Chlorobenzene	13	21				
1,1-Dichloroethene	14	22				
Toluene	13	21				
Trichloroethene	14	24				

Semi-Volatile Methods: SW-846 8270, CLP-SVOA					
Parameter	RPD, Water	RPD, Soil			
Phenol	42	35			
2-Chlorophenol	40	50			
1,4-Dichlorobenzene	28	27			
N-Nitroso-di-n- propylamine	38	38			
1,2,4-Trichlorobenzene	28	23			
4-Chloro-3-methylphenol	42	33			
Acenapthene	31	19			
4-Nitrophenol	50	50			
2,4-Dinitrotoluene	38	47			
Pentachlorophenol	50	47			
Ругепе	31	36			

TABLE 12.3 ORGANIC PRECISION CONTROL LIMITS

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Organochlorine Pesticides/PCBs: SW-846 8080, CLP-Pest							
Parameter RPD, Water RPD, Soil							
gamma-BHC (Lindane)	15	50					
Heptachlor	20	31					
Aldrin	22	43					
Dieldrin	18	38					
Endrin	21	45					
4,4'-DDT	27	50					

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QC criteria to be utilized for these QC types will be the same criteria as found in Table 5.1 for matrix specific precision and accuracy. More stringent criteria will be utilized where available. As a general rule the following criteria will apply for the Laboratory Control sample:

(LSC, DLCS, ICV02): 80-120% Recovery (Accuracy)
DLCS: 20% RPD

Some tests, particularly extractable organic analyses cannot meet this criteria, see Table 5.1.

12.5.6 Quality Control Check Samples - Performance Evaluation Samples (P.E. Samples)

Acceptance criteria for P.E. samples are set by the organization running the study (EPA, commercial vendors). All results of P.E. samples are summarized to management.

12.5.7 Quality Control Check Standards or Initial and Continuing Calibration Check Standards (ICV01, CCV)

Refer to Section 9.0, Table 9.5, <u>Instrument Calibration</u> for the acceptance criteria of the ICV01 and the CCV.

12.5.8 <u>Duplicate Samples (DUP01, DUP02)</u>

Control limits for duplicate samples are identical to those set for duplicate matrix spike samples (see section 12.6.4, Table 5.1 and Table 12.3) with one exception. Since duplicate precision is measured from analyte already present in the sample (no spike added) the level may not be detectable or may be very close to the detection limit. Concentrations of analyte less than 10 times the reporting detection limit (MDL, PQL, etc.) present in non-spiked (duplicate) samples will not be considered as out-of-control based on the QCTS data, but will be allowed a maximum RPD of 20 percent for inorganic analyses and 50 percent for organic analyses.

12.5.9 Methods Utilizing Internal Standards

The criteria which must be met for internal standard areas (when used) are found in Table 12.4, <u>Internal Standard Area Criteria</u>.

TABLE 12.4 INTERNAL STANDARD AREA CRITERIA

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Instrument	Method	Criteria, χ = Absolute Area of Daily CCV
GC/MS	EPA 524.2	$-30\% > \chi < +50\%$
GC/MS	EPA 525	$-30\% > \chi < +50\%$
GC/PID/ELCD	EPA502.2	±3 SD from CAL Standards
GC/NPD	EPA 507	30% Difference from daily CCV
GC/ECD	EPA 508	30% Difference from daily CCV
GC/ECD	EPA 515.1	30% Difference from daily CCV
HPLC	EPA 531.1	30% Difference from daily CCV
GC/MS	SW-846 8240	$-50\% > \chi < + 100\%$
GC/MS	SW-846 8260	$-50\% > \chi < + 100\%$
GC/MS	SW-846 8270	$-50\% > \chi < + 100\%$
GC/MS	CLP-VOA	$-50\% > \chi < + 100\%$
GC/MS	CLP-SVOA	-50% > χ < + 100%
GC/MS	EPA 624*	$-50\% > \chi < + 100\%$
GC/MS	EPA 625*	$-50\% > \chi < + 100\%$
GC/ECD	SW-846 8080**	$-50\% > \chi < + 100\%$

^{*}No method criteria, EMS Heritage has adopted these criteria.

^{**}Any correction of areas using Internal Standards will not exceed this criteria.

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12.5.10Surrogate Recovery Criteria

Control limit criteria for methods which do not specify specific control limits will be developed from in-house data. EMS Heritage will not adopt criteria less stringent than the published method criteria. "Advisory" criteria will not require corrective action. Refer to Table 12.5, Surrogate Recovery Criteria for method specific control criteria.

12.6 Review of Project Data

Project Managers are responsible for review of the overall project data before submission to the client. All laboratory "Certificates of Analysis" (Reports) and QA Summaries are reviewed and approved only by the QA Unit (see Reporting) prior to delivery to the Project Manager.

Project Managers are responsible for the following review prior to releasing the data report package to clients:

- 1. Review of supporting documentation including, but not limited to:
 - a. Chain of Custody.
 - b. Laboratory Analysis Request or Task Order Forms.
- 2. Review of all contract or project deliverables for completeness and accuracy including but not limited to:
 - a. Reporting forms.
 - b. Selected Raw data including chromatograms/ printouts.
 - c. Adherence to specific DQO's.
 - d. Analysis of all required QC Types at the required frequency.
- 3. Review of data for any obvious anomalous values, specifically those areas requiring knowledge of special project conditions possibly unknown to the QA Unit during review. Review large sets of data for comparability from one analytical batch or case

TABLE 12.5 SURROGATE RECOVERY CRITERIA

Volatiles: SW-846 8240, CLP-VOA					
Parameter Water, Range %R Soil, Range %R Spike Amount					
4-Bromofluorobenzene	86-115	74-113	50 μ g		
1,2-Dichloroethane-d4	76-114	70-121	50 μ g		
Toluene-d ₈	88-110	84-117	50 μ g		

Semi-Volatiles: SW-846 8270, CLP-SVOA					
Parameter	Water, Range %R	Soil, Range %R	Spike Amount		
Nitrobenzene-d ₅	35-114	23-120	50 μ g		
2-Fluorobiphenyl	43-116	30-115	50 μg		
p-Terphenyl-d ₁₄	33-141	18-137	50 μg		
Phenol-d ₆	10-94	24-113	100 μg		
2-Fluorophenol	21-100	25-121	100 μg		
2,4,6-Tribromophenol	10-123	19-122	100 μg		
2-Chlorophenol-d ₄	33-110 (advisory)	20-130 (advisory)	150 μg		
1,2-Dichlorobenzene-d ₄	16-110 (advisory)	20-130 (advisory)	100 μg		

Organochlorine Pesticide/PCBs: SW-846 8080, CLP-Pest					
Parameter Water, Range, %R Soil, Range %R					
Dibutylchlorendate (DBC) ¹	24-154 (advisory)	20-150 (advisory)			
Tetrachloro-m-xylene (TCX)	60-150 (advisory)	60-150 (advisory)			
Decachlorobiphenyl (DCB)	60-150 (advisory)	60-150 (advisory)			

¹DBC has been replaced by TCX & DCB for CLP-Pest OLMØ1.

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to another in the project (QA Unit does not generally review an entire case at one time).

4. Adherence to reporting deadlines and shipment/delivery of report packages by the best means available to meet those deadlines.

12.7 Data Reporting

The Data Entry Supervisor is responsible for supervision of all data entry operators. The Data Entry (D.E.) Supervisor (or designee other than the original data entry operator) will verify 5 percent of all data entered into LIMS. Locations having only one D.E. person will not check data in this manner; this responsibility will fall solely upon the QA Unit of that location.

Validation of data entry is performed for <u>all</u> data at a 2-5 percent frequency and is the responsibility of the QA Unit. The QA Unit may delegate this task to the QA Assistant or other staff person trained to perform that function.

The QA Unit is responsible for reviewing all final reports prior to release to clients. Reports are to be reviewed for:

- 1. Completeness All parameters, detection limits, units, dates, descriptions, etc. must be complete and correct.
- 2. Consistency All parameters must be internally consistent (CrVI \leq Cr Total; TKN \geq NH₃ N, TS \geq TDS, etc.)
- 3. Comparability All parameters and units must be reported in such a way that data sets can be evaluated relative to each other as necessary.
- 12.7.1 All modifications to samples in the LIMS system must be documented on the Sample Modification Form, Figure 12.2. Original entries or additions (comments, etc.) to original entries will not utilize this form unless the sample is a "status 8" (released) category. All modifications to test codes must be approved by the Project Manager or the QA Unit. Modifications to samples that are completed and released (LIMS status 8) are possible only by the QA Unit staff. Any report that has been sent to a client and subsequently modified in any way by the QA Unit must be identified as an "amended report" and the date of amendment and the initials of the person making the amendment will be given on the certificate of analysis.

FIGURE 12.2

REQUEST FROM:	DATE	
SAMPLE STATUS:		
SAMPLE NUMBERS:		
CLIENT:		
TEST CODE ADDED/CANCELLEI	D:	
TEST CODE CHANGED FROM _	то	
RESULTS MODIFICATIONS:		
		_
REASON FOR ADDITION OR CH	IANGE:	
MODIFIED BY:	DATE	_
PROJECT MANAGER'S APPROVA	AL INITIALS: DATE	
ACCOUNT MANAGER'S APPROV	VAL INITIALS: DATE	
OLD PRICE:	NEW PRICE:	

; =

12.7.2 Samples of final reports and computer (LIMS) QC summaries are contained in Appendix F, <u>Final Report Examples</u>.

12.7.3 Special Reporting Requirements

Sample analysis failing QC criteria due to matrix related interferences or samples requiring dilutions, etc. may have been analyzed 2 or more times. EMS Heritage will report a maximum of two analyses in those cases, or each analysis if client desires.

If a re-analysis of the sample fails the same criteria a second time and all other instrument criteria are in control, a matrix interference is assumed and both analyses may be reported. If only the reanalysis is in control, only the reanalysis will be reported.

Exceptions to this policy will be made for contract requirements and/or different client DQO's.

12.8 Data Storage

The following data is maintained for long term storage.

- bench sheets
- strip charts
- chromatograms, print-outs
- magnetic media used to store results
- invoices and financial statements
- project files
- customer data
- various raw data
- lab related periodicals
- 12.8.1 The records, hard copy and magnetic media, are kept on site for a period of six months to a year. As necessary, records may be transferred to an off-site records storage facility. The records storage

facility must provide secure, access controlled and environmentally controlled storage of records. For all GLP (40 CFR, part 160) projects, records will be kept on-site.

After the completion of the requested work, records of the raw data, quality assurance data and reports will be kept by EMS for at least five (5) years unless otherwise required by law or regulations. Magnetic tapes are maintained, however, magnetic tapes have a storage life expectancy of less than ten (10) years.

12.8.2 The following is an overview of how records are stored and retrieved at EMS Heritage.

12.8.2.1 Storing Records

To send a box off site to a records storage facility the following three steps will be followed.

- 1. A box is filled with material of a related nature.

 The box is then marked to show the contents and the related service group and destruction date.
- 2. The box is then taken to the off-site storage coordinator where the box is given a unique number. The unique number, the contents, the associated service group, and the date stored are recorded in a ledger. Once the box is entered into the ledger it is scheduled to be picked up or delivered to the off-site records storage facility.
- 3. The off-site records storage facility stores the box and gives it a unique location code. Once the location code is assigned off-site records storage facility contacts the off-site storage coordinator, who enters the location code into the box's ledger entry. This allows for the recall of individual boxes.

12.8.2.2 Recalling Records

To recall records from off-site storage requires the EMS box number and the records storage facility location code, both of which are kept in a records storage ledger.

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A copy of the ledger is kept by the off-site storage coordinator, who can also recall the box from storage.

- 12.8.3 The LIMS system located in Indianapolis is backed up on a three cycle program. Every week a backup of the system is done. Weekly backup tape sets are rotated every week. So, if the computer is backed-up with tapes set "A" this week tape set "B" is used the next week. The tape sets are rotated back and forth every other week, except the last week of the month. For the last week of the month tape set "C" is used. By using this method it is possible to restore the computer from as far back as one month. The tape sets are held at off-site storage, except when called back to do the backup. EMS Heritage also maintains over a month of "redo's" on tape that are backed up each day. "Redo's" are copies of the transactions needed to apply to an old backup of the database to bring it up to date. EMS Heritage also maintains a two week rotation of backups for the software. The software is backed up each day as well.
- 12.8.4 Data is not stored or organized per specific project unless contract or other client requirements dictate such. If storage by project is required, all raw data and final reports are copied and stored in a "central project file".

All original raw data and benchsheets are organized and stored in folders chronologically according to analytical batch and date of analysis.

13.0 Corrective Action

13.1 Introduction

Criteria levels which designate in or out of control status are specified in the applicable sections of this QA Plan. The corrective actions recommended, that will return the analysis to an "in control" status, are summarized in Table 13.1, <u>Corrective</u> Actions.

Corrective action must be undertaken whenever any number of unacceptable conditions exist.

13.2 Warning Limits

A first condition requiring corrective action occurs when external standards or otherwise specified quality control check samples are analyzed and results found to be outside established warning limits, upper or lower (UWL, LWL). Data may be acquired if the upper and lower control limits (UCL, LCL) are not exceeded in this condition, but corrective action must be undertaken. See Trends Analysis, Section 13.4.

Two successive readings outside of warning limits (but inside of control limits) are considered out of control. The analysis must be stopped and immediate corrective action is undertaken (see all Trends Analysis, below).

13.3 Control Limits

Another condition is one in which either of the <u>control</u> limits have been violated. In this case, all analyses stop and immediate corrective action is undertaken.

13.4 Trends Analysis

Analyses are out of control whenever any of the following occurs:

- 1 point is outside the control limit
- 2 consecutive points are outside the warning limits
- 7 consecutive points are on one side of the center line
- 7 consecutive points increase or decrease
- An obvious trend is observed in the distribution of points

13.5 Notification

In both of the above conditions, the determination is to be made by the analyst if the procedure is in control. If it is in control, analyses proceed; if it is out of control, all analyses are stopped and the following information is compiled immediately and in writing.

- 1. Test which is out of control.
- 2. Analyst.
- 3. Instrument.
- 4. A brief description of the problem.

At a minimum the problems will be written on the runs (bench sheets) and the QA Unit will be notified and consulted. When the seriousness of the problem warrants a more formal report, the following information will be supplied on the form in Figure 13.1.

The report is given immediately to the Group Leader, Lab Director and Quality Assurance Unit, The goal at this point is to return the analysis status to IN-CONTROL as soon as possible.

The Group Leader is responsible for returning the analysis status to in control and he or she may draw upon any or all of the resources at EMS Heritage to accomplish this in a timely fashion.

13.6 Specific Corrective Actions

The nature of the corrective actions which may be required at the bench are too numerous to document within a manual such as this one. However, there are some common procedures which are generally appropriate to a number of trouble-shooting protocols including:

- 1. Review of analytical history: Is there a sudden problem or has an analysis slowly "drifted" out of control?
- 2. Prepare fresh standards: Have the standard or stock or working solutions expired or degraded?
- 3. Prepare fresh reagents: Have any reagents expired or degraded?

4. Review the method - ask for assistance: Has an analyst introduced an un-noticed, methodical error into the procedure?

Refer to Table 13.1, Corrective Actions for detailed procedures.

EMS will subscribe to any reasonable corrective action deemed necessary by regulatory personnel. Corrective actions will also be initiated as needed as a result of systems or performance audits, split samples or data validation reviews. Exceptions to this policy must be approved by the President of the laboratory or the QAO for the purpose of negotiating a corrective action.

Following corrective action, any sample analyzed during the out of control period must be reanalyzed.

13.7 Corrective Action Documentation

Upon returning the analysis to an "in-control" status, the Group Leader is responsible for filing the following information in the method file:

- 1. Test which was out of control.
- 2. Analyst/Date.
- 3. Instrument I.D.
- 4. Brief description of the problem.
- 5. Corrective actions taken.
- 6. Current Status.
- 7. Date analysis returned to "in-control" status.
- 8. Person responsible for remedy.

Figure 13.1, <u>Analytical Control Notification</u> is the form that will be used for out of control and return to "in control" status reports.

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FIGURE 13.1 ANALYTICAL CONTROL NOTIFICATION

Test	 	
Analyst	 	
Date		
Instrument		
Brief Description of Problem		
Corrective Actions Taken		
	 -	
Command States		
Current Status		
D. A. Davis and A. J. Control		
Date Returned to In-Control	 	
Persons Responsible For Remedy_		

TABLE 13.1 CORRECTIVE ACTIONS

QC Type	Recommended Corrective Actions
Instrument Blanks	Prepare another blank, if same response, determine cause of contamination: reagents, environment, instrument equipment failure, etc. Re-analyze any samples associated with unacceptable blanks.
Calibration Standards	Reanalyze standards. If still unacceptable, then remake standards. If still unacceptable try an alternate source or prepare from neat standards.
Continuing Calibration Standards	Reanalyze standard. If still unacceptable, then recalibrate and rerun samples from the last unacceptable CCV standard.
Internal Standards	Inspect the chromatographic system for malfunctions and make corrections as required. Reanalyze affected samples. If reanalysis does not solve the problem and the system is not malfunctioning, report both analyses with comments, as required by some clients. Report only 1 analysis with comments for those clients not requiring replicate reporting.
Surrogates	Check for errors in calculations, formulation of spiking solution, internal standard responses and instrument performance. Reanalyze the samples if none of the above reveals a problem. If an undiluted analysis with acceptable surrogate recoveries is being reported, do not reanalyze diluted samples if the surrogates are outside limits. MS/MSDs outside of control limits for surrogates are never reanalyzed except for specific contractual or DQO requirements. If the sample and its associated MS/MSD show the same surrogate recovery pattern, document this in a case narrative or analysis comments. If reanalysis does not solve the problem and the system is not malfunctioning, report both analyses with comments, as required by some clients. Report only 1 analysis with comments for those clients not requiring replicate reporting. Corrective actions are not required for advisory surrogate limits. For procedures involving extraction, when surrogates are outside the limits and require corrective actions reanalyze the extract first, to determine if the problem was with the analysis. If reanalysis does not solve the problem then re-extract and analyze the second extract. If surrogate recoveries in an extracted blank do not meet specifications upon re-analysis, re-extract and reanalyze all samples and a blank. Specifically for GC/MS semi-volatile analysis, if two base/neutral or two acid surrogates are out of limits or if recovery of any one base/neutral or acid surrogate is below 10% corrective actions are required as described herein.

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TABLE 13.1 CORRECTIVE ACTIONS

K	
Method Blanks	Reanalyze blank. If still positive, determine source of contamination. If sample results are greater than 10 times the method blank value or the sample results are below the detectable level, the analysis is acceptable. If necessary, reprocess (i.e. digest or extract) the sample set.
Initial Calibration Verification Standards (ICV)	Reanalyze ICV standards. If still unacceptable, then remake ICV standards and reanalyze. If still unacceptable, then recalibrate and reanalyze ICV. If still unacceptable evaluate source and validity of all standards as appropriate.
Post Digestion/Extraction Spike (MSA)	Check for error. Refer to Figures 11.1 or 11.2 for actions regarding the Method of Standard Additions for Metals analysis. For all other post digestion spikes, reanalyze. If reanalysis does not meet the criteria, flag the results with a comment and include in the case narrative.
Pre Digestion/Extraction Matrix Spike (MS/MSD)	For %R criteria only: Check for error. If extracted standard (LCS, lab Fortified Blank, QC check sample, etc.) meets criteria, then poor recovery is considered to be a matrix interference. Extracted standard or check standard must be prepared in same prep batch where applicable. Report results with a comment and include in the case narrative. If extracted standard fails, reprep and reanalyze all affected samples. CLP protocol testing will not require reanalysis for organics testing unless a trend develops for poor accuracy.
Extracted Standards	Check for errors in preparation and analysis. Reanalyze and reprep all affected samples.
Duplicates and Duplicate Spikes	For RPD criteria only: Check for error. If extracted standard (DLCS) meets criteria, then report the result. Poor precision may be due to sample inhomogeneity; check sample appearance. If obvious sample inhomogeneity is the problem, do not enter the outlier into the QCTS, but report results with a comment and include in the case narrative. Sample RPD's exceeding 20 percent for inorganic analyses or 50 percent for organic analyses or the method specific criteria (RCRA, CLP) with no obvious sample inhomogeneity problem must be reprepped and reanalyzed for all affected samples. CLP protocol testing will not require reanalysis unless a trend develops for poor precision.

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14.0 Performance and Systems Audits

14.1 Performance Audits

A <u>performance audit</u> is an independent check to evaluate the data produced by a laboratory's analytical system, and may be categorized as a <u>quantitative appraisal of quality</u>. There are several ways that this is done:

- 1. Worksheet review
- 2. Oral worksheet review
- 3. On-site analyst work review
- 4. Independent or check sample review
- 5. Intra and interlaboratory check sample, or proficiency test (performance evaluation) sample analysis review

14.1.1 Internal Performance Audits

Internal performance audit worksheet reviews are conducted by the QA Officer and/or the QA Unit. These reviews must be conducted at a frequency necessary to assure the accuracy of either the total measurement system or its component parts. Worksheet reviews by the QAO and/or QAU will be ongoing but will occur at a minimum frequency of semi-annually for selected tests. This will be in addition to (or will be a more in-depth review of <u>all</u> data records) the 2-5 percent frequency of data entry validation performed by the OAU.

Internal performance audit worksheet reviews will consist of evaluations of all data and related supporting documentation to assure that all required QC checks are being made and evaluation criteria followed. Reports relating to internal performance audit worksheet reviews are confidential and will be released only upon approval from the President of EMS Heritage.

14.1.2 Performance Audit-Proficiency Test Samples

Performance Audits at EMS Heritage also consist of analysis of independent or commercial check samples and participation in performance evaluation sample programs.

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Refer to Table 14.1, <u>Reference Materials and Performance Evaluation</u>

<u>Participation</u> for the sources of reference materials used and the on-going performance audit participation.

All information generated from Performance Evaluation (P.E.) sample programs will be made available during systems audits or upon request. Blind samples and split samples may be submitted as deemed necessary by the QAO or the QAU.

14.2 Systems Audits

A <u>systems audit</u> is an on-site inspection and review of a laboratory's quality control system and may be categorized as a <u>qualitative appraisal of quality</u>. It will cover any or all of the operational quality control elements of the quality assurance program. Systems audits include but are not limited to the following:

- 1. Sample handling: receiving, custody, log-in, storage
- 2. Sample analysis: written SOPs and analytical methods, protocols
- 3. Records control
- 4. Documentation: bound notebooks where required; records of all sample handling and analytical procedures
- 5. Preventative maintenance: adequate records, procedures.
- 6. Proficiency testing
- 7. Personnel practices
- 8. Training
- 9. Workload
- 10. Instrumentation and facilities

14.2.1 External Systems Audits

Systems audits are performed frequently at EMS Heritage by State and Federal agencies as part of the participation in sample analyses for governmental organizations. In addition, many of our commercial clients perform routine audits of EMS Heritage. Evaluation by other appropriate

outside experts is to be performed in the event that regulatory personnel are not available. At a minimum, systems audits will occur annually. EMS Heritage Laboratories will welcome any external audit from an organization which currently or proposes to submit samples for analysis.

14.2.2 Internal Systems Audits

In order to verify that each laboratory division is performing according to the standards established by EMS Heritage Laboratories, Inc., an internal audit system has been established. Internal systems audits will be conducted no less frequently than annually. The auditor will be appointed by and directly responsible to the President of the corporation.

The President may appoint for any given audit the following staff:

- 1. Quality Assurance Officer
- 2. Quality Assurance Unit
- 3. Outside (Contracted) Auditor

EMS Heritage also encourages the individual groups at each division to perform a self-audit at periodic intervals to assess their overall operation.

These audits will be scheduled by the auditor and each laboratory director. All reports relating to internal systems audits will be treated as an internal document only; release to any outside entity will be only upon approval of the President of EMS Heritage.

To facilitate the procedure and make it consistent from division to division, a set of audit procedures and protocols has been developed and will be used consistently throughout the corporation. These forms will be used as a guideline and a tool for documentation of problems and to assure that all important areas are covered. The forms are not an end in themselves and will be subject to change/improvement by the QAO at his discretion in order to facilitate and improve the auditing process. Refer to Appendix G, Internal Systems Audit Forms.

14.2.3 Internal Systems Audits - Procedures and Protocols

Not all areas (service groups) will necessarily be audited during each visit. During each audit, at least one analytical run will be randomly selected from the service group(s) chosen. This run will be thoroughly checked using the audit checklist forms and any other pertinent audit questions.

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Upon completion of the audit, the auditor will conduct an exit interview with the QA Unit and the Laboratory Director and any other personnel deemed appropriate. A copy of the completed audit forms as well as any supporting narrative will be given to the QA Unit and the Laboratory Director at that time or immediately upon completion. A copy of all of these materials will also be sent to the President, Vice President of Operations, Quality Assurance Officer and all QA Units of all EMS Divisions. This sharing of audit information with all divisions is to provide not only a report of findings but will also serve to point out potential deficiencies which may exist at the respective divisions not audited. The auditor will retain a copy of the materials to use in the next audit. The QAO will confer with the President to prioritize items for the next audit.

Generally the audit will consist of the auditor reviewing with the group leader and the analyst all the steps involved in generating a run of reportable data. It will also include an overview of the entire lab, address issues relevant to the operation as a whole (i.e. cooler logs, log-in procedures, oven logs, etc.) Any deficiencies will be noted on the forms. Deficiencies will be discussed at the time they are noted and the auditor will explain what will be required to correct this deficiency.

Form 1 will always be completed. It addresses general laboratory practices. Form 2 will be completed during each audit for each service group audited. Forms 3-5 will be used on a service group specific basis only.

TABLE 14.1 REFERENCE MATERIALS AND PERFORMANCE EVALUATION PARTICIPATION

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Source	Туре	Frequency	Parameters	Division
APG	Unknowns	Monthly	All	All
EPA-WP	Unknowns	Semi-Annually	All	All
EPA-WS	Unknowns	Semi-Annually	All	All
North Carolina- Wastewater	Unknowns	Semi-Annually	Metals, General Chemistries	Charlotte
State of New York	Unknown	Quarterly	All	Indianapolis
State of New Jersey	Unknown	Semi-Annually	All	Indianapolis
EPA Reference	Known	On-Going	All	All
APG	Known	On-Going	All	All
NIST	Known	On-Going		All
EPA QB- Inorganics	Unknown	Quarterly	Metals	Indianapolis
EPA QB- Organics	Unknown	Quarterly	Organics	Indianapolis
Chemical Waste Management	Unknown	Quarterly	All	Indianapolis, Chicago
State of Illinois	Unknown	Yearly	All	Chicago

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15.0 Quality Assurance Reports

For each division, by the last day of the month following the completion of each fiscal quarter the Quality Assurance Unit (QAU) will provide information as required to the Quality Assurance Officer (QAO), who will, in turn, compile and summarize the information to the President, the Vice President of Operations and the Laboratory Directors. QA reports will consist of but not be limited to the following points:

15.1 QCTS Review

A statement that the QAO has reviewed the precision, accuracy and method detection limits data for the previous quarter in the QCTS data base. Included in this statement will be:

- 1. A listing of outliers Parameters for which data quality objectives have not been met. Parameters for which MDLs must be evaluated.
- 2. A brief narrative discussing the course of action to be taken to correct deficiencies identified above.

15.2 Correction of Deficiencies

A statement that the laboratory has corrected or has begun to correct deficiencies identified in the previous quarter's QA report. Included in this statement will be:

- 1. A brief narrative describing what was done.
- 2. A brief narrative describing what is yet to be done.

15.3 Performance and Systems Audits Summary

In addition, on a monthly basis the President, Vice President of Operations and the Lab Directors will be supplied with the results of all performance and systems audits most recently completed. Results of EPA WS and WP series samples will be provided to the President, the Vice-President of Operations and Lab Directors as they are received. The forms in Figures 15.1 and 15.2 to this part will be used.

15.4 Significant Problems and Solutions

Significant QA/QC problems and recommended solutions will be reported to the President on a timely basis.

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15.5 Outside Reporting

Whenever regulatory authorities require submission of QA reports, such reports will be submitted to the appropriate QA officers of those agencies at a frequency prescribed by those agencies (e.g. quarterly).

FIGURE 15.1 APG RESULTS SUMMARY

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Month ____

										Year	
Deviation from Means of Reporting La		Ch	emistrie	es	Genera	l	Metals	3	GC		GC/MS
0 - 1.0Σ											
1.01 - 2.00											
2.01 - 3.00											_
>3.00											
Total											
Values with	deviat	ion	> 2.00								
Analyte	Valu Rpte		True Val.	% R	ecovery	An	alyte	Valu Rpte		True Val.	% Recovery
N =	_							· ·			
<u>Deviation</u>			N		<u>% Pro</u>	ev. N	10 %	-		This M	10 %_
0 - 1.00 1.01 - 2.00 2.01 - 3.00 3.01 - Outlier											

of Outliers

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FIGURE 15.2 EPA PERFORMANCE EVALUATION SUMMARY

Study #	
Date Performed	

Date Reported

		<u> </u>		
% Acceptance/Group	Metals	General	GC/MS	GC
% Analytes Correct				
% Test Results Correct				
Number of Analytes				
Number of Tests				

List Unacceptable Results:

Analyte	Reported Value	True Value	% Recovery

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KEY RESUMES

List of Key Resumes

Name	Page #
Curtis Beck	7
Anne C. Bradburn	7
Susan Brotherton	5
Gregory Busch	4
Susan Bussey	14
Mel Carroll	13
J. Coplen	2
Jack Corpuz	6
David Czerny	8
Jan Dillow	18
Deborah Edwards	16
Steve Endersen	5
C. Steven Gohmann	1
Steve Guptill	21
Helmuth Janssen	21
Lisa Julian	8
Carla Keever	23
Horst Kehl	19
Gary Klingler	4
Malgorzata Kraska	23
J.T. Kurek	3
Kurt Maines	6
Scott Meeks	18
Kevin Miller	20
Linda Osborn	9
D.L. Peterson	2
Michael Piwowar	23
Margarita Rozenfeld	20
Roger Rowan	19
Michele Sakwa	21
Chris Sarkan	14
Jennifer Schott	15
Susan Sharp	15
Barbara Shrake	10
Dawn Siekerman	16
Adam West	3
Janet Williams	24
Fred Winter	13
Pen Winters	24
Ronald F. Wukasch	1

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C. Steven Gohmann, President EMS-Heritage Laboratories, Inc.

Experience:

1974 - Present, EMS Heritage Laboratories, Inc.

6 years with Other Firms

Education:

Bachelors Degree, Chemistry/1973/Analytical Chemistry, Indiana University, Bloomington,

IN

Memberships:

American Chemical Society

American Society for Testing Materials
Water Pollution Control Federation

C. Steven Gohmann has been Laboratory Director for EMS-Heritage Laboratories, Inc. (formerly Environmental Monitoring Service) since its founding in 1974 and is responsible for all activities of the lab. These duties include the purchase of equipment and major instrumentation, establishment and maintenance of quality control/assurance program to comply with EPA recommendations; the establishment and maintenance of a fulltime professional staff; laboratory cost control; and many other duties associated with the organization. Prior to this affiliation with EMS-Heritage, Mr. Gohmann was employed by a testing laboratory near Louisville, KY as an analytical chemist for the analyses of water and wastewater. Before his association with the Louisville firm, Mr. Gohmann served as a research associate at Indiana University where he assisted in the development of gas chromatographic techniques for the analyses of lunar samples for the Apollo program. He also, served as an associate instructor of chemistry while at Indiana University. He is a member of ACS, ASTM and WPCF.

Dr. Ronald F. Wukasch, Vice President EMS-Heritage Laboratories, Inc.

Experience:

1974 - Present, EMS Heritage Laboratories

20 years with Other Firms

Education:

Ph.D/1966 Environmental Engineering MS 1955 Environmental Engineering

BSCE

Active Registration: PE, Indiana 1973, Environmental Engineering

PE, Michigan 1969

Dr. Wukasch was responsible for the technical direction and overall management of the City of Indianapolis' wastewater treatment facilities improvement program. He also worked as consultant and advisor to the City-Department of Public Works on advanced wastewater treatment and resource recovery from solid waste. In addition, he has been principal investigator on numerous research projects dealing with pollutants in municipal sludge and wastewater.

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Dr. David L. Peterson, Quality Assurance Officer

EMS-Heritage Laboratories, Inc.

Experience:

1985 - present, EMS Heritage Laboratories, Inc.

9 years with Other Firms

Education:

Bachelors Degree/1966/Chemistry, Manchester College Masters Degree/1969/Biochemistry, Ball State University Doctorate/1972/Biochemistry, Purdue/Ball State University

Memberships:

American Chemical Society Indiana Academy of Science

American Society for Quality Control

Dr. Peterson serves as the Quality Assurance Officer to monitor all analytical testing. Prior to his employment with EMS-Heritage, he served as the Senior Environmental Chemist for a major engine company. Activities there included all types of analysis on water, industrial hygiene, hazardous waste, and special wastes. In addition he set up Quality control programs to fulfill requirements for laboratory accreditation by the American Industrial Hygiene Association. Dr. Peterson served six years with the Air pollution and Industrial Hygiene/Hazardous Waste laboratories at the Indiana State Board of Health. He has utilized Gas Chromatographs since 1969 and uses Atomic Absorption and GC/MS routinely. Dr. Peterson is a member of the American Chemical Society and the Indiana Academy of Science.

James Coplen, Vice President Operations

EMS-Heritage Laboratories, Inc.

Experience:

1981 - present, EMS Heritage Laboratories, Inc.

1 year with Other Firms

Education:

Bachelors Degree/1981/Environmental Science, Purdue University, West Lafayette, Indiana

Masters Degree/1983/Aquatic Toxicology, Purdue University, West Lafayette, IN

Memberships:

Mr. Coplen is responsible for coordinating resources among EMS Heritage Laboratories. He has lectured and published on laboratory operations and environmental information management. Mr. Coplen was manager of the Heritage Systems and Data Management group. His earlier work at EMS Heritage included managing the aquatic toxicity testing program and conducting acute and chromic bioassays. Mr. Coplen's graduate work included developing a model to predict the toxicity of complex wastewaters.

Dr. Joseph T. Kurek, Chief Chemist EMS-Heritage Laboratories, Inc.

Division: Indianapolis

Experience:

1988-present, EMS Heritage Laboratories, Inc.

15 years with other firms

Education:

Bachelors Degree/1967/Chemistry, Boston College

PH.D./1974/Organic Chemistry, Purdue University, West Lafayette, IN

Memberships: American Chemical Society

Dr. Kurek was on the chemistry faculty at Franklin College from 1974 to 1989, teaching mainly organic, analytical and biochemistry. Since 1976 he has been adjunct professor of chemistry at the University of Indianapolis, teaching organic and analytical chemistry. During the Fall 1982 term, he was visiting assoc. prof. of chemistry at Purdue University, W. Lafayette. He has also been a consultant to a number of corporations and government agencies in the fields of environmental and forensic analysis.

Adam West, Programmer/Systems Analyst

EMS-Heritage Laboratories, Inc.

Experience:

1986 - present, EMS Heritage Laboratories, Inc.

2 years with Other Firms

Education:

3 years toward Bachelors Degree, Computer Science

Memberships: NA

Adam West's duties include systems programming, maintenance of interactive software, laboratory information management, reporting, and programming custom software for specific use with EMS-Heritage. Assignments have included sample log-in and sample modify routines, accounting and invoicing programs and the scheduling of data security backups and data archives. He is also involved in trouble-shooting and maintaining the user environment. Prior to his work at EMS-Heritage, Mr. West was employed by a petroleum and land management company where he established and maintained tracking, accounting and payroll computing systems. He has also designed and commercially marketed programs for microcomputers.

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Gary Klingler, Indianapolis Laboratory Director

EMS-Heritage Laboratories, Inc.

Division:

Indianapolis

Experience:

1975 - present, EMS Heritage Laboratories, Inc.

3 years with Other Firms

Education:

Bachelors Degree/1975/Analytical Chemistry, Marian College, Indianapolis, IN

Memberships:

American Chemical Society

American Association for the Advancement of Science

Gary Klingler received a BS Degree in Chemistry from Marian College and has contributed significantly to research projects related to waste oil treatment while employed by an oil reprocessing facility. Mr. Klingler's duties have also included the design of water testing programs, implementation of quality control programs and the development of new treatment conditions for biological waste treatment plants. Mr. Klingler has four years experience in waste water analysis and currently serves as Laboratory Director at EMS-Heritage.

Gregory A. Busch, Quality Assurance Unit Director

EMS-Heritage Laboratories, Inc.

Division:

Indianapolis

Experience:

1988 - present, EMS Heritage Laboratories, Inc.

41/2 years with Indiana Department of Environmental Management

3½ years with other firms

Education:

Bachelors Degree/Chemistry & Biology/1980, Indiana University, Bloomington, IN

Masters Degree, Chemistry/in progress, Butler University

Memberships: American Chemical Society

Mr. Busch wrote the EMS Heritage Comprehensive Quality Assurance Plan and several SOP's related to his duties as QA Coordinator. Mr. Busch performs Internal Systems and Performance Audits at EMS Heritage. Duties include examination of 5% of the raw data on a periodic basis to verify adequacy of documentation and accuracy of reported results. Responsible for keeping up with changing environmental regulations impacting laboratory analysis and providing support and consultation on specific environmental projects. Prior to his employment with EMS, Mr. Busch was a Chemist II-Working Group Leader with the Indiana Department of Environmental Management, Office of Solid & Hazardous Waste Management and served as the RCRA QA Officer. He served as contract laboratory liaison, coordinating sampling efforts, and maintaining quality control criteria. Major responsibilities included review of lab analyses for completeness and quality, maintaining a laboratory tracking program, performing on-site lab audits, writing state laboratory contracts and writing/revising the State RCRA Quality Assurance Project Plan (QAPP). Previous experience includes work at private and state laboratories in areas of sampling and testing water, wastewater, solid & hazardous wastes utilizing wet chemistry analyses, and GC/MS.

Susan Brotherton, Senior Chemist EMS-Heritage Laboratories, Inc.

Division: Indianapolis

Experience:

1983 - present, EMS Heritage Laboratories, Inc.

6 years with Other Firms

Education:

BS/1977/Chemistry, Michigan Technological University

MS/1983/Analytical Chemistry, Michigan Technological University

Memberships: American Society for Mass Spectrometry

Since joining EMS-Heritage, Mrs. Brotherton has been responsible for initiation and automation of EPA Methods 624 and 625, SW-846 Methods 8240 and 8270 and EPA-CLP protocols for organic analysis by GC/MS. Her responsibilities also include implementing the Quality Control and Quality Assurance Programs associated with these methods as well as utilizing computer capabilities in identification and quantitation of these compounds. While employed by Michigan Technological University, she served as Laboratory Manager of the Analytical Services Lab and was involved with organic and inorganic analyses of environmental samples. She has experience with Wet Chemical Methods, AA, IC, and LC in addition to GC and GC/MS. Her graduate thesis work was on The Analysis of Polycyclic Aromatic Hydrocarbons by Chemical Ionization GC/MS.

Steven J. Endersen, Senior Chemist EMS-Heritage Laboratories, Inc.

Division:

Indianapolis

Experience:

1986 - present, EMS Heritage Laboratories, Inc.

3 years with Other Firms

Education:

Masters Degree, Analytical Chemistry, 1983, University of New Hampshire, Durham, NH

Bachelors Degree, Analytical Chemistry, 1980, Allegheny College, Meadville, PA

Memberships: NA

Steven Endersen is responsible for technical assistance and training in the Metals group at EMS-Heritage Laboratories, Inc. Duties include training new personnel on Atomic Absorption Spectrophotometers and the ICP, as well as being an analyst on these instruments. He serves as a technical specialist involved in method development, instrument maintenance, and quality control. Previously, Mr. Enderson was employed by the State of Vermont as a chemist. His duties were similar to those at EMS-Heritage. In addition, he had experience in GC/MS, GC, Microscopy, and Microbiological techniques. He is a member of Sigma Xi.

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Jack Corpuz, Senior Project Chemist

EMS-Heritage, Laboratories, Inc.

Division:

Indianapolis

Experience:

1989 - present, EMS Heritage Laboratories, Inc.

17 years with State of Indiana 2 years with other firms

Education:

BS Chemistry 1973, Purdue University, West Lafayette, Indiana

Memberships

American Chemical Society

Health Physics Society

Conference of Radiation Control Directors

Jack Corpuz' duties as a Project Chemist include customer contact, sales, consulting and technical assistance. Previous experience includes regulator for State of Indiana programs: RCRA, CERCLA, OSHA, CWA & SDWA. Mr. Corpuz has seven years bench experience in Radiochemistry support of the U.S. Department of Energy, U.S. Nuclear Regulatory Commission, and the U.S. EPA Environmental Radiation Protection Programs. Supervisory experience in OSHA, Radiological Health and RCRA.

Kurt Maines, Senior Project Chemist

EMS-Heritage, Laboratories, Inc.

Division:

Indianapolis

Experience:

1989 - present, EMS Heritage Laboratories, Inc.

4 years with Indiana Department of Environmental Management

31/2 years with other industrial/consulting firms

Education:

BS Chemistry 1984, Purdue University, West Lafayette, IN

Memberships:

American Chemical Society

Indiana Society of Hazardous Material Managers Wastewater Treatment Operator Certification

Among Kurt's responsibilities are price quotations, bid specification review, client-laboratory interface with regard to DQOs, TAT, lab capacity and regulatory compliance, and initiation of groundwater monitoring services. Previous experience includes data review and validation under RCRA and CERCLA, groundwater monitoring, hazardous waste sampling, sample collection training and coordination, wastewater treatment system start-up consulting, gas, are furnace and foundry operations analysis. Instrumentation experience includes GC, UV/VIS, particle size analyzer, AA, are spectrophotometer, and wet chemistry.

Curtis Beck, Project Chemist/QA Assistant

EMS Heritage Laboratories, Inc.

Division:

Indianapolis

Experience:

1989 - present, EMS Heritage Laboratories, Inc.

2 years with University of Florida

Education:

BS Environmental Studies/Biology 1985, Bemidji State University, Bemidji, MN

Memberships:

Indiana Society of Hazardous Materials Managers

Curtis Beck's duties as Project Chemist include client contact for price quotations, data quality objectives, lab capacity, turn around time, and regulatory compliance. As Quality Assurance Assistant, he is responsible for maintaining the LIMS database of test codes, parameters and methods and reviewing data for completeness and accuracy.

Anne C. Bradburn, GC/MS Group Leader

EMS-Heritage Laboratories, Inc.

Division:

Indianapolis

Experience:

1985 - present, EMS Heritage Laboratories, Inc.

Education:

Masters, Analytical Chemistry/1990, Butler University, Indianapolis, IN Bachelors Degree, Chemistry/1985, Indiana University, Bloomington, IN

Memberships:

American Chemical Society

American Society for Mass Spectrometry

Anne Bradburn's primary duty as GC/MS group leader is supervision and coordination of the personnel in the group. This includes training, ensuring analyses are completed efficiently with proper quality control and general problem solving. Her other duties include instrument maintenance and sample analysis.

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<u>David Czerny, Metals Group Leader</u> EMS-Heritage, Laboratories, Inc.

Division:

Indianapolis

Experience:

1989 - present, EMS Heritage Laboratories, Inc.

8 years with Other Firms

Education:

BS, Biology/1979, Wabash College, Crawfordsville, IN

M.S.E.S., Indiana University, Bloomington, IN, SPEA/1982

Memberships/Active

Registration:

Class III Wastewater Operator
American Chemical Society
Society for Applied Spectroscopy
Water Pollution Control Federation
Indiana Water Pollution Association

Mr. Czerny has been with EMS as the Metals Group Leader since December, 1989. He has a broad background in water pollution control, environmental monitoring and testing, and field work. Mr. Czerny holds a Class III Wastewater Operators Certification. Prior to coming to EMS, he was the Chief Chemist/Metals Section at the Indianapolis Dept. of Public Works (DPW). While at DPW Mr. Czerny (1983 to 1987) performed metals analyses utilizing flame and graphite furnace atomic absorption and plasma emission spectroscopy.

Lisa A. Julian, GC Group Leader

EMS-Heritage, Laboratories, Inc.

Division:

Indianapolis

Experience:

1986 - present, EMS Heritage Laboratories, Inc.

Education:

BS Chemistry/Biology 1976, Anderson University

MS Analytical Chemistry 1989, Purdue University (IUPUI), Indianapolis, IN

Memberships: American Chemical Society

Lisa joined EMS in January 1987 as part of the Masters Co-op program at Indiana-Purdue University at Indianapolis. Her initial duties included preparation of samples, analysis and calculation for PCB's, pesticides and herbicides. She was assigned and completed the task of interfacing the GC/ECD instruments to an HP Chromatography Software system and in September 1989, was promoted to group leader. Her duties include keeping the computer running efficiently, routine GB maintenance, training and supervising the technicians in the group and coordinating the regular and rush analyses.

Linda V. Osborn, General Organic Group Leader EMS-Heritage Laboratories, Inc.

Division:

Indianapolis

Experience:

1984 - present, EMS Heritage Laboratories, Inc.

6 years with Other Firm

Education:

Bachelor of Science Degree, Physical Education & Health, Secondary Education

Certification/1977, Marian College, Indianapolis, IN. Chemistry major in progress (University of Indianapolis)

Memberships: American Chemical Society

Linda Osborn has been the group leader of the General Organic Group at EMS-Heritage since April of 1989. The responsibilities include the coordination of the analyses required on the following instruments:

HPLC (with fluorescence and UV Photodiode detectors)
Purge & Trap PID/ELCD GC (in series)
GC-FID and NPD

Responsibilities also include supervision and training of personnel, familiarity with methods used, facilitating the output of quality data on a timely basis, and performing analyses when needed.

Prior to this assignment, Ms. Osborn was the Group Leader of the Gas Chromatography Group (ECD section) since February of 1985.

Specifically Ms. Osborn has HPLC experience, GC experience with emphasis on ECD (PCB/Pesticides) but including PID/ELCD, FID and NPD experience, and experience with extraction techniques including the following: liquid-liquid, sonication, soxhlet, solid phase and microextractions, with experience in the calculation and interpretation of the corresponding data.

In 1985, Ms. Osborn received certification from Hewlett Packard's Gas Chromatography 5880 training course. When first employed by EMS, Ms. Osborn performed general laboratory analyses including TOX, TOC, Oil and Grease, Surfactant, COD, Conductivity, EP Toxicity Leachates, and some AAS (for plant tanks). Prior to EMS, she was employed by D.A. Lubricant Co. in the Quality Control Department performing the oil and grease analyses including Emission Spectrophotometry, some Atomic Absorption, viscosity determinations, cold-cranking simulations, penetration points, freezing and melting points, flash points and other miscellaneous testing.

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Barbara A. Shrake, General Inorganic Chemistry, Group Leader EMS-Heritage Laboratories, Inc.

Division:

Indianapolis

Experience:

1986 - present, EMS Heritage Laboratories, Inc.

Education:

Bachelors Degree, Chemistry, 1985, Purdue University (IUPUI), Indianapolis, IN

Memberships: American Chemical Society

Mrs. Shrake has been a Group Leader at EMS Heritage, Inc. for three years and is responsible for ensuring analyses are completed efficiently and properly according to quality control criteria. Previously, she has served as an analyst in the General Inorganic Chemistry and Gas Chromatography groups at EMS Heritage.

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Frederic J. Winter, Laboratory Director, Chicago

EMS-Heritage, Laboratories, Inc.

Division:

Chicago

Experience:

1987 - present, EMS Heritage Laboratories, Inc.

1979 - present College of Lake County

19 years with other firms

Education:

AAS Mathematics 1974, Moraine Valley College

BA Environmental Chemistry 1978, Governors State University

MA Environmental Chemistry in progress

Memberships:

American Chemical Society

American Society for Testing and Materials

Water Pollution Control Federation

Central States Water Pollution Control Association

Illinois Association of Wastewater Agencies.

Mr. Winter has been with EMS Heritage Laboratories, Inc. in the position of Laboratory Director since June 1987. He has a diversified background in the environmental sciences with specialization in the fields of water/wastewater, soils and atmospheric analyses, along with specialization in organometallic chemistry. Prior to joining EMS Heritage Laboratories, Inc. he held the position of Laboratory Supervisor, Operations Laboratories, and Laboratory Director at the North Shore Sanitary District in Gurnee, Illinois, and Quality Control Chemist with Witco Chemical Corporation in Chicago, Illinois. Mr. Winter has also taught courses in Environmental Chemistry at the College of Lake County in Grayslake, Illinois since 1979.

Melody Carroll, Assistant Lab Director EMS-Heritage Laboratories, Inc.

Division:

Chicago

Experience:

1985 - present, EMS Heritage Laboratories, Inc.

2 years with other firms

Education:

M.B.A., 1991, Lewis University

B.A. Geography, 1983, University of Delaware Perkin Elmer Flame/Furnace Training Course

Memberships/Active

Registration: ACS, SAS

Melody Carroll's duties as Assistant Laboratory Director include analyst training and supervision, personnel coordination, purchasing and customer maintenance. Previous experience at EMS includes Metals Group Leader and Metals analyst with experience in ICP, Flame, Furnace and Cold Vapor. Additional experience with other firms includes microbiological and wet chemical analyses, sampling, sample log-in, training and supervision.

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Christine Sarkan, Quality Assurance Unit Director

EMS-Heritage Laboratories, Inc.

Division: Chicago

Experience:

1987 - present, EMS Heritage Laboratories, Inc.

3 years with an Environmental Laboratory

Education:

M.B.A. in process, Keller Graduate School

B.A. Biology, 1984, Knox College

Memberships: NA

Christine Sarkan's duties as Quality Assurance Unit Director include data review and validation, method development and verification, laboratory audits, coordination of performance standards and certifications and final review of all analytical reports. Previous experience with another environmental lab includes Wet Chemistry Group Leader and Training Officer; and general chemistry and Flame AA analyses.

Susan Bussey, Group Leader, Semi-Volatiles, Mass Spectroscopy and Gas Chromatography EMS-Heritage Laboratories, Inc.

Division:

Chicago

Experience:

1989-present, EMS Heritage Laboratories, Inc.

2 years with Argonne National Laboratories

1 year with other firms

Education:

B.S. Chemistry, 1986

Hewlett Packard 5890 & RTE 1000 Training Course

Memberships: American Chemical Society

Susan Bussey's duties at EMS include instrument calibration, tuning and maintenance; sample analysis, data reduction and reporting for Volatile Organic Compounds. Previous experience at Argonne includes CLP analyses and data package preparation for Volatiles. Other experience includes chemical research and development using HPLC.

Jennifer Schott, Metals Group Leader

EMS Heritage Laboratories, Inc.

Division:

Chicago

Experience:

1991-present, EMS Heritage Laboratories, Inc.

6 years with other firms

Education:

BS Biology, Minor Chemistry, 1989, Northern Illinois University

Memberships: NA

Jennifer Schott's duties at EMS include supervision of metals preparations and analyses by flame, graphite furnace and ICP. She is responsible for production scheduling, data production and review and instrument maintenance and scheduling. Previous experience includes development of SOP's, ensuring compliance of a TSD facility to EPA regulations, biodegradation of wastewaters and anaerobic degradation of hexavalent chromium insulation.

Susan Sharp, GC/MS Volatiles Group Leader

EMS Heritage Laboratories, Inc.

Division:

Chicago

Experience:

1989-present, EMS Heritage Laboratories, Inc.

2 years experience with other firms

Education:

BS Biology, Minor Chemistry, 1986, Illinois State University

Memerships:

NA

Susan Sharp's duties at EMS include supervising analyticals on ground water and soil on a packed column and a capillary column GC/MSD's. She also tests APG's and USEPA Water Pollution Studies samples.

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Dawn Siekerman, Group Leader, General Chemistry

EMS-Heritage Laboratories, Inc.

Division:

Chicago

Experience:

1988-present, EMS Heritage Laboratories, Inc.

5 months with other firms

Education:

. B.S. Biology, 1985, Illinois Benedictine College, Lisle, IL

Memberships:

American Chemical Society

Northern Illinois Technicians Association Group

Dawn Siekerman's duties as Chemist include wet chemical analyses and training and supervision of summer and part time employees. Additional experience at EMS includes Metals extractions and preparations, and PCB extractions. Experience with other firms includes Quality Control testing of foods and products.

Deborah Edwards, Project Coordinator, Chicago

EMS-Heritage Laboratories, Inc.

Division:

Chicago

Experience:

1989 - present, EMS Heritage Laboratories, Inc.

5 years with other firms

Education:

2 years undergraduate study in business

Memberships: NA

Deborah Edward's duties as Project Coordinator include client contact, price quoting, and project coordination from sampling through reporting. Project management requires ensuring regulatory compliance and appropriate methods and detection limits are reported. Previous experience includes clerical, word processing in customer service and complaints departments.

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Jan Dillow, Laboratory Director

EMS Heritage Laboratories, Inc.

Division:

Kansas City

Experience:

1990 - present, EMS Heritage Laboratories, Inc.

10 years with other firms

Education:

BS Biology and Sociology, 1969, William Woods College

Memberships:

National Water Pollution Control Federation

Missouri Wastewater Control Coalition Kansas City Chamber of Commerce

University of Missouri, Engineering Division Continuing Education Committee

National Association for Female Executives

Jan Dillow administered an office in Kansas City for six years. After completion of the present facility, she has managed the complex for four years. Mrs. Dillow currently serves as Laboratory Director at EMS Heritage.

Scott Meeks, Facility Chief Chemist

EMS-Heritage, Laboratories, Inc.

Division:

Kansas City

Experience:

1990 - present, EMS Heritage Laboratories, Inc.

9 years with Other Firms 2 years with EMS

Education:

BS Chemistry 1978, Bridgewater State College

Memberships:

American Chemical Society

American Society for Mass Spectrometry

Mr. Meeks is responsible for the technical aspects of the daily operation of the laboratory. He interfaces with production, quality control and project management staff to assure the proper and timely conduct of analyses for clients. Prior to his position as Facility Chief Chemist, Mr. Meeks supervised all organic analytical work and was responsible for the performance of GC and GC/MS analyses.

Horst Kehl, Quality Assurance Unit Director

EMS-Heritage, Laboratories, Inc.

Division:

Kansas City

Experience:

1990 - present, EMS Heritage Laboratories, Inc.

Education:

BS Chemistry, 1959, NE Missouri State University, Kirksville, MO

MS Organic Chemistry, 1963, University of Idaho, Moscow, ID

PhD Biomedical Science, 1990, Eurotechnical Research University, South Hampton, England

Memberships:

American Chemical Society

Sigma X

American Society for Pharmacology and Experimental Theraupetics (elected)

American Assoc. for the Advancement of Science

Dr. Kehl is responsible for coordinating Quality Assurance and Quality Control activities. He interfaces with production staff and management to assure the generation of superior quality data. Prior to his work at EMS, Dr. Kehl held several positions in research and teaching both in the United States and abroad. He has served as an abstractor for Chemical Abstracts and holds two US patents. Dr. Kehl has been a visiting professor at the University of Zagrub, the Toxicology Institute at the University of Munich, Germany, the University of San Marcos, Lima, Peru and the University of Autonma, Mexico City, Mexico.

Roger Rowan, GC Group Leader

EMS-Heritage, Laboratories, Inc.

Division:

Kansas City

Experience:

1989 - present, EMS Heritage Laboratories, Inc.

Education:

BS Chemistry 1989, Southern Illinois University

Memberships: NA

Mr. Rowan serves as the primary supervisor in the Gas Chromatography group. He is responsible for the timely production of quality analytical data. He reviews all raw data to insure accuracy. Mr. Rowan also served previously as the General Chemistry Group Leader, where he performed several conventional pollutant general chemistry procedures as well as metals analysis and preparation.

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Margarita Rozenfeld, Inorganic Chemistry, Group Leader

EMS-Heritage Laboratories, Inc.

Division:

Kansas City

Experience:

1990 - present, EMS Heritage Laboratories, Inc.

9.5 years with other firms

Education:

MS Chemical Engineering 1980, Moscow, USSR

Memberships: NA

Ms. Rozenfeld serves as the primary supervisor in the general chemistry and metals groups. She is responsible for the timely production of quality analytical data. Prior to this position she managed a laboratory with the Biochemicae Company, USSR.

Kevin Miller, GC/MS Group Leader

EMS-Heritage Laboratories, Inc.

Division:

Kansas City

Experience:

1990 - present, EMS Heritage Laboratories, Inc.

9 years with Other Firms

Education:

BS Chemistry 1980, Kearney State College, Kearney, NE

Memberships: NA

Mr. Miller serves as the primary supervisor/analyst in the Gas Chromatography/Mass Spectrometry group. He is responsible for the timely production of quality analytical data. He reviews all raw data to insure accuracy. Mr. Miller worked as a GC/MS operator at another environmental laboratory for $2\frac{1}{2}$ years. Prior to that, he worked as a chemist in an environmental lab in the areas of Atomic Absorption, Gas Chromatography, BOD/COD, and ASTM physical tsting. Other prior positions include an oil well drilling Fluids Engineer and a Chemical Technician.

Michele Sakwa, Laboratory Director EMS-Heritage Laboratories, Inc.

Division:

Charlotte

Experience:

1987 - present, EMS Heritage Laboratories, Inc.

9 years with Other Firms

Education:

BA Chemistry/Physics 1980, Kean College

Memberships: American Chemical Society

Ms. Sakwa serves as Lab Director and is responsible for overall operations of the laboratory. Prior to coming to EMS Heritage, she held positions as Technical Marketing Representative, Project Manager, Technical Field Engineer, and Research Scientist. She has published a technical paper and is a member of ACS. She has also received training in GLP's, OSHA Hazcom, RCRA/SDWA/CWA regulations and compliance.

Helmuth M.B. Janssen, Quality Assurance Unit Director

EMS-Heritage, Laboratories, Inc.

Division:

Charlotte

Experience:

1988 - present, EMS Heritage Laboratories, Inc.

Education:

BS Chemistry 1973, Hehgelo Institute of Technology (The Netherlands)

Memberships: American Chemical Society

Mr. Janssen is responsible for coordinating Quality Assurance and Quality Control activities. He interfaces with production staff and management to assure the generation of superior quality data. Prior to his work at EMS Heritage, he served as a Research Chemist and Laboratory Supervisor involved in data review, methods development, marketing and QA/QC protocol development.

Steven H. Guptill, Senior Chemist

EMS-Heritage, Laboratories, Inc.

Division:

Charlotte

Experience:

1986 - present, EMS Heritage Laboratories, Inc.

1 year with other firms

Education:

BS Chemistry 1985, Manhattan College, Riverdale, NY

Memberships: American Chemical Society

Mr. Guptill serves as senior chemist for EMS Heritage. He is responsible for providing technical analytical support for all groups. Prior to this position Mr. Guptill held positions as Project Manager, Organic Section Manager and a GC/MS chemist.

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Carla Keever

EMS-Heritage, Laboratories, Inc.

Project Assignment: Project Manager, Charlotte

Experience:

1987 - present

4 years with other firms

Education:

BS Chemistry 1985, University of Kentucky, Lexington, KY

Memberships: NA

Ms. Keever serves as Project Manager in the Technical Services Group. She is responsible for coordinating all facets of laboratory testing with the client. She has considerable experience in environmental monitoring and testing. Prior to coming to EMS, she was a GC/MS section leader. She has also performed flame, graphite furnace and ICP metals analyses as well as volatile and semivolatile organics analyses using GC, GC/MS and a variety of detectors.

Malgorzata Kraska, Metals Group Leader

EMS-Heritage, Laboratories, Inc.

Division:

Charlotte

Experience:

1988 - present, EMS Heritage Laboratories, Inc.

11 years with other firms

Education:

BS Chemistry 1974, Technikum Chemiczne, Kjelce, Poland

Memberships: N.A.

Ms. Kraska serves as the primary supervisor in the Metals Group. She is responsible for the timely production of quality analytical results. She has also been an analyst in the General Chemistry Group.

Michael Piwowar, GC/MS Group Leader

EMS-Heritage, Laboratories, Inc.

Division:

Charlotte

Experience:

1990 - present, EMS Heritage Laboratories, Inc.

2 years with other firms

Education:

BS Biology 1988, BS Chemistry 1988, University of South Carolina, Columbia, SC

Memberships: NA

> Mr. Piwowar serves as the primary supervisor in the GC/MS Group. He is responsible for the timely production of quality analytical data in this group. He has over 3 years of experience in volatile and semivolatile analyses on various manufacturers' instrumentation.

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Janet Williams, General Chemistry Group Leader

EMS-Heritage, Laboratories, Inc.

Division:

Charlotte

Experience:

1987 - present, EMS Heritage Laboratories, Inc.

1 year with other firms

Education:

BA Chemistry 1985, BS Biology 1985, University of North Carolina, Charlotte, NC

Memberships: NA

Ms. Williams serves as the primary supervisor in the general chemistry group. She is responsible for the timely production of quality analytical data. Prior to her group leader position she worked as an analyst.

Pen Winters, GC Group Leader

EMS-Heritage, Laboratories, Inc.

Division:

Chariotte

Experience:

1988 - present, EMS Heritage Laboratories, Inc.

3 years with other firms

Education:

BS Medical Technology 1984, Medical College of Virginia, Virginia Commonwealth

University, Richmond, VA

Memberships: American Chemical Society

American Society of Clinical Pathologists

Ms. Winters serves as the primary supervisor in the GC Group. She is responsible fo the timely production of quality analytical data in this group. Prior to coming to EMS Heritage she held positions as a Medical Technician, Mass Spectroscopist, and Clinical Microbiologist.

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APPENDIX A Selected References

Quality Control/Quality Assurance

- 1. <u>Handbook for Analytical Quality Control in Radioanalytical Laboratories</u>. EPA 600/7-77-088. 1977.
- 2. <u>Handbook for Analytical Quality Control in Water and Wastewater</u>. EPA 600/4-79-019. March 1979.
- 3. <u>Manual for Analytical Quality Control for Pesticides and Related Compounds in Human and Environmental Samples</u>. EPA 600/1-79-009. 1979.
- 4. Taylor, John K. <u>Principles of Quality Assurance of Chemical Measurements</u>, U.S. Department of commerce. 1987.
- 5. <u>Choosing Cost-Effective QA/QC Programs for Chemical Analysis</u>. EPA/600/4-85/056. 1985.
- 6. <u>Practical Statistics for Analytical Chemists</u>. R.L. Anderson, Van Nostrand Reinhold, 1987.
- 7. Annual Book of ASTM Standards. Vol. 11.01 and 11.02 (Water I and II). American Society for Testing and Materials. 1989.
- 8. Engineering Support Branch Standard Operating Procedures and Quality Assurance Manual. U.S. EPA Regions IV Environmental Services Division. April 1986.
- 9. <u>Manual for Certification of Laboratories Analyzing Drinking Water</u>. EPA 570/9-90-008. April 1990.
- 10. <u>Microbiological Methods for Monitoring the Environment</u>. EPA 600/8-78-017. 1978.
- 11. Methods for Chemical Analysis of Water and Wastes. EPA 600/4-79-020. Revised March 1983.
- 12. <u>Methods for Collection and Analysis of Aquatic Biological and Microbiological Samples.</u> Book 5, Chapter A4. U.S. Geological Survey. 1977.
- 13. Methods for Determination of Inorganic Substances in Water and Fluvial Sediments. Book 5, Chapter Al. U.S. Geological Survey. 1979.

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- 14. Methods for Measuring the Acute Toxicity of Effluents to Freshwater and Marine Organisms. Third Edition. EPA/600/4-85/019, 1985.
- 15. <u>Methods for the Determination of Organic Compounds in Drinking Water</u>. EPA 600/4-88/039, December 1988.
- 16. Methods for the Determination of Organic Substances in Water and Fluvial Sediments. Book 5, Chapter A3. U.S. Geological Survey. 1983.
- 17. Methods of Soil Analysis. Part 2, Chemical and Microbiological Properties. A.L. Page, ed., 2nd edition. 1982.
- 18. Official Methods of Analysis of the Association of Official Analytical Chemists. 2nd edition. AOAC. 1989.
- 19. <u>Procedures for Handling and Chemical Analysis of Sediments and Water Samples.</u> EPA/Corps of Engineers CE-81-1. March 1981.
- 20. RCRA Ground-water Monitoring Technical Enforcement Guidance Document, (TEGD). EPA OSWER-9950.1. September 1986.
- 21. Short-Term Methods for Estimating the Chronic Toxicity of Effluents and Receiving Waters to Freshwater Organisms. Second Edition. EPA/600/4-89/001. 1989.
- 22. Short-Term Methods for Estimating the Chronic Toxicity of Effluents and Receiving Waters to Marine and Estuarine Organisms. EPA/600/4-87/028. 1988
- 23. Standard Methods for the Examination of Water and Wastewater. 16th Edition. 1985.
- 24. Test Methods for Evaluating Solid Waste Physical/Chemical Methods. EPA/SW 846. 3rd Edition, 1986 and its Revision I Dated December 1987.
- 25. <u>USEPA Contract Laboratory Program Statement of Work for Inorganic Analysis</u>. EPA. SOW No. 7/88. July 1988.
- 26. <u>USEPA Contract Laboratory Program Statement of Work for Organic Analysis</u>. EPA. SOW No. 2/88. February 1988.

- 27. <u>USEPA Contract Laboratory Program Statement of Work for Inorganic Analysis</u>. EPA. ILM01. 1990.
- 28. <u>USEPA Contract Laboratory Program Statement of Work for Organic Analysis</u>. EPA. OLM01. 1990.
- 29. Quality Assurance Program Requirements for Environmental Programs. ANSI/ASQC-E4-19xx, September 1991.

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APPENDIX C SAFETY ISSUES

MANDATORY SAFETY PROCEDURES & PRACTICES

This section is not meant as a comprehensive manual of safety practices, but only as a indicator of some of the important ones and how they are specifically applied at EMS. The Safety Officer is responsible for responding to any questions concerning safety procedures and policies.

SAFETY GLASSES

Approved protective eyewear must be worn by all EMS Heritage Personnel at all times in all laboratory areas. Laboratory areas include all areas with quarry tile flooring and all working areas in front of fume hoods and lab benches. Non-prescription safety glasses and goggles are available free of charge to all employees. Contact lenses are not recommended for use in the laboratory and are not considered protective. All visitors should be advised to wear eye protection and should be offered a pair of safety glasses to use during their tour of designated lab areas.

Failure of an individual to wear protective eye wear in a designated area will result in the forfeiture of the next payable QA bonus to that individual.

INSPECTIONS

Periodic inspections of all laboratory areas will be performed by the Safety Officer. Inspections will assess current operating procedures and practices. Recommendations for improvement may be made or immediate remedial action may be necessary. Results of these inspections will be discussed with the Group Leaders, the Chief Chemist and the personnel directly involved with the area of inspection.

SMOKING

No smoking is allowed in any marked laboratory area, the loading dock, common hallways or the restrooms. Smoking is permitted only in the conference room, in the breakroom at designated tables, and in a private office or office area. This policy applies to visitors as well. Any individual observed smoking in a restricted area will forfeit their next payable QC bonus.

FUME HOODS

Fume hoods must be used for all procedures involving concentrated acids, bases, and solvents. Fume hoods are maintained to deliver a face velocity of 100 fpm. When an individual is not working at the hood, the face glass should be closed. The glass curtain

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should be raised only to the extent necessary for easy access. Protective eye wear must be used when working at or near a fume hood. Each hood is marked to indicate a maximum sash level for safe operation.

PROTECTIVE CLOTHING

Lab coats, gloves, and shoe covers are available for use by all EMS employees. Protective clothing should not be worn into non-laboratory areas since it may transfer hazardous material onto surfaces otherwise assumed to be clean.

It is recommended that a lab coat be worn in all laboratory areas to protect not only the individual but their clothing. Shoe covers are also recommended for procedures where splashing hazardous samples, or chemicals is probable.

Protective gloves should be used for procedures involving: handling hazardous waste samples; handling hot or cold materials; handling acids, solvents, or bases. Gloves should be made of a material which is impervious to the hazardous nature of the substance being handled.

General Lab Work - Natural Latex (Surgeons) Gloves.

Hot/Cold Work - Zetex, Leather or other non-flammable insulating material.

Acids - Nitrile, Neoprene

Bases - Neoprene, Nitrile

Solvents - Neoprene, Nitrile, Viton

Alcohol - Neoprene, Nitrile

Long hair must be tied back to avoid hazards such as fire and chemical contact.

FIRE/FIRE EXTINGUISHERS

Fire extinguishers are highly visible and located at all EMS Heritage locations. All extinguishers are type ABC or Halon. ABC extinguishers may be used on paper trash and wood, burning liquids, and electrical equipment. It is recommended however, that Halon be used on electrical fires particularly in computers or instruments. A commercial safety service checks and maintains the fire extinguishers. Powder for extinguishing flammable metal fires is available in the safety cabinet. (See Evacuation Memo)

CLEANLINESS

A very important part of safety is good housekeeping. This includes:

- l. Maintaining an organized work area.
- 2. Cleaning up every spill immediately, even water.

- 3. Keeping small objects off the floor and large objects out of traffic paths, particularly exit ways.
- 4. Returning unused chemicals, glassware, or equipment to its proper storage area.
- 5. Disposing of waste (chemicals, napkins, tongue blades, pipets) in the proper container.
- 6. Disposing of broken glassware immediately in the proper containers.
- 7. Do not place contaminated chemical transfer utensils on the bare countertop.

SPILL CLEANUP

Spills must be cleaned up immediately after they happen. Even a water spill should be removed. Special kits are available for neutralizing and absorbing various liquid spills for disposal. These kits are located in the safety cabinet. If in doubt as to the type of spill, inform the Safety Officer; always assume the worst. Ask the Safety Officer when in doubt as to the proper method for cleanup or disposal. Chemicals should never be reused following a spill. Dry chemicals should be treated with as much caution as liquids. Many times, special equipment may be necessary (i.e. respirators, splash suit, caution signs, etc.) and should be used. If in doubt as to the safety equipment required for a particular spill, consult the Safety Officer.

REPORTING ACCIDENTS

All accidents no matter how small must be reported to the Safety Officer and the Group Leader. In some cases a form may need to be filled out describing the situation and result. If medical attention is indicated, the Safety Officer may direct the individual to the emergency care center or to a hospital.

COMPRESSED GASES

All gas cylinders must be affixed to an immobile surface when in use. All cylinders must be transported with the protective valve cap in place. Cylinders in storage must be chained to the wall or placed in the protective cage. Empty cylinders should be segregated from full cylinders for easy access during exchange. Flammable gases must be segregated from oxidizers and non-flammable gases (this includes 'empty' tanks). Empty cylinders should be marked as such.

COMMON PROCEDURES

- l. Always add acid to water, never water to acid.
- 2. Never pipet by mouth, use a suction bulb.

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- 3. Never store food or drink in a sample cooler
- 4. Never eat in the laboratory, use the breakroom.
- 5. Always lubricate glass tubing before inserting in stopper.
- 6. Always dispose of waste properly, consider the janitor.
- 7. Keep balance and balance area clean.
- 8. Always read the label and MSDS.
- 9. Always report accidents to the Safety Officer.
- 10. When in doubt about an element of Safety ASK QUESTIONS
- 11. All sample and chemical containers are to be closed overnight.

SAFETY SHOWERS/EYE WASH

Safety showers should always be used in instances of gross contamination of clothing or skin. The eye wash should always be used when foreign matter is introduced into or around the eye. The safety shower and eye wash stations are periodically checked and maintained. Operation of the safety shower includes disrobing from the contaminated clothing and rinsing for at least 15 minutes. The eye wash should be used at least 15 minutes and contact lenses should be removed as soon as possible.

CHEMICAL HANDLING/STORAGE

Read the label and MSDS on each and every chemical used. The safety related item which should be understood are: Fire Hazard, Health Hazard, Reactivity Hazard, Protective Equipment, Disposal, Storage, and First Aid. If the chemical is used in making a stock solution, clearly label the new container with the date, analysts initials, expiration date, contents and any other pertinent information. Never use anything from an unlabeled container. Follow the storage recommendations. Flammables always go in an approved container and storage cabinet. Acids always go in an acid storage cabinet. Incompatible chemicals should not be stored together. Large quantities of chemicals should not be 'pigeon holed' at work locations. Properly dispose of and reorder chemicals found to be past their expiration date. Transport chemicals in protective containers if breakage is a possibility. Keep chemical containers clean with tight fitting lids and readable labels. In disposing of chemicals, samples, damaged equipment or expendable materials consider the potentialities of the action (eg. incompatible waste, exposure to the janitor).

FIRST AID

First aid kits are maintained by a commercial first aid company. Any of the materials in these kits may be used to self treat minor accidents or common ailments. All accidents, no matter how small, must be reported to the Group Leader and the Safety Officer. The instructions and labels on all medications and materials should be read and understood by the user prior to application or use. A Red Cross first aid manual is available for reference in the safety cabinet.

SAFETY CABINET

All safety related supplies are maintained in a dedicated cabinet centrally located in the laboratory. The contents includes; fire extinguishers; respirators; spill clean up kits; gloves; splash suits; shoe covers; safety manuals; and material safety data sheets (MSDS).

These items are regularly inventoried, however, all destructive uses of these items should come to the attention of the Safety Officer. All items should be replaced after use and not stored elsewhere in the laboratory. Any items routinely necessary for specific procedures should be obtained and stored separately near the area in which they are used.

EATING/DRINKING IN LAB

No eating is allowed in any laboratory area. Drinking is not recommended. Never store food or beverages in a cooler/refrigerator used for incubation, chemical or sample storage.

ALCOHOL/DRUG USE

No person under the influence of alcohol or drugs will be allowed to work in any laboratory area. Such an individual may be subject to disciplinary action which may include termination. Any individual who is required to take a prescribed medication which might affect his/her judgement should inform the Safety Officer so that non-laboratory work can be assigned.

VISITOR AREAS

We want to show off the lab in a safe and efficient manner. In general, visitors should be restricted to the hallways and non-quarry tile areas. Viewing windows are available and should be used. Anyone wishing to tour inside the lab should make prior arrangements with the Chief Chemist or the Safety Officer. Visitors should always be supplied with safety glasses/goggles. Visitors must be accompanied on tours, not only for their safety but to maintain confidentiality.

EVACUATION PROCEDURES

In case evacuation of the building is necessary, because of fire, chemical spill, or other emergency, the person discovering the accident should page "70" and inform all occupants to evacuate the building and the source or location of the accident. The Chief Chemist or designee will make an accounting of all personnel outside while the Safety Officer or designee will assure the proper authorities are notified.

MATERIAL SAFETY DATA SHEETS

Material safety data sheets are maintained for each chemical used in the lab. These sheets provide information on handling, disposal, spill cleanup, health hazards, physical/chemical data and safety measures. The MSDS should be consulted by the analyst prior to using any new or unfamiliar chemical. Each analyst should be familiar with the major hazards of each

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chemical they work with (flash point, contact hazard, etc.) The MSDS manual is kept in the safety cabinet and should be returned after each use. Each chemical supplier is required to provide a MSDS with each chemical shipped.

WORKING AFTER HOURS

Employees working after business hours or on weekends must inform their Group Leaders of when and for how long they will work. Procedures requiring the use of concentrated acids, bases or solvents should be avoided by individuals working alone.

CONTAMINATION CONTROL

All sample and chemical containers must be closed using screw top lids, ground glass stoppers, or a similar device and not allowed to stand open overnight. Before disposal all materials (rags, towels, etc.) used to clean up volatile solvent spills should be kept in an open container in a working fume hood until all traces of the solvent have evaporated. All utensils (spoons, spatulas, pipets, etc.) used to transfer materials should be cleaned weekly or as needed. All reasonable measures should be taken to avoid environmental contamination.

Soiled lab coats should be exchanged for clean coats frequently and once used should not leave the laboratory area. Soiled lab coats should be stored in the proper hamper. Personal items (coats, etc.) should not be taken into the lab areas.

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APPENDIX D DEFINITIONS AND ACRONYMS

ABSORBANCE - a measure of the decrease in incident light passing through a sample into the detector. It is defined mathematically as:

$$A = \frac{I(solvent)}{I(solution)} = \log \frac{Io}{I}$$

Where, I = radiation intensity

ACCURACY - The closeness of agreement between an observed value and an accepted reference value. When applied to a set of observed values, accuracy will be a combination of a random component and of a common systematic error (or bias) component.

ALIQUOT - a measured portion of a field sample taken for analysis.

ANALYSIS - a specific test applied to a specific sample.

ANALYSIS DATE/TIME - the date and military time (24-hour clock) of the <u>introduction</u> of the sample, standard, or blank into the analysis system.

ANALYTE - the element or ion an analysis seeks to determine; the element of interest.

ANALYTICAL SAMPLE - Any solution or media introduced into an instrument on which an analysis is performed excluding instrument calibration, initial calibration verification, initial calibration blank, continuing calibration verification and continuing calibration blank. Note the following are all defined as analytical samples: undiluted and diluted samples (EPA and non-EPA), predigestion spike samples, duplicate samples, serial dilution samples, analytical spike samples, post-digestion spike samples, interference check samples (ICS), CRDL standard for AA (CRA), CRDL standard for ICP (CRI), laboratory control sample (LCS), preparation blank (PB) and linear range analysis sample (LRS).

ANALYTICAL SET - the basic unit for analytical quality control. Also known as <u>sample</u> <u>set</u> or <u>analytical batch</u>. The analytical set is defined as samples which are analyzed (or sampled together) with the same method sequence, the same lots of reagents and with the same treatment common to all samples. The samples must have been analyzed (or collected) within the same specified time period or in continuous sequential time periods. Samples in each set should be of similar composition (same matrix).

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ANALYTICAL SPIKE - a post-digestion spike. The addition of a known amount of standard after digestion/extraction.

AUDITS - A systematic check to determine the quality of the operation of some function or activity.

<u>Performance Audits</u>: Quantitative data are independently obtained for comparison with routinely obtained data in a measurement system. Examples of these audits are EPA performance evaluation programs, commercial performance evaluation programs, split sampling program involving at least two laboratories, blind spike samples.

<u>Systems Audit</u>: These are qualitative in nature and consist of an on-site review and evaluation of a laboratory or field operations quality assurance system and physical facilities for sampling calibration and measurements.

AUTOZERO - zeroing the instrument at the proper wavelength. It is equivalent to running a standard blank with the absorbance set at zero.

AVERAGE INTENSITY - the average of two different injections (exposures).

BACKGROUND CORRECTION - a technique to compensate for variable background contribution to the instrument signal in the determination of trace elements.

BAR GRAPH SPECTRUM - a plot of the mass-to-charge ratio (m/e) versus relative intensity of the ion current.

BATCH - a group of samples which behave similarly with respect to the procedures being employed and which are processed as a unit from a specific client that are submitted on the same day and are of the same matrix. For QC purposes, if the number of samples in a group is greater than 20, then each group of 20 samples or less will all be handled as a separate batch. Also known as a "case".

BIAS - the deviation due to matrix effects of the measured value $(x_n - x_u)$ from a known spiked amount. Bias can be assessed by comparing a measured value to an accepted reference value in a sample of known concentration or by determining the recovery of a known amount of contaminant spiked into a sample (matrix spike). Thus, the bias (B) due to matrix effects based on a matrix spike is calculated as:

$$B = (x_{\bullet} - x_{\bullet}) - K$$

where:

 x_{x} = measured value for spiked sample

 x_u = measured value for unspiked sample, and

K = known value of the spike in the sample.

Using the following equations yields the percent recovery (%R). The value of %R may be used to correct the measured values for that batch of data. Thus,

$$%R = 100 (x_s - x_u)/K$$

and

$$x_c = 100 \ (x_u/\%R)$$

where $x_c = corrected value$.

Correction for bias may be applicable except when the recovery of the analyte is between 80% to 120%, when using bias correcting methods (e.g. isotope dilution), or when recovery is greater than 100%. In some cases, when the uncorrected concentration is above a regulatory threshold bias correction is not needed. The measured value will be considered the minimum amount present. In all other cases correction for bias may be used. Bias correction will only be applied when required by regulation.

BLANKS - an artificial sample of an analytical matrix designed to monitor the introduction of artifacts into the system.

a. Field Quality Control Blanks

- 1. <u>Field Blanks</u>: Blanks of analyte free water that are prepared <u>on-site</u> by filling appropriate sample containers with the water, adding appropriate preservatives, sealing the containers, and completing the appropriate documentation. These blanks should be prepared during the middle to end of a sampling event by filling sample containers with water from the equipment decontamination water transport containers. They are to be treated, stored, transported, and analyzed in a same manner as the sample group for which it was intended. These blanks may be submitted for all water parameter groups.
- 2. Equipment Blank: Blanks of analyte-free water that are prepared onsite by pouring the equipment decontamination water through decontaminated field equipment. Appropriate sample containers, for each analyte group must

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be used, preservatives added, if required, and appropriate documentation must be completed. These blanks are to be stored, transported and analyzed with the intended parameter groups. At least one equipment blank is required for each water and solid matrix analytical group, and must be collected at the beginning of the sampling episode. If field decontamination is performed on-site, additional equipment blanks must be submitted for all water and solid matrix analytical groups.

3. <u>Trip Blank</u>: These blanks are required only for water samples for VOCs. Blanks of volatile organic free water are prepared by the organization that is providing the sample containers. These are transported to the site with the empty VOC sample containers, and shipped to the analyzing laboratory in the same containers as the VOC samples. They remain <u>unopened</u> for the entire trip. Proper labeling and documentation must be completed. A trip blank must be submitted for each cooler that transports VOC samples.

b. <u>Laboratory</u>

- 1. <u>Method Blank</u>: A blank of an appropriate analyte-free matrix that is processed (digested, extracted, etc.) and analyzed with a specified sample set (See BLA02, BLA01).
- 2. Reagent Blank: An aliquot of analyte-free water or solvent that is analyzed with a sample set. (See BLA01).

CALIBRATION - process by which the correlation between instrument response and actual value of a measured parameter is determined.

- a. <u>Calibration Curve</u>: a curve which plots the concentration of known analyte standards against the instrument response to the analyte. Also known as a <u>Standard</u> Curve.
- b. <u>Calibration Standard</u>: Solutions or dilutions of a substance or material with a verifiable accuracy which are used to evaluate the sample property of an unknown sample. In analytical terms, these standards are used to establish a calibration curve or standard instrument response factors.
- c. <u>Continuing Calibration Standard</u>: Standards that are analyzed during an analytical set to verify the accuracy of the calibration curve.
- d. <u>Internal Standard</u>: a compound having similar chemical characteristics to the compounds of interest but which is not normally found in the environment or does not interfere with the compounds of interest. A known and specified concentration

of the standard is added to each sample <u>prior to analyses</u>. The concentration in the sample is based on the response of the internal standard relative to that of the calibration standard and the compound in the standard.

CASE - a finite, usually predetermined number of samples collected over a given time period from a particular site. Case numbers are assigned by the Sample Management Office. A case consists of one or more Sample Delivery Groups.

CERCLA - Comprehensive Environmental Response, Compensation, and Liability Act ("Superfund").

CLP - Contract Laboratory Program (USEPA).

CHARACTERIZATION - a determination of the approximate concentration range of compounds of interest used to choose the appropriate analytical protocol.

COEFFICIENT OF VARIATION (CV) - the standard deviation as a percent of the arithmetic mean.

CONCENTRATION LEVEL (low or medium) - characterization of soil samples or sample fractions as low concentration or medium concentration is made on the basis of the laboratory's preliminary screen, if performed or the client's designation.

CONTINUING CALIBRATION - analytical standard run at a specified frequency to verify the calibration of the system.

CONTINUOUS LIQUID-LIQUID EXTRACTION - used herein synonymously with the terms continuous extraction, continuous liquid extraction, and liquid extraction. This extraction technique involves boiling the extraction solvent in a flask and condensing the solvent above the aqueous sample. The condensed solvent drips through the sample, extracting the compounds of interest from the aqueous phase.

CONTRACT REQUIRED DETECTION LIMIT (CRDL) - minimum level of detection acceptable under a contract Statement of Work. Equivalent to Contract Required Quantitation Limit (CRQL).

CONTROL LIMITS - a range within specified measurement results must fall to be compliant. Control limits may be mandatory, requiring corrective action if exceeded, or advisory, requiring that noncompliant data be flagged.

CONTROL SAMPLE - a QC sample introduced into a process to monitor the performance of the system.

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CONFIDENCE LEVEL - the statistical probability associated with an interval of precision (or accuracy) values in a QC chart. The values of confidence intervals are generally expressed as percent probability. It is a commonly accepted convention that the result being tested is <u>significant</u> if the calculated probability is greater than 90 percent, and is <u>highly significant</u> if the probability is greater than 99 percent.

CORRELATION COEFFICIENT - a number (r) which indicates the degree of dependence between two variables (concentration - absorbance). The more dependent they are the closer the value to one. Determined on the basis of the least squares line.

CWA - Clean Water Act.

DATA QUALITY - the totality of features and characteristics of data that bears on its ability to satisfy a given purpose. The characteristics of major importance are accuracy, precision, completeness, representativeness, and comparability. These characteristics are defined as follows:

- a. Accuracy the degree of agreement of a measurement (or an average of measurements of the same thing), X, with an accepted reference or true value, T, usually expressed as the difference between the two values, X-5, or the difference as a percentage of the reference or true value, 100 (X-T)/T, and sometimes expressed as a ratio, X/T. Accuracy is a measure of the bias in a system.
- b. <u>Precision</u> a measure of mutual agreement among individual measurements of the same property, usually under prescribed similar conditions. Precision is best expressed in terms of the standard deviation. Various measures of precision exist depending upon the "prescribed similar conditions."
- c. <u>Representativeness</u> expresses the degree to which data accurately and precisely represent a characteristic of a population, parameter variations at a sampling point, a process condition, or an environmental condition.
- d. <u>Comparability</u> expresses the confidence with which one data set can be compared to another.

DATA QUALITY OBJECTIVES - a set of specifications that the environmental data must meet in order to be acceptable for its intended use in a program area. DQOs are commonly established for limits of detection and quality of data (precision, accuracy, representativeness and comparability).

DETECTION LIMITS - The smallest concentration/amount of an analyte of interest that can be measured with a stated probability of significance. Detection limits must be further defined as:

- a. Method Detection Limit the smallest concentration of an analyte of interest that can be measured and reported with 99 percent confidence that the concentration is greater than zero. The MDLs are determined from the analysis of a sample in a given matrix containing the analyte at a specified level. Determination of MDLs must be done by procedures determined in Appendix B of 40 CFR, part 136. Equivalent procedures to determine MDLs must be approved by the applicable regulatory agencies. Note: Reporting results at the MDL may result in a relative uncertainty of the result of greater than ± 100%.
- b. Practical Quantitation Limit the smallest concentration of an analyte of interest that can be reported with a specific degree of confidence. PQLs shall be determined in the same way as MDLs by using the procedures outlined in Appendix B of 40 CFR, Part 136. The standard deviation (sd) derived from the procedures will be used to calculate the PQL: PQL = 10 sd which corresponds to an uncertainty of \pm 30 percent in the measured value at the 99 percent confidence level. Also known as the "Reporting Detection Limit" or the Limit of Quantitation (LOQ).

INSTRUMENT DETECTION LIMIT - the smallest amount of an analyte of interest that generates an instrument response (signal) under prescribed conditions such that the magnitude of the signal is larger than the absolute uncertainty (error) associated with it. IDLs are determined by multiplying by 3.143 (3.0 for CLP) the average standard deviation obtained for the analysis of a standard solution (each analyte in reagent water) at a concentration of 3x-5x the estimate of the IDL on three nonconsecutive days with seven consecutive measurements per day (compute the standard deviation for each day, then average the standard deviations).

DAY - unless otherwise specified, day shall mean calendar day.

DIGESTION LOG - an official record of the sample preparation (digestion).

DISSOLVED METALS - analyte elements which have <u>not</u> been digested prior to analysis and which will pass through a 0.45 um filter.

DRY WEIGHT - the weight of a sample based on percent solids. The weight after drying in an oven.

Dry Weight = Wet Weight Basis
$$x \frac{100}{D}$$

Where D = Total Solids, %.

DUPLICATE (LAB) - a second aliquot of a sample that is treated the same as the original sample in order to determine the precision of the method. Also known as a laboratory

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replicate (See DUP01 and DUP02, Replicate Sample, Matrix Spike Duplicate, Field Duplicate).

ENVIRONMENTAL SAMPLE - any sample from a natural source or sources that reasonably may be expected to contribute pollution to or receive pollution from ground waters or surface waters. This includes, but is not limited to: receiving waters; waters used to define natural background conditions; soils; sediments; industrial, domestic or municipal discharge effluents; chemical storage or handling facilities; waste disposal facilities or areas; industrial or agricultural chemical handling or application areas; surface water run-off; and facilities for handling or applying of chemicals for weed or insect control

EPA - Environmental Protection Agency (USEPA)

EQUIPMENT BLANK - see Equipment Rinsate.

EQUIPMENT RINSATE - a sample of the media which has been used to rinse the sampling equipment. The media used for rinsing is acceptable if the concentration of any analyte of concern in the media is no higher than the highest of either:

- a. the detection limit, or
- b. five percent of the regulatory limit for that
- c. five percent of the measured concentration in that

It is collected after completion of decontamination and prior to sampling. This blank is useful in documenting adequate decontamination of sampling equipment.

EXTRACTABLE - a compound that can be partitioned into an organic solvent from the sample matrix and is amenable to gas chromatography. Extractables include semivolatile (BNA) and pesticide/Aroclor compounds.

FIELD DUPLICATES - independent samples which are collected as close as possible to the same point in space and time. They are two separate samples taken from the same source, stored in separate containers, and analyzed independently. These duplicates are useful in documenting the precision of the sampling process.

FIFRA - Federal Insecticide, Fungicide, and Rodenticide Act.

FLAME ATOMIC ABSORPTION (AA) - atomic absorption which utilizes flame for excitation.

GRAPHITE FURNACE ATOMIC ABSORPTION (GFAA) - atomic absorption which utilizes a graphite cell for excitation.

HOMOGENEITY - the degree to which a property or substance is randomly distributed throughout a material. Homogeneity depends on the size of the units under consideration. Thus, a mixture of two minerals may be inhomogeneous at the molecular or atomic level, but homogeneous at the particulate level.

INDEPENDENT STANDARD - a Contractor-prepared standard solution that is composed of analytes from a different source than those used in the standards for the initial calibration.

INDUCTIVELY COUPLED PLASMA (ICP) - a technique for the simultaneous or sequential multi-element determination of elements in solution. The basis of the method is the measurement of atomic emission by an optical spectroscopic technique. Characteristic atomic line emission spectra are produced by excitation of the sample in a radio frequency inductively coupled plasma.

IN-HOUSE - at an EMS Heritage laboratory facility.

INITIAL CALIBRATION - analysis of analytical standards for a series of different specified concentrations; used to define the linearity and dynamic range of the response of the instrument to the target compounds.

INJECTION - introduction of the analytical sample into the instrument excitation system for the purpose of measuring absorbance, emission or concentration of an analyte. May also be referred to as <u>exposure</u>.

INSTRUMENT CALIBRATION - analysis of analytical standards for a series of different specified concentrations; used to define the quantitative response, linearity, and dynamic range of the instrument to target analytes.

INTERFERENTS - substances which affect the analysis for the element of interest.

INTERNAL STANDARDS - in-house compounds added at a known concentration.

LABORATORY - synonymous with Contractor as used herein.

LABORATORY CONTROL SAMPLE (LCS) - a control sample of known composition. Aqueous and solid laboratory control samples are analyzed using the same sample preparation, reagents, and analytical methods employed for the samples received.

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LABORATORY RECEIPT DATE - the date on which a sample is received at the inventory's facility, as recorded on the shipper's delivery receipt and sample Traffic Report. Also referred to as VTSR (validated time of sample receipt).

LABORATORY SAMPLE - a sample, intended for testing or analysis, prepared from a gross sample or otherwise obtained. The laboratory sample must retain the composition of the gross sample. Often reduction in particle size is necessary in the course of reducing the quantity.

LINEAR RANGE, LINEAR DYNAMIC RANGE - the concentration range over which the analytical curve remains linear.

LOT - a quantity of bulk material of similar composition whose properties are under study.

MATRIX - the predominant material of which the sample to be analyzed is composed. Matrix is <u>not</u> synonymous with phase (liquid or solid).

MATRIX DUPLICATE - an intralaboratory split sample which is used to document the precision of a method in a given sample matrix.

MATRIX MODIFIER - salts used in AA to lessen the effects of chemical interferents, viscosity, and surface tension.

MATRIX SPIKE - an aliquot of sample spiked with a known concentration of target analyte(s). The spiking occurs prior to sample preparation and analysis. A matrix spike is used to document the bias of a method in a given sample matrix.

MATRIX SPIKE DUPLICATES - Intralaboratory split samples spiked with identical concentrations of target analyte(s). The spiking occurs prior to sample preparation and analysis. They are used to document the precision and bias of a method in a given sample matrix.

MCL - Maximum Contaminant Level (regulatory level).

METHOD BLANK (previously termed reagent blank) - an analytical control consisting of all reagents, internal standards and surrogate standards, that is carried through the entire analytical procedure. The method blank is used to define the level of laboratory background and reagent contamination.

METHOD DETECTION LIMIT (MDL) - The minimum concentration of a substance that can be measured and reported with 99% confidence that the analyte concentration is greater than zero and is determined from analysis of a sample in a given matrix type containing the analyte. See Detection Limits.

For operational purposes, when it is necessary to determine the MDL in the matrix, the MDL shall be determined by multiplying the appropriate one-sided 99% t-statistic by the standard deviation obtained from a minimum of three analyses of a matrix spike containing the analyte of interest at a concentration three to five times the estimated MDL, where the t-statistic is obtained from standard references or the table below.

No. of samples:	t-statistic
3	6.96
4	4.54
5	3.75
6	3.36
7	3.14
8	3.00
9	2.90
10	2.82

METHOD OF STANDARD ADDITIONS (MSA) - the addition of 3 increments of a standard solution (spikes) to sample aliquots of the same size. measurements are made on the original and after each addition. The slope, x-intercept and y-intercept are determined by least-square analysis. The analyte concentration is determined by the absolute value of the x-intercept. Ideally, the spike volume is low relative to the sample volume (approximately 10% of the volume). Standard addition may counteract matrix effects; it will not counteract spectral effects. Also referred to as Full Method of Standard Additions (3-Point MSA).

METHOD OF STANDARD ADDITIONS, SINGLE ADDITIONS - the simplest version of MSA is the single-addition method, in which two identical aliquots of the sample solution, each of Volume V_x , are taken. To the first (labeled A) is added a small volume V_s of a standard analyte solution of concentration c_s . To the second (labeled B) is added the same volume V_s of the solvent. The analytical signals of A and B are measured and corrected for nonanalyte signals. The unknown sample concentration c_x may be calculated as follows:

$$c_x = \frac{S_B V_s c_s}{(S_A - S_B) V_x}$$

where S_A and S_B are the analytical signals (corrected for the blank) of solutions A and B, respectively. V_s and c_s should be chosen so that S_A is roughly twice S_B on the average. It is best if V_s is made much less than V_s , and thus c_s is much greater than c_s , to avoid excess dilution of the sample matrix. If a separation or concentration step is used, the additions are best made first and carried through the entire procedure.

m/z - Mass to charge ratio, synonymous with "m/e", used in mass spectrometry

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NARRATIVE (SDG Case Narrative) - a portion of a data package which includes laboratory, contract, Case and sample number identification, and descriptive documentation of any problems encountered in processing the samples, along with corrective action taken and problem resolution.

NPDES - National Pollutant Discharge Elimination System (regulated under the CWA).

ORGANIC-FREE REAGENT WATER - For volatiles, all references to water in the methods refer to water in which an interferant is not observed at the method detection limit of the compounds of inters. Organic-free reagent water can be generated by passing tap water through a carbon filter bed containing about 1 pound of activated carbon. A water purification system may be used to generate organic-free deionized water. Organic-free reagent water may also be prepared by boiling water for 15 minutes and, subsequently, while maintaining the temperature at 90°C, bubbling a contaminant-free inert gas through the water for 1 hour.

For semivolatiles and nonvolatiles, all references to water in the methods refer to water in which an interferant is not observed at the method detection limit of the compounds of interest. Organic-free reagent water can be generated by passing tap water through a carbon filter bed containing about 1 pound of activated carbon. A water purification system may be used to generate organic-free deionized water.

PARAMETER GROUP - Is defined as a group of samples that have been preserved in the same manner, prepared by similar protocols and analyzed using instruments of similar technology (also known as <u>analyte group</u>). Examples of parameter groups are:

Volatiles (EPA methods 601, 602, and 624)
Pesticides (EPA methods 608, 614, 622)
Trace Metals (All metals except mercury)
Nutrients (Total Kjeldahl Nitrogen, Nitrate ± Nitrite, Total Phosphorous)

PERCENT SOLIDS - the proportion of solid in a sample determined by drying an aliquot of the sample.

PERCENT DIFFERENCE (%D) - As used in this QAP and elsewhere to compare two values, the percent difference indicates both the direction and the magnitude of the comparison, i.e., the percent difference may be either negative, positive, or zero. (In contrast, see relative percent difference below).

PERCENT MOISTURE - an approximation of the amount of water in a soil/sediment sample made by drying an aliquot of the sample at 105°C. The percent moisture determined in this manner also includes contributions from all compounds that may

volatilize at or below 105°C, including water. Percent moisture may be determined from decanted samples and from samples that are not decanted.

PERFORMANCE EVALUATION SAMPLES - A sample submitted for analysis whose composition and concentration are known to the submitter but unknown to the analyst. Also known as a <u>Blind Sample</u>.

PRECISION - the agreement among a set of replicate measurements without assumption of knowledge of the true value. Precision is estimated by means of duplicate/replicate analyses. These samples should contain concentrations of analyte above the MDL, and may involve the use of matrix spikes. The most commonly used estimates of precision are the relative standard deviation (RSD) or the coefficient of variation (CV),

$$RSD = CV = 100 S/\bar{x},$$

where \overline{x} = the arithmetic mean of the x_j measurements, and s = variance; and the relative percent difference (RPD) when only two samples are available.

$$RPD = 100 [(x_1 - x_2)/(s_1 + x_2)/2].$$

PREPARATION BLANK (regent blank, method blank) - an analytical control that contains distilled, deionized water and reagents, which is carried through the entire analytical procedure (digested and analyzed). An aqueous method blank is treated with the same reagents as a sample with a water matrix; a solid method blank is treated with the same reagents as a soil sample.

PROJECT - single or multiple data collection activities that are related through the same planning sequence.

PROTOCOL - a compilation of the procedures to be followed with respect to sample receipt and handling, analytical methods, data reporting and deliverables, and document control. Used synonymously with Statement of Work (SOW).

PURGE AND TRAP (device) - analytical technique (device) used to isolate volatile (purgeable) organics by stripping the compounds from water or soil by a stream of inert gas, trapping the compounds on an adsorbent such as a porous polymer trap, and thermally desorbing the trapped compounds onto the gas chromatographic column.

QUALITY ASSURANCE - a system of activities whose purpose is to provide the producer or user of environmental data the assurance that it meets defined standards of quality with a stated level of confidence.

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QUALITY ASSURANCE PLANS (QAP) - an orderly assembly of detailed and specific procedures which delineates how data of a known and accepted quality is produced.

- a. <u>Comprehensive Quality Assurance Plan (CompOAP)</u> a QA plan that outlines all the capabilities of the specified organization, the routinely used quality control measures, the routine QA targets for precision and accuracy, and all documentation, calibration and maintenance activities that are necessary to produce data of a known and acceptable quality.
- b. <u>Quality Assurance Project Plans (QAPP)</u> a QA plan that is written for a specific project outlining specific QA targets and data quality objectives as well as all protocols and QC measures needed to meet the project specific objectives.
- c. Research Quality Assurance Plans (ROAP) a special type of Quality Assurance Project Plan that is generally written as a requirement of a direct contract with the Florida DER for research activities. The specific activities are defined in section 6.0 of the guidelines. The content and format requirements are different from those of any other QAP.

QUALITY CONTROL - the overall system of activities whose purpose is to document and control the quality of environmental data so that it meets the needs of the users.

- A. Quality Control Checks standards of samples from an independent source that are analyzed at a specified frequency.
 - 1. Quality Control Check Standards standard solutions from a source other than normal calibration standards that are certified and traceable. These standards are used to check the accuracy of a calibration curve.
 - 2. Quality Control Check Sample (also known as Reference Materials) samples obtained from an independent source for which the level(s) of analytes have been validated. These samples are prepared and analyzed with a sample set of similar matrix. If these samples have been obtained from the National Institute of Standards and Testing (formerly National Bureau of Standards), these are referred to as Standard Reference Materials.
- B. <u>OCTS Quality Control Tracking System</u> the computerized system at EMS Heritage Laboratories, Inc. utilized to contain, compile and report quality control data.
 - 1. Quality Control Types
 - a. BLA01 Reagent blank, calibration blank.

An aliquot of de-ionized (DI) water containing the same reagents as the sample but is NOT taken through preparation. This sample is used as the calibration zero concentration for initial calibration purposes. The BLA01 is analyzed as a sample as frequently as required in the QAP. The BLA01 can be used to re-zero only after it has been analyzed as a sample. Instrument response is entered on bench sheets. If the instrument response is above the Control Limit when the BLA01 is run as a sample and calculated as a concentration, the samples back to the last acceptable BLA01 must be re-analyzed. If BLA01 is 0.5-0.99 of IDL, the concentration may be entered into the data base. Otherwise, enter less than 0.4X IDL. Entries into the data base MUST BE CONSISTENT, e.g., either instrument response OR concentration -NOT both, and must have units reported.

b. BLA02 - Method Blank or Preparation Blank.

Same as above but is carried through the complete steps of analysis from digestion/extraction/etc. in the exact same manner (e.g., same glassware, reagents, storage bottle) as the sample. BLA02 is matrix specific and must be run with each matrix and each prep run. A prep run consists of only 1 analyst on 1 day, utilizing the same reagents and glassware. Subtraction of the method blank is addressed by each method. A general rule applies to the BLA02: If the BLA02 is equal to or exceeds the method detection limit (MDL), samples in that group must be re-prepped. Therefore a BLA02 must be run with each sample set. Instrument response is recorded on bench sheets, but concentration is entered into the QC data base in all cases. If BDL, entered <0.4X IDL into the QC data base.

c. CAL01 - Calibration Standard.

Calibration standards (the number and frequency of which are specified in each method) are used to establish an analytical curve for that analyte based on absorbance, emission intensity, area or other type of measurable response for known standards. CAL01, CAL02, CAL03, etc., are prepared using exactly the same reagents used in the analysis of the sample. Note that some methods require a CDL standard be included as a CAL standard.

d. CCV - Continuing Calibration Verifications.

Analytical standard that is run with a frequency specified in the QAP - at a minimum frequency of 10% <u>BUT</u> may be alternated with ICV01

to meet the frequency requirements. The CCV may be from the same source as calibration standards or a different source depending on the method used. All runs must culminate with this sample's analysis except for GC/MS.

e. CDL01 - Contract Detection Limit Standard.

A reagent sample to verify analytes are quantifiable at the detection limit stated. The amount of analyte in this sample may be specified by a method, project, or client. In general, twice the analyte concentration of the regulatory detection line or CRDL is a good CDL01 amount. (Required weekly for drinking water organic analyses). Some methods mandate use of this standard for inclusion into the calibration curve.

f. DLCS - Duplicate Laboratory Control Sample (ICV02 or EPA supplied LCS).

Duplicate control sample of known analyte concentration and source analyzed by exactly the same method as the samples. DLCS must be of the same matrix as the samples but must be from a different source than the calibration standards (EPA or NIST traceable when possible). Results are expressed as % recovery and RPD. Also known as a "Laboratory Fortified Blank" when analytes are spiked into reagent water.

g. DLCS1 - Duplicate Laboratory Control Sample (ICV01 or EPA supplied LCS).

Duplicate control sample of known analyte concentration and source analyzed by exactly the same method as the samples. DLCS must be of the same matrix as the samples but must be from a different source than the calibration standards (EPA or NIST traceable when possible). Results are expressed as % recovery and RPD. Also known as a "Laboratory Fortified Blank" when analytes are spiked into reagent water. This "1" designation is a programming device used to indicate that no separable prep exists for the method.

h. DPS01 - Reagent Duplicate, Matrix Spike.

An aliquot of sample (water, oil, S/S/S) spiked with a known quantity of the analyte of interest - but added after preparation or if no preparation is involved in analysis of sample. The sample is split and

spiked with exactly the same amount of analyte. Results are expressed as Relative Percent Difference (RPD) or as required.

i. DPS02 - Duplicate Spike (Prepped).

Same as DPS01 but the sample is split in as representative a way as possible, spiked with equal amounts of the analyte, and carried through the preparation step(s). Results are expressed as RPD or as required.

j. DUP01 - Duplicate Sample Analysis (Non-prepped).

For samples not requiring digestion/extraction/etc., a homogeneous, representative aliquot (water, oil, S/S/S) is split and carried through the analytical steps to quantitation. Results are expressed as RPD.

k. DUP02 - Duplicate Sample Analysis (Prepped).

Same as DUP01 but split before any required preparation and carried through to quantitation exactly as its counterpart. Results are expressed as RPD.

l. ICV01 - Initial Calibration Verification.

This standard verifies the calibration curve, and this analyte must be from a different source as the calibration standards (EPA or NIST traceable when possible).

m. ICV02 - Initial Calibration Verification.

Same as ICV01 but added to the sample before any required preparation. May be equivalent to an LCS when reagent water is utilized as the spiking medium.

n. LCS - Laboratory Control Sample (ICV02 or EPA supplied LCS).

Control sample of known analyte concentration and source analyzed by exactly the same method as the samples. LCS should be of same matrix as samples (must utilize the same procedures) but must be from a different source that the calibration standards (EPA or NIST Appendix D Revision 0 Nov. 1, 1991 Page 18 of 27

traceable when possible). The LCS may be called a "QC Check Sample" or is also known as "Laboratory Fortified Blank" when analytes are spiked into reagent water.

o. LCS01 - Laboratory Control Sample (ICV01 or EPA supplied LCS).

Control sample of known analyte concentration and source analyzed by exactly the same method as the samples. LCS should be of the same matrix as samples (must utilize the same procedures) but must be from a different source than the calibration standards (EPA or NIST traceable when possible). Also known as a "Laboratory Fortified Blank" when analytes are spiked into reagent water. This "01" designation is a programming device used to indicate that no separable prep exists for the method.

p. SPI01 - Matrix Spike (Standard Addition).

A post digestion/extraction spike, or a method with no separable prep. An aliquot of homogeneous sample (water, oil, S/S/S) fortified (spiked) with a known quantity of specific compound(s) and carried through the analysis and quantitation steps. At least one spike per matrix and concentration must be analyzed per run or frequency specified by QAP or SOW.

q. SPI02 - Matrix Spike (Pre-digestion/extraction; prepped).

Same as a SPI01 but is used when preparations are required. Calculate as a percent recovery, unless a method or client specifies differently.

r. SUR01 - Surrogate Spike (Organic analyses only).

Surrogate standards are added to every blank, sample, LCS, MS, MSD, and standard to evaluate analytical efficiency by measuring percent recovery (unless specified to report recoveries differently). A representative sample is taken, surrogates added, analyzed, and quantitated. A SUR01 would not require preparation, or no separable prep exists for the method.

s. SUR02 - Surrogate Spike (Organic analyses only).

Same as SUR01 but surrogates are added before any preparation. Surrogates are unique compounds not normally detected in environmental samples.

RCRA - the Resource Conservation and Recovery Act.

REAGENT BLANK - see Method Blank.

REAGENT GRADE - analytical reagent (AR) grade, ACS reagent grade, and reagent grade are synonymous terms for reagents which conform to the current specifications of the Committee on Analytical Reagents of the American Chemical Society.

REAGENT WATER - water that has been generated by any method which would achieve the performance specifications for ASTM Type II water, and in which an interferant is not observed at or above the minimum quantitation limit of the parameters of interest.

RECONSTRUCTED ION CHROMATOGRAM (RIC) - a mass spectral graphical representation of the separation achieved by a gas chromatograph; a plot of total ion current versus retention time.

RELATIVE PERCENT DIFFERENCE (RPD) - as used in this QAP and elsewhere to compare two values, the relative percent difference is based on the mean of the two values, and is reported as an absolute value, i.e., always expressed as a positive number or zero. (In contrast, see percent difference above).

RELATIVE RESPONSE FACTOR (RRF) - a measure of the relative mass spectral response of an analyte compared to its internal standard. Relative Response Factors are determined by analysis of standards and are used in the calculation of concentrations of analytes in samples. RRF is determined by the following equation:

Where
$$RRF = \frac{A_x}{A_{is}} \times \frac{C_{is}}{C_x}$$

A = area of the characteristic ion measured

C = concentration

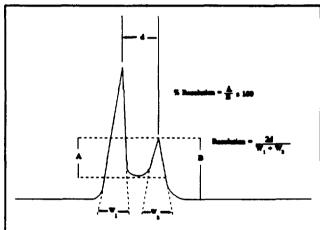
is = internal standard

x = analyte of interest

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REPLICATE SAMPLE - samples that have been collected at the same time from the same source (field replicates) or aliquots of the same sample that are prepared and analyzed at the same time (laboratory replicates). Duplicate samples are one type of replicate sample. The analytical results from replicates are used to determine the precision of a system. If the concentration of analytes in the sample are below detectable limits, Duplicate Spike Samples may be used to determine precision. Blind Replicates (Duplicates) are replicates that have been collected (field replicate) or prepared (laboratory replicate) and are submitted and analyzed as separate samples (analyst does not know they are replicates).

RESOLUTION - percent resolution is the separation between peaks on a chromatogram, calculated by dividing the depth of the valley between the peaks by the peak height of the smaller peak being resolved, multiplied by 100. Resolution is determine taking 2 times the difference in elution times between two peaks and dividing that by the sum of the baseline peak widths of the two peaks.



RUN - a continuous analytical sequence consisting of prepared samples and all associated quality assurance measurements as required by the QAP.

SAMPLE - a portion of material to be analyzed that is contained in single or multiple containers and identified by a unique sample number.

- a. <u>Sample Custody</u> all records and documentation that is required to trace a sample from point of origin through disposal after analysis. These records must include, but are not limited to:
 - 1. Field notebooks;
 - 2. Field sample ID tags;
 - 3. Laboratory transmittal forms (if applicable);

- 4. Laboratory sample receipt logs;
- 5. Sample extraction/preparation logs or worksheets;
- 6. Analytical (instrument) logs or worksheets;
- 7. Calibration and quality control data associated with a sample set;
- 8. Instrument maintenance logs;
- 9. Sample disposition logs; and
- 10. Final reports.
- b. <u>Legal Chain of Custody</u> is a special type of sample custody in which <u>all</u> events associated with a specific sample must be documented in writing. In addition to the records described above, chain of custody records must include the follows:
 - 1. Sample transmittal forms or tags that have adequate spaces for the dated, original signatures of all individuals who handle the sample (or cleaned the sample containers if obtained from a contracted laboratory) from time of collection (or container receipt) through laboratory delivery.
 - 2. Laboratory sample storage logs that identify date, time, and individuals who remove samples from storage.
 - 3. Secure, limited access storage areas.

SAMPLE DELIVERY GROUP (SDG) - a unit within a sample Case that is used to identify a group of samples for delivery. An SDG is a group of 20 or fewer samples within a Case, received over a period of up to 14 calendar days. Data from all samples in an SDG are due concurrently. A Sample Delivery Group is defined by one of the following, whichever occurs first:

- a. Case; or
- b. Each 20 samples within a Case; or
- c. Each 14-day calendar period during which samples in a Case are received, beginning with receipt of the first sample in the Case or SDG.

Samples may be assigned to Sample Delivery Groups by matrix (i.e., all soils in one SDG, all waters in another), at the discretion of the laboratory.

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SAMPLE MATRIX - means that characteristic of an environmental or laboratory sample, associated with its physical and chemical properties, which defines how such a sample is handled when subjected to the intended analytical process. The following samples matrices (major matrix groups), as defined below, should be used in QA plans whenever specifying data quality objectives:

- a. Reagent Water: A sample of water which conforms to ASTM grades II, III or IV.
- b. <u>Drinking Water</u>: Includes finished (treated) or raw source water designated as potable water. Such sources may be from surface or groundwater.
- c. <u>Surface Water</u>: Includes fresh or saline waters from streams, canals, rivers, lakes, ponds, bays and estuaries (natural or manmade).
- d. <u>Groundwater</u>: Includes all waters found below ground in confined or unconfined aquifers.
- e. <u>Wastewater</u>: Includes any influent or effluent associated with domestic or industrial waste treatment facilities.
- f. <u>Chemical Waste</u>: Includes sludges and residuals from domestic or industrial wastewater processing, and liquid or solid chemicals that are no longer used for its intended purpose.
- g. <u>Soil/Sediment</u>: Surface or subsurface soils and sediments of fresh or salt water origin.
- h. <u>Biological Tissue</u>: Includes tissues of plant or animal origin. The most common of these are shellfish, finfish and aquatic plants.
- i. Oils: Includes any oily matrix.
- j. <u>Solvents</u>: Includes waste solvents.

SAMPLE NUMBER (Client Sample Number) - a unique identification number designated by the client for each sample. The Sample Number appears on the sample Chain of Custody which documents information on that sample.

SDWA - Safe Drinking Water Act.

SEMIVOLATILE COMPOUNDS - compounds amenable to analysis by extraction of the sample with an organic solvent. Used synonymously with Base/Neutral/Acid (BNA) compounds.

SERIAL DILUTION - the dilution of a sample by a factor of five. When corrected by the dilution factor, the diluted sample must agree with the original undiluted sample within specified limits. Serial dilution may reflect the influence of interferents.

SOIL - synonymous with soil/sediment or sediment as used herein.

SPLIT SAMPLES - aliquots of sample taken from the same container and analyzed independently. In cases where aliquots of samples are impossible to obtain, field duplicate samples must be taken for the matrix duplicate analysis. These are usually taken after mixing or compositing and are used to document intra-or interlaboratory precision, recognizing their limitations.

SPIKED SAMPLES - samples fortified to a known and validated concentration of analyte. Percent recoveries are calculated for each compound in the spike.

a. <u>Field</u>: An environmental sample fortified to a known and validated concentration in the field. These may be submitted as <u>blind spike</u> (laboratory does not know they are spiked) or as identified <u>field spikes</u>. The use of this QC check <u>is not recommended</u>.

b. <u>Laboratory</u>:

- 1. <u>Reagent Spikes</u>: Samples of an appropriate analyte-free matrix (deionized water, sand, soil, etc.) that are fortified to a known and validated concentration of analyte(s) <u>before</u> sample preparation (See LCS).
- 2. <u>Sample (Matrix Spikes)</u>: Environmental sample selected from a set (not blanks) that are fortified to a known and validated concentration of analyte(s) <u>before</u> sample preparation. The concentration of each analyte in the spiking solution should be approximately 3-5 times the level expected in the sample (see SPI02, DPS02).
- 3. <u>Surrogate Spikes</u>: A compound having similar chemical characteristics to the compounds of interest, but which is not normally found in environmental samples. Known concentrations of these compounds are added to all samples in the set before sample preparation (see SUR01, SUR02).

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STANDARD ADDITION - the practice of adding a known amount of an analyte to a sample immediately prior to analysis. It is typically used to evaluate interferences. See Method of Standard Additions.

STANDARD ANALYSIS - an analytical determination made with known quantities of target compounds; used to determine response factors.

STANDARD CURVE - a plot of concentrations of known analyte standards versus the instrument response to the analyte. Calibration standards are prepared by successively diluting a standard solution to produce working standards which cover the working range of the instrument. Standards should be prepared at the frequency specified in the appropriate section. The calibration standards must be prepared using the same type of acid or solvent and at the same concentration as will result in the samples following sample preparation. This is applicable to organic and inorganic chemical analyses.

STOCK SOLUTION - a standard solution which can be diluted to derive other standards.

SUBSAMPLE - a portion taken from a sample. A laboratory sample may be a subsample of a gross sample; similarly, a test portion may be a subsample of a laboratory SVOA sample.

SURROGATES (Surrogate Standard) - for semivolatiles and pesticides/Aroclors, compounds added to every blank, sample, matrix spike, matrix spike duplicate, and standard; used to evaluate analytical efficiency by measuring recovery. Surrogates are brominated, fluorinated, or isotopically labelled compounds not expected to be detected in environmental media (see SUR01, SUR02).

SYSTEM MONITORING COMPOUNDS - compounds added to every blank, sample, matrix spike, matrix spike duplicate, and standard for volatile analysis, and used to evaluate the performance of the entire purge and trap-gas chromatograph-mass spectrometer system. These compounds are brominated or deuterated compounds not expected to be detected in environmental media. Could also be known as surrogates.

TARGET COMPOUND LIST (TCL) - a list of compounds designated at the primary targets for analysis.

TENTATIVELY IDENTIFIED COMPOUNDS (TIC) - compounds detected in samples that are not target compounds, internal standards, system monitoring compounds, or surrogates. Up to 30 peaks (those greater than 10% of peak areas or heights of nearest internal standards) are subjected to mass spectral library searches for tentative identification.

TEST PORTION (also called specimen, test specimen, test unit, aliquot) - that quantity of a material of proper size for measurement of the property of interest. Test portions may be taken from the gross sample directly, but often preliminary operations, such as mixing or further reduction in particle size, are necessary.

TIME - when required to record time on any deliverable item, time shall be expressed as Military Time, i.e., a 24-hour clock.

TRAFFIC REPORT (TR) - an EPA sample identification form filled out by the sampler, which accompanies the sample during shipment to the laboratory and which is used for documenting sample condition and receipt by the laboratory.

TRIP BLANK - a sample of media taken from the laboratory to the sampling site and returned to the laboratory unopened. The media used for the trip blank is acceptable if the concentration of any analyte of concern in the media is no higher than the highest of either:

- a. The detection limit, or
- b. Five percent of the regulatory limit for that analyte, or
- c. Five percent of the measured concentration in the sample.

A trip blank is used to document contamination attributable to shipping and field handling procedures. This type of blank is useful in documenting contamination of volatile organics samples.

TWELVE-HOUR TIME PERIOD - the twelve (12) hour time period (when applicable) for GC/MS system instrument performance check, standards calibration (initial or continuing calibration), and method blank analysis begins at the moment of injection of the DFTPP or BFB analysis that the laboratory submits as documentation of instrument performance. The time period ends after 12 hours have elapsed according to the system clock. For pesticide/Aroclor analyses performed by GC/EC, the twelve hour time period in the analytical sequence begins at the moment of injection of the instrument blank that precedes sample analyses, and ends after twelve hours have elapsed according to the system clock.

VALIDATED TIME OF SAMPLE RECEIPT (VTSR) - the date on which a sample is received at the Contractor's facility, as recorded on the shipper's delivery receipt and Sample Traffic Report.

VOLATILE COMPOUNDS - compounds amenable to analysis by the purge and trap technique. Used synonymously with purgeable compounds.

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WET WEIGHT - the weight of a sample aliquot including moisture (undried).

WIDE BORE CAPILLARY COLUMN - a gas chromatographic column with an internal diameter (ID) that is greater than 0.32mm. Columns with lesser diameters are classified as narrow bore capillaries.

COMMON LABORATORY ACRONYMS AND ABBREVIATIONS

AA	Atomic Absorption
BDL	Below Detection Limit
BED	Below Estimated Limit of Detection
BTU	British Thermal Unit
CCC	Calibration Check Compound
CVAA	Cold Vapor Atomic Absorption
DL	Detection Limit
ECD	Electron Capture Detector (GC)
ELCD	Electrolytic Conductivity Detector
EP	Extraction Procedure
FAA	Flame Atomic Absorption
FID	Flame Ionization Detector (GC)
FPD	Flame Photometric Detector
FTIR	Fourier Transform Infra-Red
GC	Gas Chromatography
GC/MS	Gas Chromatography/Mass Spectrometry
GFAA	Graphite Furnace Atomic Absorption
HOC	Halogenated Organic Compounds (Appendix III of 40 CFR Part 268)
ICP	Inductively Couple Plasma Atomic Emission Spectroscopy
IDL	Instrument Detection Limit
LIMS	Laboratory Information Management System
MDL	Method Detection Limit
mg/kg	Milligrams per Kilogram (ppm)
mg/L	Milligrams per Liter (ppm)
ND	Not Detected
NPD	Nitrogen Phosphorus Detector
OVA	Organic Vapor Analyzer
PCB	Polychlorinated Biphenyls
PID	Photoionization Detector
ppb	Parts per Billion
ppm	Parts per Million
PQL	Practical Quantitation Limits
QA/QC	Quality Assurance/Quality Control
SPCC	System Performance Check Compound
S/S/S	Soil/Sediment/Sludge

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SVOC	Semivolatile Organic Compound	
SVOA	Semivolatile Organic Analysis	
SW-846	"Test Methods for Evaluating Solid Waste: Physical/Chemical Methods	Ħ
TCLP	Toxicity Characteristic Leaching Procedure	
TOC	Total Organic Carbon	
TOX	Total Organic Halogen	
TSS	Total Suspended Solids	
TTO	Total Toxic Organics	
TX	Total Halogen	
ug/kg	Micrograms per Kilogram (ppb)	
ug/L	Micrograms per Liter (ppb)	
VOA	Volatile Organic Analysis	
VOC	Volatile Organic Compound	



Job Description

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PRESIDENT

QUALIFICATIONS

Position Education

Experience

President

BS/BA Science

10 years

Other experience, advanced education and/or proven aptitude and performance may count for education or experience. Experience may include laboratory, regulatory or other related environmental or management experience.

DUTIES/RESPONSIBILITIES

Designate a Laboratory Director at each facility and replace as necessary.

Designate a Quality Assurance Officer and replace as necessary.

Assure that the VPO, QAO and Lab Directors clearly understand their functions and responsibilities.

Establish and revise company policies and operating procedures as necessary.

Approve the Quality Assurance Program (QAP) and the Chemical Hygiene Plan (CHP).

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Job Description

VICE PRESIDENT OPERATIONS

Position

Education

Experience

VP Operations

BS/BA Science

7 years

DUTIES/RESPONSIBILITIES

Coordinate resources among locations to assure that personnel, facilities, equipment, and materials are known and available as required to produce data on time according to methods, SOP's, the QAP and client data quality objectives.

Coordinate resources among locations to assure that methods, SOP's, the QAP and the CHP are approved and followed. Assure that corrective action is taken in response to deviations when necessary.

Coordinate resources among locations to assure that appropriate and adequate training is provided and that training records are maintained.

Coordinate resources among locations to assure that laboratory waste is properly disposed.

Assure that all Laboratory Directors know, understand, and carry out their responsibilities. Communicate any deviations from policy to the President and assure any required corrective actions are taken.

Job Description

LAB DIRECTOR/ASSISTANT LAB DIRECTOR

QUALIFICATIONS

<u>Position</u>	Education	Experience	
Lab Director	BS/BA Science	7 years	
Asst. Lab Director	BS/BA Science	5 years	

Other experience, advanced education and/or proven aptitude and performance may count for education or experience. Experience may include laboratory, regulatory or other related environmental or management experience.

DUTIES/RESPONSIBILITIES

Assure that personnel, resources, facilities, equipment, and materials are known and available as required to produce data on time according to methods, SOP's, the QAP and client data quality objectives.

Assure that methods, SOP's, the QAP and the CHP are approved and followed. Assure that corrective action is taken in response to deviations when necessary.

Assure that analytical methods are as specified in the work plan. Maintain working knowledge of local, state and federal regulations as they relate to services performed by the lab.

Assure that appropriate and adequate training is provided and that training records are maintained.

Assure that raw data is appropriately archived.

Assure that laboratory waste is properly disposed.

Nominate a Quality Assurance Unit Director for approval by the QAO and the President.

Assure that the Quality Assurance Unit clearly understands their duties.

Assure that all laboratory personnel know, understand, and carry out their responsibilities.

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Job Description

QUALITY ASSURANCE OFFICER

QUALIFICATIONS

Position

Education

Experience

QA Officer

BS/BA Science

7 years

Other experience, advanced eduction and/or proven aptitude and performance may count for eduction or experience. Experience may include laboratory, regulatory or other related environmental or management experience.

DUTIES/RESPONSIBILITIES

Assure the inspection of systems and operations for conformance to methods, SOP's, the QAP, regulations and client data quality objectives at all locations. Document and report the results of the inspections and any deviations from methods, SOP's, and the QAP quarterly to the President along with corrective actions taken.

Perform follow-up inspections, etc. to assure that corrective actions have been taken. Coordinate development and revision of the Quality Assurance Plan (QAP) as required.

Coordinate and document certifications by local, state, and federal regulatory authorities at all locations.

Provide technical support and training to Quality Assurance Units and staff on methods, SOP's, regulations and the QAP. Maintain working knowledge of local, state and federal regulations and quality technology as they relate to services performed by the lab and quality control.

Coordinate and supervise all Quality Assurance Units. Assure that the Quality Assurance Units know and understand their responsibilities at each location.

Assure that all QAP, method and SOP requirements are met prior to initiation of a new procedure.

Final authority to terminate or alter any incorrect or improper analytical or measurement procedure in order to conform to requirements of the QA Plan.

Job Description

QUALITY ASSURANCE OFFICER (continued)

Assure coordination and documentation of internal and external audits and performance evaluation programs. Assure the coordination of responses to external QA audits or performance evaluations. Assure the development of responses to other QA questions.

Advise purchasing on the quality of materials, supplies and equipment.

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Job Description

OUALITY ASSURANCE UNIT

QUALIFICATIONS

Position

Education

Experience

OA Unit

BS/BA Science

4 years

Other experience, advanced education and/or proven aptitude and performance may count for education or experience. Experience may include laboratory, regulatory or other related environmental or management experience.

DUTIES/RESPONSIBILITIES

Inspect systems and operations adequately and perform performance evaluation studies to assure conformance to methods, SOP's, the QAP, regulations and client data quality objectives. Audit data entry and other data management systems to assure accuracy and completeness.

Document and report to the Quality Assurance Officer and Lab Director results of inspections and any deviations from methods, SOP's, and the QAP along with suggested corrective action.

Review and accept or reject client specific QAP's. Maintain documentation trail of all methods, SOP's and QAP's.

Provide technical support and training to staff on methods, SOP's and the QAP. Maintain working knowledge of local, state and federal regulations and quality technology as they relate to services performed by the lab and the related quality assurance issues.

Document that all QAP, method and SOP requirements are met prior to initiation of a new procedure. Maintain records of detection limit studies.

Review the Certificate of Analysis and other data deliverables for conformance with methods, SOP's, client data quality objectives and the QAP. Respond to QA questions on Certificate of Analysis and other deliverables.

Coordinate and document internal and external audits and performance evaluation programs. Host and respond to QA audits or external performance evaluation studies.

Advise purchasing on the quality of materials, supplies and equipment.

Job Description

QUALITY ASSURANCE UNIT (continued)

Function as QA Unit for GLP projects.

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Job Description

CHIEF CHEMIST/SENIOR SCIENTIST

QUALIFICATIONS

<u>Position</u>	<u>Education</u>	Experience
Chief Chemist	BS/BA Science	5 years
Senior Chemist	BS/BA Science	3 years

Other experience, advanced education and/or proven aptitude and performance may count for education or experience. Experience may include laboratory, regulatory or other related environmental or management experience.

DUTIES/RESPONSIBILITIES

Provide technical support and training on instruments, methodologies, regulations and the QAP to staff. Assist in troubleshooting analytical problems.

Plan and assist in the implementation of new methods, instruments and equipment to meet changing regulatory or production requirements. Assure compliance with methods, SOP's, the QAP and CHP prior to startup.

Perform and manage special or non-routine projects requiring new or unique methods, SOP's or data quality objectives.

Maintain working knowledge regulations and new developments in analytical and environmental chemistry as they relate to services performed by the lab.

Write and revise SOP's and methods as necessary.

Job Description

PROJECT MANAGER/PROJECT COORDINATOR

OUALIFICATIONS

<u>Position</u>	Education	<u>Experience</u>	
Project Manager	BS/BA Science	2 years	
Project Coordinator	HS	3 years	

Other experience, advanced education and/or proven aptitude and performance may count for education or experience. Experience may include laboratory, regulatory or other related environmental or management experience.

DUTIES/RESPONSIBILITIES

Provide primary interface between the customer and the lab. Assure that the customers regulatory and data quality objectives, schedule and workload volume are known to the lab. Maintain contact and document correspondence with customers through all phases of project.

Prepare quotations, proposals and work plans to match the clients regulatory and data quality objectives. Assure that the lab's methods, pricing and capacity are known to the client. Obtain all appropriate business related information.

Maintain working knowledge of local, state and federal regulations as they relate to services performed by the lab.

Review sample submission documentation and appropriate LIMS reports to insure correctness prior to initiation of work. Review data deliverables for completeness and conformance to customer requirements.

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Job Description

GROUP LEADER/ASSISTANT GROUP LEADER

OUALIFICATIONS

<u>Position</u>	Education	<u>Experience</u>
Group Leader	BS/BA Science	3 years
Asst. Group Leader	BS/BA Science	2 years

Other experience, advanced education and/or proven aptitude and performance may count for education or experience. Experience may include laboratory, regulatory or other related environmental or management experience.

DUTIES/RESPONSIBILITIES

Schedule personnel, equipment and resources to produce data which complies with methods, SOP's, the QAP and customer data quality objectives and customer due dates.

Insure training of analysts on methods, SOP's, the CHP and the QAP prior to performing procedures. Document training and maintain training files. Maintain methods, SOP's, the QAP and the CHP on file and ensure document control.

Assure instruments and equipment are inspected, cleaned, maintained and as applicable, calibrated and standardized.

Review raw data for conformance with methods, SOP's, the QAP and client data quality objectives. Document all deviations from methods, SOP's and the QAP. Provide case narratives and supporting analytical documentation for data packages when necessary. Forward data to Data Management Group. Archive raw data files following data entry.

Inform the Lab Director immediately of any personnel, instrument, equipment or materials problems. Inform the Lab Director, Project Managers and QA Unit(s) of any deviations from schedule, methods, SOP's or the QAP.

Respond to and undertake corrective action to external and internal audits.

Job Description

LAB ANALYST/LAB TECHNICIAN/FIELD TECHNICIAN

QUALIFICATIONS

Position

Education

Lab Analyst

Degree Preferred

Field Technician

Degree Preferred

Other experience, advanced education and/or proven aptitude and performance may count for education or experience. Experience may include laboratory, regulatory or other related environmental or management experience.

DUTIES/RESPONSIBILITIES

Perform sample collection, preparation and analysis in conformance with methods, SOP's and the QAP to meet client due dates and data quality objectives.

Obtain training on and read and understand all methods, SOP's and the QAP prior to performing new procedures.

Inspect, clean, calibrate, standardize and maintain sampling, analysis and measurement equipment, glassware and instrumentation. Document these activities as required by SOP's, methods and the QAP.

Accurately and neatly document all calculations, observations and deviations as required by methods, SOP's and the QAP and forward raw data to Group Leader.

Handle lab chemicals and samples per custody requirements, SOP's, QAP and the CHP. Properly dispose of laboratory waste. Maintain organized work area.

Inform the Group Leader or Lab Director immediately of any personnel, instrument, equipment or material problems or shortages. Inform the Group Leader or Lab Director of any deviations from schedule, methods, SOP's or the QAP.

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Job Description

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SAMPLE CUSTODIAN/SHIPPING COORDINATOR

QUALIFICATIONS

<u>Position</u> <u>Education</u>

Sample Custodian

BS/BA Science

Shipping Coordinator

HS

Other experience, advanced education and/or proven aptitude and performance may count for education or experience. Experience may include laboratory, regulatory or other related environmental or management experience.

DUTIES/RESPONSIBILITIES

Receive, handle, ship, store and dispose of samples and sample containers in conformance with methods, SOP's, the CHP, the QAP and shipping regulations to meet client due dates and data quality objectives. Coordinate shipping and receiving with client and Project Manager.

Maintain inventory of all supplies.

Validate and maintain legal chain of custody of samples and necessary documentation. Document and inform Project Managers immediately of any deviations from anticipated work volume or schedule, methods, SOP's or the QAP.

Obtain training on and read and understand all methods, SOP's and the QAP prior to performing new procedures.

Maintain organization of sample and waste storage areas. Coordinate disposal of samples and lab waste.

Assure that samples are logged into the LIMS timely and accurately.

Job Description

DATA PACKAGE COORDINATOR/DATA ENTRY TECHNICIAN/DOCUMENT CONTROL OFFICER

QUALIFICATIONS

Position

Education

Data Package Coordinator HS Document Control Officer HS Data Entry Technician HS

Other experience, advanced education and/or proven aptitude and performance may count for education or experience. Experience may include laboratory, regulatory or other related environmental or management experience.

DUTIES/RESPONSIBILITIES

Accurately and expediently record all required analytical, sample, and accounting data in LIMS in conformance with methods, SOP's and the QAP.

Obtain training on and read and understand all relevant methods, SOP's and the QAP prior to performing new procedures.

Assemble, copy and distribute to the Project Manager, data packages and data deliverables per client specified requests, SOP's and the QAP to meet clients' due dates.

Coordinate archiving of raw data for GLP, CLP and other client specified projects.

Inform the Lab Director immediately of any personnel or equipment problems. Inform the Lab Director of any deviations from schedule, methods, SOP's or the QAP.



Appendix F Revision 0 Nov. 1, 1991

FINAL REPORT EXAMPLES

CERTIFICATE OF ANALYSIS

Service Location		ceived	Lab I	
EMS HERITAGE LABORATORIES, INC.	<u> </u>	APR-91	A228	
7901 W. MORRIS ST.	ļ	mplete	PO Nun	
INDIANAPOLIS, IN 46231 (317)243-8305		JUN-91	9060768	
(317)243-6303		JUN-91	Sample	ea
	13-	JUN-91		
Report To		Bill To		
•				
Sample De	escription			
FAA OR ICP ACID DIGESTION OF AQUEOUS SAMPLES	SW846-3005			
Analyst: J. CARSON Analysis Date: 30-APR-			Test: P130.4.	0
Parameter	Result		Det. Limit	Units
INITIAL WEIGHT OR VOLUME	50			mL 1
FINAL WEIGHT OR VOLUME	50			_mL
BARIUM ICP SW846-6010 Analyst: M. JAO Prep: FAA OR ICP ACID DIGESTION OF AQUEOUS S		j .	Test: M104.3.	0
Parameter BARIUM	Result 0.049		Det. Limit 0.010	Units mg/L
CADMIUM ICP SW846-6010 Analysis H. JAO Prep: FAA OR ICP ACID DIGESTION OF AQUEOUS S		<u> </u>	Test: M108.3.	0
Parameter	Result		Det. Limit	Units
CADMIUM	BDL		0.0050	mg/L
CHROMIUM ICP SW846-6010 Analyst: M. JAO Analysis Date: 01-MAY- Prep: FAA OR ICP ACID DIGESTION OF AQUEOUS S	-91 Instrument: ICP AMPLES SW846-3005	5	Test: M110.3.	0
Parameter CHROMIUM	Result BDL		Det. Limit 0.010	Units mg/L
NICKEL ICP SW846-6010 Analysis Date: 01-MAY Analysis Date: 01-MAY			Test: M122.3.	0
Prep: FAA OR ICP ACID DIGESTION OF AQUEOUS S) 		<u> </u>
Parameter NICKEL	Result BDL		Det. Limit 0.010	Units
NIUNEL	DUL		0.010	mg/L
SILVER ICP SW846-6010 Analyst: M. JAO Analysis Date: 01-MAY Prep: FAA OR ICP ACID DIGESTION OF AQUEOUS S		5	Test: M130.3.	0
Parameter	Result	·	Det. Limit	Units
SILVER	BOL		0.010	_
<u> </u>				

Frep. TAA OF	R ICP ACID DIGESTION OF A	Result	Det. Limit Units
RON	rai alleter	BDL	0.020 mg/L
SODIUM ICP SI Analyst: M. JAO Prep: FAA OI	Analysis Da	ate: 01-MAY-91 Instrument: ICP QUEOUS SAMPLES SW846-3005	Test: M131.3. 0
SODIUM	Parameter	Result 6.7	Det. Limit Units 0.20 mg/L
MANGANESE ICI Analyst: M. JAO Prep: FAA OI	Analysis Da	ate: 01-MAY-91 Instrument: ICP QUEOUS SAMPLES SW846-3005	Test: M119.3. 0
MANGANESE	Parameter	Result BDL	Det. Limit Units 0.010 mg/L
GFAA ACID DIG	GESTION OF AQUEOUS SAMPLE Analysis Da	S SW846-3020 ate: 04-MAY-91	Test: P130.6. 0
INITIAL WEIGH FINAL WEIGHT (Result 50 50	Det. Limit Unit: mL mL
ARSENIC GFAA	SW846-7060		
·		ate: 10-MAY-91 Instrument: GFAA SAMPLES SW846-3020 Result BDL	Test: M103.2. 0 Det. Limit Unit: 0.0050 mg/L
Prep: GFAA ARSENIC LEAD GFAA SWA	Analysis Da ACID DIGESTION OF AQUEOUS Parameter 846-7421	Result BDL ate: 09-MAY-91 Instrument: GFAA	Det. Limit Unit
Prep: GFAA ARSENIC LEAD GFAA SWAANALYST: M. BAUER Prep: GFAA	Analysis Da ACID DIGESTION OF AQUEOUS Parameter 846-7421 Analysis Da	Result BDL ate: 09-MAY-91 Instrument: GFAA	Det. Limit Unit: 0.0050 mg/L
Prep: GFAA ARSENIC LEAD GFAA SWA Analyst: M. BAUER Prep: GFAA LEAD SELENIUM GFAA Analyst: K. KEHOE	Analysis Da ACID DIGESTION OF AQUEOUS Parameter 846-7421 ACID DIGESTION OF AQUEOUS Parameter A SW846-7740	Result BDL ate: 09-MAY-91 Instrument: GFAA S SAMPLES SW846-3020 Result BDL ate: 07-MAY-91 Instrument: GFAA	Det. Limit Unit: 0.0050 mg/L Test: M116.2. 0
Prep: GFAA ARSENIC LEAD GFAA SWANALYST: M. BAUER Prep: GFAA ARSENIUM GFAA ANALYST: K. KEHOE Prep: GFAA	Analysis Da ACID DIGESTION OF AQUEOUS Parameter 846-7421 ACID DIGESTION OF AQUEOUS Parameter A SW846-7740 Analysis Da Analysis Da Analysis Da	Result BDL ate: 09-MAY-91 Instrument: GFAA S SAMPLES SW846-3020 Result BDL ate: 07-MAY-91 Instrument: GFAA	Det. Limit Units 0.0050 mg/L Test: M116.2. 0 Det. Limit Unit 0.0050 mg/L
Prep: GFAA ARSENIC LEAD GFAA SWA Analyst: M. BAUER Prep: GFAA LEAD SELENIUM GFAA Analyst: K. KEHOE Prep: GFAA	Analysis Da Parameter B46-7421 ACID DIGESTION OF AQUEOUS Parameter A SW846-7740 ACID DIGESTION OF AQUEOUS Parameter ACID DIGESTION OF AQUEOUS Parameter ACID DIGESTION OF AQUEOUS	Result BDL ate: 09-MAY-91 Instrument: GFAA S SAMPLES SW846-3020 Result BDL ate: 07-MAY-91 Instrument: GFAA S SAMPLES SW846-3020 Result BDL Result BDL	Det. Limit Unit: 0.0050 mg/L Test: M116.2. 0 Det. Limit Unit 0.0050 mg/L Test: M128.2. 0
Prep: GFAA ARSENIC LEAD GFAA SWA Analyst: M. BAUER Prep: GFAA LEAD SELENIUM GFAA Analyst: K. KEHOE Prep: GFAA MERCURY CVAA Analyst: J. WARE INITIAL WEIGH	Analysis Danalysis Danalysis Danameter Analysis Danameter AACID DIGESTION OF AQUEOUS Parameter A SW846-7740 ACID DIGESTION OF AQUEOUS Parameter ACID DIGESTION OF AQUEOUS Parameter ACID DIGESTION OF AQUEOUS Parameter	Result BDL ate: 09-MAY-91 Instrument: GFAA SAMPLES SW846-3020 Result BDL ate: 07-MAY-91 Instrument: GFAA SAMPLES SW846-3020 Result BDL Result BDL	Det. Limit
Prep: GFAA ARSENIC LEAD GFAA SWI Analyst: M. BAUER Prep: GFAA LEAD SELENIUM GFAA Analyst: K. KEHOE Prep: GFAA MERCURY CVAA Analyst: J. WARE INITIAL WEIGH FINAL VOLUME MERCURY CVAA Analyst: J. WARE	Analysis Danalysis Danalysis Danameter 846-7421 ACID DIGESTION OF AQUEOUS Parameter A SW846-7740 ACID DIGESTION OF AQUEOUS Parameter ACID DIGESTION OF AQUEOUS Parameter T OR VOLUME SW846-7470 Analysis Dana	Result BDL ate: 09-MAY-91 Instrument: GFAA SAMPLES SW846-3020 Result BDL ate: 07-MAY-91 Instrument: GFAA SAMPLES SW846-3020 Result BDL SSAMPLES SW846-7470 ate: 08-MAY-91 Result 100	Det. Limit

CHLORIDE (COLORIMETRIC, AUTO			······································	
Analyst: C. BRODERICK	Analysis Date: 15-MAY-91	Instrument: AUTO-ANALYZER	Test: G102.9.	,
Parameter		Result 2	Det. Limit	Units
CHLORIDE			1.0	mg/L
PHENOLS DISTILLATION SW846-9	065			
Analyst: D. JOSEPH	Analysis Date: 24-APR-91		Test: P405.7.	0
Parameter		Result	Det. Limit	Units
INITIAL WEIGHT OR VOLUME		100		mL
FINAL VOLUME		100		mL
PHENOLS 4AAP (AUTOMATED) SW84 Analyst: J. GRIFFIN Prep: PHENOLS DISTILLATION	Analysis Date: 25-APR-91	Instrument: AUTO-ANALYZER	Test: 0405.7.	0
Parameter		Result	Det. Limit	Units
PHENOLS		BDL	0.01	mg/L
CHI FATE TURDIDINETRIC METHOD	CUOAS 0020			
SULFATE TURBIDIMETRIC METHOD Analyst: K. RILEY	Analysis Date: 16-MAY-91		Test: G108.6.	0
Parameter		Result	Det. Limit	Units
SULFATE		200	125	mg/L
PH (AQUEOUS) SW846-9040 Analyst: M. BAUER	Amelijais Debis 27 ADD 04		T 0/07 F	•
Parameter	Analysis Date: 23-APR-91	Result	Test: G607.5.	Units
PH PH allecter		7.2 Result	0.1	Std. Unit
		1		
SPECIFIC CONDUCTANCE SW846-9				
Analyst: L. MATTINGLY	Analysis Date: 24-APR-91	1	Test: G604.4.	
Parameter CONDUCTIVITY		Result 870	Det. Limit	Units umHOS/cm
CONDUCTIVITY		1070	1.0	L dim 105/ Cili
TOTAL ORGANIC CARBON SW846-9	060			
Analyst: L. BUTLER	Analysis Date: 25-APR-91	Instrument: TOC	Test: 0401.0.	0
Parameter		Result	Det. Limit	Units
OTAL ORGANIC CARBON (TOC)		BDL	3	mg/L
TOTAL ORGANIC HALIDES SW846-	9020		<u></u>	
Analyst: S. HALLORAN	Analysis Date: 25-APR-91	Instrument: TOX	Test: 0404.0.	0
Parameter		Result	Det. Limit	Units
TOTAL ORGANIC HALOGEN (TOX)		BDL	0.04	mg/L
DISCOLVED SOLIDE EDA 160 1				
DISSOLVED SOLIDS EPA 160.1 Analyst: P. ANDERSON	Analysis Date: 24-APR-91		Test: G402.7.	0
Parameter		Result	Det. Limit	Units
SOLIDS		570	10	
TOTAL SOLIDS EPA 160.3	Analysis Bak: 38 and 54			•
Analyst: P. ANDERSON	Analysis Date: 30-APR-91		Test: G401.7.	
=		Result	Det. Limit	Units
Parameter SOLIDS		570	10	mg/L

Parameter

INITIAL WEIGHT OR VOLUME FINAL VOLUME

Page	3	

Units

mL mL

Det. Limit

Result

250 250

CYANIDE, TOTAL (AUTOMATED) SW846-9012 Analysis Date: 24-APR-91 Test: G101.4. 0 Analyst: J. GRIFFIN Instrument: AUTO-ANALYZER Prep: CYANIDE DISTILLATION SW846-9010 Units Parameter Result Det. Limit CYANIDE **BDL** 0.01 mg/L **SULFIDE SW846-9030** Analysis Date: 30-APR-91 Test: G110.4. 0 Analyst: L. HETTICH Parameter Result Det. Limit Units SULFIDE **BDL** 1.0 mg/L

3011101	T DDC		<i>AI</i>
VOLATILE ORGANICS TARGET COMPOUND LIST		T	•
Analyst: H. WILLIAMS Analysis Date: 02-M		Test: 0530.1.	
Parameter ACETONE	Result BDL	Det. Limit 20	Units ug/L
ACROLEIN	BDL	50	
	BDL		ug/L
ACRYLONITRILE	BDL	70	ug/L
BENZENE BROMORICHI OROMETHANE		5	ug/L
BROMODICHLOROMETHANE	BDL	5	ug/L
BROMOFORM	BDL	5	ug/L
BROMOMETHANE	BDL	10	ug/L
CARBON DISULFIDE	BDL	5	ug/L
CARBON TETRACHLORIDE	BDL	5	ug/L
CHLOROBENZENE	BDL	5	ug/L
CHLOROETHANE	BDL	10	ug/L
CHLOROFORM	BDL	5	ug/L
CHLOROMETHANE	BDL	10	ug/L
DIBROMOCHLOROMETHANE	BDL	5	ug/L
CIS-1,3-DICHLOROPROPENE	BDL	5	ug/L
DICHLORODIFLUOROMETHANE	BDL	5	ug/L
1,1-DICHLOROETHANE	BDL	5	ug/L
1,2-DICHLOROETHANE	BDL	5	ug/L
1,1-DICHLOROETHENE	BDL	5	
1,2-DICHLOROPROPANE	BDL	5	ug/L
			ug/L
ETHYLBENZENE EL HODOTPAGUL ODOMETHANE	BDL	5	ug/L
FLUOROTRICHLOROMETHANE	BDL	5	ug/L
2-HEXANONE	BDL	10	ug/L
METHYLENE CHLORIDE	BDL	5	ug/L
METHYL ETHYL KETONE	BDL	10	ug/L
4-METHYL-2-PENTANONE	BDL	10	ug/L
STYRENE	BDL	5	ug/L
1,1,2,2-TETRACHLOROETHANE	BDL	5	ug/L
TETRACHLOROETHENE	BDL	5	ug/L
TETRAHYDROFURAN	BDL	25	ug/L
TOLUENE	BDL	5	ug/L
1,2-DICHLOROETHENE (TOTAL)	BDL	5	ug/L
TRANS-1,3-DICHLOROPROPENE	BDL	5	ug/L
1,1,1-TRICHLOROETHANE	BDL	5	ug/L
1,1,2-TRICHLOROETHANE	BDL	5	ug/L
TRICHLOROETHENE	BDL	5	
VINYL ACETATE	BDL	r .	ug/L
		10	ug/L
VINYL CHLORIDE	BDL	10	ug/L
XYLENE (TOTAL)	BDL	5	ug/L
2-CHLOROETHYLVINYLETHER	BDL	10	ug/L
DIETHYLETHER	BDL	5	ug/L
1,1,2-TRICHLORO-1,2,2-TRIFLUOROETHANE	BDL	5	ug/L
ETHYL ACETATE	BDL	5	ug/L

Lab	Sample	: ID:	A228	370
-----	--------	-------	------	-----

Parameter METHYL-T-BUTYL ETHER	Result BDL	Det. Limit 5	Units ug/L
SURROGATE RECOVERY			
DICHLOROETHANE-D4 TOLUENE-D8 BROMOFLUOROBENZENE	93 101 101		% Rec % Rec % Rec

GC/MS SEPARATORY FUNNEL LIQUID-LIQUID E	XTRACTION SW846-3510		
Analyst: M. FRANK Analysis Date:	: 27-APR-91	Test: P233.4	. 0
Parameter INITIAL WEIGHT OR VOLUME	Result	Det. Limit	Units Liters
FINAL VOLUME	1		mL

SEMI-VOLATILE TARGET COMPOUND LIST : Analyst: K. STONER Prep: GC/MS SEPARATORY FUNNEL LIQUID-LI	21-MAY-91 Instrument: GC/MS SVOA	Test: 0531.1.	0
Parameter	Result	Det. Limit	Units
CENAPHTHENE	BDL	10	ug/L
CENAPHTHYLENE	BDL	10	ug/L
INTHRACENE	BDL	10	ug/L
BENZ (A) ANTHRACENE	BDL	10	ug/L
BENZÔ(Å)PYRENE	BDL	10	ug/L
BENZO(B) FLUORANTHENE	BDL	10	ug/L
ENZO(G,H,I)PERYLENE	BDL	10	ug/L
ENZO(K)FLUORANTHENE	BDL	10	ug/L
ENZYL ÁLCOHOL	BDL	10	ug/L
BENZYLBUTYLPHTHALATE	BDL	10	ug/L
SIS(2-CHLOROETHOXY)METHANE	BDL	10	ug/L
BIS(2-CHLOROETHYL)ÉTHER	BDL	10	ug/L
BIS(2-CHLOROISOPROPYL)ETHER	BDL	10	ug/L
IS(2-ETHYLHEXYL)PHTHÁLATE	13	10	ug/L
-BROMOPHENYLPHENYLETHER	BDL	10	ug/L
ARBAZOLE	BDL	10	ug/L
-CHLOROANILINE	BDL	10	ug/L
-CHLORONAPHTHALENE	BDL	10	ug/L
-CHLOROPHENYLPHENYLETHER	BDL	10	ug/L
HRYSENE	BDL	10	ug/L
DIBENZ(A,H)ANTHRACENE	BDL	10	ug/L
IBENZOFURAN	BDL	10	ug/L
,2-DICHLOROBENZENE	BDL	10	ug/L ug/L
,3-DICHLOROBENZENE	BDL	10	ug/L ug/L
,4-DICHLOROBENZENE	BDL	10	ug/L ug/L
3'-DICHLOROBENZIDINE	BDL	20	T -
1;3 -DICHLOROBENZIDINE DIETHYLPHTHALATE	BDL	10	ug/L
DIMETHYLPHTHALATE	BDL		ug/L
		10	ug/L
OI-N-BUTYLPHTHALATE	BDL	10	ug/L
DINITROBENZENES	BDL	50	ug/L
2,4-DINITROTOLUENE	BDL	10	ug/L
2,6-DINITROTOLUENE	BDL	10	ug/L
OI -N-OCTYLPHTHALATE	BDL	10	
LUORANTHENE	BDL	10	ug/L
LUORENE	BDL	10	ug/L
IEXACHLOROBENZENE	BDL	10	ug/L
IEXACHLOROBUTADIENE	BDL	10	ug/L
HEXACHLOROCYCLOPENTADIENE	BDL	10	ug/L
HEXACHLOROETHANE	BDL	10	ug/L

Lab Sample ID: A228370

	EMS HERITAGE LABORATURIES, INC.	L	ab Samble II): AZZ63/U
	Parameter	Result	Det. Limit	Units
	INDENO(1,2,3-CD)PYRENE	BDL	10	ug/L
	ISOPHORONE	BDL	10	ug/L
	2-METHYLNAPHTHALENE	BDL	10	ug/L
i	NAPHTHALENE	BDL	10	ug/L
	2-NITROANILINE	BDL	50	ug/L
-	3-NITROANILINE	BDL	50	ug/L
1	4-NITROANILINE	BDL	50	ug/L
	NITROBENZENE	BDL	10	ug/L
_ :	N-NITROSO-DIPHENYLAMINE	BDL	io	ug/L
	N-NITROSO-DI-N-PROPYLAMINE	BDL	10	ug/L
	PHENANTHRENE	BDL	10	ug/L
	2-PICOLINE	BOL	50	ug/L
	PYRENE	BDL	10	ug/L
	PYRIDINE	BDL	50	ug/L
	TETRACHLOROBENZENES	BDL	10	ug/L
-	TOLUENEDIAMINE	BDL	50	ug/L
	1,2,4-TRICHLOROBENZENE	BDL	10	ug/L
	BENZOIC ACID	BDL	50	ug/L
	4-CHLORO-3-METHYLPHENOL	BDL	10	
_	2-CHLOROPHENOL	BDL	10	ug/L
	2,4-DICHLOROPHENOL	BDL	10	ug/L
		BDL		ug/L
-	2,4-DIMETHYLPHENOL		10	ug/L
	4,6-DINITRO-2-METHYLPHENOL	BDL	50	ug/L
	2,4-DINITROPHENOL	BDL	50	ug/L
_	2-METHYL PHENOL	BDL	10	ug/L
	4-METHYLPHENOL	BDL	10	ug/L
	2-NITROPHENOL	BDL	10	ug/L
	4-NITROPHENOL	BDL	50	ug/L
	PENTACHLOROPHENOL	BDL	50	ug/L
	PHENOL	BDL	10	ug/L
	TETRACHLOROPHENOL	BDL	10	ug/L
	2,4,5-TRICHLOROPHENOL	BDL	10	ug/L
	2,4,6-TRICHLOROPHENOL	BDL	10	ug/L
	1,2-DIPHENYLHYDRAZINE	BDL	10	ug/L
	N-NITROSODIMETHYLAMINE	BDL	10	ug/L
_	1,2,3,4-TETRACHLOROBENZENE	BDL	10	
	1,2,4,5-TETRACHLOROBENZENE	BDL	10	ug/L
_	SURROGATE RECOVERY			
	2 CLHODODUCNOL	50	1	ov Doc
	2-FLUOROPHENOL	52		% Rec
	PHENOL - D5	32		% Rec
	NITROBENZENE-D5	94		% Rec
	2-FLUOROBIPHENYL	78		% Rec
	2,4,6-TRIBROMOPHENOL	78		% Rec
	TERPHENYL - D14	82	I	% Rec

Samo	e	Comments

BDL Below Detection Limit

QUALITY ASSURANCE REPORT

Service Location	Received	Lab ID
EMS HERITAGE LABORATORIES, INC.	23-APR-91	_ A228370 -
7901 W. MORRIS ST.	Complete	PO Number
INDIANAPOLIS, IN 46231	05-JUN-91	90607684-59A
(317)243-8305	Printed	Sampled
	13-JUN-91	<u> </u>

	_		
Sample	Descr	10	it i on

BARI	UM ICP S	W846-601	0			··		·-··	
Analys	st : M. JAO		Analysis Date: 01-MAY-9	1 Instrum	ent: ICP		Test: M104.	3.0	
	er: S. ENDE p: FAA O		Review Date: 02-MAY-9 ID DIGESTION OF AQUEOUS SA		-		Run: R1198	38	
QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
ICV01	0239253	WP1083	BARIUM	2.00		2.06	mg/L	103	
DPS02	0231974	A228372	BARIUM	0	2.00	1.92	mg/L	96	4 -
SP102	0231973	A228372	BARTUM	0	2.00	1.84	mg/L	92	
ccv	9239277	INORGANIC	BARTUN	5.00		4.52	mg/L	90	
BLA01	9239276	NA	BARTUM			< 0.004	mg/L		1.
LCS	9231971	NA	BARIUM	20.0		18.9	mg/L	94	
BLA02	0231972	NA	BARIUM	[į	< 0.004	mg/L		
SAMPLE	A228370		See Certificate of Analysis				-	1.0	
CCV	9239275	INORGANIC	BARIUM	5.00		4.81	mg/L	96	
BLA01	9239274	NA	BARIUM			< 0.004	mg/L		

CADMIUM ICP SW846-6010			
Analyst : M. JAO	Analysis Date: 01-MA	Y-91 Instrument: ICP	Test: M108.3.0
Reviewer: S. ENDERSEN	Review Date: 02-MA		Run: R119838

	;		ID DIGESTION OF AQUEOUS		r	r	11-1-	T# 5	T
QC Type	Identifier	Source	Par meter	True/Sampi	Spike Val	0bserved	Units	% Rec	RPD
ICV01	Q239254	WP1083	CADMIUM	0.400		0.425	mg/L	106	1
I CV01	Q239259	WP1083	CADMIUM	0.400		0.407	mg/L	102	1
DPS02	Q231974	A228372	CADHIUM	0	0.050	0.0415	mg/L	83	4 -
SP102	Q231973	A228372	CADMIUM	0	0.050	0.0398	mg/L	80	1
CDL01	0239279	NA	CADMIUM	0.0100	İ	0.0116	mg/L	116	
CCV	Q239277	INORGANIC	CADHIUM	5.00		4.48	mg/L	90	-
BLA01	Q239276	NA	CADMIUM			< 0.002	mg/L	1	1
LCS	Q231971	NA	CADNIUM	0.500		0.407	mg/L	81	
BLA02	9231972	NA	CADMIUM			< 0.002	mg/L		_
SAMPLE	A228370	1	See Certificate of Analysis						1
CCV	Q239275	INORGANIC	CADMIUM	5.00		4.80	mg/L	96	
BLA01	9239274	NA	CADMIUM			< 0.002	mg/L	}	
CDL01	9239270	NA	CADMIUM	0.0100		0.0122	mg/L	122	-

Analys Review	MIUM ICP st: M. JAO eer: S. ENDE p: FAA O	RSEN	6010 Analysis Date Review Date CID DIGESTION OF AQU	: 02-MAY-91 File !			Test: M110.: Run: R1198			- Т
QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD	
1CV01	Q239254	WP1083	CHROMIUM	0.400		0.421	mg/L	105		Τ
ICV01	9239259	WP1083	CHROMIUM	0.400		0.400	mg/L	100		- 1
DPS02	9231974	A228372	CHROMIUM	0	0.200	0.193	mg/L	97	3	
SP102	Q231973	A228372	CHROMIUM	0	0.200	0.188	mg/L	94		Т
CDL01	9239279	NA	CHROMIUM	0.0200		0.0260	mg/L	130		

Page 1

Lab Sample ID: A228370

QC Type	Identifier	Source	Parameter ·	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
ccv	Q239277	INORGANIC	CHROMIUM	5.00		4.54	mg/L	91	
BLA01	9239276	NA	CHROMIUM			< 0.004	mg/L		
LCS	Q231971	NA	CHROMIUM	2.00		1.85	mg/L	92	
BLA02	Q231972	NA	CHROMIUM			< 0.004	mg/L		
SAMPLE	A228370		See Certificate of Analysis				1		
CCV	9239275	INORGANIC	CHROMIUM	5.00		4.83	mg/L	97	
BLA01	9239274	NA	CHROMIUM			< 0.004	mg/L		
CDL01	9239270	NA	CHROMIUM	0.0200		0.0210	mg/L	105	

NICKEL ICP SW846-6010

Analyst: M. JAO Analysis Date: 01-MAY-91 Instrument: ICP Test: M122.3.0
Reviewer: S. ENDERSEN Review Date: 02-MAY-91 File ID: Run: R119838

Prep: FAA OR ICP ACID DIGESTION OF AQUEOUS SAMPLES SW846-3005

	ch. IVV C	IL TOL MO	TO DIGESTION OF MOLOGS	JAN LLJ JA	040 3003				
OC Typ	e Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
1CV01	9239254	WP1083	NICKEL	0.400		0.419	mg/L	105	
ICV01	9239259	WP1083	NICKEL	0.400		0.408	mg/L	102	
DPS02	9231974	A228372	NICKEL	0	0.500	0.468	mg/L	94	3
SP102	9231973	A228372	NICKEL	0	0.500	0.457	mg/L	91	
CDL01	9239279	NA	NICKEL	0.0200		0.0229	mg/L	115	
ccv	9239277	INORGANIC	NICKEL	5.00		4.50	mg/L	90	
BLA01	9239276	NA	NICKEL			< 0.004	mg/L		
LCS	9231971	NA	NICKEL	5.00		4.62	mg/L	92	
BLA02	9231972	NA	NICKEL			< 0.004	mg/L		
SAMPLE	A228370		See Certificate of Analysis		1				1
CCV	9239275	INORGANIC	NICKEL	5.00		4.79	mg/L	96	1
BLA01	9239274	NA	NICKEL		1	< 0.004	mg/L		
CDL01	9239270	NA	NICKEL	0.0200		0.0199	mg/L	99	

SILVER ICP SW846-6010

Analyst: M. JAO Analysis Date: 01-MAY-91 Instrument: ICP Test: M130.3.0 Reviewer: S. ENDERSEN Review Date: 02-MAY-91 File ID: Run: R119838

Prep: FAA OR ICP ACID DIGESTION OF AQUEOUS SAMPLES SW846-3005

110	P. IAA C	IN IOI AC	TO PIGESTION OF AQUEOUS	SAM LLS SH	040-3003				
QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
[CV01	Q239253	WP1083	SILVER	2.00	1	2.05	mg/L	102	
DPS02	9231974	A228372	SILVER	0	0.050	0.0492	mg/L	98	2
SP102	9231973	A228372	SILVER	0	0.050	0.0479	mg/L	96	
CDL01	9239252	NA	SILVER	0.0200		0.0200	mg/L	100	
CCV	9239277	INORGANIC	SILVER	1.00	1	0.912	mg/L	91	
BLA01	9239276	NA	SILVER		ŀ	< 0.004	mg/L		
LCS	9231971	NA	SILVER	0.500		0.468	mg/L	94	
BLA02	9231972	NA	SILVER		İ	< 0.004	mg/L		
SAMPLE	A228370		See Certificate of Analysis						
CCV	Q239275	INORGANIC	SILVER	1.00		0.950	mg/L	95	1
BLA01	9239274	NA	SILVER		1	< 0.004	mg/L		
CDL01	9239271	NA	SILVER	0.0200	}	0.0208	mg/L	104	1

TDA	81 '	7 ^	n .	CI	104		c 0	10
IRO	N.	ı L	P :	3 H	184	- a	ดบ	10

Analyst: M. JAO Analysis Date: 01-MAY-91 Instrument: ICP Test: M115.3.0 Reviewer: S. ENDERSEN Review Date: 02-MAY-91 File ID: Run: R119838

Prep: FAA OR ICP ACID DIGESTION OF AQUEOUS SAMPLES SW846-3005

110	p. TAN O	it Iti At	TD BIGESTION OF	MOLOUS SMITELS ON	0.0 0000				_
QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
1CV01	Q239254	WP1083	IRON	0.400		0.425	mg/L	106	
1CV01	9239259	WP1083	IRON	0.400		0.413	mg/L	103	1
DPS02	Q231974	A228372	IRON	0	1.00	0.968	mg/L	97	3
SP102	9231973	A228372	IRON	0	1.00	0.942	mg/L	94	1
CDL01	9239279	NA	IRON	0.0400		0.0407	mg/L	102	
CCA	0239277	INORGANIC	IRON	5.00		4.59	mg/L	92	

Lab Sample ID: A22837-

QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
BLA01	9239276	NA	IRON			< 0.008	mg/L		ļ
LCS	9231971	NA	IRON	10.0	1	9.40	mg/L	94	_
BLA02	9231972	NA	IRON			< 0.008	mg/L		
SAMPLE	A228370	1	See Certificate of Analysis	ļ	1			1	1
CCV	9239275	INORGANIC	IRON	5.00		4.89	mg/L	98	_
BLA01	9239274	NA	IRON			< 0.008	mg/L		
CDL01	9239270	NA	IRON	0.0400	<u> </u>	0.0453	mg/L	113	<u> </u>

SODIUM ICP SW846-6010

Analyst : M. JAO

Analysis Date: 01-MAY-91 Instrument: ICP

Test: M131.3.0

Reviewer: S. ENDERSEN

Review Date: 02-MAY-91 File ID:

Run: R119838

Prep: FAA OR ICP ACID DIGESTION OF AQUEOUS SAMPLES SW846-3005

QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
DPS02	ŀ	A228372	SODIUM	0.202	100	94.7	mg/L	94	4
SP102	Q231973	A228372	SODIUM	0.202	100	91.5	mg/L	91	1
ccv	9239277	INORGANIC	SODIUM	50.0		45.2	mg/L	90	-
BLA01	9239276	NA	SODIUM			< 0.08	mg/L		1
LCS	Q231971	NA	SODIUM	100		95.7	mg/L	96	
BLA02	9231972	NA .	SODIUM			< 0.08	mg/L		-
SAMPLE	A228370		See Certificate of Analysis						İ
CCV	9239275	INORGANIC	SODIUN	50.0		47.6	mg/L	95	
BLA01	9239274	NA	SODIUM		ļ	< 0.08	mg/L	ļ	

MANGANESE ICP SW846-6010

Analyst : M. JAO

Analysis Date: 01-MAY-91 Instrument: ICP

Test: M119.3.0

Reviewer: S. ENDERSEN

Review Date: 02-MAY-91 File ID:

Run: R119838

Pren: FAA OR TCP ACID DIGESTION OF ADUFOUS SAMPLES SW846-3005

QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPC
ICV01	Q239254	WP1083	MANGANESE	0.400		0.423	mg/L	106	-
CV01	9239259	WP1083	MANGANESE	0.400	İ	0.414	mg/L	103	
PS02	Q231974	A228372	MANGANESE	0	0.500	0.485	mg/L	97	2
SP102	Q231973	A228372	MANGANESE	0	0.500	0.473	mg/L	95	١.
DL01	9239279	NA	MANGANESE	0.0200	}	0.0211	mg/L	105	
ccv	9239277	INORGANIC	MANGANESE	5.00	ļ	4.54	mg/L	91	
BLA01	Q239276	NA	MANGANESE	ļ		< 0.004	mg/L		
_cs	9231971	NA	MANGANESE	5.00		4.79	mg/L	96	'
BLA02	9231972	NA	MANGANESE		1	< 0.004	mg/L		1
SAMPLE	A228370		See Certificate of Analysis						-
CCV	9239275	INORGANIC	MANGANESE	5.00		4.82	mg/L	96	.
BLA01	Q239274	NA	MANGANESE			< 0.004	mg/L		1
CDL01	9239270	NA	MANGANESE	0.0200		0.0211	mg/L	105	

ARSENIC	CEAA	CHOVE.	7060
MK SPAIL	13 F M M	3804D=	, i imi i

Analyst : K. KEHOE

Analysis Date: 10-MAY-91 Instrument: GFAA

Test: M103.2.0

Reviewer: S. ENDERSEN

Review Date: 13-MAY-91 File ID: 020782

Run: R120843

Prep: GFAA ACID DIGESTION OF AQUEOUS SAMPLES SW846-3020

110	p. ul AA	MOID DI	acorton of Aqueou	S SAIT LES SHOTO SO	LU				
QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
CAL01	9245956		ARSENIC	0.000		0.000	mg/L		1
CAL01	9245957		ARSENIC	0.0050		0.034	mg/L		_
CAL01	9245958		ARSENIC	0.0100		0.059	mg/L		
CALO1	9245959	ļ	ARSENIC	0.0200		0.111	mg/L		1
CALO1	9245960		ARSENIC	0.0400		0.219	mg/L		1
ICV01	9245962		ARSENIC	0.0100	•	0.0104	mg/L	104	-
DPS02	9239846	A228368	ARSENIC	0.00	0.0400	0.0357	mg/L	89	1.1
SPI02	9239845	A228368	ARSENIC	0.00	0.0400	0.0359	mg/L	90	1
BLA01	Q245961		ARSENIC			< 0.0020	mg/L		_
CCV	9245963		ARSENIC	0.0200		0.0201	mg/L	101	1

Page 3

Lab Sample ID: A228370

QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
CDL01	9245964		ARSENIC	0.0100		0.0101	mg/L	101	
LCS	9239843		ARSENIC	0.0200		0.0183	mg/L	92	1
BLA02	9239844		ARSENIC			< 0.0020	mg/L	1	
SAMPLE	A228370		See Certificate of Analysis					İ	
BLA01	Q245965		ARSENIC			< 0.0020	mg/L		
CCV	9245966		ARSENIC	0.0200	1	0.0196	mg/L	98	
CDL01	9245969		ARSENIC	0.0100		0.0087	mg/L	87	<u> </u>

LEAD GFAA SW846-7421			
Analyst : M. BAUER	Analysis Date: 09-MAY-91	Instrument: GFAA	Test: M116.2.0
Reviewer: S. O'NEAL	Review Date: 10-MAY-91	File ID: 020779	Run: R120734

Pre	p: GFAA	ACID DIG	ESTION OF AQUEOUS SAMPLES	SW846-30	20				
QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
CAL01	Q245452	ŧ	LEAD	0.000		0.000	mg/L		
CAL01	Q245453	ŀ	LEAD	0.0050		0.026	mg/L	1	1
CAL01	Q245454		LEAD	0.010		0.068	mg/L		
CAL01	Q245455		LEAD	0.020		0.122	mg/L		
CAL01	Q245456		LEAD	0.030		0.174	mg/L		
ICV01	9245457		LEAD	0.0100		0.0095	mg/L	95	
DPS02	9239846	A228368	LEAD	0.00	0.0200	0.0224	mg/L	112	9.35
SP102	9239845	A228368	LEAD	0.00	0.0200	0.0204	mg/L	102	
BLA01	Q245458	1	LEAD			< 0.0020	mg/L		
ccv	9245459	1	LEAD	0.0200		0.0190	mg/L	95	1 1
CDL01	9245460	l	LEAD	0.0100		0.0095	mg/L	95	
LCS	9239843	Ì	LEAD	0.0200]	0.0189	mg/L	95	
BLA02	9239844		LEAD			0.0021	mg/L		
SAMPLE	A228370		See Certificate of Analysis		ł				1
CCV	Q245461	-	LEAD	0.0200		0.0201	mg/L	101	
BLA01	Q245462	1	LEAD			< 0.0020	mg/L		
CDL01	Q245463		LEAD	0.0100		0.0103	mg/L	103	

	t : K. KEHOE	•	7-MAY-91 Instrum			Test: M128.		
	er: S. ENDERSEN p: GFAA ACII	Review Date: 00 DIGESTION OF AQUEOUS SAM	8-may-91 File 1D PLES SW846-30			Run: R1203	61	
СТуре	Identifier Sour	ce Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RF
AL01	9243258	SELENIUM	0.000		0.000	mg/L		
ALO1	9243259	SELENTUM	0.0050		0.016	mg/L		
ALO1	9243260	SELENIUM	0.0100		0.031	mg/L		1
ALO1	9243261	SELENIUM	0.0200		0.063	mg/L		
ALO1	9243262	SELENIUM	0.0400		0.119	mg/L		
CV01	9243264	SELENIUM	0.0100		0.0103	mg/L	103	
PS02	Q239846 A228	368 SELENIUM	0.0033	0.0100	0.0135	mg/L	102	3
P102	9239845 A228	368 SELENIUM	0.0033	0.0100	0.0139	mg/L	106	
DL01	Q243266	SELENIUM	0.0100		0.0110	mg/L	110	
LA01	Q243271	SELENIUM			< 0.0020	mg/L		
cv	Q243272	SELENIUM	0.0200		0.0205	mg/L	103	
cs	Q239843	SELENIUM	0.0200		0.0200	mg/L	100	
LAO2	Q239844	SELENIUM			< 0.0020	mg/L		
AMPLE	A228370	See Certificate of Analysis						
LA01	9243273	SELENIUM			0.0021	mg/L		
cv	9243274	SELENIUM	0.0200		0.0212	mg/L	106	
DLO1	9243279	SELENIUM	0.0100		0.0124	mg/L	124	-

Lab Sample ID: A22837_

	er: s. o'NE p: MERCU		Analysis Date: 08-M/ Review Date: 10-M/ ACID DIGESTION OF AQUEOU	AY-91 File ID	: 020776 SW846-74	70	Run: R1206	68	•
C Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
ALO1	Q2449 8 1		MERCURY	0.0000		0.000	mg/L		İ
ALO1	9244982		MERCURY	0.00020		0.009	mg/L		'
ALO1	9244983		MERCURY	0.00050		0.021	mg/L		
ALO1	9244984		MERCURY	0.0010		0.039	mg/L		1
AL01	Q244985		MERCURY	0.0020		0.079	mg/L		'
ALO1	9244986		MERCURY	0.0030		0.105	mg/L		
ALO1	9244987		MERCURY	0.0050		0.178	mg/L		
CV01	9244989		MERCURY	0.00200	1	0.00191	ing/L	96	
PS02	9237814	A228366	MERCURY	0.00	0.0010	0.00105	mg/L	105	21
P102	9237813	A228366	MERCURY	0.00	0.0010	0.00085	mg/L	85	
LA01	9244991		MERCURY			< 0.000020	mg/L		1
CA	9244992		MERCURY	0.00300		0.00290	mg/L	97	
.£S	Q237811		MERCURY	0.00300		0.00273	mg/L	91	
LA02	9237812		MERCURY			< 0.000020	mg/L		
AMPLE	A228370		See Certificate of Analysis						}
LA01	9244993		MERCURY	-		< 0.000020	mg/L		1
CV	9244994		MERCURY	0.00300		0.00290	mg/L	97	

Revie	er: B. SHRA	KE	Analysis Date: 15-M Review Date: 16-M		: 91051501		Test: G102. Run: R1213		
ос туре	Identifier	Source	Parameter	True/Sampt	Spike Val	Observed	Units	% Rec	RPD-
CALO1	9249487	İ	CHLORIDE	100		99.75	mg/L	100	
CALO1	9249488		CHLORIDE	80	1	81.01	mg/L	101	ł
CAL01	Q249489		CHLORIDE	60		59.57	mg/L	99	-
CALO1	9249490		CHLORIDE	40		41.71	mg/L	104	
CALO1	9249491		CHLORIDE	20	İ	19.96	mg/L	100	ł
CALO1	9249492	ļ	CHLORIDE	8		8.27		103	1 .
CALO1	9249493		CHLORIDE	4		3.898	mg/L	97	
CV01	9249495		CHLORIDE	34.4	ŧ	36.5	mg/L	106	1
CV01	9249504		CHLORIDE	34.4		35.6	mg/L	103	1
ICV01	9249521	ļ	CHLORIDE	34.4	-	38.0	mg/L	110	_
SP101	9249501	A228357	CHLORIDE	15	20	36	mg/L	105	}
PS01	9249502	A228357	CHLORIDE	15	20	36	mg/L	105	0
BLA01	9249494		CHLORIDE			-0.529	mg/L		_
CCV	9249496		CHLORIDE	40		40.68	mg/L	102	
SAMPLE	A228370		See Certificate of Analysis			}	1	l	
CCV	9249503	}	CHLORIDE	40	1	40.62	mg/L	101	

	er: B. SHRAKE p: PHENOLS DIS	Review Di 111100 SW846-906	mte: 29-APR-91 File ID 5	: 356		Run: R1194	28	
ас Туре	Identifier Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
ALO1	Q236757	PHENOL	0.90		0.90	mg/L	100	
ALO1	9236758	PHENOL	0.70		0.70	mg/L	100	-
ALO1	9236759	PHENOL	0.50		0.50	mg/L	100	
ALO1	Q236760	PHENOL	0.30		0.30	mg/L	100	}
AL01	9236761	PHENOL	0.10		0.10	mg/L	100	1
AL01	9236762	PHENOL	0.00		-0.001	mg/L		
CV01	9236765	PHENOL	0.400		0.382	mg/L	96	
CV01	9236782	PHENOL	0.40		0.390	mg/L	98	
CV01	9236793	PHENOL	0.400		0.384	mg/L	96	

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Lab Sample ID: A228370

QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
1CV01	9236797		PHENOL	0.40		0.410	mg/L	103	
1CV02	9236767		PHENOL	0.40	1	0.37	mg/L	93	
DUP01	9236779	A228361	PHENOL	0.00		0.00	mg/L		0
SPI01	9236780	A228362	PHENOL	0.00	0.40	0.40	mg/L	100	
DPS02	9236790	A228370	PHENOL	0.00	0.20	0.18	mg/L	90	5
SP102	9236783	A228370	PHENOL	0.00	0.20	0.19	mg/L	95	1
DUP02	9236777	A228356	PHENOL	0.00		0.00	mg/L		0
BLA01	9236763		PHENOL			-0.001	mg/L		
CDL01	9236764		PHENOL	0.01		0.01	mg/L	100	
ccv	9236781		PHENOL	0.30	ļ	0.30	mg/L	100	
SAMPLE	A228370		See Certificate of Analysis						
ccv	9236785		PHENOL	0.30		0.28	mg/L	93	
BLA01	9236798		PHENOL			-0.001	mg/L		
BLAOZ	9236766		PHENOL			0.001	mg/L		

•	st : K. RILE wer: B. SHRA		Analysis Date: 16-M/ Review Date: 17-M/		o: 801-803		Test: G108.		
С Туре	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPC
ALO1	Q250109	1	SULFATE	0.00		0.51	mg/L		
ALO1	Q250110		SULFATE	5		4.3	mg/L	86	
AL01	9250111		SULFATE	10	1	9.7	mg/L	97	
AL01	9250112	1	SULFATE	20	1	20.8	mg/L	104	
ALO1	9250113		SULFATE	30		29.7	mg/L	99	
CV01	Q250117		SULFATE	16.8		16.7	mg/L	99	
CV01	Q250125		SULFATE	16.8		18.3	mg/L	109	
SP101	9250122	A228360	SULFATE	7.9	10	17.6	mg/L	97	
PS01	9250132	A228360	SULFATE	7.9	10	17	mg/L	91	3.
BLA01	9250116	†	SULFATE	· · · · · · · · · · · · · · · · · · ·	-	0.505	mg/L		İ
ccv	Q250120		SULFATE	20		21.1	mg/L	106	
SAMPLE	A228370		See Certificate of Analysis			1			
ccv	9250124	1	SULFATE	20		20.48	mg/L	102	

Analys	AQUEOUS) st : M. BAUE wer: A. DATT	ER	Analysis Date: 23-AP	Analysis Date: 23-APR-91 Review Date: 24-APR-91 File ID:			Test: G607.5.0 Run: R119190			
QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD	
CALO1	9235059		PH	10.0		9.99	Std. Units			
ALO1	9235060		PH	7.0		7.03	Std. Units	1		
CALO1	9235061		PH	4.0		4.02	Std. Units			
CV01	9235062	,	PH	7.80	1	7.87	Std. Units	101		
UP01	0235068	A228356	PH	6.9		6.9	Std. Units		0	
cv	9235071		PH	7.00		7.08	Std. Units	101		
AMPLE	A228370		See Certificate of Analysis							
cv	Q235073	.	РН	7.00		7.00	Std. Units	100		

SPEC	IFIC CON	DUCTANC	E SW846-9050						
Analys	t : L. MATT	INGLY	Analysis Date: 24-AP	R-91			Test: G604.4.	0	
Review	er: B. SHRA	KE	Review Date: 26-AP	R-91 File ID	: 0856-0858		Run: R119306		
QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
ICV01	9235845		CONDUCTIVITY	340		314	umHOS/cm	92	
OUP01	9235846	A228363	CONDUCTIVITY	750		730	umHOS/cm		2.7
BLA01	9235844		CONDUCTIVITY			1.14	umHOS/cm		
SAMPLE	A228370		See Certificate of Analysis					-	-

•	it : L. BUTL ver: B. SHRA		Analysis Date: 25-AF Review Date: 26-AF		nent: TOC): 1059	Test: 0401.0.0 Run: R119531			
QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPC
CALO1	9237419		TOTAL ORGANIC CARBON (TOC)	1000		1005	mg/L	100	
1CV01	Q237420		TOTAL ORGANIC CARBON (TOC)	100		104.5	mg/L	105	1
1CV01	9237432		TOTAL ORGANIC CARBON (TOC)	100		107.8	mg/L	108	
DUP01	9237429	C129570	TOTAL ORGANIC CARBON (TOC)	18	1	18	mg/L		٥
SPI01	9237421	A228359	TOTAL ORGANIC CARBON (TOC)	4	10	14	mg/L	100	ļ
DPS01	9237422	A228359	TOTAL ORGANIC CARBON (TOC)	4	10	14	mg/L	100	0
BLA01	9237418		TOTAL ORGANIC CARBON (TOC)			1.544	mg/L		
ccv	9237424		TOTAL ORGANIC CARBON (TOC)	200		211.2	mg/L	106	
SAMPLE	A228370		See Certificate of Analysis			}			
ccv	Q237431		TOTAL ORGANIC CARBON (TOC)	200	Í	212.6	mg/L	106	

Analys	L ORGANI st : S. HALL Her: B. SHRA	ORAN	ES SW846-9020 Analysis Date: 25-APR- Review Date: 30-APR-		ent: TOX		Test: 0404.		
	Identifier		Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
CAL01	9240124		TOTAL ORGANIC HALOGEN (TOX)	30.00		34.63	mg/L	115	_
1CV01	9240125		TOTAL ORGANIC HALOGEN (TOX)	0.15		0.175	mg/L	117	
1CV01	9240141		TOTAL ORGANIC HALOGEN (TOX)	0.15		0.165	mg/L	111	
1CV01	9240196		TOTAL ORGANIC HALOGEN (TOX)	0.15		0.1607	mg/L	107	
1CV01	9240197		TOTAL ORGANIC HALOGEN (TOX)	0.15		0.1619	mg/L	108	-
DUP01	9240175	A228370	TOTAL ORGANIC HALOGEN (TOX)	0.00		0.00	mg/L		0
DPS01	9240169	A228366	TOTAL ORGANIC HALOGEN (TOX)	0.00	0.025	0.0269	mg/L	108	0.93
SPI01	9240168	A228366	TOTAL ORGANIC HALOGEN (TOX)	0.00	0.025	0.0268	mg/L	107	-
ccv	9240166		TOTAL ORGANIC HALOGEN (TOX)	30.00		31.69	mg/L	107	1
SAMPLE	A228370		See Certificate of Analysis				į.	Ì	ł
ccv	Q240193		TOTAL ORGANIC HALOGEN (TOX)	31.00		35.08	mg/L	113	

Analys	OLVED SO st : P. ANDE ver: B. SHR/	RSON	A 160.1 Analysis Date: 24-AP Review Date: 29-AP		: 1071-1073		Test: G402.7		-
ас туре	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
1CV01	9236961		SOLIDS	131.2		145	mg/L	111	
DUP01	9236963	A228357	SOLIDS	1300		1300	mg/L		0 -
BLA01	9236960	[SOLIDS			0.0006	mg/L] 1
SAMPLE	A228370		See Certificate of Analysis						

Analys	L SOLIDS st : P. ANDE er: B. SHR/	RSON			Test: G401.7.0 Run: R119879				
QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
ICV01	Q239627		SOLIDS	10.0794		10.0784	mg/L	100	
1001	9239630		SOLIDS	12.4845		12.4832	mg/L	100	
DUP01	9239628	A228356	SOLIDS	7800	1	7900	mg/L		1 .
SAMPLE	A228370		See Certificate of Analysis		1			j	1

SAMPLE	A228370		see Certi	Ticate of Analysis			<u> </u>	1			1
Analys Review	it : J. GRIF Her: B. SHRA	FIN Ke	•	SW846-9012 Analysis Date: Review Date: SW846-9010	_		ment: AUTO-A	NALYZER	Test: G101.4 Run: R1195		
QC Type	Identifier	Source	Parameter			True/Sampi	Spike Val	Observed	Units	% Rec	RPD
CAL01	Q237268	ļ,	CYANIDE			0.45		0.44	mg/L	98	
CAL01	9237269	Į,	CYANIDE			0.30		0.31	mg/L	103	1
CAL01	9237270		CYANIDE			0.20		0.21	mg/L	105	-
CAL01	9237271		CYANIDE			0.10		0.098	mg/L	98	

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Lab Sample ID: A228370

EMS	HERITA	GE L/	ABORAT	ORIES	, INC.
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QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
CAL01	9237272		CYANIDE	0.00		-0.006	mg/L		
ICV01	9237274		CYANIDE	2.40		2.05	mg/L	85	
CAL01	9237275		CYANIDE	0.40		0.40	mg/L	100	
ICV01	9237306		CYANIDE	2.40		2.10	mg/L	88	
1CV01	9237308		CYANIDE	2.40		2.06	mg/L	86	
1CV01	923 <i>7</i> 338		CYANIDE	2.40		2.18	mg/L	91	
1CV02	9237320		CYANIDE	0.30		0.31	mg/L	103	
ICV02	9237332		CYANIDE	0.30		0.32	mg/L	107	
1 CV02	9237426		CYANIDE	0.30		0.29	mg/L	97	
DUP01	9237296	A228062	CYANIDE	0.00		0.00	mg/L	-	0
SPI01	9237301	A228068	CYANIDE	0.00	0.40	0.39	mg/L	98	
DPS02	9237327	A228358	CYANIDE	0.00	0.10	0.10	mg/L	100	9.5
SP102	9237325	A228358	CYANIDE	0.00	0.10	0.11	mg/L	110	
DUP02	9237324	A228331	CYANIDE	0.02	1	0.02	mg/L		0
BLA01	9237273		CYANIDE			-0.006	mg/L		ì
CCV	9237330		CYANIDE	0.30		0.33	mg/L	110	}
SAMPLE	A228370		See Certificate of Analysis						}
CCV	9237337		CYANIDE	0.30		0.29	mg/L	97	1
BLA01	o237339		CYANIDE			-0.006	mg/L		1
BLA02	9237321		CYANIDE			-0.003	mg/L		

SULFIDE SW846-9030 Analyst : L. HETTICH Reviewer: B. SHRAKE			•	Analysis Date: 30-APR-91 Review Date: 01-MAY-91 File ID: 673			Test: G110.4.0 Run: R119834				
QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD		
I CV01	Q239227		SULFIDE	46.8		47.1	mg/L	101			
DUP01	9239242	A228320	SULFIDE	0.00		0.00	mg/L		0		
DPS01	9239237	A228362	SULFIDE	0.00	23.4	22	mg/L	94	5		
SPI01	9239235	A228362	SULFIDE	0.00	23.4	21	mg/L	90	1		
BLA01	9239226	İ	SULFIDE			0.23	mg/L				
SAMPLE	A228370		See Certificate of Analysis	1		{			1		

	VOLATIL	E ORGAN	ICS TARGET COMPOUND LIST S	N846-8240					
Analys	st : H. WILI	IAMS	Analysis Date: 02-MAY-	91 Instrum	ment: GC/MS	VOA	Test: 0530.1	.0	
Review	er: C. KOL/	NOWSKI	Review Date: 07-MAY-	91 File ID	: >4713C	_	Run: R12031	6	
QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
ccv	9242750		See Attached Report g4710c.ind						
BLA01	9242764		See Attached Report g4712c.ind				1		
SAMPLE	A228370	ĺ	See Certificate of Analysis				•		ļ

Kevie	ver: C. KOLA	ANOWSKI	Review Date: U/-MAY-5	T File ID	: >4/13C		Run: R1205	16	
QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
ccv	9242750		See Attached Report g4710c.ind				1		
BLA01	9242764		See Attached Report g4712c.ind						
SAMPLE	A228370		See Certificate of Analysis	1					
	SEMI-VO	LATILE 1	TARGET COMPOUND LIST SW846-	8270					

Analyst : K. STONER Analysis Date: 21-MAY-91 Instrument: GC/MS SVOA Test: 0531.1.0 Reviewer: S. BROTHERTON Review Date: 23-MAY-91 File ID: >1448B
Prep: GC/MS SEPARATORY FUNNEL LIQUID-LIQUID EXTRACTION SW846-3510 Run: R121927

110	Trep. de/file service to the end of the end										
ас туре	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD		
DPS02	9229907	A228362	PHENOL	0.00	200	58	ug/L	29	0		
DPS02	Q229907	A228362	2-CHLOROPHENOL	0.00	200	170	ug/L	85	6		
DPS02	9229907	A228362	1,4-DICHLOROBENZENE (P-DICHLOROBENZE	0.00	100	71	ug/L	71	4		
DPS02	9229907	A228362	N-NITROSO-DI-N-PROPYLAMINE	0.00	100	110	ug/L	110	10		
DPS02	9229907	A228362	1,2,4-TRICHLOROBENZENE	0.00	100	74	ug/L	74	7		
DPS02	9229907	A228362	4-CHLORO-3-METHYLPHENOL	0.00	200	170	ug/L	85	0		
DPS02	9229907	A228362	ACENAPHTHENE	0.00	100	98	ug/L	98	2		
DPS02	9229907	A228362	4-NITROPHENOL	0.00	200	110	ug/L	55	15		
DPS02	9229907	A228362	2,4-DINITROTOLUENE	0.00	100	* 120	ug/L	120	6		
DPS02	9229907	A228362	PENTACHLOROPHENOL	0.00	200	170	ug/L	85	13		
DPS02	9229907	A228362	PYRENE	0.00	100	74	ug/L	74	4		
SP102	9229906	A228362	PHENOL	0.00	200	58	ug/L	29			
								D			

Lab Sample ID: A22837C-

QC Type	Identifier	Source	Parameter	True/Sampl	Spike Val	Observed	Units	% Rec	RPD
SP102	9229906	A228362	2-CHLOROPHENOL	0.00	200	160	ug/L	80	
SP102	9229906	A228362	1,4-DICHLOROBENZENE (P-DICHLOROBENZE	0.00	100	68	ug/L	68	-
SP102	9229906	A228362	N-NITROSO-DI-N-PROPYLAMINE	0.00	100	100	ug/L	100	
SPI02	Q229906	A228362	1,2,4-TRICHLOROBENZENE	0.00	100	68	ug/L	68	'
SP102	9229906	A228362	4-CHLORO-3-METHYLPHENOL	0.00	200	170	ug/L	85	
SP102	9229906	A228362	ACENAPHTHENE	0.00	100	96	ug/L	96	
SP102	9229906	A228362	4-NITROPHENOL	0.00	200	85	ug/L	43	İ
SP102	9229906	A228362	2,4-DINITROTOLUENE	0.00	100	* 110	ug/L	110	
SP102	9229906	A228362	PENTACHLOROPHENOL	0.00	200	150	ug/L	75	_
SP102	9229906	A228362	PYRENE	0.00	100	77	ug/L	77	}
CCV	Q253419		See Attached Report g1442b.ind]					
BLA01	Q253420	Ì	See Attached Report g1452b.ind	1	1				_
BLA02	9229904		See Attached Report g1454b.ind					1	
SAMPLE	A228370		See Certificate of Analysis						L '

Comments

Q229907 NOTE: * OUT OF QC LIMITS. Q229906 NOTE: * OUT OF QC LIMITS

Notes

∢ Less Than Lower Detection Limit

See Note for Parameter

Continuing Calibration Check HSL Compounds

Case No:	Calibration Date: 05/02/91
Contractor: EMSIN	Time: 09:04
Contract No:	Laboratory ID: >4710C
Instrument ID: GCMS#3	Initial Calibration Date: 03/11/91

Minimum RF for SPCC is .30

Maximum % Diff for CCC is 25%

Compound	RF	RF	%Diff	CCC	SPCC	
Acrolein Acrylonitrile Chloromethane Bromomethane Vinyl Chloride	.11587 .29668 .63154 .65134 .75090	.10133 .33265 .58524 .87492 .92693	12.55 12.12 7.33 34.33 23.44	*	**	(Conc=1250.00) (Conc=1750.00) (Conc=50.00) (Conc=50.00) (Conc=50.00)
Dichlorodifluoromethane Chloroethane Methylene Chloride Acetone Carbon Disulfide	1.01974 .63228 1.61904 .33471 3.29353	.40335 3.28110	46.46 .85 7.63 20.51 .38			(Conc=50.00) (Conc=50.00) (Conc=50.00) (Conc=50.00) (Conc=50.00)
Trichlorofluoromethane 1,1-Dichloroethene 1,1-Dichloroethane [Tetrahydrofuran] 1,2-Dichloroethene (total)	1.93412 1.18569 1.72263 .26479 1.29508	1.84894 .35856 1.35569	73.23 5.85 7.33 35.41 4.68	*	**	(Conc=50.00) (Conc=50.00) (Conc=50.00) (Conc=125.00)
Diethyl Ether Chloroform Trichloro-trifluoroethane 1,2-Dichloroethane-d4 2-Butanone	.75636 2.32848 2.33924 1.36023 .63507	1.40512	7.27 15.14 1.96 3.30 18.19	*		(Conc=50.00) (Conc=50.00) (Conc=50.00) (Conc=50.00)
1,2-Dichloroethane Methyl-t-butyl ether 1,1,1-Trichloroethane Carbon Tetrachloride Vinyl Acetate	1.69415 .71056 .60935 .64329 .01637	.72126 .54319 .55867 .01512	5.05 1.51 10.86 13.15 7.67			(Conc=50.00) (Conc=50.00) (Conc=50.00) (Conc=50.00) (Conc=50.00)
Bromodichloromethane 1,2-Dichloropropane cis-1,3-Dichloropropene Trichloroethene Diisopropyl Ether 2-Chloroethyl Vinyl Ether	.21128	.69331 .42102 .88405 .47329 1.26629 .23324	6.86 11.89 10.12 3.22 14.76 10.39	*		(Conc=50.00) (Conc=50.00) (Conc=50.00) (Conc=50.00) (Conc=50.00)
Dibromochloromethane	.65723	.67803	3.16			(Conc=50.00)

- Response Factor from daily standard file at 50.00 ug/L RF

RF - Average Response Factor from Initial Calibration Form VI
 %Diff - % Difference from original average or curve
 CCC - Calibration Check Compounds (*) SPCC - System Performance Check Compounds (*)

Continuing Calibration Check HSL Compounds

Case No: Calibration Date: 05/02/91

Contractor: EMSIN Time: 09:04

Contract No: Laboratory ID: >4710C

Initial Calibration Date: 03/11/91 Instrument ID: GCMS#3

Minimum RF for SPCC is .30

Maximum % Diff for CCC is 25%

Compound	RF	RF	%Diff	CCC	SPCC	
1,1,2-Trichloroethane Benzene trans-1,3-Dichloropropene Bromoform 4-Methyl-2-Pentanone 2-Hexanone Tetrachloroethene 1,1,2,2-Tetrachloroethane Toluene-d8	.34707 .80989 .24801 .59242 .43460 .32002 .46675 .67351	.40198 .91181 .25932 .58411 .55971 .01998 .47916 .81681	15.82 12.58 4.56 1.40 28.79 93.76 2.66 21.28 25.34	•••	**	(Conc=50.00) (Conc=50.00) (Conc=50.00) (Conc=50.00) (Conc=50.00) (Conc=50.00) (Conc=50.00) (Conc=50.00)
Toluene Chlorobenzene Ethylbenzene Bromofluorobenzene Styrene Xylene (total) 1,3-Dichlorobenzene 1,2-Dichlorobenzene 1,4-Dichlorobenzene	.59101 .85636 .41045 .75389 .84683 .49361	.67996 .99806 .46285 .82421 .99194 .56280	15.05 16.55 12.77 9.33 17.14 14.02	*	**	(Conc=50.00) (Conc=50.00) (Conc=50.00) (Conc=50.00) (Conc=100.00) (Conc=50.00) (Conc=50.00) (Conc=50.00)

Response Factor from daily standard file at 50.00 ug/L

Form VII Page 2 of 2

Average Response Factor from Initial Calibration Form VI

[%]Diff - % Difference from original average or curve CCC - Calibration Check Compounds (*) SPCC - System Performance Check Compounds (_

EMS LABORATORIES, INC VOLATILES ANALYSIS-WATER METHOD BLANK ANALYSIS METHOD: SW846-8240

MASS SPEC FILE: >4712C ANALYSIS DATE/TIME: 5/2/91 11:31

TARGET COMPOUND LIST Acetone	BDL BDL BDL BDL BDL BDL BDL BDL BDL BDL	UNITE DU UU UU UU UU UU UU UU UU UU UU UU UU	DET. LIMIT 20 50 70 5 5 10 5 5 10 5 5 5 10 5 5 5 10 10 5 5 5 10 10 5 5 5 5
SURROGATE LIST Dichloroethane-d4 Toluene-d8 Bromofluorobenzene () = ESTIMATED CONCENTRATION	95 100	·	pike conc) (50) (50) (50)

Continuing Calibration Check HSL Compounds

Case No:		Calibr	ation Da	te: 05/21/91		
Contractor: EMSIN		Time: 10:15		_		
Contract No:		Labora	tory ID:	>1442B		
Instrument ID: 8702		Initia	l Calibr	ation Date: (05/17/91	· -
Minimum RF for SPCC is	0.05	Maxim	um % Dif	f for CCC is	25.0%	_
Compound				CCC SPCC		_
Benzoic acid bis(2-Chloroethoxy)methane 2,4-Dichlorophenol 1,2,4-Trichlorobenzene Naphthalene 4-Chloroaniline Hexachlorobutadiene 4-Chloro-3-methylphenol 2-Methylnaphthalene Toluenediamine Hexachlorocyclopentadiene RF - Response Factor from RF - Average Response Fac %Diff - % Difference from or CCC - Calibration Check Con	1.21605 .87910 .37546 .87388 .48043 .49557 .23870 .46888 .18165 .25501 .26185 .12175 .33428 .28626 .30170 .88128 .36021 .16956 .19495 .55158 .10407 .38031	1.37808 .89638 .54768 .88904 .52000 .57938 .24652 .54378 .20085 .28795 .28206 .14259 .38039 .30624 .34484 1.01269 .37813 .20393 .20068 .62951 .06917 .44859 -ndard fi	13.33 1.97 45.87 1.73 8.24 16.91 3.28 15.97 10.57 12.92 7.72 17.11 13.79 6.98 14.30 14.91 4.97 20.26 2.93 14.13 33.53 17.95 		Check Comp	ounds ('

Continuing Calibration Check HSL Compounds

Case No:	Calibration Date: 05/21/91
Contractor: EMSIN	Time: 10:15
Contract No:	Laboratory ID: >1442B
Instrument ID: 8702	Initial Calibration Date: 05/17/91

Minimum RF for SPCC is 0.05

Maximum % Diff for CCC is 25.0%

Response Factor from daily standard file at 50.00 UG/KG RF

Average Response Factor from Initial Calibration Form VI
 f - % Difference from original average or curve
 Calibration Check Compounds (*) SPCC - System Performance Check Compounds (**)

Form VII Page 2 of 3

Continuing Calibration Check **HŠL** Compounds

Case No: Calibration Date: 05/21/91

Time: 10:15 Contractor: EMSIN

Contract No: Laboratory ID: >1442B

Instrument ID: 8702 Initial Calibration Date: 05/17/91

Minimum \overline{RF} for SPCC is 0.05 Maximum % Diff for CCC is 25.0%

Compound	RF	RF	%Diff	CCC	SPCC
Pyrene Butylbenzylphthalate 3,3'-Dichlorobenzidine Benzo(a)anthracene Chrysene bis(2-Ethylhexyl)phthalate Terphenyl-dl4 Di-n-octylphthalate Benzo(b)fluoranthene Benzo(k)fluoranthene Benzo(a)pyrene Indeno(1,2,3-cd)pyrene Dibenz(a,h)anthracene Benzo(g,h,i)perylene	.56314 .25126 .98093 .84411 .70335 1.03844 1.74984 1.21802 1.07580	.28365 1.21182 1.09577 .89543 1.50580 1.88885 1.32625 1.22572 1.14897 .64396 .69495	53.25 43.35 12.89 23.54 29.81 27.31 45.01 7.94 8.89 13.94 19.75 21.46 21.25 21.64	*	

Response Factor from daily standard file at 50.00~UG/KG Average Response Factor from Initial Calibration Form VI % Difference from original average or curve

%Diff -

CCC - Calibration Check Compounds (*) SPCC - System Performance Check Compounds (*

Form VII Page 3 of 3

EMS LABORATORIES, INC SEMI-VOLATILES ANALYSIS-INSTRUMENT BLANK ANALYSIS METHOD: SW846-8270

MASS SPEC FILE: >1452B

ANALYSIS DATE/TIME: 5/21/91 12:09

TARAST COMPOUND LICT	NECUL T	UNITE	DET LIMIT
	RESULT	UNITS	DET. LIMIT
Acenaphthene		mG/L	10
Acenapthylene		mG/L	10
Anthracene		mG/L	10
Benz(a)anthracene		mG/L	10
Benzo(a)pyrene	. BDL	mG/L	10
Benzo(b)fluoranthene	. BDL	mG/L	10
Benzo(ghi)perylene	. BDL	mG/L	10
Benzo(k)fluoranthene	. BDL	mG/L	10
Benzyl Álcohol		mG/L	10
Benzylbutylphthalate		mG/L	10
Bis(2-chloroethoxy)methane		mG/L	10
Bis(2-chloroethyl)ether		mG/L	10
Bis(2-chloroisopropyl)ether		mG/L	10
Bis(2-ethylhexyl)phthalate		mG/L	10
4-Bromophenylphenylether		mG/L	10
Carbazole	. BDL	mG/L	10
4-Chloroaniline		mG/L	10
2-Chloronaphthalene		mG/L	10
		mG/L	10
4-Chlorophenylphenylether			10
Chrysene		mG/L	10
Dibenz(a,h)anthracene		mG/L	10
Dibenzofuran		mG/L	
1,2-Dichlorobenzene		mG/L	10
1,3-Dichlorobenzene	. BDL	mG/L	10
1,4-Dichlorobenzene	. BDL	mG/L	10
3,3' -Dichlorobenzidine	. BDL	mG/L	20
Diethylphthalate	. BDL	mG/L	10
Dimethylphthalate	. BDL	mG/L	10
Di-n-butylphthalate		mG/L	10
Dinitrobenzenes		mG/L	50
2,4-Dinitrotoluene		mG/L	10
2,6-Dinitrotoluene	. BDL	mG/L	10
Di-n-octylphthalate		mG/L	10
Fluoranthene		mG/L	10
Fluorene		mG/L	10
Hexachlorobenzene	. BDL	mG/L	10
Hexachlorobutadiene		mG/L	10
Hexachlorocyclopentadiene		mG/L	10
Hexachloroethane		mG/L	10
<pre>Indeno(1,2,3-cd)pyrene</pre>		mG/L	10
Isophorone		mG/L	10
2-Methylnaphthalene		mG/L	10
Naphthalene		mG/L	10
•		•	

2-Nitroaniline	BDL	mG/L	50
3-Nitroaniline	BDL	mG/L	50
4-Nitroaniline	BDL	mG/L	50
Nitrobenzene	BDL	mG´/L	10
N-Nitroso-diphenylamine	BDL	mG/L	10
N-Nitroso-di-n-propylamine	BDL	mG/L	10
Phenanthrene	BDL	mG/L	10
2-Picoline	BDL	mG/L	50
Pyrene	BDL	mG/L	10
Pyridine	BDL	mG/L	50
Tetrachlorobenzenes	BDL	mG/L	10
Toluenediamine	BDL	mG/L	50
1,2,4-Trichlorobenzene	BDL	mG/L	10
Benzoic Acid	BDL	mG/L	50
4-Chloro-3-methylphenol	BDL	mG/L	10
2-Chlorophenol	BDL	mG/L	10
2,4-Dichlorophenol	BDL	mG/L	10
2,4-Dimethylphenol	BDL	mG/L	10
4,6-Dinitro-2-methylphenol	BDL	mG/L	50
2,4-Dinitrophenol	BDL	mG/L	50
2-Methylphenol	BDL	mG/L	10
4-Methylphenol	BDL	mG/L	10
2-Nitrophenol	BDL	mG/L	10
4-Nitrophenol	BDL	mG/L	50
Pentachlorophenol	BDL	mG/L	50
Phenol	BDL	mG/L	10
Tetrachlorophenol	BDL	mG/L	10
2,4,5-Trichlorophenol	BDL	mG/L	50
2,4,6-Trichlorophenol	BDL	mG/L	10

EMS LABORATORIES, INC SEMI-VOLATILES ANALYSIS-WATER METHOD BLANK ANALYSIS METHOD: SW846-8270

MASS SPEC FILE: >1454B

QC#- A229904

ANALYSIS DATE/TIME: 5/21/91 13:57

PREP DATE: 4/27/91

TARGET COMPOUND LIST Acenaphthene		SULT BDL BDL	UNITS uG/L uG/L	DET. LIMIT 10 10
Anthracene		BDL	uG/L	10
Benz(a)anthracene		BDL	uG/L	10
Benzo(a)pyrene		BDL	uG/L	10
Benzo(b)fluoranthene		BDL	uG/L	10
Benzo(ghi)perylene		BDL	uG/L	10
Benzo(k)fluoranthene		BDL	uG/L	10
Benzyl Alcohol		BDL	uG/L	10
Benzylbutylphthalate		BDL	uG/L	10
Bis(2-chloroethoxy)methane		BDL	uG/L	10
Bis(2-chloroethyl)ether		BDL	uG/L	10
Bis(2-chloroisopropyl)ether.		BDL	uG/L	10
Bis(2-ethylhexyl)phthalate		BDL	uG/L	10
4-Bromophenylphenylether		BDL	uG/L	10
Carbazole		BDL	uG/L	10
4-Chloroaniline		BDL	uG/L	10
2-Chloronaphthalene		BDL	uG/L	10
4-Chlorophenylphenylether		BDL	uG/L	10
Chrysene		BDL	uG/L	10
Dibenz(a,h)anthracene		BDL	uG/L	10
Dibenzofuran		BDL	uG/L	10
1,2-Dichlorobenzene		BDL	uG/L	10
1,3-Dichlorobenzene		BDL	uG/L	10
1,4-Dichlorobenzene		BDL	uG/L	10
3,3' -Dichlorobenzidine		BDL	uG/L	20
Diethylphthalate		BDL	uG/L	10
Dimethylphthalate		BDL	uG/L	10
Di-n-butylphthalate		BDL	uG/L	10
Dinitrobenzenes		BDL	uG/L	50
2,4-Dinitrotoluene		BDL	uG/L	10
2,6-Dinitrotoluene		BDL	uG/L	10
Di-n-octylphthalate		BDL	uG/L	10
Fluoranthene		BDL	uG/L	10
Fluorene		BDL	uG/L	10
Hexachlorobenzene		BDL	uG/L	10
Hexachlorobutadiene		BDL	uG/L	10
Hexachlorocyclopentadiene		BDL	uG/L	10
Hexachloroethane		BDL	uG/L	10
<pre>Indeno(1,2,3-cd)pyrene</pre>		BDL	uG/L	10
Isophorone		BDL	uG/L	10
2-Methylnaphthalene		BDL	uG/L	10
Naphthalene	• • •	BDL	uG/L	10

2-Nitroaniline. 3-Nitroaniline. 4-Nitroaniline. Nitrobenzene. N-Nitroso-diphenylamine. N-Nitroso-di-n-propylamine. Phenanthrene. 2-Picoline. Pyrene. Pyridine. Tetrachlorobenzenes. Toluenediamine. 1,2,4-Trichlorobenzene Benzoic Acid. 4-Chloro-3-methylphenol. 2,4-Dichlorophenol. 2,4-Dimethylphenol. 4,6-Dinitro-2-methylphenol. 4,6-Dinitrophenol. 2-Methylphenol. 4-Methylphenol. 4-Nitrophenol. 4-Nitrophenol. Pentachlorophenol. Pentachlorophenol. 2,4,5-Trichlorophenol. 2,4,5-Trichlorophenol. 2,4,6-Trichlorophenol.	BDL BDL BDL BDL BDL BDL BDL BDL BDL BDL	uG/L uG/L uGG/L uGG/L uGG/L uGG/L uG/L u	50 50 10 10 10 10 50 10 10 10 50 10 10 50 10
SURROGATE LIST 2-Fluorophenol Phenol-d5 Nitrobenzene-d5 2-Fluorobiphenyl 2,4,6-Tribromophenol Terphenyl-d14 () = ESTIMATED CONCENTRATION	62 39 91 88 91 76	(sp % Rec % Rec % Rec % Rec % Rec % Rec	ike conc) (100) (100) (50) (50) (100) (50)

Appendix G Revision 0 Nov. 1, 1991 Page 1 of 12

FORM	4 1	
DATI	E	
LOCA	ATION	
AUD	TTOR	
	GENERAL INFORMATION	ATTO NO COLOTENTO
1)	Does the laboratory have copies of SOPs/methods manuals and QA Plans available to all personnel?	YES NO COMMENTS
2)	Are training records maintained and up to date? Is training by group leader or above?	
3)	Is a Sample Custodian designated?	
4)	Are written Standard Operating Procedures (SOPs) developed for receipt and storage of samples?	
5)	Are samples stored in such a way that their preservation is maintained?	
6)	Are refrigerator/freezer logs maintained, checked daily and up to date?	
7)	Are excursions in cooler temperature noted and appropriate actions taken as required?	
8)	Are volatiles stored separately from semi-vols?	
9)	Does someone responsible review and initial the sample log and log sheets daily?	
10)	Are contamination free areas provided for trace level analytical work?	
11)	Does the facility appear clean and safe?	
12)	Are toxics handled in fume hoods? Are fume hoods checked & documented quarterly?	
13)	Are the toxic chemical handling areas either a stainless steel bench or an impervious material covered with absorbent material?	
91GB114	18.153	

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DA	CE	7	OF	EA	DA	1
ΓA	UL	4	Or	rv	RM	1

		YES NO COMMENTS
14)	Is there documented "trace-free" water available for preparing standards and blanks?	
15)	Is the conductivity of "trace-free" water routinely checked and recorded?	
16)	Is (Are) the analytical balance(s) correctly located (free from drafts and rapid temperature changes) and checked semi-annually by a certified technician and documented?	
17)	Are the balances checked with class S weights and documented at least weekly?	
18)	Are solvent storage cabinets properly vented in order to prevent possible laboratory contamination?	
19)	Are reagent grade (or higher purity) chemicals used to prepare standards and reagents?	
20)	Are reagents dated upon receipt and upon opening? Is First-in, First-out method used?	
21)	Are reagents standardized before use?	
22)	Generally, are reference materials properly labeled with concentrations, preparation date, solvent, preservative expiration date and name of person preparing?	
23)	Are spike/calibration stock standards preparation and tracking logbook(s) maintained?	
24)	Are the primary standards approved by the EMS QAP?	
25)	Are bench sheets filed so that they are readily accessible?	
26)	Are standards and samples (& extracts) stored separately?	
27)	Are samples checked for proper preservation upon arrival? Is this noted within LIMS?	
28)	Are logs maintained for all ovens and incubators?	
29)	Is an NBS thermometer available?	

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PAGE 3 OF FORM 1		YES NO COMMENTS
30)	Are copies of EPA - PEs and others such as APG on file?	
31)	Is the distilled/deionized water system functioning properly? Is it checked daily and noted in a logbook?	
32)	Are samples requiring chain of custody documentation properly checked before signing chain of custody papers?	
33)	What is done with lab waste?	
34)	Are chemical waste disposal policies/procedures adequate?	
35)	Are blind samples introduced into the lab by the QAO?	
36)	Are corrections on bench sheets, lab books, etc. made in such a way that initial entries are legible?	
37)	Are such corrections initialed and dated?	

Revi Nov.	endix G sion 0 1, 1991 4 of 12				
FOR	M 2				
DAT	E				
SER	VICE GROUP AUDITOR			_	
LOC	ANALYST ANALYST		 -		
	TEST	YES	NO	COMMENTS	
1)	Are methods manuals & QAP's available to the analysts?				
2)	Is the SOP for glassware washing posted at the cleaning stations?			Ι	
3)	Are the types and numbers of required Blanks being checked?				
	Bla 01? Bla 02? Cal 00?				
4)	Are blank data logged to the QA sheet and transferred to LIMS?			Γ	
4a)	Does blank data appear to be "in control"?]	
4b)	Do original instrument outputs agree with what was reported?				
5)	Are calibration curves maintained for all analytes?]	
6)	Are calibrations verified (1 point) or performed (5 points) prior to the analysis of samples CAL01 + CAL00? Comment:			<u> </u>	
7)	Is an EPA, NIST or other approved external reference used to check and verify concentrations of CAL standards and is the result recorded on the QA sheet?			J	
7a)	Are the results recorded in LIMS?				
8)	Do the analysts record bench data in a neat and accurate manner?				
		1			

91GB1148.I53

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PAGI	E 2 OF FORM 2	YES NO COMMENTS
8a)	Is all quality assurance data correctly transferred from the Bench/QA sheets into LIMS?	
9)	Is a calibration verification performed at the correct frequency?	
9a)	Documented to LIMS?	
10)	Is a prepped performance check (LCS) analyzed once per run or as required by the QAP? (eg one per set of 10)	
11)	Is a prepped sample spike (SPI02) analyzed with each run or as required by the QAP? (eg one per set of 10)	
12)	Is a prepped sample replicate analyzed (or a duplicate spike?) at the appropriate frequency according to the QAP? (eg one per set of 10)	
13)	If any problems occur, has corrective action been taken and documented?	
	Comments:	
14)	Is a calibration verification solution analyzed at the end of each run?	
15)	Is the method of internal standards used for GC/MS?	
16)	List internal standards used for VOA. (Attach copy of report)	
	Are these correct for the method?	
17)	List internal standards used for SVOA's. (attach copy of report)	
	Are these correct for the method?	
18)	Are surrogates properly analyzed and reported? Are surrogates within CLP acceptance ranges?	

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PAGE 3 OF FO	R۸	N	2
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IAGI		YES	NO	COMMENTS
19)	Are standard recoveries correctly calculated? Are spike recoveries correctly calculated? Are duplicate recoveries correctly calculated? Are they all documented in LIMS?			
20)	Does a manual recalculation agree with concentrations otherwise calculated and/or reported? (ie verify electronic integrator performing and correct peaks were chosen for GC, GC/MS)			
21)	For GC/MS, are unknowns (non-target compounds) correctly searched in the NBS library and documented as present with an estimated concentration?			
22)	For ICP, are interelement interference check solutions properly analyzed and documented prior to each run?			
23)	Have detection limits been empirically determined according to 40CFR136 for the analytes determined?			_
24)	Are standards correctly Labelled? Logged & documented?			
25)	Are instrument operating manuals readily available?			
26)	Are data acceptance criteria developed and used for Blanks? Duplicates? Verification standards? Spikes?			
27)	Are instrument maintenance logs in place and maintained?			
28)	Is the mercury analyzer operational and well maintained (i.e. properly vented)?			
29)	Is preventive maintenance applied and documented?			
30)	Are data calculations spot-checked by a second person? Does this person initial these checked calculations?			<u> </u>
		-		

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PAGE	4 OF FORM 2	YES NO COMMENTS
31)	Do supervisory personnel review bench sheets and initial them?	
32)	Do records indicate that corrective action was taken as necessary?	
33)	Has a cooperative attitude been displayed?	
34)	Has corrective action indicated during previous visits been implemented?	
35)	Are extractions/digestion performed within holding times?	
36)	Are analyses performed within holding times?	
37)	Are sample preparation methods correct?	
38)	Is a background corrector properly used? (AA, GFAA, ICP)	
39)	Is an approved methodology used for the runs being examined?	
40)	Have acceptance criteria for start-up QC been met?	
41)	Do instrument outputs (strip charts, print-outs, bench sheets, etc.) agree with what is in the final report?	
42)	Are labs data validation procedures adequate?	
43)	Based upon blank, standard, spike and duplicate data, are goals of data accuracy and precision being met?	
44)	Is the method of standard additions being used correctly for HGA analyses?	
45)	Is analytical sensitivity adequate?	
46)	Are instrument guidelines and/or acceptance criteria used at the bench and at data review which determine the acceptability and reportability of data?	
47)	Are Blanks carried through entire process?	
48)	Are desiccants changed and documented?	

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PAGE 5 OF FORM 2

- 49) Are TOC, TOX and BTU instrument logs current?
- 50) Are normalities of titrants routinely checked?
- 51) Are sufficient cyanide distillation apparatus available to routinely analyze all samples within the required holding period?

YES	NO	COMMENTS	
			-
		Γ	
	<u> </u>	1	-
			_
			_

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	LAB			
	DATE			
	AUDITOR			
	METALS GROUPS			
		YES	NO	COMMENTS
)	Are samples for mercury analysis collected in glass and preserved with HNO ₃ and Permanganate?			
)	Are mercury determinations performed within 28 days of sample collection?			
)	Are proper sample digestion techniques used and documented?			
)	Is background correction used a) For all HGA work? b) As required for flame AA work?			J
)	Are correct matrix modifiers used for HGA?		1	1
)	Is an ionization suppressor used as needed for flame AA?		<u> </u>	<u> </u>
)	For ICP are interelement corrections checked?		L	
)	Do prep sets of 10 include proper QA and acceptance criteria?		.l T	<u></u>
)	Are wavelengths used recorded on bench sheets?		<u> </u>	
)	Are flame/furnace/ICP programs adequate? a) Integration times adequate? b) Furnace temperature programs correct? c) Sample aliquot or deposition times adequate?			
)	Are sensitivity guidelines available at instruments?			

Appen Revision Nov. 1 Page 1	on 0	-
FORM	1 4	
LAB		_
DATE	E	_
AUDI	TTOR	
	GAS CHROMATOGRAPHY	YES NO COMMENTS
1)	Are samples collected in amber glass bottle with teflon-lined lids?	
2)	Are samples kept at 4°C	
3)	Are samples extracted within 7 days?	
3 a)	Analyzed within 40 days?	
4)	Are standards properly documented?	
5)	Are standards prepared fresh regularly?	
6)	Is there a well maintained log, documenting stability of all detectors?	
7)	Are Arochlor 1221 and 1232 standards run at the proper frequency and the data maintained for on-site inspection?	

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FORM	M 5	
LAB		
	E	
AUD	ITOR	
	GC/MS	YES NO COMMENTS
1)	Are samples extracted and/or analyzed within all required holding times?	
2)	Are control limits established and used for CCC and SPCC samples?	
3)	Is QA data properly transferred to LIMS?	
4)	Are Blanks, Spikes and Standards analyzed at the appropriate frequency?	
5)	Is Spike/spike duplicate data being generated?	
6)	Are surrogates properly analyzed with all samples?	
7)	Are surrogate results tabulated?	
8)	Are both 5 and 25 ml purge vessels available and properly used?	
9)	Are correct internal standards being used?	
10)	Is data reviewed prior to issuing final report?	
11)	Has QAO reviewed 2-5% of the runs?	
12)	Are reviewed bench sheets initialed?	
13)	Are all tuning criteria met?	
14)	Is the trap for the purge and trap filled with the proper adsorbent?	
91 GB 11	148.153	

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PAGE 2 OF FORM 5

- 15) Is raw data being archived and documented properly (i.e magnetic tape)?
- 16) Is a split/splitless capillary injector in place?

YES	NO	COMMENTS	
			}
	7	1	
	L	j	

DO NOT DUPLICATE

STANDARD OPERATING PROCEDURE AND DOCUMENTATION OF LABORATORY PERSONNEL TRAINING

It is the responsibility of the Group Leader or designee to conduct preliminary training to insure assimilation of the new employee into the mainstream of analytical methodology at EMS-Heritage.

To assist EMS-Heritage, the employee, and the Group Leader, various training forms are kept in each employee's file.

- 1. Form One contains the steps taken during the employees first days to accomplish orientation and start training.
- 2. Form Two contains a checklist, to be completed by each laboratory employee and their group leader to assure familiarity with basic laboratory equipment and technologies.
- 3. Form Three is a data review sheet containing a check list. The analyst must use during review of their own data.
- 4. Form Four is provided to document on-going training including in-house training seminars, group meetings, outside seminars, additional classes, accomplishments, etc.
- 5. The remaining forms include documentation of the employee's training on specific methodologies. At the onset of training, the employees are assigned to a trainer. The employee's trainer is the group leader or his/her designee, who is an experienced and audited chemist for that methodology.

The first step in the employee's training is the reading of the approved method. Employees will observe and assist their trainer for a period of time dependent on the complexity of the analysis. At such time, as the trainer deems appropriate, training employees will run the analysis independently as the trainer observes and instructs. Information regarding the first run is recorded on the training document. The trainer will continue to review the employee's data and technique until the trainer considers the employee trained. The last step in the training process is the performance audit, to be given by the Group Leader, or designee. The performance audit will cover adherence to method technique, documentations and knowledge of interferences, chemistry, etc. A blind standard may also be issued at this time. Successful completion of a performance audit must occur before analyst may operate independently.

6. Retraining verification will occur formally just prior to each employees winter review, on an annual basis. The verification will occur through one of the following mechanisms; a Blind Performance Evaluation sample, a Written Test, or a Performance Audit.

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FORM I

NEW EMPLOYEE ORIENTATION CHECK LIST

NAME:		
DATE OF E	MPLOYMENT:	-
POSITION:_		_
LAB GROU	P:	
		_
	Fill out all necessary forms through the Personnel Department	
	Get fitted for lab coats through our Purchasing Department	
	Obtain safety glasses and/or goggles. Prescription wearers should set up an eye appointment to obtain prescription safety glasses.	-
	General tour of facility and introduction to key personnel	-
	Meeting with the QA/QC Office. At this time employee should sign our Signature List and Confidentiality Agreement	_
	Introduction to the LIMS system through the Computer Department. Ananalyst parameter number should be issued, if appropriate.	
	Schedule New Employee Orientation through HERITAGE Personnel Department	-
	Safety tour and orientation from Safety Team member	-
	Obtain building entry code from Lab Director with a demonstration and explanation of our security system	
************	Obtain explanation of the following:	
	Time Cards (How to fill out and where to keep) Pay Day	
	Dress Code Break Time Policy/Smoking Hours	_
	Lunch Hour	
	Overtime Policy	_
	Obtain copy of the Personnel Manual and Standard Operating Procedures for your respective group, and a job description	
Company	Tairinia	-
Supervisor's	Initials Date	

FORM TWO

NAME:						
TRAINING TECHNIQU		FOR	BASIC	LABORATORY	EQUIPMENT	ANE
	Analytical Balar	nce				
	Use of and Clea	aning P	rocedure	for Pipettes		
	Fume Hood					
	Operation of C	entrifug	ge			
	Waste Solvent	informa	ition and	Disposal		
	Use of pH pape	er				
	Gas Cylinder T	rain i ng				
	Glassware Desc	ription	and Clear	ning S.O.P.		
	Disposal of Bro	ken Gl	assware			
	Water Types as	nd Usag	ges			
	Sampling a San	nple				
	Cooler Arrange	ments	- Sample	and Extract Storage	;	
	Tour of Stock	Room				
	Standards and	Resoen	ıts (Taheli	ing and Storage)		

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FORM THREE

DATA REVIEW

I.	Com	pleteness
	A.	Name
	B.	Date
	C.	Instrument
	D.	Units
	E.	DL
		QA Types
	G.	
II.	Accu	гасу
	A.	Spot-check calculations for samples and QA
	B.	Are dilutions accounted for?
	C.	Are weight/volume factors accounted for?
	D.	Are absorbance or results within the linear range?
Ш.	QA	
	A.	Are all QA requirements met?
	B.	Proper frequency and recoveries?
	C.	Are QA IDs and true values recorded?
	ח	Are cal curve units on a intercent and slone recorded

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FORM FOUR

ON-GOING TRAINING FORM

NAME:
DATE OF EMPLOYMENT:
POSITION:
LAB GROUP:
(Include Date, Topic, Description and Trainer):

EMS HERITAGE- INDIANAPOLIS, INDIANA

Analyst:	Education:			 		
Date Employed:	Prior Experience:				Page	of
General Organic Group Starting	Date:					
Analytical Method	Method Referenced	Date Trained	Trained By	Approved	EPA %Recov	Run
	· · · · · · · · · · · · · · · · · · ·					
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Analyst				Education		
Date Employed				Prior Experience _		
Metals Group						
Analytical Method	Method Referenced	Date Trained	Trained By	Approved	EPA Recovery	Run #
QC, Calculations, Paperwork	NA				NA	NA
General FAA	NA				NA	NA
IL S12 FAA	NA					
PE 2380 FAA	NA					
PE 5100 FAA	NA					
General GFAA	NA				NA	NA
PE 3030 GFAA	NA					
PE 5100 GFAA	NA NA					
General CVAA	SW846-7470 and EPA 245.1				NA.	NA
PE 403 CVAA	n					
PE 2380 CVAA	Ħ					
PE 5100 CVAA	11					
IL S12 CVAA	11					
88SE1537.B					Appendix H Revision O Nov. 1, 19 Page 7 of	91

Analytical Method	Method Referenced	Date Trained	Trained By	Approved	EPA Recovery	Run #
General ICP	EPA 200.7 and				NA	NA
IL ICAP-61	and SW846-6010					
E.P. Toxicity	SW846-1310				NA	NA
T.C.L.P.	Federal Register				NA	
pH - Water	SW846-9040					
pH - Soil	SW846-9045					
General Sample Prep	NA.				NA.	NA
ICP, FAA Water Prep	SW846-3005, 3010, EPA 200.0					
GFAA Water Prep	SW846-3020, EPA 200.0					
Extract Preparation	<i>S</i> W846-3010, 3020					
Solid Preparation	SW846-3050					
0il Prep - Acid Digestion	SW846-3030					
Oil Prep - MIBK Dissolution	<i>S</i> W846-3040					
Hg In Soils Prep	SW846-7471, EPA 245.5					

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Analytical Method	Method Referenced	Date Trained	Trained By	Approved	EPA Recovery	Run #
Cation Exchange Capacity	SW846-9081				NA	
Paint Filter Test	SW846-9095				NA	
Mobile Metals (Oily Waste)	SW846-1330				NA	NA
Glassware	EPA 200.0				NA	NA

CONTINUED TRAINING

Type of Training Date Completed Location Instructor Comments

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TRAINING CHECK LIST - EXTRACTION PREPARATION

AREA OF TRAINING	DATE	TRAINED BY
Read Methods SW846-3010, 3020, EPA 200.0		
Sample Volumme		
	ADDITIONAL TRAINING	
	,	
		

TRAINING CHECK LIST - T.C.L.P.

AREA OF TRAINING	DATE	TRAINED BY
Read Method - Federal Register		
Purpose of Test		
Extraction Fluid Preparation		
Selecting Representative Sample		
Extractor		
pH Meter Calibration		
Contamination and Safety Hazards		
Filtration of Extract		
Paperwork		
Routine Maintenance		
Test Passed		
	ADDITIONAL TRAINING	
Troubleshooting		
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TRAINING CHECK LIST - OC. CALCULATIONS, PAPERWORK

AREA OF TRAINING	DATE	TRAINED BY
Types and Frequency of QC Samples		
Acceptance Criteria For QC Samples		
Instrument Log Books		
Reagent Prep Log Book		
Glassware Types		
Sources of PERs		
DL Determinations		***************************************
Significant Figures		
Reporting Units		
Linear Regression		
Dilutions		
Reproducibility/Spike Recovery Calculations		
Test Code Organization		
Data Sheet Corrections		
Raw Data ID/Filing		
Rush Samples		
Test Passed	***************************************	
	ADDITIONAL TRAINING	

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TRAINING CHECK LIST - GENERAL GRAA

ARFA OF TRAINING	DATE	TRAINED BY
Theory of GFAA		
STPF System		
Tube Installation		
Inert Gas/Stop Gas		
EDL Operation		
Hazards		
Contamination Sources		
Common Interferences		
Zeeman Background Correction		
Matrix Modifiers		
Characteristic Mass		
Calibration		
Theory of Standard Additions		
GFAA Quantitation		-
Peak Height vs. Peak Area		
Paperwork		
Test Passed		
	ADDITIONAL TRAINING	
Trouble Shooting		
End Cone Installation		

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TRAINING CHECK LIST - GENERAL ICP

AREA OF TRAINING	DATE	TRAINED BY
Read Method EPA 200.7		
Principles of ICP		
Simultaneous/Sequential		
Linear Range		
Interferences		
a. Chemical		
b. Physical		
c. Spectral		
Background Correction		
Interfering Element Corrections		
Multielement Standards		
Paperwork		
Test Passed		
	ADDITIONAL TRAINING	
		
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TRAINING CHECK LIST - PE 5100 CVAA

AREA OF TRAINING	DATE	TRAINED BY
Computer Functions/Log In		
Recalling CVAA Program		
Instrument Set Up		
CVAA Apparatus Set Up		
Cell Alignment		
Energy Maximization		
Sample Analysis		
Leachate Analysis		
Routine Maintenance		
Instrument Shutdown		
General CVAA Test Passed		
	ADDITIONAL TRAINING	

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TRAINING CHECK LIST - DH - WATER

AREA OF TRAINING	DATE	TRAINED BY
Read Method SW846-9040		
Meter Calibration		
Sample Analysis		
Electrode Cleaning		
Electrode Storage		
Paperwork		
Test Passed		
	ADDITIONAL TRAINING	
Troubleshooting		
		······································

TRAINING CHECK LIST - GLASSWARE

AREA OF TRAINING	DATE	TRAINED BY
Read Glassware Washing Procedure		
Principles of Steps Involved		
Contamination		
Reagent Preparation		
Acid Bath Use and Cleaning		
Pipet Washing		
Hg Bottles		
Safety Hazards		
Test Passed		-
	ADDITIONAL TRAINING	

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TRAINING CHECK LIST - PE 3030 GPAA

AREA OF TRAINING	DATE	TRAINED BY
Keyboard Functions		
Instrument Settings		
Load Floppy Disks		
User Index		
File Storage		
Energy Maximization		
Premix Method		
Autosampler/Tip Alignment		
Sample Analysis		
Printer Operation		
Instrument Shutdown		
Routine Maintenance		
Test Passed		
	ADDITIONAL TRAINING	
Write GFAA Programs		
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TRAINING CHECK LIST - PE 5100 FAA

AREA OF TRAINING	DATE	TRAINED BY
Computer Functions/Log In		
Recalling FAA Programs		
Energy Maximization		
D ₂ Lamp Alignment		
Burner Alignment		
Gas Box/Ignition		
Absorbance Maximization		
Nitrous Oxide Procedures		
Atomic Emission Procedures		
Calibration		
Sample Analysis		
Instrument Shutdown		
Routine Maintenance		
Test Passed		
	ADDITIONAL TRAINING	
D ₂ Lamo Replacement		
Data File Management		
Automated Runs		
Multielement Runs		
Write FAA Programs		

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TRAINING CHECK LIST - PE 2380 FAA

AREA OF TRAINING	DATE	TRAINED BY
Keyboard Functions		
Instrument Settings		
Energy Maximization		
D ₂ Lamp Alignment		
D ₂ Energy Setting		
Burner Alignment		
Gas Box/Ignition		
Absorbance Maximization		
Nitrous Oxide Procedures		
Atomic Emission Procedures		
Calibration		
Sample Analysis		
Instrument Shutdown		
Routine Maintenance		
Test Passed		
	ADDITIONAL TRAINING	
D ₂ Lamp Replacement		
		

TRAINING CHECK LIST - GENERAL FAA

AREA OF TRAINING	DATE	TRAINED BY
Principles of Atomic Absorption and Emission		
Single Beam/Double Beam		
Modulation		
D ₂ Background Correction		
Matrix Modifiers		
Flow Spoiler/Impact Bead		
Hollow Cathode Lamps		
Sample Introduction System		
Air-Acetylene/Nitrous Oxide-Acetylene Flames		
Common Interferences		
Theory of Standard Additions		
Paperwork		
Test Passed		
	ADDITIONAL TRAINING	
Troubleshooting	`	
		

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TRAINING CHECK LIST - OIL PREPARATION (ACID DIGESTION)

AREA OF TRAINING	DATE	TRAINED BY
Read Method SW846-3030		
Modifications in SW846- 3030		
Sample Preparation		
General Prep Test Passed		
	ADDITIONAL TRAINING	
	<u> </u>	
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TRAINING CHECK LIST - He IN SOILS PREPARATION

AREA OF TRAINING	DATE	TRAINED BY
Read Methods SW846-7471, EPA 245.5		
Aqua Regia Preparation		- <u></u>
Sample Preparation		
Safety Hazards		
Sample Digestion		
General Prep Test Passed		
	ADDITIONAL TRAINING	
	,	

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TRAINING CHECK LIST - CATTON EXCHANGE CAPACITY

AREA OF TRAINING	DATE	TRAINED BY
Read Method SW846-9081		
Reagent Preparation		
Centrifuge		
Calculations		
Paperwork		
General Prep Test Passed		
	ADDITIONAL TRAINING	
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TRAINING CHECK LIST - PAINT FILTER TEST

AREA OF TRAINING	DATE	TRAINED BY
Read Method SW846-9095		
Sample Size		
Duration of Test		
Paperwork		
Test Passed		
	ADDITIONAL TRAINING	

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TRAINING CHECK LIST - MOBILE METALS (OILY WASTE)

AREA OF TRAINING	DATE	TRAINED BY
Read Method SW846-1330		
Soxhlet Extraction		
Preparation of Solvent Phase		
Calculations		
Safety Hazards		
Paperwork		
Test Passed		
	ADDITIONAL TRAINING	

TRAINING CHECK LIST - SOLID PREPARATION

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AREA OF TRAINING	DATE	TRAINED BY
Read Method SW846-3050		
Sample Preparation		
GFAA vs. ICP, FAA Prep		
General Prep Test Passed		
	ADDITIONAL TRAINING	

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TRAINING CHECK LIST - GPAA WATER PREPARATION

AREA OF TRAINING	DATE	TRAINED BY
Read Methods SW846-3020, EPA 200.0		
Sample Volume/Dilution		
Total, Dissolved, and Suspended Metals		
Sample Preparation		
General Prep Test Passed		
1		
	ADDITIONAL TRAINING	
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TRAINING CHECK LIST - GENERAL SAMPLE PREPARATION Page 29 of 64

AREA OF TRAINING	DATE	TRAINED BY
Principles of Sample Preparation		
Multiphase Samples		
Representative Samples		
GFAA/ICP, FAA Preps		
Glassware Types		
Contamination Sources		
Spikes and Duplicates		
Safety Hazards		
EMS Prep Codes		
Paper Work		
Test Passed		
•		
	ADDITIONAL TRAINING	

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TRAINING CHECK LIST - pH - SOIL

AREA OF TRAINING	DATE	TRAINED BY
Read Method SW846-9045		
Meter Calibration		
Sample Analysis		
Electrode Cleaning		
Electrode Storage		
Paperwork		
Test Passed		
	ADDITIONAL TRAINING	
Troubleshooting		

TRAINING CHECK LIST - E.P. TOXICITY

AREA OF TRAINING	DATE	TRAINED BY
Read Method SW846-1310		
Purpose of Test		
Selecting Representative Sample		
Extractor		
pH Meter Calibration		
Contamination and Safety Hazards		
Filtration of Extract		
Paperwork		
Routine Maintenance		
Test Passed		
	ADDITIONAL TRAINING	
Troubleshooting		

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TRAINING CHECK LIST - IL S12 CVAA

AREA OF TRAINING	DATE	TRAINED BY
Instrument Set Up		
CVAA Apparatus Set Up		
Cell Alignment		
Energy Maximization		
Sample Analysis		
Leachate Analysis		
Routine Maintenance		
General CVAA Test Passed		
	ADDITIONAL TRAINING	
	1	

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TRAINING CHECK LIST - OIL PREPARATION (MIBK DISSOLUTION)

AREA OF TRAINING	DATE	TRAINED BY
Read Method SW846-3040		
Limitations of Method		
Sample Preparation		
Safety Hazards		
General Prep Test Passed		
Dissolution Test Passed		
	ADDITIONAL TRAINING	
		

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TRAINING CHECK LIST - PE 2380 CVAA

AREA OF TRAINING	DATE	TRAINED BY
Instrument Set Up		
CVAA Apparatus Set Up		
Cell Alignment		
Energy Maximization		
Sample Analysis		
Leachate Analysis		
Routine Maintenance		
General CVAA Test Passed		
	ADDITIONAL TRAINING	

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TRAINING CHECK LIST - PE 403 CVAA

AREA OF TRAINING	DATE	TRAINED BY
Instrument Set Up		
CVAA Apparatus Set Up		
Cell Alignment		
Energy Maximization		
Sample Analysis		
Leachate Analysis		
Routine Maintenance		
General CVAA Test Passed		
	,	
	ADDITIONAL TRAINING	

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TRAINING DOCUMENT

ANALYST		DATE HIRED	
TEST	DATE	TRAINER	
METHOD READ	1ST RUN	% RECOVERY	_
	AUDIT	BLINDSTD	
TEST	DATE	TRAINER	_
METHOD READ	IST RUN	% RECOVERY	
1	AUDIT	BLINDSID.	
TEST	DATE	TRAINER	
METHOD READ	IST RUN	% RECOVERY	
	AUDIT	BLINDSTD	
TEST	DATE	TRAINER	_
METHOD READ	1ST RUN	% RECOVERY	
	AUDIT	BLIND STD.	
			_

MASS SPEC TRAINING CHECKLIST - SAMPLE PREP

AREA OF TRAINING	DATE	TRAINED BY
VOLATILES		
Soils		
ZHE		
FID Screening		
SEMI-VOLATILES		
Water		
QC		
Emulsions		
Soil-Sonication		
QC		
Soil-Soxhlet		
QC		
oil		
Making Surrogate		
Making Spike Solution		
	ADDITIONAL 1	PRAINTNG
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MASS SPEC TRAINING CHECKLIST - LOGBOOKS

AREA OF TRAINING DATE TRAINED BY

Corrections			
Class S Weights			
Soil Prep			
Standard Prep			
Instrument; Maintenance			
Instrument; Run Sequences			
	ADDITION	AL TRAINING	

MASS SPEC TRAINING CHECKLIST - SAMPLE ANALYSIS

AREA OF TRAINING	DATE	TRAINED BY
Methods		
Read SW846- 8240/8270		
Read EPA 624/625		
Read SDWA 524/525		
General		
Glassware Cleaning		
Tuning		
QC Frequency, Reand Preparation	equirements	
Tune Check		
Calibration		
Cal. Checks		
Blanks		
MS/MSD		
ISTD and Surr		
Calculations		
Response Factor		
Concentration		
Dilutions		
Units		
Sig Figs & Rounding		

AREA OF TRAINING DATE
Data Review

TRAINED BY

IDFILE Update	
FORMS	
Cover Sheet	
Initial Calibration	
Peak Clipping	

ADDITIONAL TRAINING

MASS SPEC TRAINING CHECKLIST - GENERAL MAINTENANCE

AREA OF TRAINING DATE TRAINED BY

Column Replacement		
Capillary		
Packed		
Tank Changing		
Changing Septum and Liner		
Cleaning the Source- including venting and pumpdown	Initial training	
MS	solo	
MSD	solo	
Cleaning the Quadrupole		
Replacing the Multiplier		
Cleaning and/or Replacing Jet Separator		
Changing trap		
Checking Tekmar flows		

ADDITIONAL TRAINING				

MASS SPEC TRAINING CHECKLIST - SYSTEM MAINTENANCE

DATE	TRAINED BY
ADDITIO	NAL TRAINING

EMS Laboratories, Indianapolis IN

Pre-Employment Resources Data

NAME:_	
DATE O	F EMPLOYMENT:
List o	f previous applicable work experience:
	
List o	f previous applicable academic training - both institutional nd work-shop:
	- .
List a	ny additional information:
Data f	orm completed:

TRAINING DOCUMENT - PAPERWORK AND LOG BOOKS

The trainer should explain where each of the following can be found. Also the use and the purpose of each should be explained.

AREA OF TRAINING	DATE	TRAINED BY
CALCULATION FORMS: Prep/Bench sheets Calculating sheets Cover sheets Surrogate Recovery forms Blank-QC forms Run sheets Order sheet log book Response factor log book Concentration/Dilution log Verbal Result log Report packet forms		
OTHER LOG BOOKS: Log of Logs Standards log book Standards verification log Gases log book Verification log for repipeter and balances Refrigeration temperature log Extract log Instrument maintenance log Power Failure log Report Packet log Computer Tape log Log of Vaulted data	S	
OTHERS: Worklists ER forms Over time Forms Methods Manuals and SOPs Instrument Manuals Computer Manuals		

The trainer should explain the use and/or the purpose of each of the following.

AREA OF TRAINING	DATE	TRAINED BY
GLASSWARE: Identification of glassware Disposal of broken glassware Read glassware cleaning procedure Washing extraction glassware Washing microglassware Pipet washing Washing Herbicide glassware Safety hazards		
PREPARATION OF FLORISIL COLUMNS: Read procedure Difference between "jumbo" and 5 1/4" columns Storage of florisil, pipets, silica gel and glass wool Preparartion of florisil columns		
PREPARATION OF ALUMINA COLUMNS: Read procedure Storage of alumina, pipets and glasswool Deactivation - to activity III alumina Alumina verification Preparation of Alumina columns		
PREPARATION OF SODIUM SULFATE Purpose for baking Storage of sodium sulfate Use of the muffle furnace Dessicator/desicant Storage of sodium sulfate after baking		
STANDARDS TRAYS: Use of different standards Storage of the standards and the standards trays Organization of the standards refrigerator Organization of the ASV trays Filling the ASVs		

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ANALY	yst			
DATE	HIRED			
PREI	P	DESCRIPTI	Ю	Method
P229	.1/P232.1	PCBs in C	oi1	SW846-3580
	Difference bet	ween a "clea	n" and "waste	oil
	Clean Oil Prep Trainer Method read	Dat	e of training	
	Prep Book			
	Standard BLA	Run	* Recovery	Appoval by:
				
	Waste Oil Prep Trainer Method read Prep Book	Dat	e of training	
	Standard BLA	Run	1 Recovery	Appoval by:
	Acetone Preps			
	Trainer	 	e of training	
	Standard BLA	Run	* Recovery	Appoval by:

ANALY	ST				
DATE	HIRED				
PREP		DESCRIPTION			Method
P231.	1	PCBs in Soils			SW846-3550
	Soil Prep				
	Trainer Method read Prep Book			of training	
	Standard BLA	Run			Appoval by:
	Swipe Prep				
	Trainer Method read Prep Book			of training	
	Standard BLA	Run		* Recovery	
•	Air Sample Prep	-			
1	Trainer Method read Prep Book			of training	
	BLA	Run	-	* Recovery	Appoval by:
			_		

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ANAL!	YST			<u> </u>	
DATE					
PRE	P	DE	SCRIPTION	I	Method
P235	. 1	PCBs in	Soils/so	xhlet	SW846-3550
	Soxhlet 1	Prep			
	Trainer Method re	ead	Date	of training	
	Prep Book Soxhlet (Thimble s	Glassware storage	torage _	or	-
	obing on		o cherace		
	Standa:		<u>un</u>	* Recovery	Appoyal by:
		· -			
					
P230	.1	PCBs in Wa	ater/sep.	funnel extr	SW846-3510
	Trainer _	ractions -	Date	of training	
					_
	<u>Standar</u> BLA		in	* Recovery	Appoval by:

YNYT,							
DATE	HIRED						
PREP		DESCRIPTIO	ИС	Method			
P233	.1 PCB/Pes	ts in Water/se	ep. funnel extr.	SW846-3510			
	Water extrac	tion - PCB/Pes	st				
	Trainer	Dat	te of training _				
	Sodium sulfa		ol filters				
	BLA						
P236		·	Soil/SPP!	SW846-3550			
	SPP! extractions - PCB/Pests						
	Trainer	Dat	ce of training				
-	Prep Book Using the ultrasonic probe Concentrating the sample Splitting the extract for semi-vols Dilutions for GPC clean ups GPC clean up training on page 8						
	Standard BLA	Run	* Recovery	ppoval by:			
							

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ANALYST
DATE HIRED
GPC training - The trainer should explain the purpose of the GPC clean up and the principle of the clean up. The trainer should also give an explantion of the flow through the system and the purpose of the various solvents.
Trainer Date of training Method read
AREA OF TRAINING
Pre-run:
Reservoirs Air supply
Vacuum pump
Instrument settings & Run parameters Checking sample pump flow rate
Starting the run:
Filtering samples
Loading input vials
Loading sample extract vials - crimp cap Starting the instrument
Shut down - at completion of the run:
Vacuum & sample pump - shut off automatically
Reservoir switch to fill & reservoirs removed Air tank turned off
Others (put date of training):
Calibration
Repacking the column
Back flushing the valves

ANALY DATE	ST		·		
PREP		Method			
P238. P201.	1 4	SW846-3590 SW846-8150			
	TrainerMethod re Prep Book Using the	ad	Dat	es in Water e of training n ecations to be	
	Standard BLA	\$ +2-2000 p	Run	Recovery	_
	Method rea	ad	Date	e of training	
	Prep Book Diazometha Inclu Calabraha	ude all sa	afety pro	n ecations to be	_ taken
	Standard		Run	Recovery	Appoval by:
	Herbica	de Weter	rs liether	rextraction	

ANALYST							
DATE HIRED							
PREP		DESCRIPTI	ON	Method			
P201.4	Herbicides in Water Diazomethane Derivitization			SW846-3590 SW846-8150			
Traine	r	extractionsDa	- Herbicides	in Water			
Prepar	Prep Book Preparation of Reagents Acid washed Glassware						
Acid w Concen	e & Glass wool						
	Diazomethane derivatization Include all safety precations to be taken						
<u>Stan</u> BLA	dard_	Run	* Recovery	Appoval by:			
							

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ANAL'	YST				
DATE	HIRED				
PREP		DESCRIPTIO	N	Method	
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	Water extracti Trainer Method read Prep Book KD Glassware Salt & filter Concentrating	Dat	e of training		
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	Soil extraction Trainer Method read	Dat	e of training		
	Standard BLA	Run	Recovery	Appoval by:	

ANALYS DATE H					
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P K S	later extraction rep Books D Glassware _ Sodium sulfate concentrating t	& Glass woo			
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		Run	Recovery		
T	oil extraction rainer ethod read	Dat	e of training		
	Standard BLA	Run	Recovery	Appoval by:	

ANALY	'ST				
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	Water extraction - Prep Books KD Glassware Sodium sulfate & G Concentrating the	lass wool			
	Trainer				
	BLA	Run	* Recovery	Appoval b	Y:
	Soil/SPP! Prep book Using the ultrason Concentrating the Splitting the extr Dilutions for GPC GPC clean up SPP! extractions -	sample	emi-vols	_	
	TrainerMethod read	_ Date	of training _	· · · · · · · · · · · · · · · · · · ·	ء المراجعة المرجعة (المراجعة المرجعة (المرجعة (المرجعة (المرجعة (المرجعة (المرجعة (المرجعة (المرجعة (المرجعة (المرجعة (الم
	<u> </u>	Run	* Recovery	Appoval b	Y:
					

Appendix H
Revision 0
Nov. 1, 1991
Page 56 of 64

ANALY DATE	ST HIRED						
PREP			DESCRI	PTION		Metl	nod
P425.	1	EDB/DB	SCP - mic	roext	ractions	EPA	504
	Storage	ok of 40 of bak		m chlo	oride		
	Trainer Method			Date	of training	·	
	Stand BLA	ard	Run		Recovery	Appoval	by:

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ANALYST	<u> </u>		
DATE RIKED			
Computer training	I		
Trainer	_		
SOP read		-	
TRAINING CHECK			
explained by the	trainer, and und	derstood before	e checking off.
Check off by putt	ing date of tra	iining.	
Identification of the			
	trument ID	Device	LU
	E_ECD_1 (1)	Tl	1
	_ECD_2 (2)	T2	58
	E_ECD_3 (3)	T3	59
	_ECD_4 (4)	T4	60
	_ECD_5 (5)	T5	50
	_ECD_6 (6)	L1	51
	_ECD_7 (7)	L2	52
	_ECD_8 (8)	L3	53
	D_9 (9)	L5	55 54
	CD_10 (10)	L4	54
	D (11)	L6	56 53
NPD NP	D (12)	L7	57
Logging onto the compu	ter station		
	code and working	na directory es	stablished
	rd established		
	LDS, CI and LPL	.OT	
Preparing a run for th			
	ing the instrum		
	ing a new seque	ence	
	g the sequence		
Assign	ing the sequence	e to the instr	rument
Copying files		•	
Copyin	g a sequence		
Copyin	g a method		
Modifying a sequence			
Freein	g the sequence		ng
Using	the modify sequ	ence command	•
Modify	ing the assigne	ed method	
Modify	ing the sample	list	i can day sage.
Modify	ing the result	files	and the second

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ANALYST
DATE HIRED
Computer training
Trainer
SOP read
TRAINING CHECK LIST - Each of the following should be explained by the trainer, and understood before checking off. Check off by putting date of training.
Modifying a method
Modifying the timed events table
Modifying the run time
Modifying the timed events table Modifying the run time Modifying the plot scale Modifying the output device
Modifying the output device
Modifying the data input - ie: minimum area and threshold
threshold
Unlocking a locked file
Running the CI mode Reseting a file
Unlocking a file
Other CI commands
Up a downed device
Stopping the reanalysis of a sequence
Checking the system status with IO listing
Archiving files
When to archive
Initializing storage tapes Directory listing of files
Directory listing of files
Using the archive utility
Purging files
NOTE: Do not purge any files without asking first!!
When to purge
What files can be purged
Using the purge utility
Note: These method and sequence files are set up for the
GC/ECDs and should never be purged.
METHODS:
PCB1,PCB2,PCB3,PCB4,PCB5, and PCB6
CLP1, CLP2, CLP3, CLP4, CLP5, and CLP6
Sequences:
RECAL1 AND RECAL2
Approved by
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TRAINING DOCUMENT - GC/ECD CALCULATIONS

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	Run order & QC requirements Location of calculating forms, run forms, c
	sheets & DCB recovery forms
	Identification of Aroclors
	Identification of baseline problems and base
	modifications (using the computer)
	Calculation of response factors
	Calculation of the % RSD (linearity)
	Calculation of the %D (between beginning & en
	standards)
	Calculation of extract concentration
	Filling out the run sheets
	Calculation of the sample result including an
	dilution factors
	Calculation of the Standard recovery
	Calculation of the Spike recovery Calculation of the Dual Purpose Spike (
	* recovery and the * replicability
	Calculation of the % recovery for the RCS (IC
	Calculation of the % recovery for the CCV (en
	Medium)
_•	, ,
	k Run
Date	
	nel
Nati 1	(dithat
Revi	ewed by
	Comments

Approved by _

TRAINING DOCUMENT - GC/ECD CALCULATIONS

ANALY	ST
DATE	HIRED
	Herbicide calculations
	Trainer
	SOP read
	TRAINING CHECK LIST - Each of the following should be
	explained by the trainer, and understood before checking off.
	Check off by putting date of training.
	Run order & QC requirements
	Location of calculating forms & run forms
	Identification of the Herbicides
	Calculation of the retention time windows
	Identification of baseline problems and baseline
	modifications (using the computer)
	Calculation of response factors
	Calculation of the % RSD (linearity)
	Calculation of the %D (between beginning & ending
	standards)
	Calculation of extract concentration
	Filling out the run sheets
	Calculation of the sample result including any
	dilution factors
	Calculation of the Standard recovery
	Calculation of the Spike recovery
	Calculation of the Dual Purpose Spike (DPS)
	% recovery and the % replicability
	Calculation of the % recovery for the RCS (ICV02)
	Calculation of the % recovery for the CCV (ending
	Medium)
	Second column confirmation of the herbicides
	Packet + Forms it necount First Run
	Pirst Run Date
	Channel
	Run Number
	Rull Number
	Reviewed by
,	Comments
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TRAINING DOCUMENT - GC/ECD CALCULATIONS

PCB/Pest calculations Trainer SOP read TRAINING CHECK LIST - Each of the following should explained by the trainer, and understood before checking of Check off by putting date of training. Run order & QC requirements Location of calculating forms, run forms & QC for Identification of Pesticides and Arcolors Calculation of the retention time windows Identification of baseline problems and baseling modifications (using the computer) Calculation of response factors Calculation of the % RSD (linearity) Calculation of the % RSD (linearity) Calculation of the % RSD (linearity) Calculation of the sumple result including any dilution factors Calculation of the sample result including any dilution factors Calculation of the Standard recovery Calculation of the Spike recovery Calculation of the Dual Purpose Spike (DF % recovery and the % recovery for the RCS (ICVC Calculation of the % recovery for the RCS (ICVC Calculation of the % recovery for the RCS (ICVC Calculation of the % recovery for the CCV (ending Nedium) Second column confirmation of pesticides and PC factor + form Pirst Run Number Reviewed by	LYST	·
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Run Number		
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Approved by _____

TRAINING DOCUMENT - GC/ECD CALCULATIONS

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	calculations			
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	_ Identification			aceline
		ons (using the		GSETTHE
	_ Calculation of			ione
	_ Calculation of	response facto	re	10115
	_ Calculation of	the & RSD (lin	nearity)	
	Calculation of	the &D (between	n beginning &	ending
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	_ Calculation of	extract concen	tration	
	_ Filling out the			
	_ Calculation of		ult including	anv
<u> </u>	dilution fa		•	•
	_ Calculation of	the Standard r	ecovery	
	_ Calculation of	the Spike reco	very	
···	_ Calculation of			(DPS)
		and the % repl		
	_ Calculation of			
	_ Calculation of	the 1 recovery	for the CCV	(ending
	Medium)			
	_ Second column c	onfirmation of	EDB & DBCP	
First Dun				
Channel				
Run Number	r			
Reviewed	by			
Comme	ents			
		<u> </u>		
Run Number	Second column c	onfirmation of	EDB & DBCP	

TRAINING DOCUMENT - GC/ECD INSTUMETNATION

DATE HIRED	
Trainer	
SOP read	AINING CHECK LIST - Each of the following should be plained by the trainer, and understood before checking off. ock off by putting date of training. a Run
explained	by the trainer, and understood before checking off.
Starting a Run	
Trainer	
SOP read	
	Checking the rinse bottles Loading the autosampler trays Autosampler control on the 5880 - starting the autosampler program for a single channel run
	Autosampler control on the 5880 - starting the autosampler program for a dual channel run Autosampler control on the 5890
Trainer	Approved by
	Changing the syringe
	Septa and liner change
	Checking flows
	Sylonizing the injection ports
	Replacing a column
	Replacing the chemical filters
	Replacing the detector
Trainer	
SOP read	mpproved by
501 1000	
	Changing the syringe
	Septa and liner change
 	
<u> </u>	
**************************************	Replacing the chemical filters
	Replacing the detector
	, <u></u>

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TRAINING DOCUMENT Recertification on Prep Methods

NALYST ATE HIRED				
Date of Recert.	Test Code	Test Method	Test score ¹ Std. Recovery ²	Approved by

^{&#}x27;Tests are based on the procedures as listed in the method

²Standard are given as blind standards. These standards are to be prepared by the group leader.



EMS HERITAGE LABORATORIES, INC.

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Indianapolis, Indiana 46231

Attn: Ramona Eaton

1992 UPDATE SERVICE

for the Comprehensive Quality Assurance Plan

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